Paper II

X-ray structure analysis of isocitrate dehydrogenase from the hyperthermophile *Archaeoglobus fulgidus*: thermal stability and domain swapping

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Abbreviations used are: IDH, isocitrate dehydrogenase; GDH, glutamate dehydrogenase; ApIDH, Aeropyrum pernix IDH; EcIDH, Escherichia coli IDH; BsIDH, Bacillus subtilis IDH; TmIDH, Thermotoga maritima IDH; AfIDH, Archaeoglobus fulgidus IDH; PfIDH, Pyrococcus furiosus IDH; melting temperature, $T_{\rm m}$

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Summary

With the aim of investigating thermal stability comparatively within one protein family we solved the crystal structure of hyperthermostable isocitrate dehydrogenase (IDH) from Archaeoglobus fulgidus (AfIDH) at 2.5 Å resolution. To identify heat adaptive mechanisms in AfIDH a detailed structural comparison was performed with mesophilic IDH from Escherichia coli (EcIDH). AfIDH was strikingly similar to EcIDH and displayed almost the same number of ion pairs and ionic networks. Two unique inter-domain networks were however present in AfIDH; one three-membered ionic network between the large and the small domain and one four-membered ionic network between the clasp and the small domain. An aromatic cluster was found towards the N-terminal as well as one unique inter-subunit aromatic cluster in the clasp domain, indicating that aromatic clusters may be important for thermal stabilization of AfIDH. Three small loop deletions were observed in AfIDH of which one was located in the clasp domain. The unique features of the clasp-like domain of AfIDH and its ionic interactions with the small domain seem to be important for thermal stabilization as the apparent melting temperature (T_m) decreased by 18 °C when this domain was swapped with that of EcIDH. By contrast, EcIDH was only stabilized by 4 °C when the clasp domain of AfIDH was introduced, a result probably due to the introduction of the four-membered aromatic inter-subunit cluster and loop shortening. Common and unique heat adaptive traits of AfIDH with those recently observed for hyperthermostable IDH from Aeropyrum pernix (ApIDH) and Thermotoga maritima (TmIDH) are discussed herein.

Introduction

Hyperthermophilic organisms are defined by a temperature optimum for growth at or above 80 °C (Stetter 1999). All hyperthermophiles known so far are prokaryotes, although a few eukaryotic organisms, such as the Pompeii worm (Alvinella *pompejana*), have been found in the hot waters of hydrothermal vents where they experience a thermal gradient of 60 °C or more over its body length (Cary et al. 1998). Most of the hyperthermophilic microorganisms belong to the archaeal domain although hyperthermophiles have been found in the two bacterial orders, Thermotogales and Aquificales (Blöchl et al. 1995). Enzymes synthesized by (hyper)thermophiles are typically thermostable, or resistant to irreversible inactivation at high temperatures, and thermophilic, i.e. optimally active at elevated temperatures between 60 and 125 °C (Vieille and Zeikus 2001). In order to reveal main adaptive strategies used for protein stabilization, numerous three-dimensional structures of hyperthermophilic proteins have been obtained. Comparative studies between hyperthermophilic and mesophilic enzymes have demonstrated that interactions such as hydrogen bonds, disulfide bonds, ion-pairs, salt bridges, hydrophobic interactions and compactness are of importance for stability (Scandurra et al. 2000; Vieille and Zeikus 2001). No universal basis of stability has been recognized and the major reason for this is the relatively small free energy difference between the folded and the unfolded state and the complex way in which the small number of weak forces, determining protein stability, interplay with each other (Karlström et al. 2005). However, the most common determinants for increased thermal stability are in the first line a statistical prevalence of ionic interactions at the protein surface, increased formation of large ionic networks, electrostatic optimisation and the reduction of repulsive charge-charge interactions (Spassov et al. 1997; Vetriani et al. 1998; Karshikoff and Ladenstein 2001; Karlström et al. 2005).

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Despite a great availability of data on thermophiles and thermophilic proteins, most of the information generated on how proteins from hyperthermophilic organisms have adapted in their natural environments are based on case studies (Karshikoff and Ladenstein 2001) or structural and biochemical comparisons of single mesophilic/hyperthermophilic enzyme pairs.

We have chosen the enzyme isocitrate dehydrogenase (IDH) as a model enzyme for studying environmental adaptations of proteins to extreme temperatures. IDH catalyses the oxidative decarboxylation of D-isocitrate to 2-oxoglutarate and CO_2 with NAD⁺ (EC 1.1.1.41) or NADP⁺ (EC 1.1.1.42) as cofactor and comprises a diverse enzyme family with regard to cofactor specificity, primary structure, and oligomeric state.

Aeropyrum pernix (Ap), Pyrococcus furiosus (Pf), Archaeoglobus fulgidus (Af) and *Thermotoga maritima (Tm)* are hyperthermophilic microorganisms growing optimally at 95, 100, 83 and 80 °C, respectively (Fiala and Stetter 1986; Huber et al. 1986; Sako et al. 1996). We have previously estimated the thermal stability of *Ap- Pf-, Af-* and *Tm*IDH by determination of the apparent melting temperatures (*T*_m) using differential scanning calorimetry (DSC) (Steen et al. 2001). It was found that *Ap*IDH has highest thermal stability with a *T*_m of 109.9 °C. *Pf, Af* and *Tm*IDH were less stable with a *T*_m of 103.7, 98.5, and 98.3 °C, respectively. However, each of the hyperthermophilic IDHs showed a significantly higher *T*_m than the one determined for IDH from *Escherichia coli* (*T*_m = 52.6 °C) (*Ec*IDH) (Karlström et al. 2005) and pig (*T*_m = 59 °C) (Karlström et al. 2006). Recently, the three-dimensional structures of *Ap*IDH (Karlström et al. 2005) and *Tm*IDH (Karlström et al. 2006) have been resolved and the structural properties important for their high thermal stability have been identified. Here we report the crystallization and structure determination of the hyperthermophilic *Af*IDH including a comparative structural study with other known IDH homologs. Furthermore, chimeras between the hyperthermophilic *Af*IDH and the mesophilic *Ec*IDH were constructed to investigate the contribution of the clasp domain to the thermal stability of wild-type *Af*IDH.

Results and Discussion

Quality and description of the model

The final model of *Af*IDH comprises all 412 amino acid residues for both molecules in the asymmetric unit, although some side-chains are rather poorly defined in electron density. In addition, 9 Zinc atoms, 3 Chloride atoms and 94 water molecules were modelled in electron density. The final model and structure factors have been deposited in the Protein Data Bank with accession codes PDB2IV0. For an overview of the refinement statistics, see Table 1.

Overall structure and active site of AfIDH

As observed for all other IDHs, AfIDH consists of three domains; a large domain, a small domain and a clasp domain. Residues 1-120 and 312-412 belong to the large domain, residues 121-153 and 196-311 form the small domain and the remaining residues 154-195 form the clasp-like domain. AfIDH formed a homo-dimer in the crystal structure and the inter-subunit relationship and interface contacts in AfIDH are similar to those in other reported IDH structures (Hurley et al. 1989; Singh et al. 2001; Karlström et al. 2005). Apart from the formation of the clasp domain, the dimer associates through helices h and i in both subunits creating a stable four helix bundle. The dimer has two active sites that are located in a deep cleft formed by the large and the small domains of one subunit and the small domain of the adjacent subunit. AfIDH was crystallized and its structure solved without substrate and cofactor in the active site, however two Zn^{2+} -ions were found tightly bound to Asp301, Asp305, Asp277' (' corresponding to second subunit) in the active site of both subunits (Fig. 1a and 1b). In subunit B a water molecule (W81) was also found in contact with Zn^{2+} (Fig. 1b). In both *Ec*IDH and *Ap*IDH binding of Ca^{2+} has been shown to involve conserved water molecules together with three conserved aspartic residues equivalent to those found in

AfIDH (Stoddard et al. 1993; Karlström et al. 2005). Furthermore, AfIDH shares the highly conserved residues involved in substrate binding and catalysis as shown for ApIDH, EcIDH and Bacillus subtilis IDH (BsIDH) (Dean 1993; Singh et al. 2001; Karlström et al. 2005) (Fig. 2). Thus, it is believed that AfIDH has similar interactions with substrate and cofactor in the active site. To this date several crystal structures of different NADP⁺-dependent homo-dimeric IDHs have been reported: *Ec*IDH (Hurley et al. 1989) (PDB-codes 3ICD; closed form and 1SJS; open form), ApIDH (Karlström et al. 2005) (PDB-codes 1XGV, 1TYO and 1XKD), BsIDH (Singh et al. 2001) (PDBcode 1HQS), porcine heart mitochondrial IDH (Ceccarelli et al. 2002) (PcIDH, PDBcode 1LWD), human cytosolic IDH (Xu et al. 2004) (HcIDH, PDB-code 1T0L; closed form and 1T09; open form) and TmIDH (Karlstrom et al. 2006). Overlay structure analysis of AfIDH with other resolved structures of IDH revealed that it was most similar to EcIDH. A comparison with the open and closed form of EcIDH revealed AfIDH as more similar to that of the open conformation (Root-Mean-Square Distance (RMSD) of 0.99 Å) than to the closed conformation (RMSD of 1.78 Å). The RMS difference between the small domain and large domain of AfIDH versus that of the open structure of *Ec*IDH was 0.57 Å (using 170 C^{α}-atoms) and 0.96 Å (using 208 $C^{\alpha}\text{-atoms}),$ respectively (Fig. 3a and 3b) . Secondary structure elements were given the nomenclature as implemented in *Ec*IDH (Hurley et al. 1989).

Overall, helix and strand regions were conserved in *Af*IDH compared to *Ec*IDH, however, some local differences were observed in *Af*IDH. An alignment based on secondary structural assignment (Fig. 2) revealed that a helix, g2, replaced the loop between strand *L* and strand *K* in *Af*IDH compared to *Ec*IDH. This g2 helix has previously been observed in the structure of *Ap*IDH (Karlström et al. 2005).

Thermal stability

The $\Delta T_{\rm m}$ between *Ec*IDH and *Af*IDH is 45.4 °C. As *Af*IDH showed highest structural similarity to *Ec*IDH, a comparative study was performed between these enzymes to reveal heat adaptive traits in the hyperthermophilic enzyme.

Accessible surface area (ASA)

Analysis of *Af*IDH shows a significant increase of ASA contributed by polar residues and also a significant decrease of ASA contributed by hydrophobic residues compared to *Ec*IDH. However, the percentage of ASA contributed by charged residues was slightly lower in *Af*IDH compared to *Ec*IDH, which is also the case for the hyperthermophilic IDH from *Ap*IDH (Karlström et al. 2005). The distribution of hydrophobic, polar and charged surface area of *Af*IDH and *Ec*IDH is shown in Table 2. The dimer interface of *Af*IDH buried 5564 Å² of the 37970 Å² molecular surface area (MSA) of the dimer, giving a solvent-accessible surface area of 32406 Å² for the dimer. In contrast, the *Ec*IDH dimer buries 6020 Å² of the 38547 Å² of the dimer with a total solvent-accessible surface of 32527 Å², i.e the buried inter-subunit surface comprise 14.7 % and 15.6 % of *Af*IDH and *Ec*IDH, respectively.

Amino acid composition

The *Af*IDH subunit is made up of 412 amino acids of which 30.1 % are polar residues, 27.2% charged residues and 42.7% hydrophobic residues. Compared to *Ec*IDH, the distribution of amino acids was similar in the two proteins with a slightly higher fraction of polar residues in *Af*IDH and a slightly higher fraction of hydrophobic residues in *Ec*IDH (Table 2). However, *Af*IDH has a significantly higher fraction of aromatic residues of 11.2% compared to 8.7% in *Ec*IDH. It has previously been shown that *Af*IDH and other archaeal hyperthermophilic IDHs have a decreased

number of Cys residues compared with EcIDH, 0.20 % and 1.40 %, respectively (Steen et al. 1997; Steen et al. 2001), hence, following the trend observed for thermophilic proteins whereby Cys residues tend to be avoided.

Ionic interactions

Comparative studies between hyperthermophilic and mesophilic homologs have shown a clear tendency for the total number of ion pairs and large ionic networks to increase with the optimal growth temperature of the organisms as well as with the $T_{\rm m}$ of the proteins (Yip et al. 1995; Knapp et al. 1997; Yip et al. 1998). Surprisingly, few differences were found between the mesophilic EcIDH and the hyperthermophilic AfIDH in the total amount of charges and ionic interactions. The number of ionic networks at different cut off distance, are summarized in Table 2. Overall, the number and size of ionic networks did not differ significantly between AfIDH and EcIDH, however the location of the ionic networks seem to play an important role for the increased thermal stability of the hyperthermphilic enzyme. In AfIDH, a threemembered inter-domain network was found at 4.0 Å cut off between Asp119 (helix d, large domain), Arg201 (helix f, small domain) and Asp324 (loop region between β strand E and D, large domain). Although the amino acids involved in this network are conserved in EcIDH, only a single ion pair was found in the mesophilic enzyme at 4.0 Å cut off. These amino acids are also conserved in ApIDH and part of a sevenmembered ionic network. Furthermore, a four-membered ionic network in AfIDH between the clasp domain, Arg163 (loop in clasp domain), and the small domain of the second subunit, Glu196' (helix f), Lys200' (helix f) and Glu240' (helix g1), was observed as low as 3.5 Å cut off distance, hence, making this a possible strong saltbridge interaction. No ionic network was observed in the clasp domain of EcIDH with a 3.5 Å cut off, however, a three-membered ionic network was observed when

analysed at 4.0 Å cut off. Hence, a stronger ionic contribution to stability was present in the clasp domain of *Af*IDH compared to *Ec*IDH.

Loop shortening

Shortening of small loops was observed in *Af*IDH compared to *Ec*IDH: by three residues in the loop between helix *a* and strand *A*; two residues between helix *b* and *c* and finally, three residues between helix *e* and strand *N* in the clasp domain (Figures 2 and 3). The two latter loop deletions were also reported in the structure of *Ap*IDH. Loop deletion or loop shortening is often seen in hyperthermophilic proteins and in many cases reflected by a higher content of secondary structure. Loop regions and random coil structures, particularly in solvent-exposed regions are usually flexible areas and most likely to collapse at elevated temperatures. Hence, loop-loop stabilization by ionic interactions can be crucial for maintaining the functional structure at high temperatures. In *Af*IDH, two loop-loop interactions were observed in each subunit closer than 4.0 Å: subunit A; Asp26-Lys59 (3.65 Å) and Arg215-Asp291 (2.64 Å), subunit B; Asp26-Lys59 (3.06 Å) and Lys216-Glu266 (3.0 Å). In *Ap*IDH, one loop-loop interaction was observed in each subunit equivalent to Asp26-Lys59 in *Af*IDH: subunit A; Asp34-Arg70 (3.73 Å), subunit B; Asp34-Arg70 (3.88 Å). However, no loop-loop interactions were observed in the mesophilic *Ec*IDH.

Aromatic clustering

The presence of aromatic clusters has earlier been suggested to contribute to thermal stability of proteins (Kannan and Vishveshwara 2000; Dalhus et al. 2002). A large number of aromatic residues were observed in the N-terminal of *Af*IDH compared to *Ec*IDH (Fig. 4). An aromatic cluster of three residues; Tyr3 (N-terminus), Phe64 (β -strand *A*) and Phe91 (helix *c*), was observed in this region and is likely to protect the

N-terminus from thermal unfolding. In the clasp domain of AfIDH, the conserved aromatic interaction between Phe174 and Phe174' (in subfamily I IDHs (Steen et al. 2001)) on helix *e* was extended to an aromatic cluster by Phe179 and Phe179' on helix *e* which, in addition involved Trp161 and Trp161' on strand *M*, resulting in a 6-residue aromatic cluster (Fig. 5). The aromatic cluster in the clasp domain has previously been observed in *Ap*IDH and is believed to provide stabilization of the interface. In *Ec*IDH, Phe179 in *Af*IDH is substituted with Met183.

Hydrogen bonds

The total number of hydrogen bonds was slightly higher in the structure of *Ec*IDH compared to *Af*IDH, 700 and 687, respectively. However, a small increase in the fraction of hydrogen bonds per residue was observed in *Af*IDH (Table 2). The fraction of main-chain/main-chain and main-chain/side-chain hydrogen bonds were similar in the two structures, however a small increase of side-chain/side-chain hydrogen bonds was observed in *Af*IDH. Furthermore, the number of interface hydrogen bonds was 23 and 22 in *Af*IDH and *Ec*IDH, respectively. Although the structures of *Af*IDH and *Ec*IDH have similar resolution, the determination of hydrogen bonds is highly resolution-dependent and in this case therefore be an inaccurate element to compare in terms of thermal stability.

Stability effects of domain swapping

Three features likely to contribute to the higher thermal stability in *Af*IDH *contra Ec*IDH were identified in the clasp-like domain; an aromatic cluster, a four-membered ionic network from both subunits to the small domain of the other subunit and a loop shortening; Lys186-Phe190 (*Ec*IDH numbering). In order to investigate the

implications of these features on the thermal stability of *Af*IDH, chimeric proteins between *Af*IDH and *Ec*IDH were constructed. Each of the two chimeras was composed of the large and small domain from one enzyme with the clasp-like domain from the second enzyme (*Ec*IDH/*Af*IDH_{clasp} and *Af*IDH/*Ec*IDH_{clasp}). The presence of an aromatic cluster and a loop-shortening could be formed in the chimera *Ec*IDH/*Af*IDH_{clasp} since these are present in the *Af*IDH-clasp. However, the fourmembered ionic network can not be formed since the network is not conserved in *Ec*IDH. Thus, a possible increase in stability of this chimera would be related to the formation of these two factors. In the chimera *Af*IDH/*Ec*IDH_{clasp}, a likely decrease in stability would be addressed to the removal of all three putative stabilizing factors mentioned above.

 $T_{\rm m}$ of the chimeras were monitored by DSC and was as previously observed for the wild-type enzymes (Steen et al. 2001; Karlström et al. 2005) an irreversible process. A considerable decrease in $T_{\rm m}$ was observed for the chimer *AfIDH/EcIDH*_{clasp} ($\Delta T_{\rm m}$ - 18 °C) as compared to wild-type *AfIDH*, followed by a reduction in half-life ($t_{1/2}$) and a lower apparent temperature optimum ($T_{\rm opt}$) (Table 4). In contrast, swapping the *AfIDH* clasp into the *EcIDH* increased $T_{\rm m}$ by 4 °C corresponding to an increase in $t_{1/2}$ at 50 °C and a slightly increased $T_{\rm opt}$ (Table 4). Taken together, these data indicate that the inter-domain ionic network in *AfIDH* contributes more to the thermal stability of the native enzyme than the aromatic cluster and the loop-shortening. Interestingly, disruption of an aromatic cluster in *TmIDH* reduced the apparent $T_{\rm m}$ by 3.5 °C (Karlstrom et al. 2006), a result comparable with the increased $T_{\rm m}$ for *EcIDH*/*AfIDH*_{clasp}.

It should be noted that the mutations performed by domain-swapping involve several amino acids and could lead to both favourable and unfavourable interactions with other parts of the enzyme. However, amino acids from both subunits in the dimer contribute to the binding of substrate, allowing any major differences in k_{cat} , as a control of conformational changes between wild-type and chimeric enzymes, to be easily detected. Kinetic properties (Table 3) revealed that swapping of the clasp-like domains did not have a significant effect on the affinities for substrates of the chimeras compared to their respective wild-type proteins. Furthermore, catalytic efficiencies (k_{cat}/K_m) of the chimeras were similar to those determined for wild-type proteins, suggesting that no major conformational changes had occurred upon folding (Table 3). Hence, changes in apparent T_m of the chimeras may thus be related to the differences in the clasp domain between *Af*IDH and *Ec*IDH.

Comparison of heat adaptive traits in hyperthermophilic IDHs

Previously, we have shown that *Af*IDH has a T_m of 98.5 °C (Steen et al. 2001) which is similar to the T_m of *Tm*IDH (98.3 °C) (Steen et al. 2001) but lower than the one determined for *Ap*IDH (109.9 °C) (Steen et al. 2001). Recently, we have succeeded in determining the three-dimensional structure of *Ap*IDH (Karlström et al. 2005) and *Tm*IDH (Karlstrom et al. 2006). An important factor contributing to the thermal stability of *Ap*IDH was a disulfide bond at the N-terminus of each subunit between Cys9 and Cys87. *Af*IDH lacks the distinct N-terminal extension involved in formation of this disulphide bound in *Ap*IDH and only one cys residue was observed in each subunit of the *Af*IDH structure. Hence, no disulphide bridges could be identified in the *Af*IDH structure. The *Tm*IDH has a shorter primary sequence than *Af*IDH (412) and *Ap*IDH (435) with 399 amino acids. Furthermore, as described above for the *Af*IDH, the inter-subunit assembly in the clasp domain is different in the *Tm*IDH from subfamily II compared to the IDHs in subfamily I (Karlstrom et al. 2006). The difference is also evident in the location of the C- and N-terminal in the structure from *Tm*IDH where the side chains in the two termini are separated by a distance of 4.2 Å

(Karlstrom et al. 2006), whereas in AfIDH, ApIDH and EcIDH from subfamily I the termini are separated on the large domain by 40 Å, 42.7 Å and 41.5 Å, respectively (subunit A). Stabilization of the N-terminus was seen in the structure from *Tm*IDH by electrostatic compensation between three lysines (3, 5 and 7) and two aspartic acid residues (36 and 341). In the structure from TmIDH an ionic interaction between Asp389, located on a helix in the C-terminus, and Lys29 (close to N-terminus), was found to be important for the stability of the protein. Site-directed mutagenesis resulted in a considerable decrease in apparent T_m of almost 22 °C compared to the recombinant wild-type of TmIDH (Karlstrom et al. 2006). The ionic interaction was also shown to be conserved in *Pc*IDH and *Hc*IDH and due to the high impact of the mutation in *Tm*IDH, believed to be involved in the protection of both termini from thermal unfolding. As described above, an aromatic cluster was observed in the Nterminal region of AfIDH and is most likely an alternative strategy for this protein to protect the flexible region of the N-terminus from heat degradation. Furthermore, an ionic interaction was observed in the N-terminus of AfIDH between Lys14 and Glu90 (helix c), however, this interaction was also found to be conserved in EcIDH. In the C-terminus no ionic interaction was observed in AfIDH or EcIDH at 4.0 Å, however, at 6.0 Å a four-membered ionic network was observed in AfIDH involving Arg48 (helix a), Glu403 (helix m), Arg399, (helix m) and Glu400 (helix m) (subunit B). Ionic interactions in the N-and C-terminus to prevent heat denaturation have also been observed in studies from other hyperthermophilic proteins, in particular the extensive work performed on citrate synthase (Bell et al. 2002).

In ApIDH, more ion pairs and larger ionic networks were present in the protein contra the mesophilic *EcIDH* homolog. A major determinant conferring the increased thermal stability in ApIDH, confirmed by mutational studies was a seven-membered inter-domain ionic network with many neighbouring charged residues extending the network to 23 and 15 members if a cutoff of 6 Å instead of 4.2 Å was used. In *Af*IDH only three of these seven amino acid residues were conserved. In the structure of *Tm*IDH no inter-domain ionic networks were identified (Karlstrom et al. 2006).

As previously found for ApIDH, AfIDH contained a small number of inter-subunit ion pairs. At 4.0 Å distance cut off, only four inter-subunit ion pairs were found in AfIDH (between Arg163 in the clasp domain and Glu196 from the small domain from both subunits, and the conserved interaction between Lys233 and Asp301 in each active site). There were no differences in the number of inter-subunit ion pairs when compared to the mesophilic EcIDH, which also contained four inter-subunit ion pairs at 4.0 Å. Increasing the cut off to 6.0 Å did not reveal any major differences in the number of inter-subunit ion pairs in the hyperthermophilic ApIDH and AfIDH compared to the mesophilic EcIDH (12, 11 and 10 ion pairs, respectively). In contrast, TmIDH was found to have 8 and 11 inter-subunit ion pairs at 4.0 and 6.0 Å cut off, respectively (Karlström et al. 2006).

In conclusion, the structural comparison between the three hyperthermophilic IDH homologs has revealed the importance of stabilization of the N-terminus, although different strategies have been employed by the different enzymes. The size and positioning of ionic networks differs among the model enzymes, a result in common with previous observations for hyperthermophilic glutamate dehydrogenases (GDH) (Britton et al. 1999; Bhuiya et al. 2005). As may be expected from the exceptional high T_m of ApIDH, this enzyme has larger networks compared with both TmIDH and AfIDH. The importance of electrostatic contribution to the stability of hyperthermophilic proteins has previously been discussed by Xiao and Honig (Xiao and Honig 1999). Their study revealed that in all instances the electrostatic interactions were more favourable in the hyperthermophilic proteins compared to mesophilic homologs. Furthermore, the electrostatic free energy was found not to be

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correlated with the number of ionisable amino acids, ion pairs or ion pair networks in a protein structure, but rather the location of these groups within the structure. It has previously been noted that each of the hyperthermophilic IDHs under investigation have net charges towards zero as opposed to mesophilic *Ec*IDH and porcine IDH. Furthermore, when the cut off was increased to 6 and 8 Å the ionic networks were substantially increased in *Ap*IDH, indicating a electrostatic optimization of the surface (Karlström et al. 2005). In common with *Tm*IDH this was not observed for *Af*IDH, indicating a less optimized surface in these enzymes. It should be noted that the *T*_m of both *Tm*IDH and *Af*IDH is ~ 10 °C lower than that of *Ap*IDH.

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Materials and Methods

Strains

E. coli strain EB106 (icd-11 dadR1 trpA62 trpE61 tna-5 λ^{-}) was originally obtained from the *E. coli* Genetic Stock Center, MCD Biology Dept.,Yale University, in courtesy of Dr. Mary K.B. Berlyn. In order to express proteins in this strain, *E. coli* EB106 was lysogenized using the λ DE3 Lysogenization Kit from NOVAGEN.

Crystallization and data collection

*Af*IDH was crystallised using the hanging drop vapour diffusion technique with reservoir solution consisting of 0.6 M ZnSO₄ and 0.1 M Na Cacodylate, pH 6.3, which are optimised from the Nextal EasyXtal Cations suite condition number 60 (QIAGEN GmbH, Hilden, Germany). Crystals were grown by mixing equal volumes (2 μ l) of a 9 mg/ml protein solution with the reservoir solution. The drops were equilibrated at 8°C. 20% (v/v) glycerol added to the reservoir solution sufficed as a cryo-protectant for flash-cooling the crystals in liquid nitrogen. A data set (see Table 1) was collected at the macromolecular crystallography beamline X06SA at the Swiss Light Source (SLS). The data was collected with a MAR225 CCD detector and allowed the determination of the crystal structure using molecular replacement techniques.

Structure determination and refinement

The collected data set was indexed and integrated using MOSFLM (Leslie 1992). The crystals were monoclinic, with unit cell parameters of a=81.6Å, b=65.4Å, c=87.2Å, β =95.28°. The data was scaled, merged and the intensities were converted into structure factors using the CCP4 programs SCALA and TRUNCATE (CCP4 1994). A summary of the data collection statistics is presented in Table 1. The systematic

absences in the collected data set indicated the presence of a two-fold screw axis along the b-axis, with the only possible space group being $P2_1$. The solvent content was estimated to be around 52%, with a Matthews Coefficient of 2.6 $Å^3 Da^{-1}$, assuming two protein molecules per asymmetric unit. The crystal structure of AfIDH was determined by molecular replacement methods using MOLREP (CCP4 1994). The crystal structure of ApIDH was used as search model and the automated program functions in MOLREP were applied in order to create the model that presumably had the best fit to the sequence of AfIDH. Reflections up to a high-resolution limit of 3.5 Å were used. One well-resolved solution for the two molecules in the asymmetric unit could be found, having a correlation coefficient of 0.459 and an R_{factor} of 45.1%. A rigid-body fitting of the model using a high-resolution cut-off at 2.5 Å resulted in R_{work} of 44.5% (R_{free} of 46.0%). After a manual intervention using O (Jones et al. 1991), the model was refined in REFMAC5 (Murshudov et al. 1999), resulting in Rfactors of 24.0% and 29.1% for the working and test sets of reflections, respectively. Subsequent cycles of refinement interspersed with manual rebuilding, gave final R_{work} and R_{free} values of 19.6% and 25.4%, respectively, with acceptable protein geometry. In general, water molecules were added using the embedded functions found in REFMAC and manually checked using O. The four residues found in disallowed regions of the Ramachandran plot are all located in flexible loops.

Structure analysis

Sequence alignment was performed using the program STAMP (Russell and Barton 1992) based on the C^{α}-atom coordinates and secondary structural assignments using the program DSSP (Kabsch and Sander 1983). Potential salt-bridge formation and

ionic networks were analysed using the program CONTACT (CCP4 1994) with varying maximum distances from 3.5 to 8.0 Å. Accessible surface areas were calculated using CNS with Ala, Ile, Leu, Met, Phe, Pro, Trp, and Val defined as hydrophobic residues, Asn, Gln, Ser, Thr, Tyr, Cys and Gly regarded as polar residues and Asp, Glu, Arg, Lys and His as charged residues. The water probe radius was set to 1.4 Å and the accuracy of the numerical integration was set to 0.12. Water and ions were excluded from the model.

Hydrogen bonds were calculated using HBPLUS (McDonald and Thornton 1994) v3.15 and the following default parameters: maximum distances for D-A, 3.9 Å and for H-A, 2.5 Å; minimum angels for D-H-A, D-A-AA and H-A-AA was 90°. Ion pairs that were counted as hydrogen bonds by the program were excluded from the analysis.

Figure preparations

Figure 1 and 3-5 were made using the program PYMOL (PyMol 2005) and figure 2 was prepared using the program ALSCRIPT (Barton 1993).

Cloning and domain swapping

The wild-type *Ec idh* gene was amplified from *E. coli* K12 (DSMZ - Literature Reference No. 4684) genomic DNA by PCR using the primers P1-P2 (Table 5) and ligated into pET101/D-TOPO (Invitrogen Corporation, Carlsbad, CA 92008, USA). Domain-swapping of the clasp-like domain from *Ec idh* with the clasp domain from *Af idh* was performed by using the ExSiteTMPCR-Based Site-Directed Mutagenesis Kit (Stratagene, La Jolla, CA) with the pET101/D-TOPO-*Ec idh* as template. Mutagenesis was performed as described in the manufacturer protocol. Because of the large size of the mutagenic primers the mutagenesis was performed in a two-step

reaction with primers P3-P6 (Table 5). Swapping of the clasp-like domain from *Af idh* with the clasp-like domain from *Ec idh* (*Af idh/Ec idh*) was performed by overlap extension by PCR (Warrens et al. 1997) on the template pET11-a/*Af idh* (Steen et al. 2001) using primers P7-P14 (Table 5). For expression, the chimer construct was sub-cloned into pET101/D-TOPO vector using primers P15-P16 (Table 5) to amplify the gene.

Expression and purification

and purification of wild-type AfIDH and AfIDH/EcIDH_{clasp} were Expression performed as previously described for wild-type AfIDH (Steen et al. 2001). Expression of wild-type *Ec*IDH and *Ec*IDH/*Af*IDH_{clasp} was performed at 37 °C in *E*. coli strain EB106 (DE3) in LB broth containing ampicillin (100µg/ml). Isopropyl-βthiogalactopyranoside was added to 1.0 mM concentration to induce expression at A_{600} $_{nm} = 0.7 - 0.8$, and the incubation was continued for 4-5 hours at 37 °C. Cells were harvested by centrifugation (5000 x g for 15 min) and frozen at -20°C until used. Cells carrying expressed wild-type *EcIDH* or *EcIDH/AfIDH*_{clasp}, were resuspended in 20 mM sodium-phosphate buffer, pH 7.0, containing 1 mM EDTA and disrupted using a French pressure cell. After removal of cell debris by centrifugation (10000 x g for 30 min) the cell extracts were applied to a Red Sepharose column (Millipore) equilibrated with 20 mM sodium phosphate buffer, pH 7.0 containing 1 mM EDTA. The column was first washed with the sodium phosphate buffer until A_{280} was zero and then washed with the sodium-phosphate buffer containing 0.25 mM NADP⁺. Protein was eluted with 20 mM sodium phosphate buffer, pH 7.0 containing 1.0 mM EDTA, 0.25 mM NADP⁺, 10 mM isocitrate and 10 mM MgCl₂. Purified enzymes were stored in 20 mM Tris/HCl, pH 7.5, 50% glycerol and 1.0 mM DTT at -20° C.

Temperature dependence and thermal stability

Thermal inactivation experiments were performed at 50 °C for wild-type *Ec*IDH and *Ec*IDH/*Af*IDH_{clasp} and at 65 °C and 70 °C for wild-type *Af*IDH and *Af*IDH/*Ec*IDH_{clasp}. The 150 μ l incubation solution contained 20 mM Tris/HCl, pH 7.5, 50% glycerol, 1.0 mM DTT and 3 μ g of enzyme. Samples were collected at given times and remaining activity was measured using assays for the respective enzymes.

Differential scanning calorimetry was carried out with a VP-DSC MicroCalorimeter (MicroCalTM). The samples were dialyzed against the reference buffer (50 mM potassium-phosphate buffer, pH 7.5, 0.1 M NaCl). A protein concentration of 1.2 mg/ml was used for all enzymes. The calorimetric scans were carried out between 20 and 120 °C for wild-type *Af*IDH and *Af*IDH/*Ec*IDH_{clasp} and between 20 and 90 °C for wild-type *Ec*IDH and *Ec*IDH/*Af*IDH_{clasp} with a scan rate of 1 K/min. A second scan was run to estimate reversibility. Apparent T_{mS} were determined from the transition midpoint upon unfolding, due to the irreversible nature of the enzymes.

Enzyme assay and measurement of kinetic properties

Enzyme activity of IDH was measured photometrically at 40 °C (wild-type *Ec*IDH and *Ec*IDH/*Af*IDH_{clasp}) or 60 °C (wild-type *Af*IDH and *Af*IDH/*Ec*IDH_{clasp}) by monitoring the formation of NADPH at 340 nm ($\varepsilon_{340} = 6.22 \text{ mM}^{-1} \text{ cm}^{-1}$). The 1 ml reaction contained 50 mM Tricine-KOH, pH 8.0 and 10 mM MgCl₂ at assay temperature, 1 mM isocitrate for wild-type *Ec*IDH and *Ec*IDH/*Af*IDH_{clasp} or 3 mM and 5 mM for wild-type *Af*IDH and *Af*IDH/*Ec*IDH_{clasp}, respectively. Cofactor was added at a concentration of 0.25 mM. One Unit of enzyme activity is referred to as the reduction of 1 µmol of NADP⁺ per minute. For determination of *K*_m and *V*_{max} values for isocitrate, the NADP⁺ concentration was kept fixed at 0.25 mM while varying the isocitrate concentration. For determination of $K_{\rm m}$ values for NADP⁺, the isocitrate concentration was kept fixed at 1 mM for *Ec*IDH and *Ec*IDH/*Af*IDH_{clasp}, 3 mM for *Af*IDH and 5 mM for *Af*IDH/*Ec*IDH_{clasp} while varying the cofactor concentration. Kinetic data were analyzed by the Direct linear plot using the Enzpack 3 software package (Biosoft, Cambridge, UK). Protein concentrations were measured by the method of Bradford (Bradford 1976).

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Tables

Data collection	
Wavelength (Å)	0.92
Space group	P2 ₁
Unit cell parameters (Å, °)	81.6, 65.4, 87.2, 90, 95.28, 90
Resolution range (Å)	40-2.5 (2.64-2.5)
No. observed reflections	130993 (19123)
No. unique reflections	31867 (4612)
Redundancy	4.1 (4.1)
Completeness (%)	99.8 (100.0)
Mean I/σI	11.1 (3.5)
R _{merge} (%)	11.4 (47.6)
Refinement	
Non-hydrogen atoms	6553
Metal ions	9Zn/3Cl
No. solvent molecules	94
Overall B-factor (Å ²)	16.80
R _{cryst} (%)	19.6
R_{free} (%)	25.4
Deviation from ideal geometry	
Bond lengths (Å)	0.016
Bond angles (°)	1.785
Ramachandran plot	
Most favoured region (%)	87.7
Allowed regions (%)	11.7
Disallowed regions (%)	0.6

 Table 1. Data collection and refinement statistics for AfIDH.

	<i>Af</i> IDH	<i>Ec</i> IDH _{open}
PDB code	2IV0	1SJS
<i>T</i> _m (° C)	98.5	52.6*
Resolution (Å)	2.5	2.4
RMSD (Å) C_{α} -atoms small domain/large domain/overall	0.57/0.	96/0.99
	(170 atoms/208 a	atoms/378 atoms)
RMSD (Å) C_{α} -atoms of clasp-domain	4.424 (total domain)	/0.693 (without loop)
	(42 atoms	/40 atoms)
No. residues per subunit	412	416
Charged residues (%) ^a	27.2	27.2
Polar residues (%) ^b	30.1	28.6
Hydrophobic residues (%) ^c	42.7	44.9
Aromatic residues (%) ^d	11.2	8.7
No. hydrogen bonds	687	700
No. hydrogen bonds (MM) ^e /per residues	0.62	0.63
No. hydrogen bonds (MS) ^f /per residues	0.15	0.16
No. hydrogen bonds (SS) ^g /per residues	0.061	0.048
No. inter-subunit hydrogen bonds	23	22
No. ion pairs per dimer (4/ 6/8 Å)**	53/104 /175	58/110 /181
No. ion pairs per residue in dimer $(4/6/8 \text{ Å})$	0.129/0.252/0.425	0.139/0.264/0.435
Volume $(x10^4 \text{\AA}^3)$	8.5	8.4
Accessible surface area of dimer (\AA^2)	32406	32527
Buried inter-subunit surface (% of MSA) ^h	14.7	15.6
No. residues forming two ion pairs (4/6/8 Å)	14/41/60	18/50/75
No. residues forming three ion pairs (4/6/8 Å)	0/10/42	0/6/22
No. 2/3/4 membered intra-subunit networks (4 Å)	27/8/0	24/12/0
No. 2/3/4 membered inter-subunit networks (4 Å)	0/2/2	0/2/2
% of charged residues forming ion pairs (4/6/8 Å)	46/78/92	45/76/89
Distribution of hydrophobic/polar/charged	20.7/30.1/49.3	25.6/23.1/51.3
residues at accessible surface (%)		
Distribution of hydrophobic/polar/charged	47.0/27.3/25.7	50.4/19.7/29.9
residues at interface (%)		

Table 2. Characteristics of AfIDH and EcIDH

^{*} Data from (Karlström et al. 2005)

- ** Å cutoff
- ^a Charged residues: R, K, H, D, E
- ^b Polar residues: G, S, T, Y, N, Q, C
- ^c Hydrophobic residues: A, V, L, I, W, F, P, M
- ^d Aromatic residues: W, F, Y, H
- ^e MM, main-chain/main-chain hydrogen bonds
- ^f MS, main-chain/side-chain hydrogen bonds
- ^g SS, side-chain/side-chain hydrogen bonds
- ^h MSA, Molecular Surface Area

Table 3. Kinetics of chimeras and wild-type enzymes. Kinetic parameters were determined at 40°C (wild-type *Ec*IDH and *Ec*IDH/*Af*IDH_{clasp}) or 60 °C (wild-type *Af*IDH and *Af*IDH/*Ec*IDH_{clasp}).

		Kr	n	kca	at	$k_{\rm cat}/$	K _m
		(μΝ	(1)	(s ⁻¹)*	(µM⁻	¹ s ⁻¹)
Enzyme	kDa	Isocitrate	NADP ⁺	Isocitrate	NADP ⁺	Isocitrate	NADP ⁺
EcIDH	45.8	40.5	39.2	106.3	88.1	2.6	2.2
EcIDH/AfIDH _{clasp}	45.4	26.1	56.8	71.7	86.6	2.7	1.5
AfIDH/EcIDH _{clasp}	46.2	519	15.7	219.0	155.1	0.4	9.8
<i>Af</i> IDH	45.8	332	16.5	254.6	219.3	0.7	13.3

* per catalytic site

	T _m	t_{1}	₂ (min)	Topt
Enzyme	(°C)			(°C)
EcIDH	52.6*	24.2	(at 50 °C)	50
EcIDH/AfIDH _{clasp}	56.4	36.5	(at 50 °C)	55
AfIDH/EcIDH _{clasp}	80.0	108	(at 65 °C)	70
		Α	(at 70 °C)	
<i>Af</i> IDH	98.5 ^{**}	167	(at 65 °C)	~ 90
		106	(at 70 °C)	

Table 4. Thermal properties of chimeras and wild-type IDH.

^A 6.2 % remaining activity after 5 min of incubation.

* Data from (Karlström et al. 2005)

**Data from (Steen et al. 2001)

Name	Primer sequence	Product
P1	5 caccatggaaagtaaagtagttgttc-3.	Wild-type Ec idh (K12) in pET101/D-TOPO
P2	5' - cattacatgttttcgatgatcgcg-3'	
	Step 1	
P3	5 - gacgtgtacgccggtatagaggcctcatgacagccctgaggctgcgaggattaggagg ttcctgcgtgaagagatg- 3	
P4	5'P - ttccgagttttcacggaagataacc - 3'	<i>Ec idh/Af idh</i> in pET101/D-TOPO
	Step 2	
P5	5 - ttcctcgcagaggaggtcgggatatcgataagggaggagggcgcggcataggggttaagccg tgttcggaagaaggc - 3	
P6	5.P - cctcctaatcctcgcagcctcagg - 3.	
P7	5' - cgcataaatgtcttcggtgttctccctgaaaataacg - 3'	
P8	5' - gagaacaccgaagacatttatgcgggtatcgaatgg - 3'	
6d	5' - aaactegetgateggettaatacegataceacaatgtte - 3'	
P10	5' - ggtattaagccgatcagcgagtttgccaccaagag - 3'	Af idh/Ec idh by overlap extension PCR
P11	5 - ccagtagcgctcgaaggagg - 3	
P12	5' - cattacatgttttcgatgatcgcg - 3'	
P13	5 - acacaagcttgaaggagatatacatatgcagtacgagaaggtcaaacctcc - 3	
P14	5' - caggatcctcatagcgactgcaggttttcaacc - 3'	
P15	5' - caccatgcagtaggaggtcaaacctcc - 3'	Af idh/Ec idh
P16	5' - caggatcctcatagcgactgcaggttttcaacc - 3'	in pET101/D-TOPO

بنام لم $^{\prime}$ L^{\prime} vild_tw Ę . ŧ d foi Table 5. Prin

Figure Legends

Figure 1. The structure of *Af*IDH was crystallized and resolved without substrate and cofactor in the active site, however, a high concentration of zinc was present in the crystallisation condition. Two Zn^{2+} -ion was found tightly bound to Asp301, Asp305, Asp277' in the active site of subunit A (A) and subunit B (B) of *Af*IDH. Amino acids from subunit A is coloured blue and amino acids from subunit B is coloured red. Amino acids from both subunits are involved in binding of zinc.

Figure 2. Structure-based sequence alignment based on the secondary structural assignments of *Af*IDH with *Ec*IDH (PDB code 1SJS), *Ap*IDH (PDB code 1TYO) and *Bs*IDH (PDB code 1HQS). Helices and strands appear as cylinders and arrows. Conserved residues are coloured green and residues showing conservation of hydrophobic character are in yellow. Conservation of small size has smaller font. Amino acids involved in binding of isocitrate in *Ec*IDH, *Ap*IDH and *Bs*IDH are indicated by red squares, and loops and helices involved in binding of NADP⁺ are underlined by orange bars. Residues involved in the clasp-domain are underlined by a red bar. Residues within structural equivalent regions are boxed. Secondary structure elements were given the nomenclature as implemented in *Ec*IDH (Hurley et al. 1989). Sequence numbering according to *Af*IDH is red.

Figure 3. Overlay of the small (A) and large (B) domains of AfIDH and EcIDH_{open}.

Figure 4. Overlay of large domain from *Af*IDH and *Ec*IDH showing the difference in aromatic residue content in the N-terminus. Red and Blue; *Af*IDH (Tyr3, Phe64 and Phe91), Green; *Ec*IDH (Tyr95).

Figure 5. Clasp-domain from AfIDH (blue) and EcIDH (green) showing an aromatic cluster in the AfIDH structure compared to EcIDH. EcIDH contained a Met residue in the corresponding position of Phe179 in AfIDH.

Figures



Fig.2









Fig.4



