

NMR Characterization of LtL oil

by

Lu Zhang



Master thesis in Chemistry

University of Bergen

June 2020

ACKNOWLEDGMENTS

I would like to express deep gratitude to my supervisor Professor Tanja Barth for her constant support and encouragement without which this study would not have reached fruition. She has provided great support both on my research and my life. I would also like to thank Camilla Løhre (PhD), Hilde Vik Halleraker and Solmaz Ghoreishi (PhD) for helping me with the lab work, and our research group for inspiring presentations during group meetings.

I would also like to direct gratitude towards the staff of the Department of Chemistry, and especially Unni Lange Buanes for the consultant of extension of master study. I would further like to thank Jarl Underhaug and Jose Carlos Reyes Guerrero from NMR team for their great guidance with required software and making sure my NMR samples were run in a timely fashion.

Most importantly, I would like to thank my parents and my husband Ren for believing in me and for their understanding and unconditional support. Last, but not least, I would like to thank my little girl Aria who was born during my master study period, being a mother is a wonderful gift that is unimaginable to me.

ABSTRACT

The need for energy is increasing sharply due to the rapid increase in the world's population and developing technologies, while the current energy resources with limited reserves are decreasing. Lignocellulosic biomass is the most viable choice for producing high value chemicals, liquid fuels, and energy from a renewable source. Lignin, in particular, is of interest because it serves as the only significant source of renewable aromatic carbon.

Lignin can be converted into bio-oil by Lignin-to-Liquid solvolysis, which includes addition of lignin, a hydrogen donor (formic acid) and a solvent into a high pressure/high temperature reactor and heating the reaction mixture, as a closed system.

The bio oil is a very complex mixture containing phenolic compounds, carbohydrates, furans, ketones, aldehydes, carboxylic acids, and water. In this thesis, nuclear magnetic resonance (NMR) spectroscopy, including ^1H , ^{13}C , 2D HSQC spectra, was used to characterize the functional groups in LtL oil converted with different solvolysis conditions.

^{13}C NMR suffers from low sensitivity due to its low natural abundance and long relaxation times. The determination of optimum relaxation reagent $\text{Cr}(\text{acac})_3$ concentrations for quantitative ^{13}C has also studied.

ABBREVIATIONS

NMR	Nuclear magnetic resonance
Cr(acac) ₃	Chromium (III) acetylacetonate
D1	Relaxation delay
DMSO-d ₆	Deuterated dimethyl sulfoxide
FA	Formic acid
FT	Fourier Transform
FID	Free Induction Decay
FT-IR	Fourier transform infrared spectroscopy
GC	Gas chromatography
GC-MS	Gas chromatography -Mass spectrometry
HSQC	Heteronuclear Single Quantum Coherence
LtL	Lignin-to-Liquid
ns	Number of scans
NOE	Nuclear Overhauser Enhancement
ppm	Parts per million ($\times 10^{-6}$)
TMS	Tetramethylsilane

CONTENTS

ACKNOWLEDGMENTS	II
ABSTRACT	III
ABBREVIATIONS	IV
Chapter 1 Introduction	1
1.1 Potential of biomass	1
1.2 Lignin	4
1.3 Thermal conversion process	9
1.4 Lignin-to-Liquid (LtL) solvolysis	10
1.5 NMR characterization of bio-oil	12
1.5.1 Analysis methods of bio-oil	12
1.5.2 NMR analysis of bio-oil	15
Chapter 2 Method	20
2.1 Nuclear magnetic resonance introduction	20
2.1.1 Basic principle of NMR	20
2.1.2 Chemical shift	22
2.2 T1 Relaxation	23
2.3 The NOE effect	26
2.4 The Free Induction Decay (FID)	28
2.5 Quantitative ¹³ C NMR experiment	28
2.6 2D HSQC NMR experiment	30
2.7 Objective	32
Chapter 3 Experiment	34
3.1 Materials	34
3.2 Sample preparation	36
3.2.1 Standard compound samples	36
3.2.2 Mixture samples	37
3.2.3 Samples with relaxation reagent	37

3.2.4 LtL-oil samples	38
3.3 NMR experiments	38
3.3.1 NMR spectrometers	38
3.3.2 Procedure of running a sample on NMR instrument	41
3.3.3 Pulse program	42
3.3.4 Data processing in Topspin	47
Chapter 4 Results	48
4.1 Deuterated solvent for NMR samples	48
4.2 T1 relaxation NMR experiment	49
4.3 Standard compound NMR experiments	50
4.4 Bio-oil based mixture NMR experiments	55
4.5 LtL-oil NMR experiments	56
4.5.1 Two LtL- oils with and without formic acid ¹³ C	57
4.5.2 Four LtL- oils converted under different conditions	57
Chapter 5 Discussion	60
5.1 The choice of deuterated solvent	60
5.2 The choice of relaxation agent	62
5.3 Comparison of NMR spectra acquired from different magnets	64
5.4 Functional groups in LtL-oils	66
Chapter 6 Conclusion	71
Chapter 7 Further work	72
Appendix A NMR spectra	73
Appendix B NMR spectra integrations	88
References	116

Chapter 1 Introduction

1.1 Potential of biomass

Environmental concerns and growing worldwide energy demands impel scientists to investigate alternative and renewable sources of energy, fuels, and materials to shift dependence away from the fixed supply of fossil fuels. The world is now in a transition state between an energy economy that, in most nations, has an overwhelming dependence on petroleum, natural gas, and coal, to a new energy economy that will be based heavily on alternative, renewable sources of energy, including fuels derived from plants[1]. As world demand for liquid transportation fuels grows, the surplus production capacity of traditional petroleum reservoirs is approaching a vanishingly small margin, resulting in upward pressure on the global price of petroleum. Coupling this with increasing volatility in the price and uncertainty of supply, many national programs encouraging domestic production of fuels have arisen, typically focused on the use of domestic biomass resources to meet demand for liquids[2].

Biomass can be utilized for the production of process heat, steam, motive power, and electricity, and can be converted by thermal or biological routes into a range of useful energy carriers such as liquid fuels and synthesis gas. Figure 1-1 illustrates a fully integrated biorefinery cycle. The term biomass is used to describe any material of recent biological origin and includes plant materials such as trees, grasses, and agricultural crops, as well as animal manure and municipal biosolids. As a raw material, biomass is a nearly universal feedstock due to its domestic availability and renewability. Indeed, until the widespread utilization of crude oil as an energy source in the 19th century, biomass supplied the majority of the world's energy needs. In one sense, the situation has now come full circle: concern over the environmental effects of fossil-fuel

combustion, as well as disquiet about dwindling petroleum reserves - coupled with increasing global energy demand - have brought about a resurgence of interest in the utilization of biomass as an energy source[2].

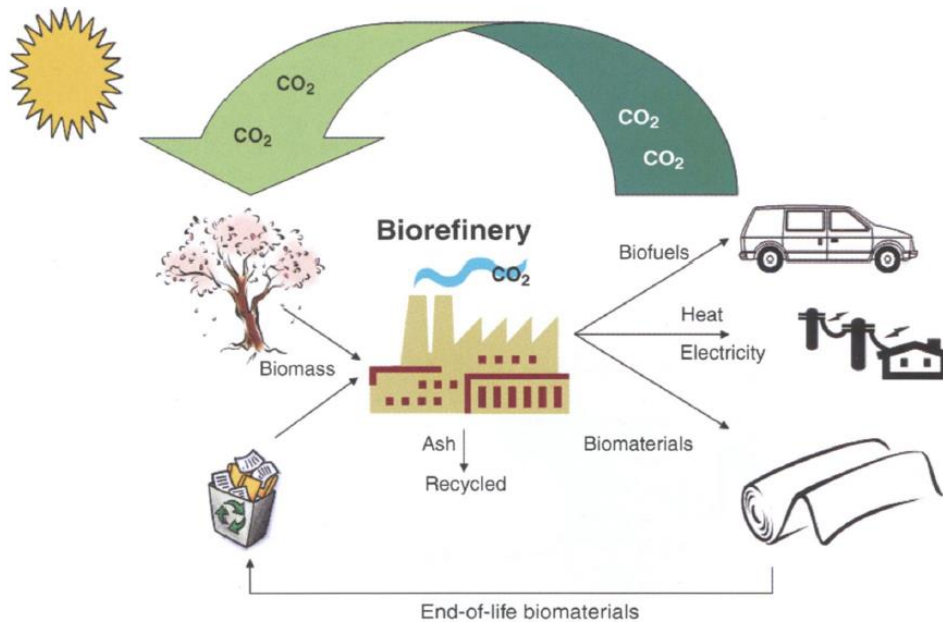


Figure 1-1 The fully integrated agro-biofuel-biomaterial-biopower cycle for sustainable technologies[3]

The benefits most frequently cited for biomass utilization as an energy source can be summarized as follows[4, 5]:

- Fuel-supply diversification: biomass is a widespread resource, the utilization of which can diversify the fuel supply and in turn lead to an energy supply that is more secure, i.e. less subject to geopolitical constraints.
- Reduction of greenhouse gases: substituting fossil fuels with biofuels can lead to a reduction in greenhouse gases, principally CO₂. This stems from the fact that the amount of CO₂ released from combustion of the biomass is inherently equal to that fixed by the plant during its growth. However, the extent of greenhouse-gas reduction is dependent on a number of factors, including the degree to which fossil fuels are used in the production and distribution of biofuels. Another consideration is the extent to which stored CO₂ is released to the atmosphere during the clearing

and tilling of grassland and forests that are to be used for biomass cultivation.

- Increased rural income: the increased agricultural activity associated with the production of energy crops can generate employment in rural areas and result in increased farm income, thereby reversing the trend of rural depopulation prevalent in many countries.
- Restoration of degraded land: If appropriate crops are selected, degraded land (currently unsuited for the production of food crops) can be utilized for biofuels production.

Given its abundance and affordability, lignocellulosic biomass (Figure 1-2) that consists of cellulose, hemicellulose and lignin has been extensively investigated and utilized to obtain fuels, such as bioethanol, biodiesel, hydrocarbon fuels, and chemicals[6].

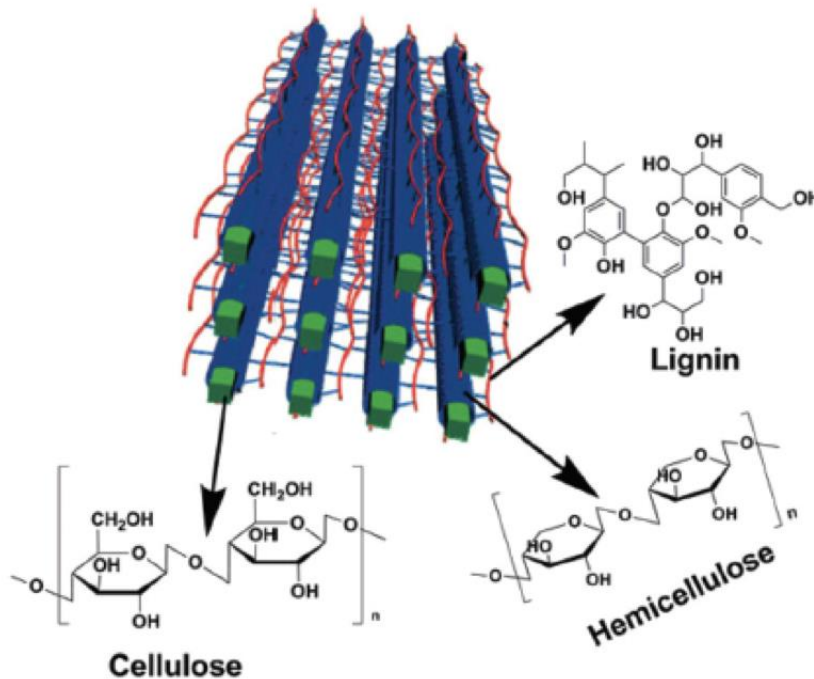


Figure 1-2 Schematic analysis of the location and structure of three components in the lignocellulosic biomass[7]

Cellulose is a high molecular weight polymer that consists of linear β -1,4

glycoside linked chains of D-glucose units[8], whereas hemicellulose is a heterogeneous polymer that comprises several different types of monosaccharides. Lignin is one of the most abundant natural polymers (only second to cellulose), which constitutes 20%-30% of the weight of lignocellulosic biomass[9]. From the perspective of high energy content and polymer structure, lignin is considered a promising source of renewable fuels and chemicals.

1.2 Lignin

Lignin is present in most of the terrestrial plants with content varies from 10-40% and to lend rigidity to cell walls[10, 11]. Currently, lignin is a by-product mostly from pulp and paper industry, and the so-called “black liquor” which is the discarded lignin stream has long been underutilized [12]. Also, with the development of biorefinery industry, biorefinery lignin also becomes more available in recent years. Therefore, finding value-added applications for biorefinery lignin is of great importance. Lignin provides all vascular plants, such as woody plants and trees, their rigidity and water-impermeability and protects them against microbial and fungal destruction of cellulosic fibres [13]. Lignin has aromatic structures and relatively high carbon content, which make them a good candidate for producing carbon-based materials. Lignin is also biodegradable and renewable that is desired to maintain an environmentally friendly, sustainable and carbon-neutral community[14].

However, lignin does not have well-defined structures and properties. The complex structure of lignin contains numerous ether linkages, –OH groups, and methoxy groups. This cross-linked macromolecule composed of three types of monolignols, also known as phenolic monomers, including p-coumaryl alcohol (H), coniferyl alcohol (G) and sinapyl alcohol (S) as shown in Figure 1-3 [15].

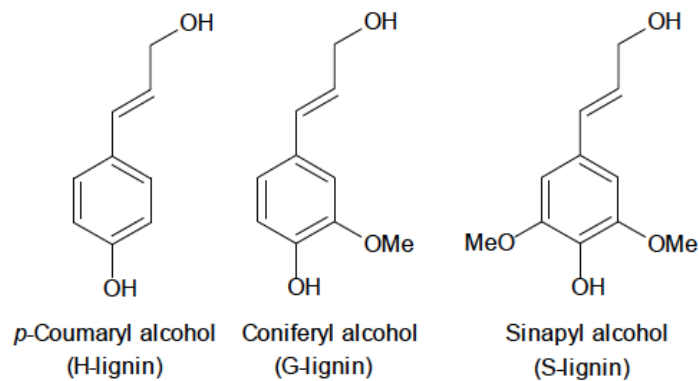


Figure 1-3 The three types of monolignols in lignin

These three units are inter-crosslinked via in-situ free radical polymerization to form a three dimensionally complex network. Figure 1-4 shows these units are interconnected through various interunit linkages α -O-4, β -O-4, β - β , β -1, β -5, 5-5', and 4-O-5 macromolecular lignin.

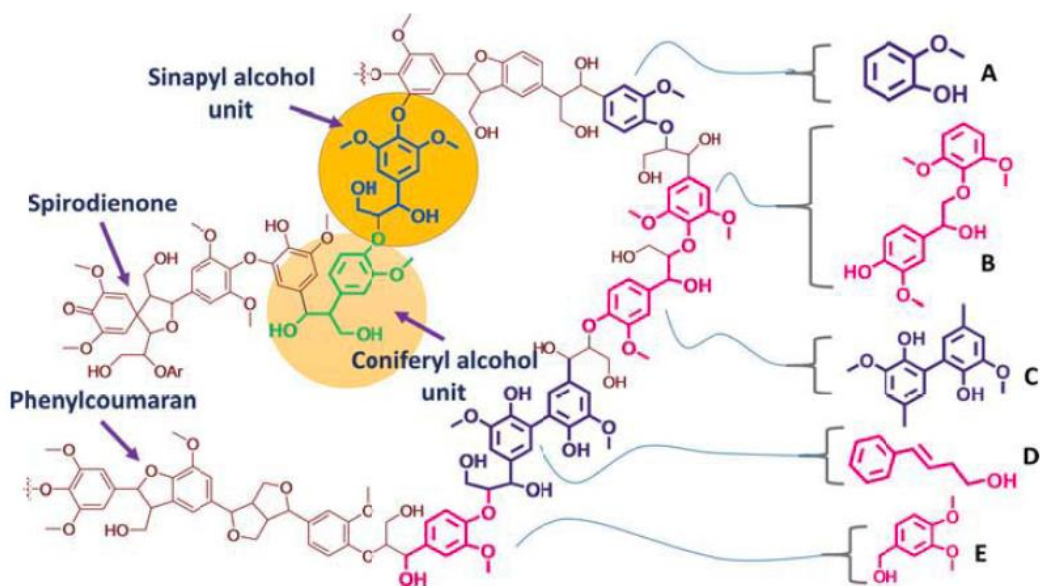


Figure 1-4: Schematic depiction of lignin showing various linkages and lignin model compounds. Model (A) phenol and methoxy functionalities, (B) a β -O-4 linkage, (C) a 5-5' linkage, (D) a propyl side chain, and (E) a benzylic group. [16]

Lignification typically occurs after the polysaccharides have been laid down in the cell wall[17]. The units in the lignin polymer are linked by a variety of chemical bonds

that have different chemical properties. Lignin composition varies among plant species:

- with some exceptions, lignin from gymnosperms is composed of G-units only (with minor amounts of H-units);
- angiosperm dicot lignin is composed mainly of G- and S-units (with very little H);
- lignin from grasses (monocots) contains G- and S-units at comparable levels, and more H-units than dicots. [18]

The majority of secondary wall formation in most plants occurs in the water conducting cells of the xylem, which tend to be enriched in G lignin, and in the structural fibers, which in angiosperms typically contain higher levels of S subunits[19]. Lignified cell walls are also found in sclereid cells, endodermal tissue in roots, and in specialized cells of anthers and some seed pods, where they are important for the dehydration-driven release of pollen and seeds, respectively. The wood of both gymnosperm and angiosperm trees is composed largely of the secondary cell walls of vascular tissue and associate fibers, and as such, of lignin. It is important to note that lignin biosynthesis is also of crucial importance in herbaceous (non-woody) species.

Less abundant units than the three main monolignols have been identified from a variety of species and may be incorporated into the polymer at varying levels[18]. Particularly common, less abundant subunits include ferulates (which form crosslinks between lignin and hemicelluloses), conifer aldehyde, and acylated monolignols containing acetate, p-coumarate, or p-hydroxybenzoate moieties[19]. Lignin composition also varies among cell types and can even be different in individual cell wall layers. Lignin composition is also influenced by environmental conditions; for example, lignin in compression wood is enriched in H-units[17]. Hence, both developmental and environmental parameters influence the composition and, thus, the structure of the lignin polymer.

Following their synthesis in the cytoplasm, lignin monomers are exported to the extracellular space (the apoplast) for incorporation into the lignin polymer[19]. After reaching the apoplast, monolignols undergo single-electron oxidation by wall-bound

laccases and/or peroxidases to form radical species. Then bond formation can occur between two such radicals (any combination of two monolignols, a monolignol and an elongating polymer end, or two polymer ends) in a process known as bimolecular radical coupling. Alternatively, activated monolignols can be involved in a radical transfer reaction.

The three different monolignols, and the subunits derived from them can freely form bond with one another[19]. Depending on the ring-substitution pattern of the molecules (H, G and S) and their location (free or attached to the polymer), multiple sites may be available for bond formation, resulting in a variety of potential coupling patterns, mainly including β -O-4, β -5 and β - β linkages as the central (β) carbon of the monolignol tail is most frequently involved, see an example of lignin structure predicted by NMR analysis in Figure 1-5.

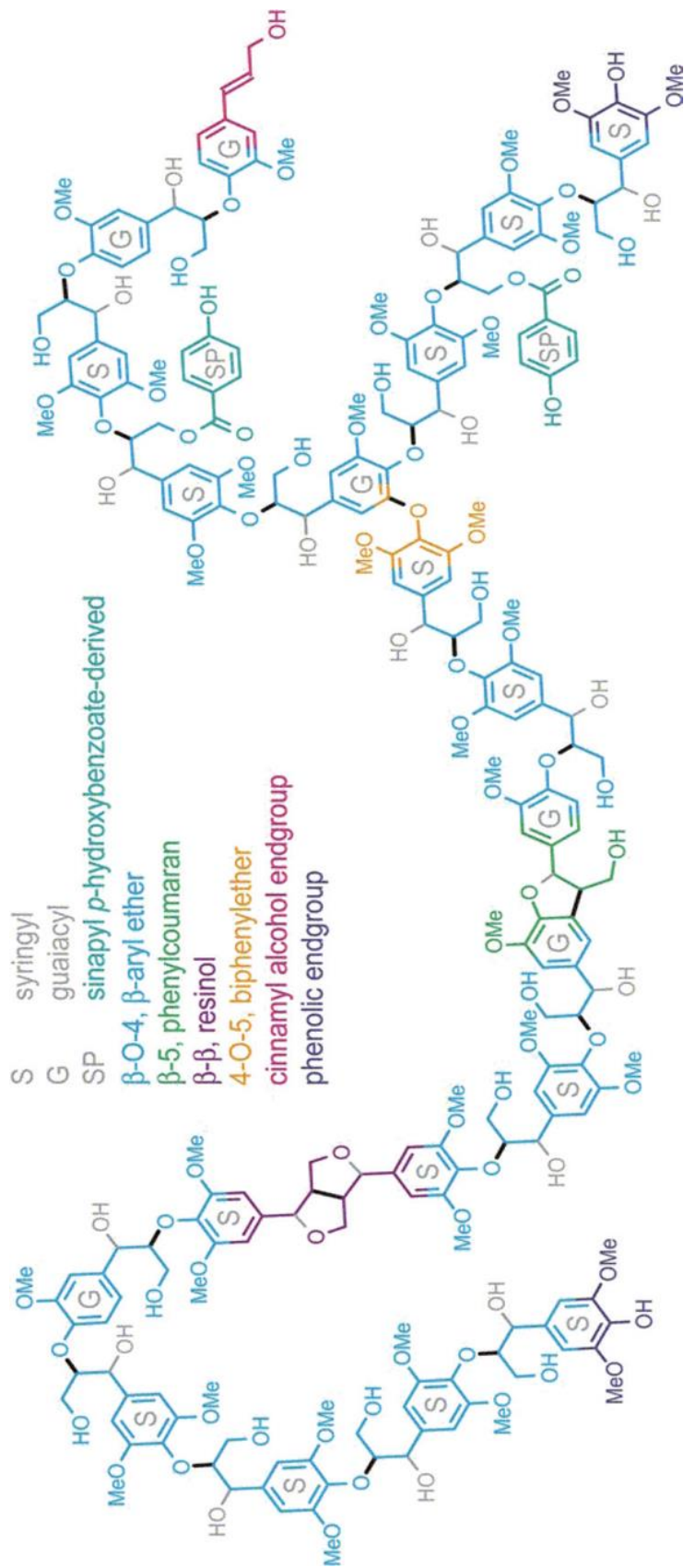


Figure 1-5 Lignin polymer from poplar, as predicted from NMR-based lignin analysis. [18]

1.3 Thermal conversion process

Lignocellulosic biomass feedstock can be converted to biofuels and chemicals through multiple processes. Conversion technologies fall into two main categories: biochemical conversion and thermochemical conversion. The thermochemical platform for conversion of lignocellulosic biomass aims to efficiently produce biofuels and chemicals via processes that use heat and chemistry[20].

There are three main thermal processes available for converting biomass to a more useful energy form – pyrolysis, gasification and combustion. Figure 1-6 summarizes their products and applications[2]. monoxide and hydrogen known as synthesis gas or syngas.

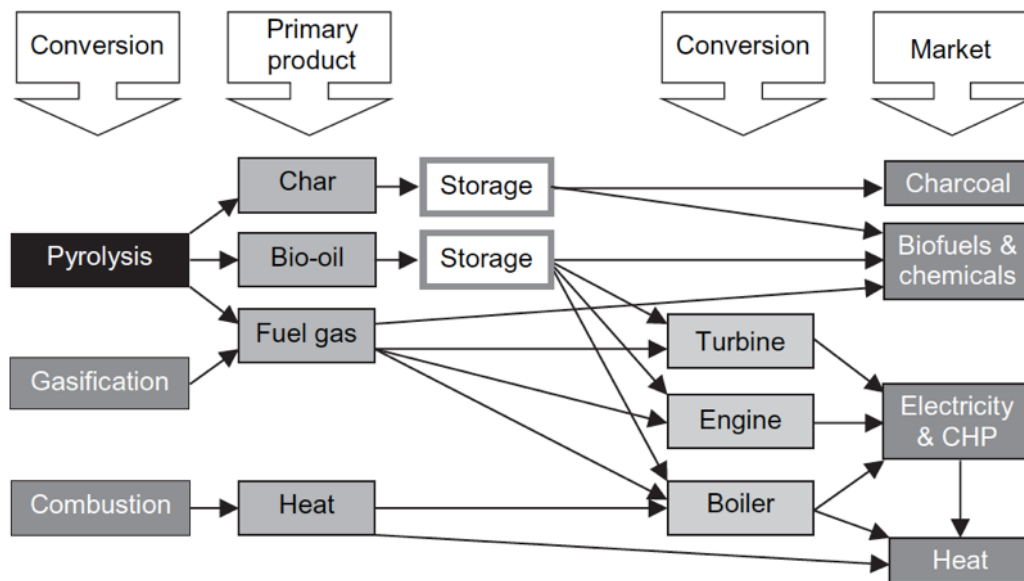


Figure 1-6 Products from thermal biomass conversion.[2]

Combustion is the thermal conversion of organic matter with an oxidant (generally oxygen) to produce primarily carbon dioxide and water. The oxidant is in stoichiometric excess, i.e. complete oxidation. Biomass combustion mainly leads to heat energy which can be utilized for steam and/or electricity production[21].

Gasification is the thermal conversion of organic matter at elevated temperatures

and in oxygen-deficient conditions (reducing conditions) to produce primarily permanent gases, with char, water, and condensable as minor products. The main permanent gases resulting from biomass gasification are the mixture of carbon

Pyrolysis is thermal decomposition occurring in the absence of oxygen. It is always also the first step in combustion and gasification, but in these processes, it is followed by total or partial oxidation of the primary products. Lower process temperatures and longer vapor residence times favor the production of charcoal. High temperatures and longer residence times increase biomass conversion to gas, and moderate temperatures and short vapor residence time are optimum for producing liquids. Three products are always produced, but the proportions can be varied over a wide range by adjustment of the process parameters.

1.4 Lignin-to-Liquid (LtL) solvolysis

Solvolysis is pyrolysis in the presence of a solvent. The process is known as direct liquefaction although this overlooks the prospects for also fractionating or gasifying biomass via solvolysis [2]. A variety of solvents can be employed, their physical and chemical properties determining the temperature and pressures at which the processes operate. Because water is a major constituent of biomass, it is frequently used as a solvent, in which case solvolysis is referred to as hydrothermal processing. In comparison to atmospheric pyrolysis, solvolysis provides the advantages of milder conditions and a single-phase environment due to the miscibility of the organic products in the (supercritical) solvent. Anthracene, water and ethanol have been shown to have suitable solvent properties in this context. Especially ethanol has been reported to have a very high solvency for biomass and a low critical temperature which makes it attractive for lignin depolymerization[22].

A promising and relatively new lignin reductive conversion methodology is the Lignin-to-Liquid (LtL) process. In the LtL process the lignin is converted at moderate temperatures in the presence of two additional compounds: (i) formic acid (FA) as a

substitute of molecular hydrogen, and (ii) a solvent, mostly ethanol or water. Figure 1-7 depicts the main conversion route during LtL solvolysis, excluding aliphatic reaction products from the equation, as the most recent publications report aromatics to be the dominating product structures. solvolysis, excluding aliphatic reaction products from the equation, as the most recent publications report aromatics to be the dominating product structures.

Formic acid (FA) acts, together with a co-solvent, as an in situ hydrogen donor in the liquid reaction medium during LtL-solvolysis, providing depolymerization and hydrodeoxygenation of the lignin copolymer. Thermal degradation products of FA provide CO and H₂O or CO₂ and H₂ with the latter as the preferred decomposition pathway[23]. FA has shown to deliver reactive hydrogen upon degradation during LtL-solvolysis, together with CO₂, in a more reactive way than for H₂ gas[22].

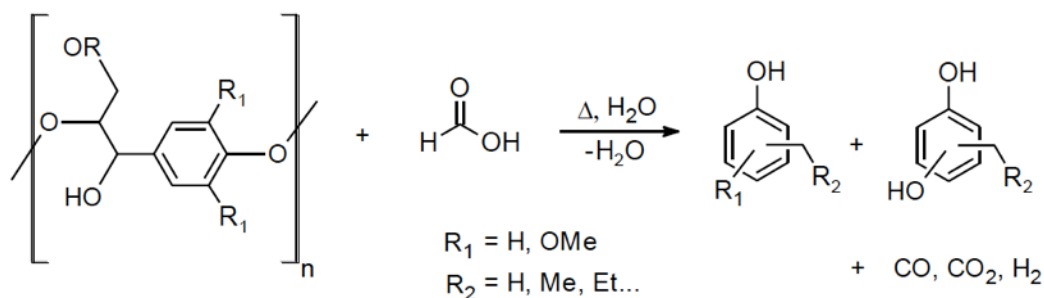


Figure 1-7 :Overview of the main conversion route; Lignin is a methoxylated, phenolic polymer which in course of the solvolytic reaction is degraded to phenol monomers with different substitution patterns and aliphatic compounds. Simultaneously a hydrodeoxygenation occurs in which formic acid (FA) serves as the hydrogen donor. Water is generated during the reaction[24].

The solvent is added to create a more homogeneous reaction environment and increase the mass and heat transfer rates. This allows operating at milder reaction conditions and increases the selectivity towards the production of bio-oil. Water is considered as a promising reaction media due to its abundance, low cost and green nature, although it has a low ability to solubilize most type of lignin. Ethanol has also been identified as one of the most promising alternatives due to its very good solvent properties for biomass and low critical temperature; furthermore, it is one of the main

fuel type products obtained from the conversion of lignocellulosic feedstock[25].

1.5 NMR characterization of bio-oil

1.5.1 Analysis methods of bio-oil

To understand the possible decomposition pathways of lignin and to find an effective upgrading method, several analytical techniques have been employed to analyze bio-oil. Popular analytical techniques which have been widely employed for bio-oil analysis are gas chromatography (GC), high performance liquid chromatography (HPLC), GPC, GC–MS, liquid chromatography electrospray ionization mass spectrometry (LC-ESI-MS), Fourier transform infrared spectroscopy (FTIR), NMR and thermo gravimetric analysis (TGA)[26]. In bio-oil analysis, results from many analytical techniques complement each other leading to unequivocal identification of its chemical components. Important prerequisite of GC based detection is the volatility of analyses due to which gases, volatile non-polar and chemical derivatives of polar organic compounds in bio-oils are largely amenable. HPLC is usually associated with the analysis of polar and thermo-labile organic compounds. Mass spectrometer is a powerful detector to chromatographic systems as well as an independent technique for studying behavior of molecules and advanced gaseous phase reactions. GC–MS and LC–MS have immensely contributed in the identification of bio-oil components. FTIR analysis reveals presence of functional groups in compounds and helps in establishing molecular structure. NMR is essential in determining the nature and types of covalently bonded nuclei (hydrogen or carbon) within a molecule. It gives an estimated number of such nuclei under observation through peak area integrations.

GC-MS analysis

Mass spectrometry (MS) provides crucial elemental and structural composition of organic molecules. Key technological advancements in sample introduction, ionization, and analyzers in the past two decades have revolutionized the research which has

largely focused upon biological perspectives. Bio-oils have comparable degree of compositional complexity; hence sufficient scope exists in the investigations utilizing higher mass resolving power of MS to extract accurate molecular level information. Researchers in of the area of bio-oil analysis have extensively involved quadrupole analyzer based mass spectrometers. This analytical approach is quite suitable for characterization of small molecular weight organic compounds from aliquots obtained after sample preparation and offers labor and cost effectiveness. The analysis may involve GC to introduce sample to achieve chromatographic separations. The obtained results are conclusive in electron ionization (EI) mode since mass spectral database exists.

GC-MS library search gave a tentative identification of the compounds present in the bio-oils. There are approximately a hundred compounds were detected in lignin bio oil by GC-MS and almost all of them contain a phenol structure. Furthermore, phenol, acetovanillone, cresols, guaiacol, 4-ethylphenol, syringaldehyde, acetosyringone, 4-methylguaiacol, catechol, 3-methylcatechol, 4-methylguaiacol, 4-vinylguaiacol, vanillin, syringol, eugenol isoeugenol, and acetovanillone have been reported.[27]

Figure 1-8 depicts an example of GC-MS chromatogram showing a representative structural distribution for the LtL-oils[28].

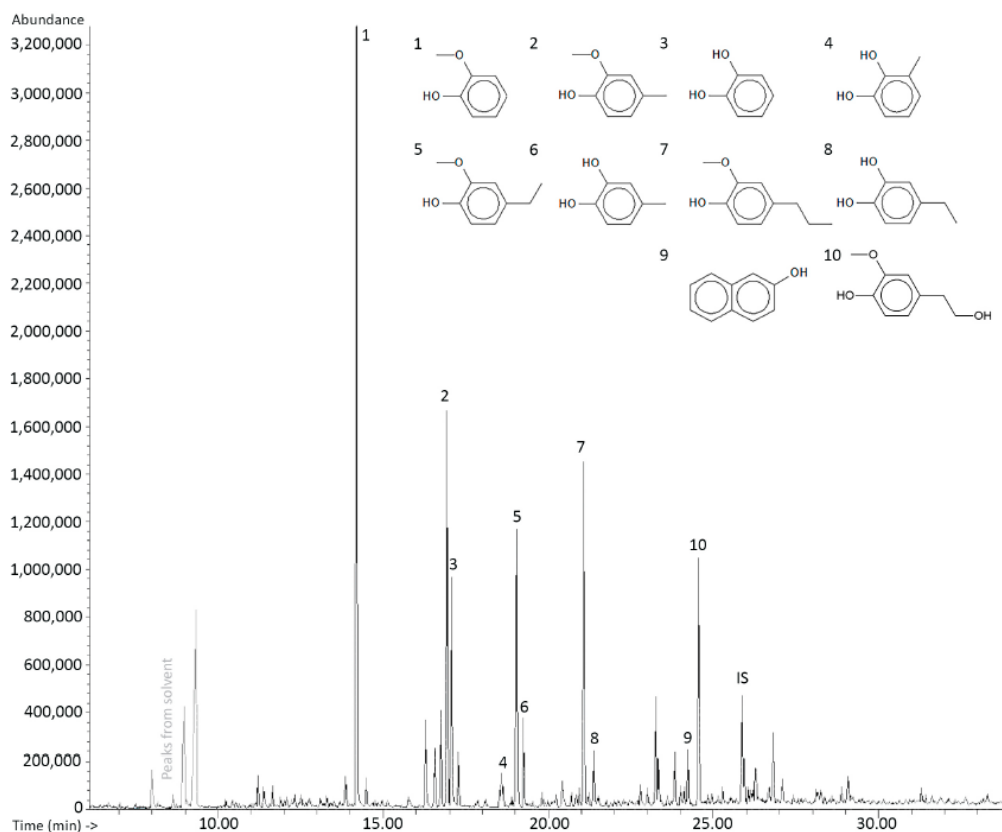


Figure 1-8 Chromatogram of a LtL-oil representing a typical compositional distribution.

FTIR analysis

Due to the limitation of volatility of high molecular weight components in the lignin bio-oil, it has been indicated that there are only about 40 % of lignin bio-oil that could be detected by GC[13]. Therefore, many researchers also try to find alternative characterization method which could analyze the whole portion of pyrolysis oil, such as FTIR. In FTIR analysis, mid-infrared region is extensively utilized to reveal the presence of various functional groups in molecules. FTIR analysis is of vital significance to study the fundamental vibrations and associated rotational–vibrational structure within a molecule. Scientific literature contains huge information on updates and upcoming trends in modern infrared spectroscopy.

1.5.2 NMR analysis of bio-oil

Nevertheless, the complexity of the lignin bio-oil limits the use of these instruments due to different issues and consequently many chemical components could not be identified. Therefore, the main goal of the present study was to investigate the structures of lignin bio-oil under different conversion conditions. Most recently, some research workers introduce NMR, including quantitative ^1H , ^{31}P , and ^{13}C -NMR, and semi-quantitative HSQC-NMR to characterize bio oils[29]. Such investigations improve understanding about lignin conversion mechanisms, which leads to more control in the degradation pathways.

Nuclear magnetic resonance (NMR) is one of the important spectroscopic techniques employed in analytical and bioanalytical chemistry. NMR spectrometers are sophisticated and largely present in research and academic institutions across the world. The operating frequencies of modern NMR spectrometers have extraordinarily increased to nearly 1 GHz. The technique is nowadays essentially employed in range of experiments on solid, liquid, and gaseous samples, over a wide temperature span. Inherent limitation associated with NMR is its relatively low sensitivity as compared to other spectroscopic techniques. Hence, trace and ultra-trace level quantitation are always represented by poor limits of detection that is several orders of magnitude poorer than mass spectrometry or emission/absorption/fluorescence based techniques. NMR is exceptionally used in the structural analysis of petroleum fractions. It is an essential tool for carrying out group type analysis and directly measures aromatic and aliphatic carbon and hydrogen distributions. It also provides important information on the carbon and hydrogen structural groupings in a molecule. ^{13}C NMR studies of coal and asphaltenes are the pioneer NMR applications in petroleum analysis[30].

In bio-oil analysis, major advantages of NMR are that the whole bio-oil can be dissolved in a suitable solvent and a qualitative assessment of the oxygen containing functionalities can be determined simply from integration of appropriate regions of the

^{13}C and ^1H NMR spectra. Bio-oils have been largely studied with NMR and the ^{13}C NMR spectra are particularly instructive because of their large chemical shift regions. ^{13}C and ^1H NMR offer a reasonable trade-off between functional group identification and analytical measurement effort. Accuracy however depends upon selection of chemical shift regions, baseline compensation, and correction for incomplete longitudinal relaxation effects.

NMR techniques are used to quantitatively identify functional groups in bio-oil. NMR methods previously used for biomass and products from thermochemical conversion of biomass[29]. Proton NMR is widely applied in bio-oil characterization. Figure 1-9 presents typical assignments of an ^1H NMR spectrum for bio-oil from pinewood[29].

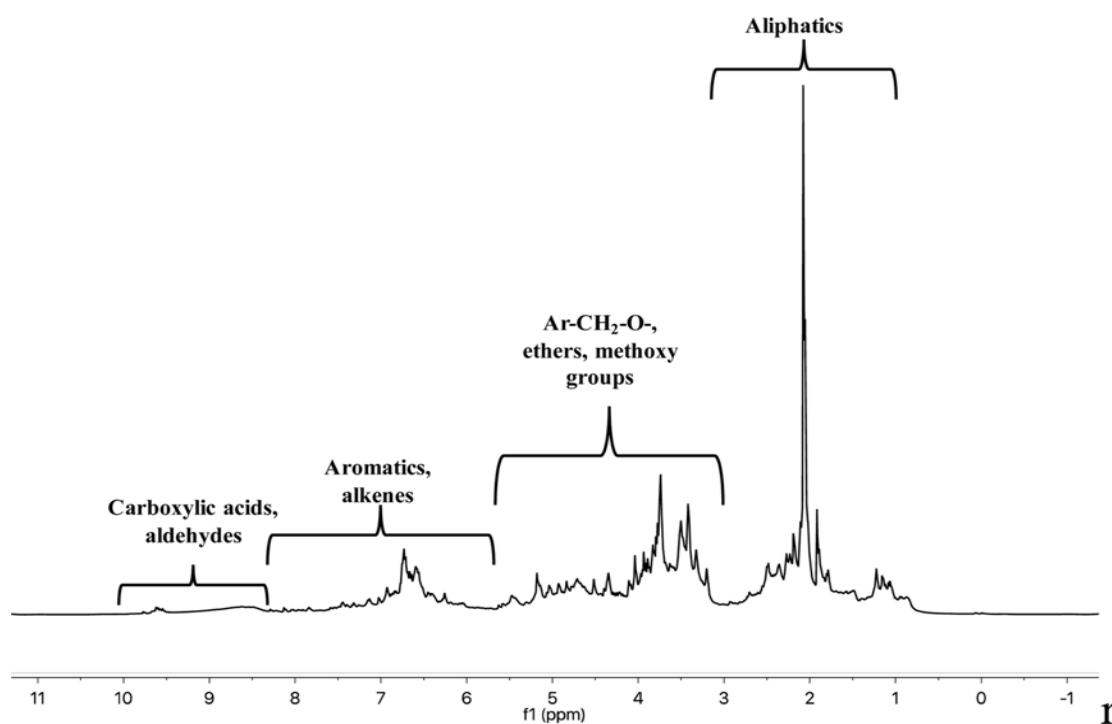


Figure 1-9 ^1H NMR spectrum of the water-insoluble fraction of pine wood pyrolysis oil measured in DMSO-d_6 [29]

^{13}C NMR spectroscopy provides carbon information on bio-oil components. In comparison to a ^1H NMR spectrum, a ^{13}C NMR spectrum benefits from a broader chemical shift range, which means less spectral overlap. The limitation of quantitative ^{13}C NMR is its low sensitivity and long experiment time due to the low natural

abundance of ^{13}C nuclei and pulse delay times. An example of the characterization of phenolic compounds of the bio-oil obtained from spruce wood in ^{13}C NMR spectra are shown in Figure 1-10 [31]. A typical structural assignment range of functional groups was developed, and this analysis is summarized Table 1-1 [32].

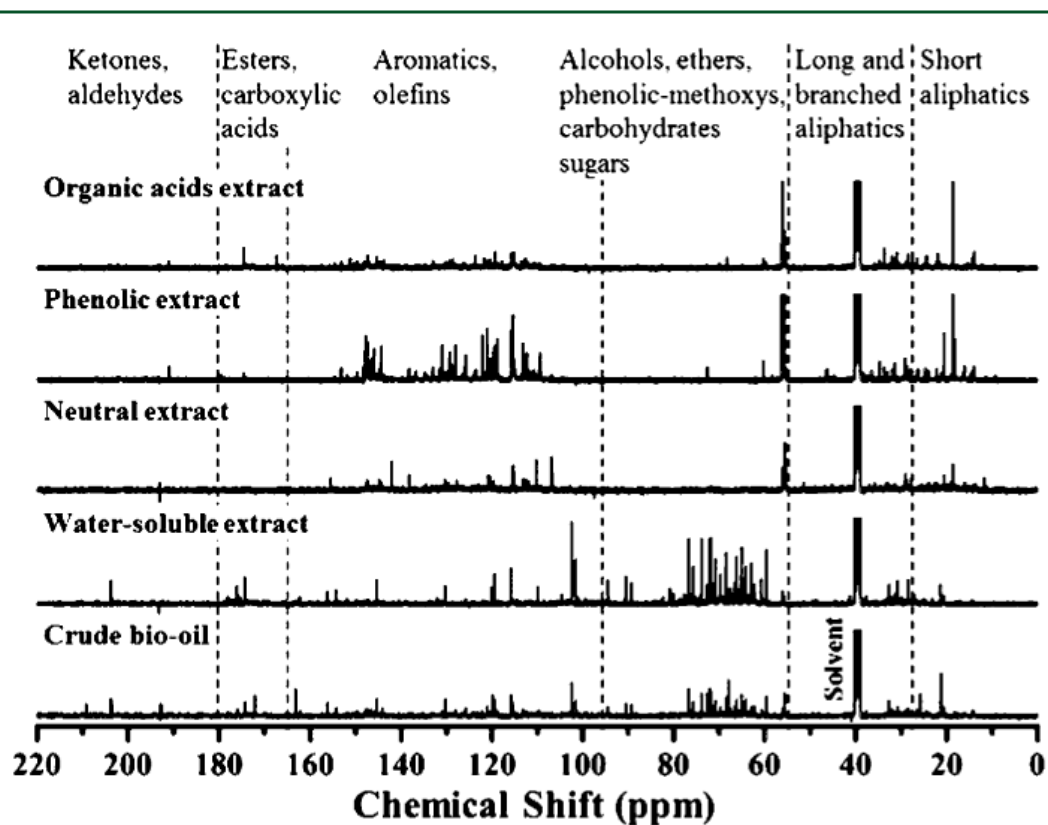


Figure 1-10 ^{13}C NMR spectra of the crude bio-oil, water-soluble extract, neutral extract, phenolic extract, and organic acids extract.[31]

Functional group	Integration region (ppm)	Examples	
Carbonyl or Carboxyl bond	215.0 – 166.5		
Aromatic C-O bond	166.5 – 142.0		
Aromatic C-C bond	142.0 – 125.0		
Aromatic C-H bond	125.0 – 95.8		
Aliphatic C-O bond	95.8 – 60.8		
Methoxyl-Aromatic bond	60.8 – 55.2		
Aliphatic C-C bond	General	55.2 – 0.0	
	Methyl – Aromatic	21.6 – 19.1	
	Methyl – Aromatic at ortho position of a hydroxyl or methoxyl group	16.1 – 15.4	

Table 1-1 A ^{13}C NMR Chemical Shift Assignment Range of Lignin bio-oil [32]

The chemical components of bio-oil are very complex, and only about 40 wt % could be analyzed by GC[13]. In addition, FT-IR has a limited ability to quantitatively analyze complex pyrolysis mixtures. In contrast, NMR has a much higher resolution and provides detailed structural information of lignin bio-oils. The possible pathway of primary decomposed functional groups- aliphatic hydroxyl, carboxyl, and methoxyl groups and ether bond, in lignin bio-oil conversion is shown in Figure 1-11[30].

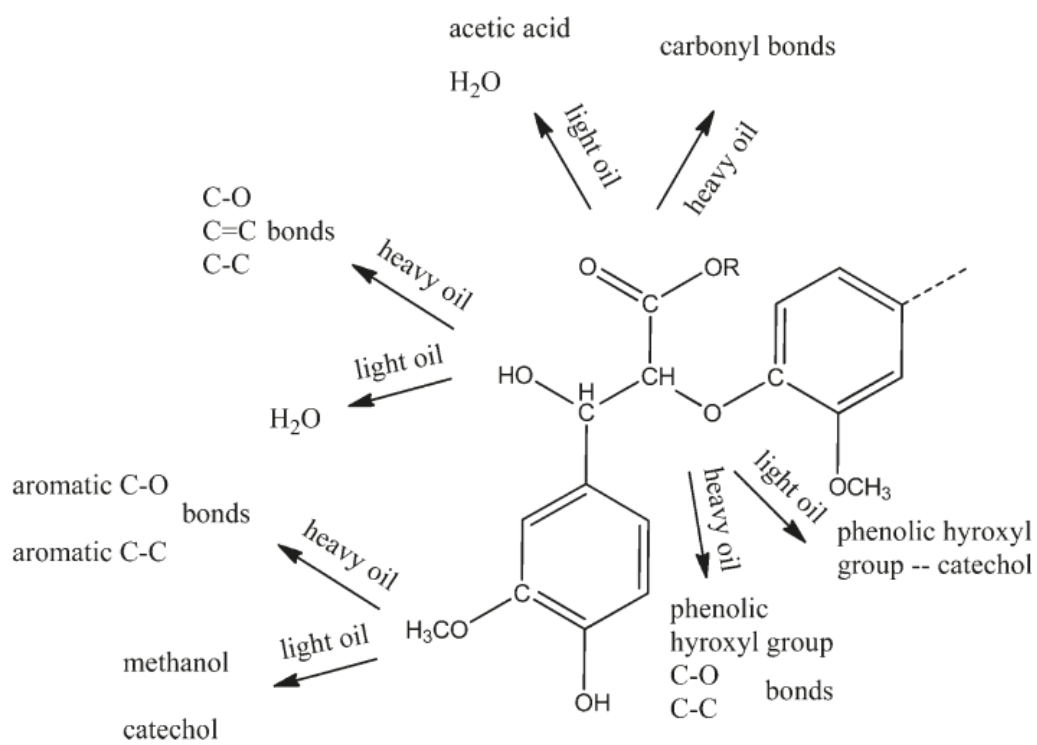


Figure 1-11 Possible pathway of primary decomposed functional groups in lignin bio-oil.[30]

Chapter 2 Method

Various NMR experiments have been employed to better understand the components and structures of thermally generated bio-oils. ^1H and ^{13}C NMR analyses have been widely used to investigate the structural hydrogen-carbon framework of bio-oils. Moreover, selective analysis of functional groups in the bio-oils through NMR analysis techniques allows a deep understanding of the characteristics of bio-oils.

2.1 Nuclear magnetic resonance introduction

2.1.1 Basic principle of NMR

Nuclear magnetic resonance, or NMR as it is abbreviated by scientists, is a phenomenon which occurs when the nuclei of certain atoms are immersed in a static magnetic field and exposed to a second oscillating magnetic field. Some nuclei experience this phenomenon, and others do not, dependent upon whether they possess a property called spin. Most of the matter the composed of molecules can be examined with NMR. Molecules are composed of atoms which can be thought of as a small magnetic field and will cause the nucleus to produce an NMR signal.

The nuclei of many elemental isotopes have a characteristic spin (**I**). Some nuclei have integral spins (e.g. $I = 1, 2, 3 \dots$), some have fractional spins (e.g. $I = 1/2, 3/2, 5/2 \dots$), and a few have no spin, $I = 0$ (e.g. $^{12}\text{C}, ^{16}\text{O}, ^{32}\text{S}, \dots$). Isotopes of particular interest and use to organic chemists are ^1H , ^{13}C , ^{19}F and ^{31}P , all of which have $I = 1/2$. Since the analysis of this spin state is straightforward, our discussion of nmr will be limited to these and other $I = 1/2$ nuclei. The following features lead to the nmr phenomenon:

- 1) A spinning charge generates a magnetic field, as shown by the animation on the right. The resulting spin-magnet has a magnetic moment (μ) proportional to the spin shows in Figure 2-1.

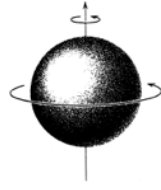


Figure 2-1 Spinning creates a magnetic moment.

2) In the presence of an external magnetic field (B_0), two spin states exist, $+1/2$ and $-1/2$. The magnetic moment of the lower energy $+1/2$ state is aligned with the external field, but that of the higher energy $-1/2$ spin state is opposed to the external field.

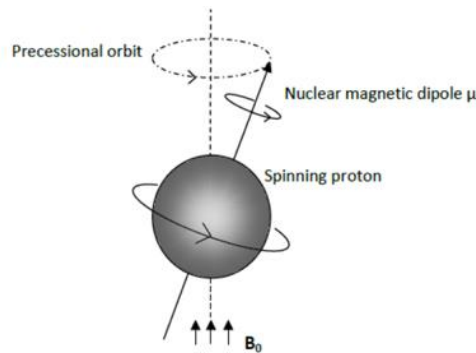


Figure 2-2 Representation of a precessing proton in a stationary magnetic field B_0

3) The difference in energy between the two spin states is dependent on the external magnetic field strength, and is always very small. The Figure 1-3 illustrates that the two spin states have the same energy when the external field is zero, but diverge as the field increases.

The energy difference ΔE between the spin energy states can be rewritten in terms of frequency units ν :

$$\Delta E = h\nu \tag{Eq.2-1}$$

where h is Planck's constant equal to $6.6256 \times 10^{-34} \text{ J s}$

Transitions between the spin energy states occurs at the Larmor frequency ν_0 :

$$\nu_0 = \gamma B_0 / 2\pi$$

Eq.2-2

where the proportionality factor gamma is a constant for the isotope of each element and is called the magnetogyric (or gyromagnetic) ratio.

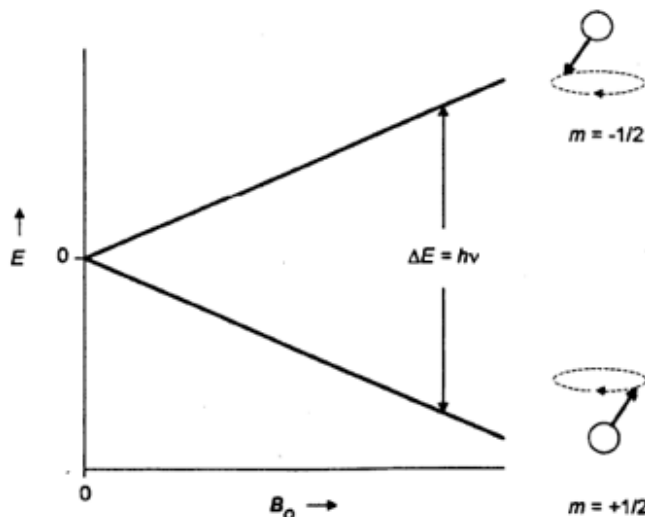


Figure 2-3 The energy difference ΔE between the spin energy states

The signal in NMR spectroscopy results from the difference between the energy absorbed by the spins which make a transition from the lower energy state to the higher energy state, and the energy emitted by the spins which simultaneously make a transition from the higher energy state to the lower energy state. The signal is thus proportional to the population difference between the states. NMR is a rather sensitive spectroscopy since it is capable of detecting these very small population differences. It is the resonance, or exchange of energy at a specific frequency between the spins and the spectrometer, which gives NMR its sensitivity.

2.1.2 Chemical shift

All proton resonances do not occur at the same position. The Larmor precession frequency (ν_0) varies because the actual magnetic field B at the nucleus is always less than the external field B_0 . The origin of this effect is the "superconducting" circulation of electrons in the molecule, which occurs in such a way that a local magnetic field B_e is created, which opposes B_0 (B_e is proportional to B_0). Thus $B = B_0 - B_e$. We therefore say that the nucleus is shielded from the external magnetic field. The extent of shielding

is influenced by many structural features within the molecule, hence the name chemical shift.

The scale most commonly used to assign peaks in an NMR spectrum is the ppm-scale. Chemical shift (δ) is defined in terms of a ratio of frequency difference between the resonance and that of TMS and operating frequency:

$$\delta = \frac{(\nu_s - \nu_{TMS}) \times 10^6}{\nu_{spectrometer}} \quad \delta \text{ (ppm: part per million)} \quad \text{Eq.2-3}$$

2.2 T1 Relaxation

Relaxation is the process by which the bulk magnetization reaches its equilibrium Value. Figure 1-4 illustrates that the z-magnetization grows back along the applied field axis (z-axis). Any transverse magnetization and other coherences decay over time. Phenomenologically, relaxation is categorized into Longitudinal relaxation (or T1 relaxation) that describes the return of the z component of longitudinal (z-)magnetization to its equilibrium value. Transverse relaxation (or T2 relaxation) that describes the decay of transverse (x,y) magnetization.

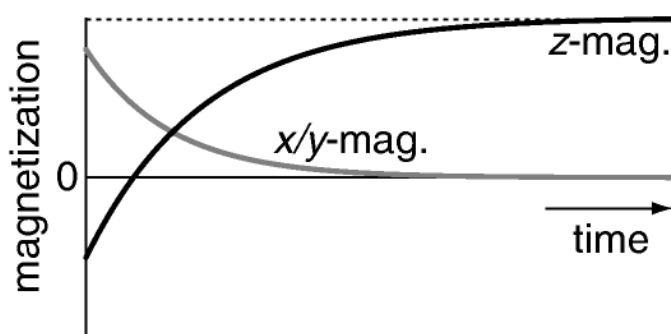


Figure 2-4 The z-magnetization grows back along the applied field axis (z-axis) over time

T₁ Relaxation (Spin-Lattice Relaxation)

The equilibrium distribution of α vs. β -state is governed by the Boltzman distribution:

$$\frac{\#(\alpha)}{\#(\beta)} = e^{-(E_\alpha - E_\beta)/kT}$$

Eq.2-5

Since α and β -states correspond to different energies changing the relative populations of these states changes the energy of the system. Therefore, T₁ relaxation is an enthalpic process. The energy is transferred to or taken from other spins. The surrounding of spins is often called the lattice and hence T₁ relaxation is often referred to as spin-lattice relaxation.

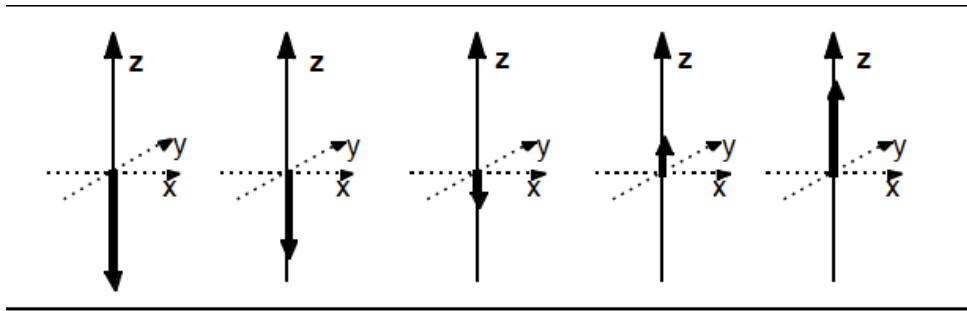


Figure 2-5 Return of the z-component of magnetization to the equilibrium value. The initial state may have been created through population inversion arising from an 180° pulse.

The vectors shown in the figure 2-5 above represent the magnetization vectors rather than magnetic moments from individual spins. Longitudinal magnetization decays exponentially according to:

$$M_z(t) = M_z(t_0) e^{-t/T_1}$$

Eq.2-6

The T₁ relaxation time determines the pulse repetition rate, the delay that must be inserted between individual scans. T₁ times for protons are in the range of 0.5 to a few seconds. T₁ times for quadrupolar nuclei ($I > 1/2$) can be rather short (ms range).

T1 relaxation time measurement

T1 relaxation times are usually determined from the inversion recovery sequence:
180 deg pulse-delay-90 deg pulse-delay-acquisition

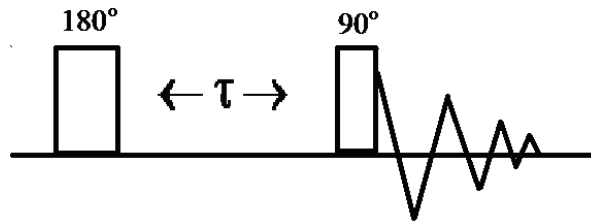


Figure 2-6 Pulse sequence used to measure T_1 using inversion recovery

In this sequence, magnetization is inverted by application of a 180-degree pulse. A delay follows during which T1 relaxation takes place bringing the $-z$ magnetization back towards $+z$. Afterwards a 90-degree pulse along y turns the magnetization onto the x -axis into observable signal (Figure 2-7). The magnitude and sign of the x -magnetization at the end of the sequence depends on the T1 constant and the relaxation delay Δ .

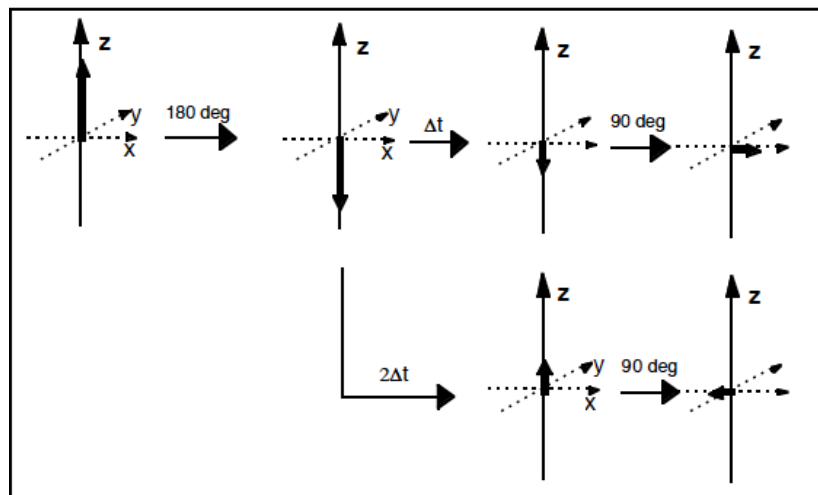


Figure 2-7 Magnetization vectors illustration in inversion recovery pulse sequence

The value for T1 is determined with the help of TopSpin using the fitting function:

$$f(t) = I_0 \cdot [1 - 2 \exp(-t/T_1)]$$

Eq.2-7

2.3 The NOE effect

The Nuclear Overhauser Enhancement (NOE) effect increases the intensities of the signals. Two nuclei close in space exhibiting a dipolar coupling will help each other to relax. The NOE effect works through space, not necessarily through scalar couplings. The system for two dipolar coupled nuclei A and X with no scalar coupling is illustrated Figure 2-8 :

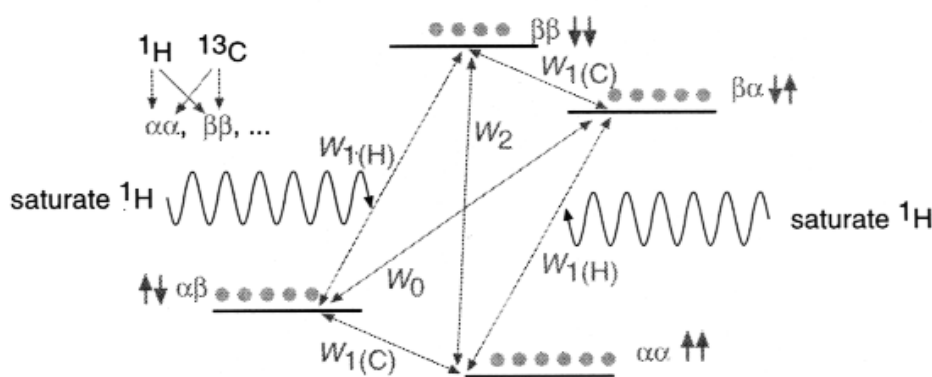


Figure 2-8 The system for two dipolar coupled nuclei ^1H and ^{13}C with no scalar coupling [33]

Two nuclei A and X (illustrated with H and C) are dipolar coupled and help relax each other through the pathways W_0 and W_2 . The usual relaxation pathway W_1 is irradiated.

Figure 2-8 describes the possible pathways for relaxation. W_2 is a double quantum transition and W_0 a zero-quantum transition. W_0 and W_2 are determined mostly by dipole-dipole relaxation. When non-dipolar mechanisms are relaxing the system, this affects mainly W_1 .

Irradiation of the sample with the resonance frequency of A equalizes (i.e. saturates) the populations in the N_1 - N_3 and N_2 - N_4 levels. The population difference for X has not changed and there seems to be no signal enhancement.

The irradiation shifts the system out of equilibrium, and because the W_{1A} pathway is irradiated and the W_{1X} has not changed from equilibrium, the only pathways to relaxation are W_0 and W_2 . When relaxation along W_0 occurs, populations in N_2 are increased and N_3 decreased, thus reducing the difference in populations $N_1 - N_2$ and $N_3 - N_4$, thereby reducing the signal intensity from X. Relaxation along W_2 , on the other hand, increases population N_1 at the expense of N_4 , seemingly increasing the difference between populations $N_1 - N_2$ and $N_3 - N_4$ giving enhanced signal intensity from X. As NOE is a relaxation process it needs some time to build up.

The system is not relaxed by W_0 or W_2 , but the two pathways in combination. Both W_0 and W_2 relax the system and are therefore competing pathways. What decides whether W_0 or W_2 dominates in the relaxation process is the correlation time, T_c . Small molecules tumble faster than large ones and thus have shorter T_c . W_2 dominates at short correlation times and W_0 at long correlation times. Thus, small molecules give positive NOE effect and large molecules give negative NOE effect.

The frequency of the applied magnetic field is also an important factor. The fluctuating magnetic fields needed to induce the double quantum transition must contain frequencies close to the sum of the Larmor frequencies of A and X. The zero-quantum transition requires much lower frequencies. Thus, it is more likely to find negative NOE amplifications in a high field.

The NOE effect will also depend on the distance between the cross-relaxing molecules and will decrease in inverse proportion to the sixth power of the distance. This is an important tool for calculating distances between molecules through space.

Maximum NOEs cannot be achieved, since the dipole-dipole relaxation process is not the only process relaxing the nuclei. In theory the maximum NOE intensity enhancement is given by:

$$\eta + 1 = 1 + \frac{1}{2} * \gamma_X / \gamma_A$$

Eq.2-8

2.4 The Free Induction Decay (FID)

The signal registered in the apparatus arises from the magnetization in the $x'y'$ -plane. The relaxation of these magnetization vectors induces a signal picked up by the receiver coil in the apparatus. This results in a decaying time dependent signal. This signal is called the Free Induction Decay (FID), it contains all the information from the sample but is rather hard to analyze directly. The FID is therefore Fourier Transformed to obtain a frequency dependent spectrum which is possible to analyze (illustrated in Figure 2-9).

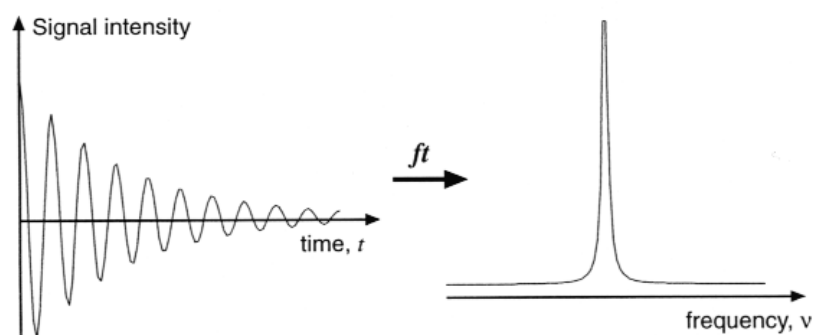


Figure 2-9 From FID to spectrum by Fourier Transformation

The Fourier transformation is described mathematically by:

$$F(\nu) \propto \int_{-\infty}^{\infty} f(t) e^{-i2\pi\nu t} dt$$

Eq.2-9

and this is the mathematical basis of the Fourier Transform (FT) NMR method.

2.5 Quantitative ^{13}C NMR experiment

Many of the molecules studied by NMR contain carbon. Unfortunately, the

carbon-12 nucleus does not have a nuclear spin, but the carbon-13 nucleus does due to the presence of an unpaired neutron. Carbon-13 nuclei make up approximately one percent of the carbon nuclei on earth. Therefore, ^{13}C NMR spectroscopy will be less sensitive than proton NMR spectroscopy.

With the appropriate concentration, field strength, and pulse sequences, however, carbon-13 NMR spectroscopy can be used to supplement the previously described proton NMR information and has a much larger chemical shift range. Its low natural abundance and proton decoupling means that spin-spin couplings are seldom observed. This greatly simplifies the spectrum and makes it less crowded. ^{13}C is a low sensitivity nucleus that yields sharp signals and has a wide chemical shift range. A typical analysis of a ^{13}C NMR spectrum consists of matching expected chemical shifts to the expected moieties. Each type of signal has a characteristic chemical shift range that can be used for assignment, shows in Figure 2-10.

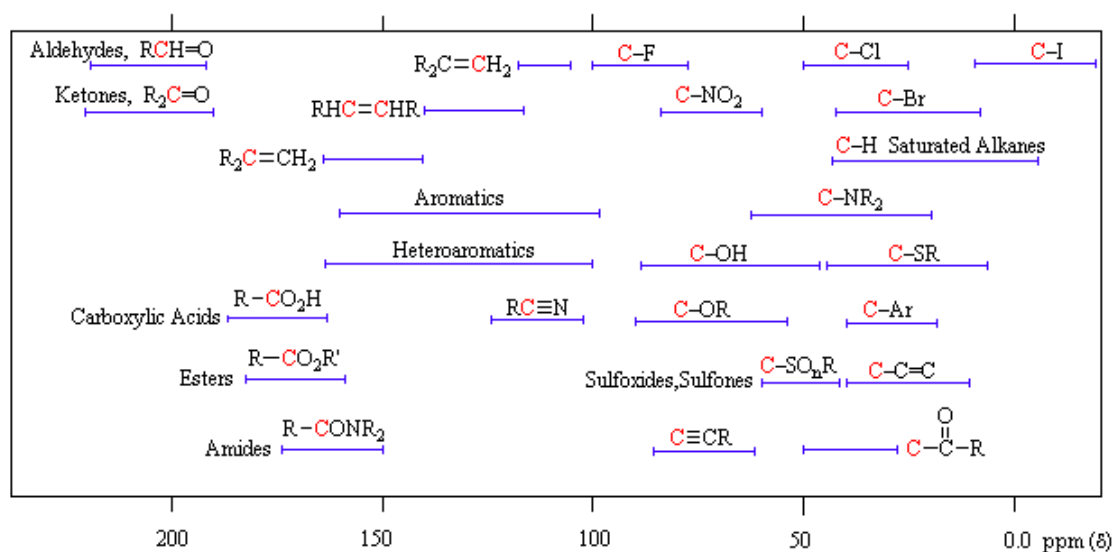


Figure 2-10 The chemical shift range in ^{13}C NMR spectrum[34]

Integration is almost useless in a regular ^{13}C NMR spectrum because of uneven nuclear Overhauser effect (NOE) enhancement of the signals by decoupling and long T1 relaxation times. Quantitative spectra may be obtained by inverse gated decoupling and long delays in the region of 10 minutes between pulses.

An inverse gated decoupling experiment can be carried out for quantitative carbon calculation. Figure 2-11 illustrates the pulse scheme for inverse gated decoupled. The BB decoupling pulses are only switched on when the acquisition of the ^{13}C spectrum starts and stays on for the time of acquisition. The NOE effect will then not have time to build up, but the spectrum will still be decoupled. The ^{13}C spectrum is now not distorted by the NOE enhancement, and can now be integrated to the relative number of ^{13}C atoms in the molecule. When integration is required for a ^{13}C spectrum, this experiment can be used with a long relaxation time, although the sensitivity is dramatically reduced as compared to measuring carbon with regular decoupling.

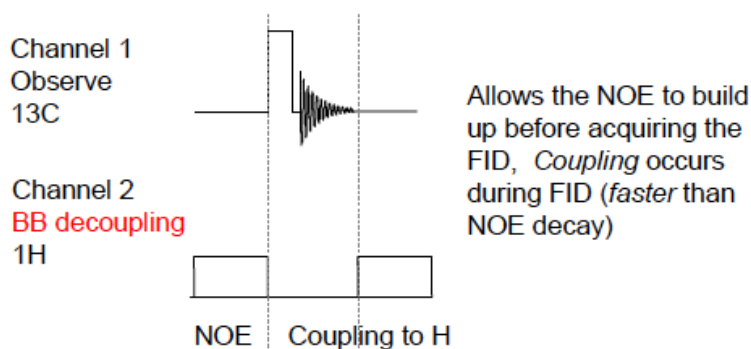


Figure 2-11 Pulse scheme of Inverse gated decoupled ^{13}C NMR experiment[35]

2.6 2D HSQC NMR experiment

The construction of a 2D experiment is simple: In addition to preparation and detection which are already known from 1D experiments the 2D experiment has an indirect evolution time t_1 and a mixing sequence. This scheme (Figure 2-12) can be viewed as:

- Preparation: do something with the nuclei
- Evolution: let them process freely
- Mixing: do something else
- Detection: detect the result

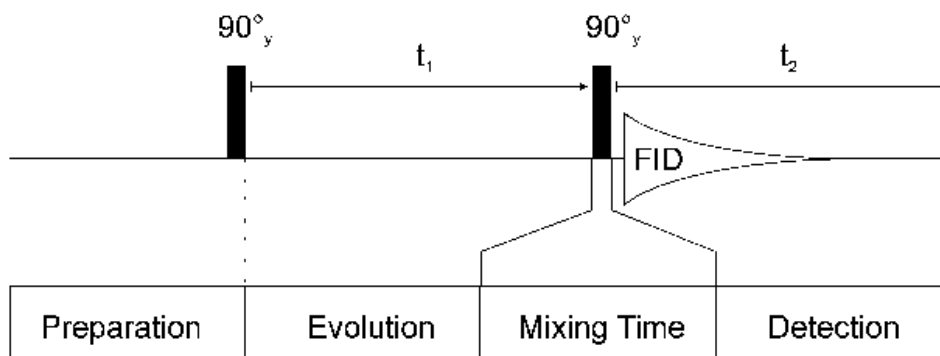


Figure 2-12 Scheme of pulse program of 2D NMR experiment

After preparation the spins can precess freely for a given time t_1 . During this time the magnetization is labelled with the chemical shift of the first nucleus. During the mixing time magnetization is then transferred from the first nucleus to a second one. Mixing sequences utilize two mechanisms for magnetization transfer: scalar coupling or dipolar interaction (NOE). Data are acquired at the end of the experiment (detection, often called direct evolution time); during this time the magnetization is labelled with the chemical shift of the second nucleus.

Two-dimensional FT yields the 2D spectrum with two frequency axes. A diagonal of signals (A and B) divides the spectrum in two equal halves. Symmetrical to this diagonal, there are more signals (X), called cross signals. The diagonal results from contributions of the magnetization that has not been changed by the mixing sequence (equal frequency in both dimensions) i.e. from contributions which remained on the same nucleus during both evolution times. The cross signals originate from nuclei that exchanged magnetization during the mixing time (frequencies of the first and second nucleus in each dimension, respectively). They indicate an interaction of these two nuclei. Therefore, the cross signals contain the really important information of 2D NMR spectra. Figure 2-13 illustrates the scheme of 2D NMR spectrum.

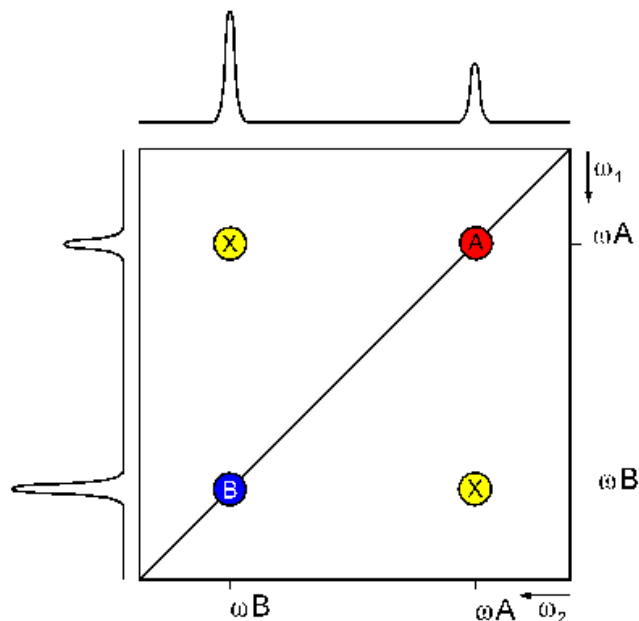


Figure 2-13 Scheme of a 2D NMR spectrum

^1H - ^{13}C Heteronuclear Single Quantum Coherence Spectroscopy (HSQC) spectrum correlates chemical shifts of heteronucleus X (F1 dimension) and protons (F2 dimension) via the direct heteronuclear coupling. The HSQC experiment is widely used for recording one-bond correlation spectra between carbon and proton, with the proton being the observed nucleus (i.e. it is an inverse experiment). A ^1H - ^{13}C HSQC pulse sequences is shown in Figure 2-14.

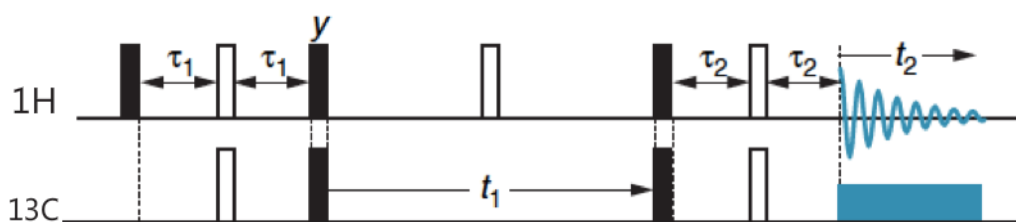


Figure 2-14 A pulse sequences for ^1H and ^{13}C correlation using the HSQC experiment[36]

2.7 Objective

LtL oils are complex mixtures of degradation products from the thermal breakdown of lignin. Considerable efforts are being made to characterize the oil and

quantify individual components. However, such analysis is very strenuous and time-consuming. NMR techniques have the potential to characterize mixtures in terms of structural groups, rather than individual compounds, and thus provide quantitative information on the oil composition without requiring complete separation and quantification of individual compounds.

In this Master's project, the aim is to establish analytical protocols for NMR characterization of LtL-oils. The purpose is to quantify the major structural groups like aromatic rings, aliphatic chains, phenol groups, carboxylic groups, ketones and alcohols. ^{13}C NMR will be the most central technique, but proton NMR and 2-D methods of derivatized oils will also be evaluated.

In a pilot phase, standard compounds that correspond to major components based on GC-MS analysis results of the bio-oils will be prepared as mixtures with known concentrations, and the conditions for quantitative NMR spectroscopy including the choice of solvent and relaxation agent will be established. The use of a quantitative internal standard will be considered. When suitable NMR conditions are established, analysis of LtL oils will be attempted and quantification of functional groups of LtL oils converted under different conditions will be calculated.

Chapter 3 Experiment

3.1 Materials

6 standard compounds showed in Figure 3-1 which are compared from the GC-MS analysis results are chosen during the experiment[28]. 3 solvents and 1 relaxation agent listed in Table 3-1 are used for preparation of NMR samples. All the solvents and compounds were purchased from Sigma-Aldrich and used without further upgrading.

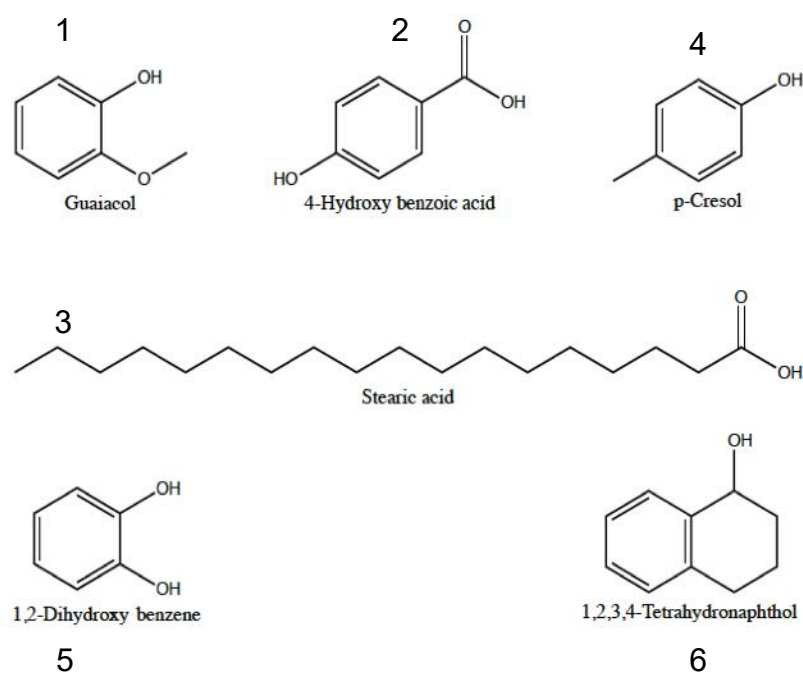


Figure 3-1 Compounds used in the study

Table 3-1 Solvent used in the study

Solvent	Source	Use
CDCl_3	Sigma-Aldrich	Solvent for the standard compounds
DMSO-D6	Sigma-Aldrich	Solvent for mixture and LtL-oil
Chromium(III) acetylacetonate	Sigma-Aldrich	Relaxation agent

The LtL-oils are produced by solvolysis method in UiB lab, detailed descriptions of solvolysis system and conditions have been previously published, a simplified flow sheet for LtL-solvolysis laboratory workup is displayed in Figure 3-2[37]. The biomass used in the experimental setups consisted of industrially obtained wood shavings from softwood, comprised by ~90% Norway spruce (*Picea abies*) and limited amounts of Pine (*Pinus sylvestris*). It was kindly supplied by Weyland AS in Norway, produced from locally grown wood in the district of Hardanger, Norway, and processed at Granvin Bruk in Hardanger[38]. The LtL oils were produced by Camilla Løhre and Hilde Vik Halleraker, the production conditions of LtL-Oils 1-6 were listed in Table 3-2. The LtL oils were used as received.

Table 3-2 Conditions of the production of LtL oil

LtL-oil samples	Lignin g	Water mL	Formic acid mL	Time h	Temperature °C
1	0.5	4	0.75	2	340
2	0.5	4	0.75 (¹³ C)	2	340
3	0.5	4	0.5	2	320
4	0.5	4	1	2	360
5	0.5	4	0.5	2	320
6	0.5	4	0.5 (¹³ C)	2	320

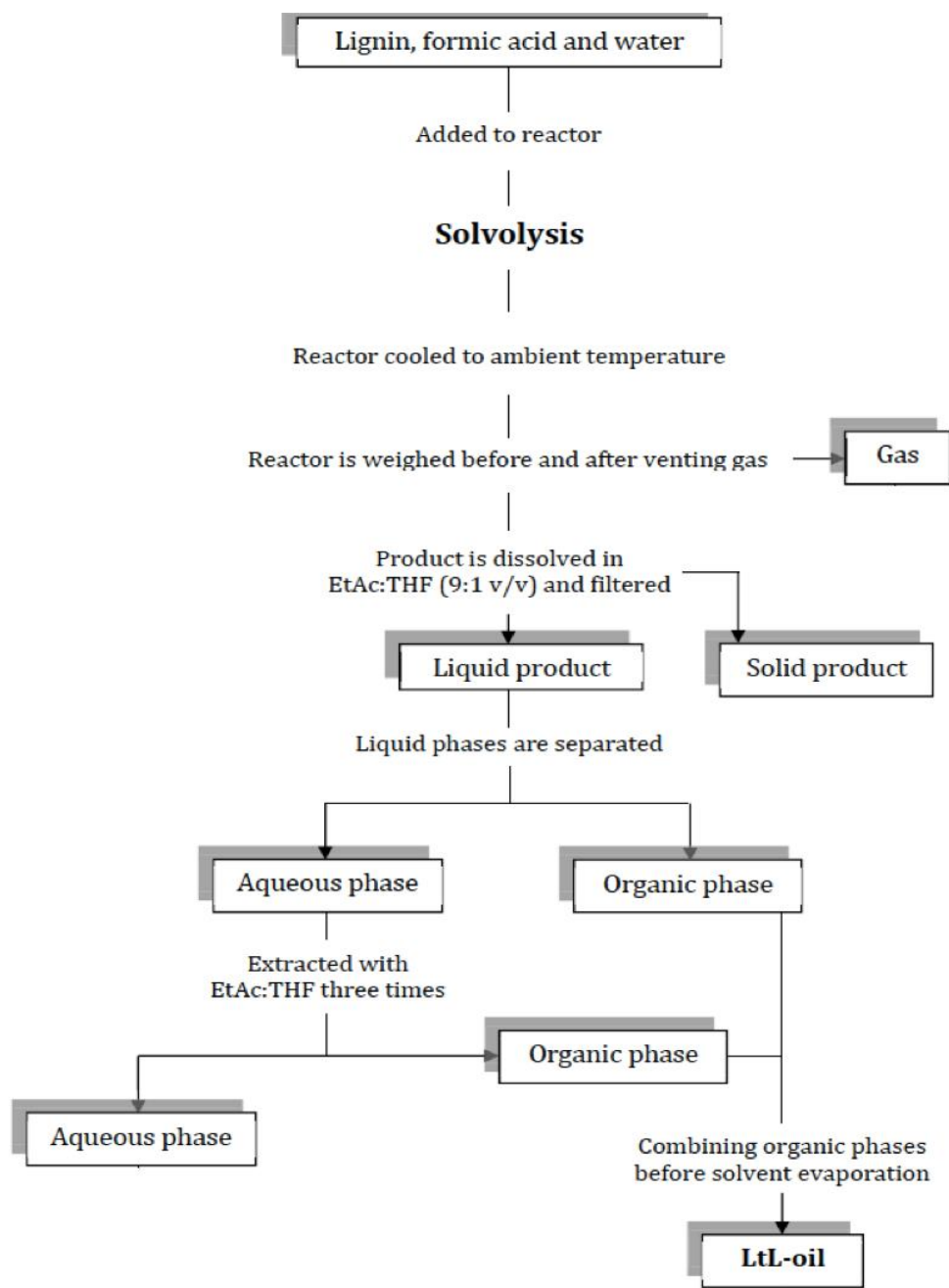


Figure 3-2 Flow sheet for LtL-solvolysis laboratory workup[37]

3.2 Sample preparation

3.2.1 Standard compound samples

Each standard compound showed in Figure 3-1 was weighed 0.06 g by an analytical balance then dissolved in 0.6ml solvent, which made the concentration were

around 10%. The samples were shaken vigorously and transferred to NMR tubes. The samples were named Standard 1-7.

3.2.2 Mixture samples

5 mixture samples were prepared with standard compounds according to Table 3-3. The standard compounds were weighed by a total weight of 0.06g, and the weights of each compound in the mixture were equal. Then they were dissolved in 0.6ml DMSO-d6 and transferred to NMR tubes.

Table 3-3 Mixture samples preparation

Sample Name	Standard compounds	Weight of each compound/g	DMSO-d6 /ml
Mix 1	1,2	0.03	0.6
Mix 2	1,2,3	0.02	0.6
Mix 3	1,2,3,4	0.15	0.6
Mix 4	1,2,3,4,5	0.12	0.6
Mix 5	1,2,3,4,5,6	0.10	0.6

3.2.3 Samples with relaxation reagent

Two samples of 10% guaiacol (standard compound 1) with different concentration of Cr(acac)₃ were prepared according to Table 3-4:

Table 3-4 Samples with relaxation reagent preparation

LtL-oil samples	Weight of LtL oils /mg	Cr(acac) ₃	DMSO-d6 /ml
Standard1-2	0.06	0.01M	0.6
Standard1-2	0.06	0.05M	0.6

3.2.4 LtL-oil samples

Each LtL oil showed in Table 3-2 was weighed 0.06 g by an analytical balance and 0.1M Cr(acac)₃ were added as relaxation reagent, then dissolved in 0.6ml DMSO-d₆. The samples were shaken vigorously and transferred to NMR tubes. The weight of samples LtL-oil 1-6 is shown in Table 3-5.

Table 3-5 LtL-Oil samples preparation

Sample Name	Weight of LtL-Oils /mg	Cr(acac) ₃	DMSO-d ₆ /ml
LtL-Oil 1	60.4	0.01M	0.6
LtL-Oil 2	60.1	0.01M	0.6
LtL-Oil 3	60.1	0.01M	0.6
LtL-Oil 4	59.5	0.01M	0.6
LtL-Oil 5	60.2	0.01M	0.6
LtL-Oil 6	60.2	0.05M	0.6

3.3 NMR experiments

3.3.1 NMR spectrometers

Three NMR spectrometers were used for this study are listed in Table 3-6.

Table 3-6 NMR Spectrometers and probes

Spectrometers	Resonance frequency for protons	Probe	Automatic sample changer
Bruker Biospin 850 Ascend	850MHz	TCI CryoProbe	Yes
Bruker Biospin AV600	600MHz	TCI Cryo Probe	No
Bruker Biospin AV500	500MHz	Broadband Observe (BBO) probe	Yes

The Bruker Biospin AV 500 instrument is located at the Department of Chemistry in UiB, it is equipped with BBO probe to perform ^1H and ^{13}C experiments. Broadband observe – BBO- means that there are two rf-coils (double resonance, plus lock channel), one for X nucleus (where X can be a different nuclei, including ^{13}C), and one for ^1H . The X rf-coil is located inside the ^1H rf-coil. Automatic sample changer is applied on AV500 NMR, and samples are put in queue using the IconNMR software.



Figure 3-3 Bruker Biospin AV500

The Bruker Biospin AV600 instrument is located at Department of Chemistry in UiB, it is equipped with a TCI Cryo Probe, and is mainly used for high resolution NMR studies of proteins and macromolecules in the liquid state. The TCI Cryo Probe is a proton-optimized triple resonance (three rf-coils) NMR inverse (the ^1H coil on the inside) probe, featuring three fully independent channels (plus lock channel) for simultaneous decoupling on multiple nuclei such as ^{13}C and ^{15}N . Sample is manually inserted for each measurement. It's preferable to shim the sample manually on AV600. However, the AV600 was discharged in December 2017.



Figure 3-4 Bruker Biospin AV600

The Bruker Biospin 850 Ascend is located at NNP (Norwegian NMR platform), it is also equipped with a TCI Cryo Probe. An automatic sample changer is applied, samples are put in queue using the IconNMR software.



Figure 3-5 Bruker Biospin 850 Ascend



Figure 3-6 NMR samples for AV500 and AV600



Figure 3-7 NMR samples for 850MHz

3.3.2 Procedure of running a sample on NMR instrument

Each spectrometer is connected to a PC running Windows XP and controlled by a program called Topspin. The procedure to set up and perform the experiment is as following:

- Edit the specific acquisition parameters for selected pulse program by typing *ased*. The detailed information about pulse program are stated in chapter 3.3.3.

- Insert the sample to NMR instrument.
- Perform tuning and matching after the sample had been inserted typing *atma* (followed by *atmm* only if the automatic tuning and matching was insufficient).
- Turn on and adjust the deuterium lock to lock the signal before the solvent is selected by typing *lock*.
- Correct the phase by using *autophase*.
- Perform shimming on the sample by typing *topshim gui* and the configuration shown in Figure 3-8:

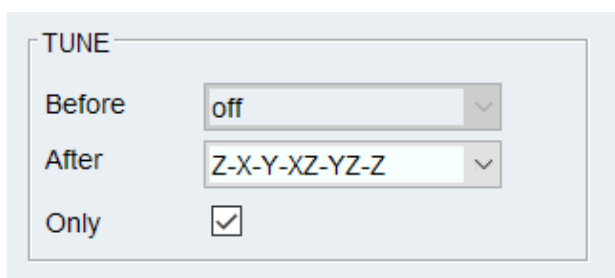


Figure 3-8: Configurations for topshim gui.

- Adjust the receiver gain by typing *rga*.
- Check the estimated experimental time by typing *expt*.
- Start acquisition by typing *zg*.

3.3.3 Pulse program

The following pulse sequences were used:

- **¹H NMR Experiments:**

Zg: The Standard ¹H NMR pulse program.

Zg30: The Standard ¹H NMR 30° pulse program.

1D-sequence, using a 90°(zg) or 30°(zg30) flip angle. Result is a routine proton NMR spectrum. A typical list of acquisition parameters for a standard 1D ¹H experiment is shown in Table 3-7, the specific parameters used in the experiments are record in result in chapter 4.

Table 3-7 Acquisition parameters in ^1H NMR experiment*

Parameter	Description	Typical name/value
PULPROG	pulse program used	zg
TD	number of points in FID	32768
NS	number of scans	32
DS	number of dummy scans	4
SWH	spectral width (Hz)	6000-12000Hz Hz
AQ	acquisition time	3s
RG	receiver gain	10
DW	dwell time	85 μs
DE	Pre scan delay	6 μs
D1	Relaxation delay	1.0 s
TD0	How often write data	1
NUC1	Nucleus for channel 1	^1H
P1	90° pulse width	5-12 μs
PL1	Relative power level for pulse	0.00 dB
PL1W	Power level for pulse	5-60 W
SFO1	Transmitter frequency	600.1728208

*Acquisition parameters setting is acquired from course “KJEM 251 NMR Spectroscopy I” in UiB, John Georg Seland, Sept.2014

- **^{13}C NMR Experiments:**

Zgig: Inverse gated ^1H -decoupled ^{13}C NMR pulse program.

Zgig30: Inverse gated ^1H -decoupled ^{13}C NMR 30° pulse program.

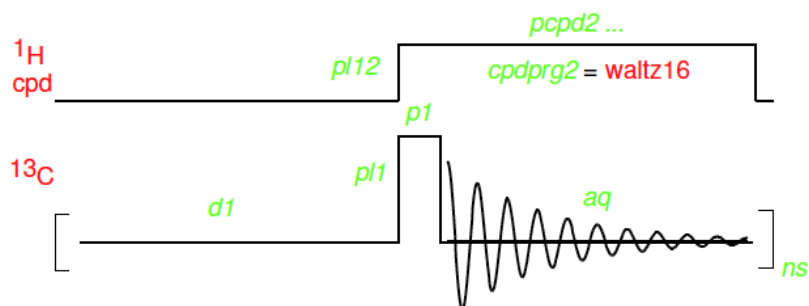


Figure 3-9: Inverse gated ^1H -decoupled ^{13}C NMR pulse program
(cpd: composite pulse decoupling sequence)

1D-sequence with inverse gated decoupling, using a 90° (zgig) or 30° (zgig30) pulse. This experiment yields ^1H -decoupled NMR spectra of ^{13}C -nuclei without signal enhancement by the nuclear overhauser effect. A typical list of acquisition parameters for a standard 1D ^1H -decoupled ^{13}C NMR experiment is shown in Table 3-8, the specific parameters used in the experiments are record in result in chapter 4.

Table 3-8 Acquisition parameters in 1D ^{13}C experiment*

Parameter	Description	Typical name/value
PULPROG	pulse program used	zgdc
TD	number of points in FID	32768
NS	number of scans	32 or more
DS	number of dummy scans	0
SWH	spectral width (Hz)	100-300 KHz
AQ	acquisition time	2s
RG	receiver gain	1000
DW	dwelt time	45 μs
DE	Pre scan delay	50 μs
D1	Relaxation delay	5.0 s or higher
TD0	How often write data	1
NUC1	Nucleus for channel 1	^{13}C
P1	90° pulse width for ^{13}C	10-15 μs
PL1	Relative power level for pulse	0.00 dB (AV500)
PL1W	Power level for pulse	5-60 W
SFO1	Transmitter frequency (nucleus 1)	125.7682745 (AV500)
CPDPRG2	Filename for BB decoupling	waltz16
NUC2	Nucleus for channel 2	^1H
PCPD2	90° pulse width for BB decoupling	60-100 μs
PL2	Relative power level for pulse	0.00 dB (AV500)
PL2W	Power level for pulse	20-30 W
PL12	Relative power level for BB pulse	15 dB
PL12W	Power level for BB pulse	0.5W
SFO2	Transmitter frequency (nucleus 2)	500.1328208 (AV500)

*Acquisition parameters setting is acquired from course “KJEM 251 NMR Spectroscopy I” in UiB, John Georg Seland, Sept.2014

- **T1 relaxation NMR experiments:**

T1irpg: Inversion recovery ^{13}C NMR pulse program

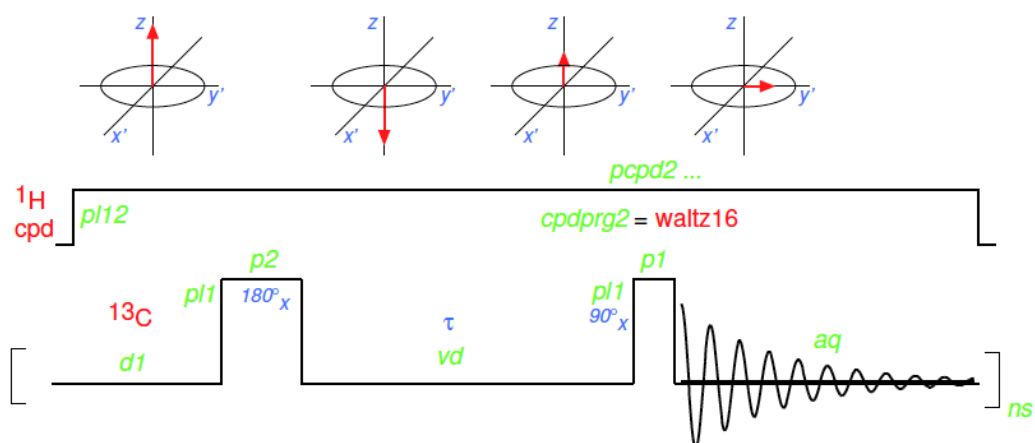


Figure 3-10: Inverse recovery ^{13}C NMR pulse program

Table 3-9 Acquisition parameters in inversion recovery ^{13}C NMR experiment*

Parameter	Description	Typical name/value
PULPROG	pulse program used	t1irpg
TD	number of points in FID	32768
NS	number of scans	4 or more
DS	number of dummy scans	4
SWH	spectral width (Hz)	100-300 KHz
AQ	acquisition time	2s
RG	receiver gain	1000
DW	dwell time	45 μs
DE	Pre scan delay	6.5 μs
D1	Relaxation delay	15 s
d11	time delay	30 ms
d12	time delay	20 μs
VDLIST	List with variable delay (τ)	T1_kjem251
NUC1	Nucleus for channel 1	^{13}C
P1	90° pulse width for ^{13}C	7-15 μs
P2	180° angle pulse width for ^{13}C	14-30 μs
PL1	Relative power level for pulse	0.00 dB (AV500)
PL1W	Power level for pulse	5-60 W
SFO1	Transmitter frequency (nucleus 1)	125.7682745 (AV500)
CPDPRG2	Filename for BB decoupling	waltz16
NUC2	Nucleus for channel 2	^1H
PCPD2	90° pulse width for BB decoupling	60-100 μs
PL2	Relative power level for pulse	0.00 dB (AV500)
PL2W	Power level for pulse	20-30 W
PL12	Relative power level for BB pulse	20 dB
PL12W	Power level for BB pulse	0.5W
PL13	Relative power level for second BB pulse	20 dB
PL13W	Power level for second BB pulse	0.5W
SFO2	Transmitter frequency (nucleus 2)	500.1328208 (AV500)

*Acquisition parameters setting is acquired from course “KJEM 251 NMR Spectroscopy I” in UiB, John Georg Seland, Sept.2014

T_1 relaxation times for ^{13}C is measured using the Inversion Recovery method. The T_1 data acquisition is somewhat similar to obtaining simple ^1H -decoupled ^{13}C NMR spectra. However, a series of spectra for different values of a time delay parameter t were recorded to make it possible to plot each of the ^{13}C signal intensities as functions of time. A typical list of acquisition parameters for an inversion recovery ^{13}C NMR experiment is shown in Table 3-9, the specific parameters used in the experiments are record in result in chapter 4.

- **2D ^1H - ^{13}C HSQC experiments:**

Hsqcetgpsi2: Phase-sensitive gs-HSQC with sensitivity enhancement pulse program

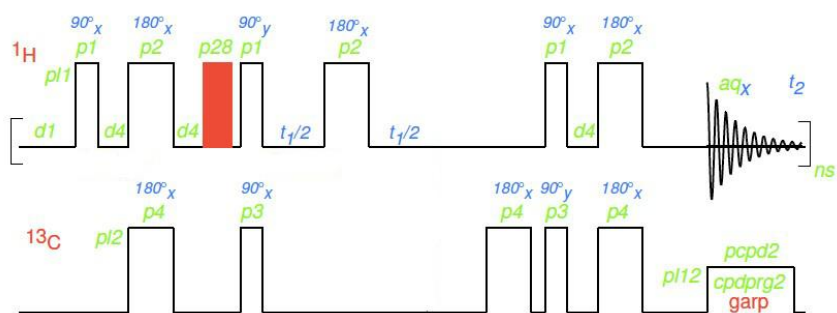


Figure 3-11: 2D gs-HSQC ^1H - ^{13}C NMR pulse program

2D pulse sequence with HSQC (Heteronuclear Single Quantum Coherence) method performs the H, C correlation via the ^{13}C chemical shift evolution of a single quantum coherence. In this case it is a gradient-selected correlation using echo/antiecho selection method. A typical list of acquisition parameters for a standard 2D HSQC NMR experiment is shown in Table 3-10, the specific parameters used in the experiments are record in result in chapter 4.

Table 3-10 Acquisition parameters in 2D HSQC experiment*

Parameter	Description	Typical name/value
PULPROG	pulse program used	hsqcetgpf
TD	number of points in FID	2048-4096
NS	number of scans	2
DS	number of dummy scans	8 or more
SWH	spectral width (Hz)	15000-30000 Hz
AQ	acquisition time	0.5-2s
RG	receiver gain	10
DW	dwelt time	30-200 μs
DE	Pre scan delay	6 μs
CNST2	C-H coupling constant	145 Hz
CNST11	degree of enhancement	3
D0	incremented delay	1-10 μs s
D1	Relaxation delay	1-2 s
d4	$\frac{1}{4J}$ value	$1/(4\text{CNST2})$
D11	delay	30 ms
D13	delay	4 μs
D16	delay for gradient recovery	0.0005 s
Delta	delay	1.5 ms
Delta1	delay	0.7 ms

in0	increment between FIDs (=INF1)	100-500 μ s
INF1	increment between FIDs	100-500 μ s
ST1CNT	Loop counter	64
NUC1	Nucleus for channel 1	1 H
P1	90° pulse width for 1 H	5-12 μ s
P2	180° angle pulse width for 1 H	14-30 μ s
PL1	Relative power level for pulse	0.00 or 3.00 dB
P28	trim pulse duration	3 μ s
PL1W	Power level for pulse	5-60 W
SFO1	Transmitter frequency	600.1728208
CPDPRG2	Filename for BB decoupling	garp
NUC2	Nucleus for channel 2	13 C
P3	90° pulse width for 13 C	7-15 μ s
P4	180° angle pulse width for 13 C	14-30 μ s
PCPD2	90° pulse width for BB decoupling	60-100 μ s
PL2	Relative power level for pulse	0.00 or -1.80 dB
PL2W	Power level for pulse	20-30 W
PL12	Relative power level for BB pulse	15 dB
PL12W	Power level for BB pulse	0.5W
SFO2	Transmitter frequency (nucleus 2)	150.1328208 (AV600)
GPNAM1	Shape of pulsed field gradient 1	SINE.100
GPNAM2	Shape of pulsed field gradient 2	SINE.100
GPZ1	Strength of pulsed field gradient 1	80 %
GPZ2	Strength of pulsed field gradient 2	20.1 %
P16	Length of pulsed field gradient	1000 μ s

*Acquisition parameters setting is acquired from course “KJEM 251 NMR Spectroscopy I” in UiB, John Georg Seland, Sept.2014

3.3.4 Data processing in Topspin

The list of processing parameters was accessed by typing *edp*. In order to obtain an NMR spectrum, the FID (time domain signals) had to be transformed to the frequency domain by means of the Fourier transform. Besides, exponential multiplication was applied before Fourier transform to improve upon the signal-to-noise ratio. These two steps were done by typing *ef*. Then the spectrum was phased to obtain pure absorption spectra, i.e. all resonances should have a positive phase. Automatic phase correction was performed by typing *apk*. And an extra manually phasing was always performed using phase correction in toolbox. For LtL-oil sample NMR spectrums, baseline correction was performed. For quantitative 13 C NMR spectrums, peaks were manually picked and integration the peaks were manually selected.

Chapter 4 Results

4.1 Deuterated solvent for NMR samples

The standard compounds 1-6 were dissolved in 2 solvents, the solubility results are shown in table 4-1.

Table 4-1 Solubility of standard compounds 1-6 in CDCl₃ and DMSO-d₆:

Compounds	1	2	3	4	5	6
CDCl ₃	Y	Y	N*	Y	Y	Y
DMSO-d ₆	Y	Y	Y	Y	Y	Y

*Compound 3 –stearic acid forms a gel at room temperature in CDCl₃

The ¹H NMR experiments was performed on the samples of 10% guaiacol (compound 1) in CDCl₃ and 10% guaiacol in DMSO-d₆ on Bruker AV600 (pulse program: zg, ns=32, sw=12ppm). The spectra in Figure 4-1 and Figure 4-2 show that the peak from Ar-OH in guaiacol is 5.6006 ppm in CDCl₃ while it is 8.8744 ppm in DMSO-d₆. It is also observed the Ar-OH signal in DMSO-d₆ is stronger than in CDCl₃.

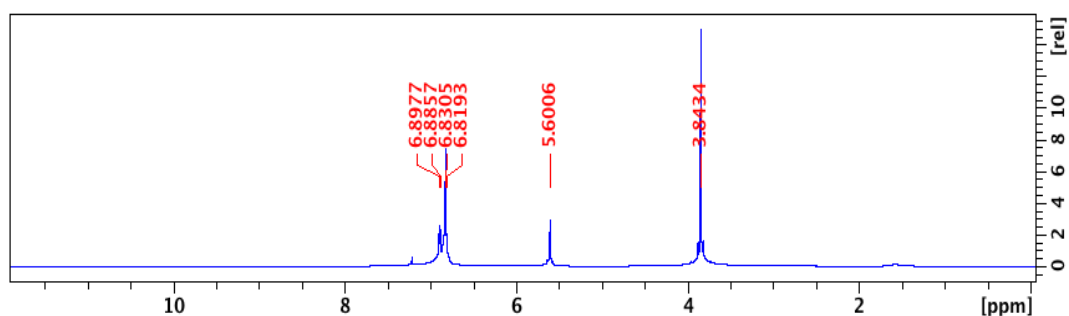


Figure 4-1 ¹H NMR spectrum of 10% guaiacol in CDCl₃

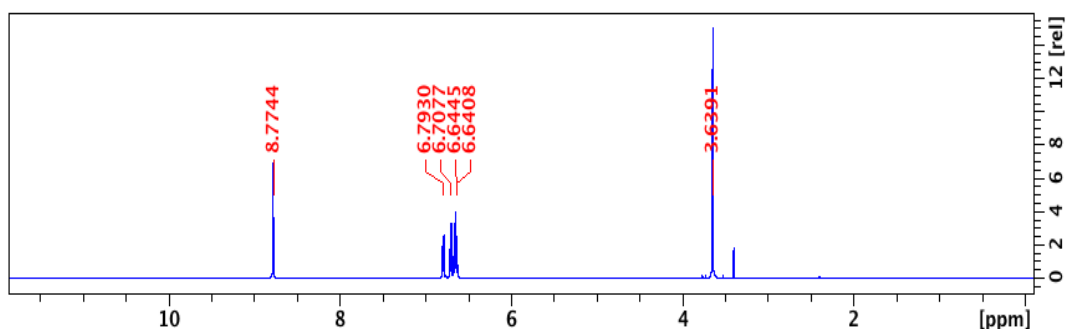


Figure 4-2 ^1H NMR spectrum of 10% guaiacol in DMSO-d_6

4.2 T1 relaxation NMR experiment

T1 relaxation affects the relative integration between signals, since each nucleus (individual spins) in a molecule has a different T1 value, for quantitative purposes is mandatory to wait at least 5 times the longest T1 in the sample between scans in order to recover 99% of the equilibrium magnetization.

T1 relaxation NMR experiment of 10%Guaiacol in DMSO-d_6 is performed on Bruker AV600 (pulse program: t1irpg, D1=15, TD=16, ns=4). T1 was calculated by Topspin software and the results shows the longest T1=8.816s, see Figure 4-3. In this case, the d1 should be set higher than $5 \cdot T1$ (44s), which will result a 12 h experiment time for quantitative ^{13}C NMR experiment of one sample.

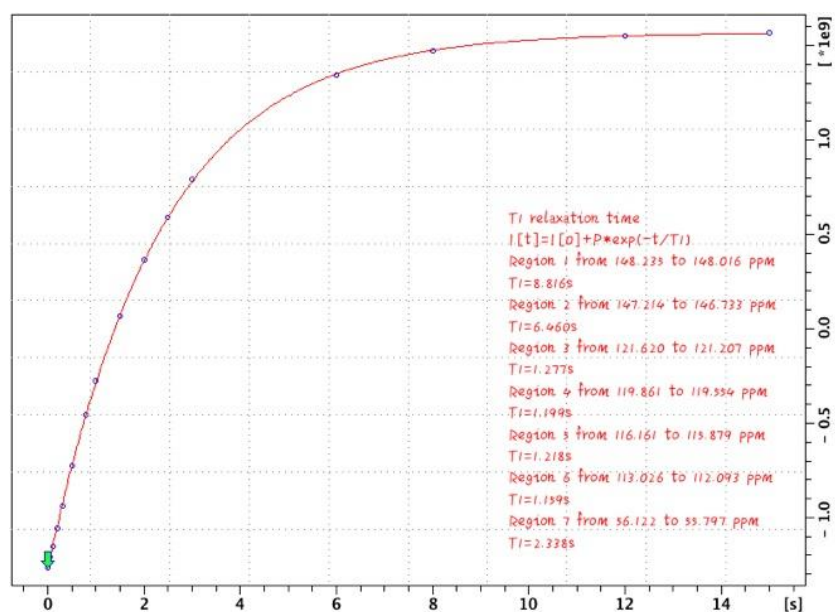


Figure 4-3 ^{13}C T1 measurement of guaiacol

Quantitative ^{13}C NMR spectra for 3 guaiacol samples were acquired from Bruker AV600 (pulse program: zgig30, d1=10, sw=220ppm, ns=1024), and the experiment time for each sample is 3h 16min, see the spectra in Figure 4-4.

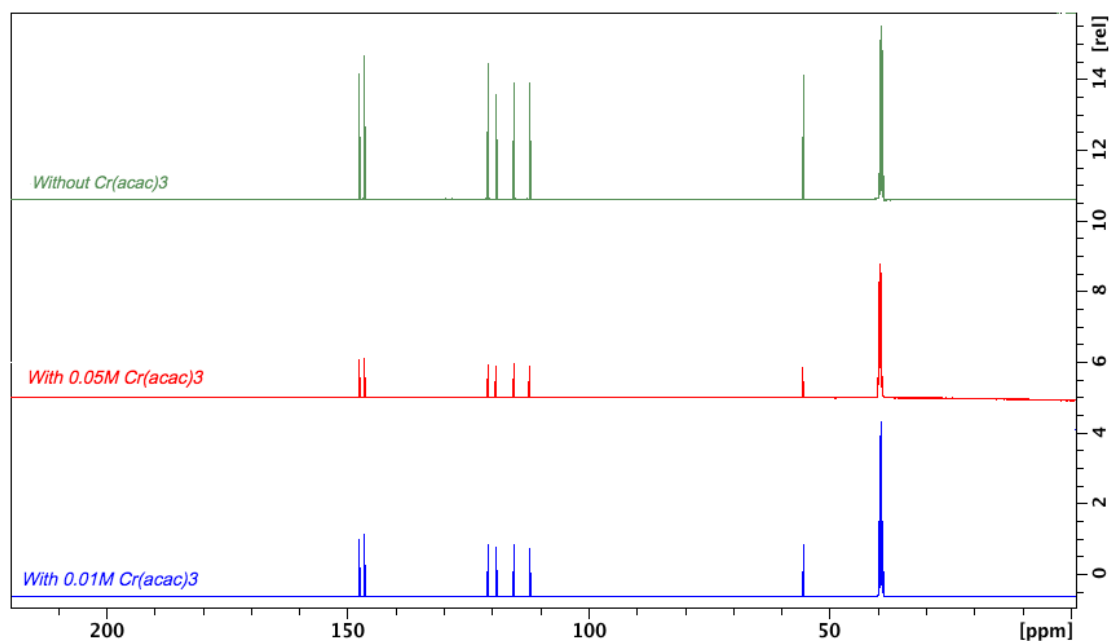


Figure 4-4 Quantitative ^{13}C NMR spectra of Guaiacol (with and without relaxation agent)

4.3 Standard compound NMR experiments

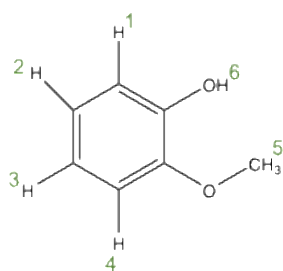
^1H NMR (pulse program: zg30, ns=32, sw=16ppm), ^{13}C NMR (pulse program: zgig30, ns=1024, sw=220ppm), 2D HSQC (pulse program: hsqcetgpsi2, td= 256 in F1/1024 in F2, ds=16, ns=8) of the samples of 10% standard compounds in DMSO- d_6 were performed on Bruker AV500. (See spectra in Appendix A1-A9)

The 2D HSQC pulse sequence utilizes several polarization transfers steps that increase the overall sensitivity of the experiment. The experiment employs pulsed field gradients to select the proper magnetization components to produce 2D ^1H - ^{13}C HSQC spectra. Such a sequence effectively removes residual ^1H signals from hydrogens connected to ^{12}C -containing carbon atoms and suppresses the solvent signal. This 2D experiment correlates the chemical shift of proton with the chemical shift of the directly bonded carbon. On the bottom axis is a proton spectrum and on the other is a carbon.

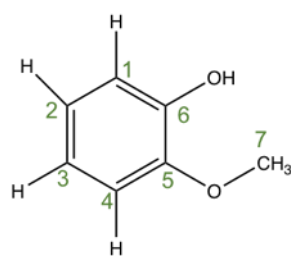
Cross peaks give the shift of the corresponding proton and carbon.

The following results shows the peak identification of each compounds, ^1H chemical shift and ^{13}C chemical shift are assigned according to the connection between H and C in 2D HSQC NMR spectra:

Standard 1 - Guaiacol
(Spectra in appendix A1-A5)

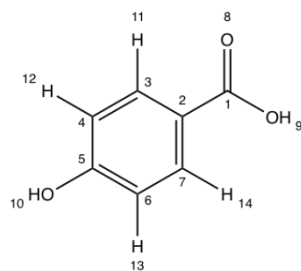


Assign	^1H shift/ppm	^1H shift/ppm
	CDCL ₃	DMSO-d ₆
1	6.90	6.79
2	6.89	6.70
3	6.83	6.64
4	6.81	6.64
5	3.84	3.64
6	5.60	8.77



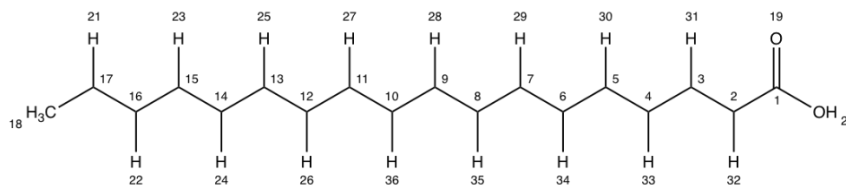
Assign	^{13}C shift/ppm	^{13}C shift/ppm
	CDCl_3	DMSO-d_6
1	114.5	116.0
2	121.5	121.4
3	120.1	119.7
4	110.7	112.8
5	146.6	148.1
6	145.7	147.0
7	55.9	55.9

Standard 2 - 4-Hydroxy benzoic acid
(Spectra in appendix A6-A8)



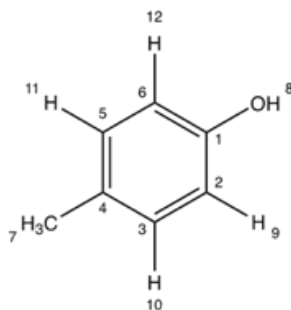
Assigned	^1H shift/ppm	Assign	^{13}C shift/ppm
H Atom number	DMSO	C Atom number	DMSO
11	7.80	1	167.1
12	6.83	2	121.3
10	10.24	3	131.5
13	6.81	4	115.1
14	6.81	5	161.6
8	12.39	6	115.1
		7	131.5

Standard 3 - Stearic acid
(Spectra in appendix A9-A11)



Assigned H Atom number	¹ H shift/ppm DMSO	Assign C Atom number	¹³ C shift/ppm DMSO
18	0.88	1	180.4
20	11.48	2	34.1
21-30, 33-36	1.26	3	24.7
31	1.63	4-15	29.7
32	2.34	16	31.9
		17	22.7
		18	14.1

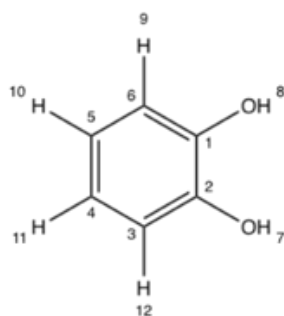
Standard 4 – p-Cresol
(Spectra in appendix A12-A14)



Assigned H Atom number	¹ H shift/ppm DMSO	Assign C Atom number	¹³ C shift/ppm DMSO
7	2.19	1	155.0
8	9.09	2	115.0
9	6.68	3	127.1
10	6.95	4	129.7
11	6.95	5	127.1
12	6.68	6	115.0
		7	20.0

Standard 5 - 1,2-Dihydroxy benzene

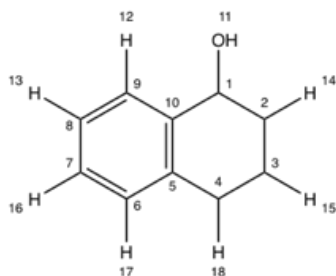
(Spectra in appendix A15-A16)



Assigned H Atom number	¹ H shift/ppm DMSO	Assign C Atom number	¹³ C shift/ppm DMSO
7	8.79	1	145.7
8	8.79	2	145.7
9	6.72	3	116.0
10	6.72	4	119.8
11	6.60	5	119.8
12	6.60	6	116.0

Standard 6 - 1,2,3,4-Tetrahydronaphthol

(Spectra in appendix A17-A19)



Assigned H Atom number	¹ H shift/ppm DMSO	Assign C Atom number	¹³ C shift/ppm DMSO
1	4.58	1	66.8
11	5.09	2	32.8
12	7.40	3	19.5
13,16	7.14	4	29.3
17	7.06	5	136.8
18	2.72	6	128.8
4	2.67	7	126.0
3,15	1.71	8	127.0
2,14	1.90	9	128.7

1	4.58	10	140.8
---	------	----	-------

4.4 Bio-oil based mixture NMR experiments

^1H NMR(pulse program: zg30, ns=128, sw=20ppm), ^{13}C NMR (pulse program: zgig30, ns=1024, sw=220ppm), 2D HSQC NMR (pulse program: hsqcetgpsi2, td= 256 in F1/1024 in F2, ds=16, ns=8) experiments for the samples of mixtures 1-5 in DMSO-d6 are performed on Bruker AV600. (See spectra in Appendix A20-A24)

The peaks of ^1H chemical shift and ^{13}C chemical shift were assigned in chapter 4.3. Functional group are identified by chemical shift range and 2D HSQC spectra . The standard compounds 1-6 can be identified in 2D HSQC spectra of Mixtures, see the example of Mixture 5 of all the standard compounds in Figure 4-5. Figure 4-6 &4-7 shows the summary of functional group assignment in ^1H & ^{13}C spectra.

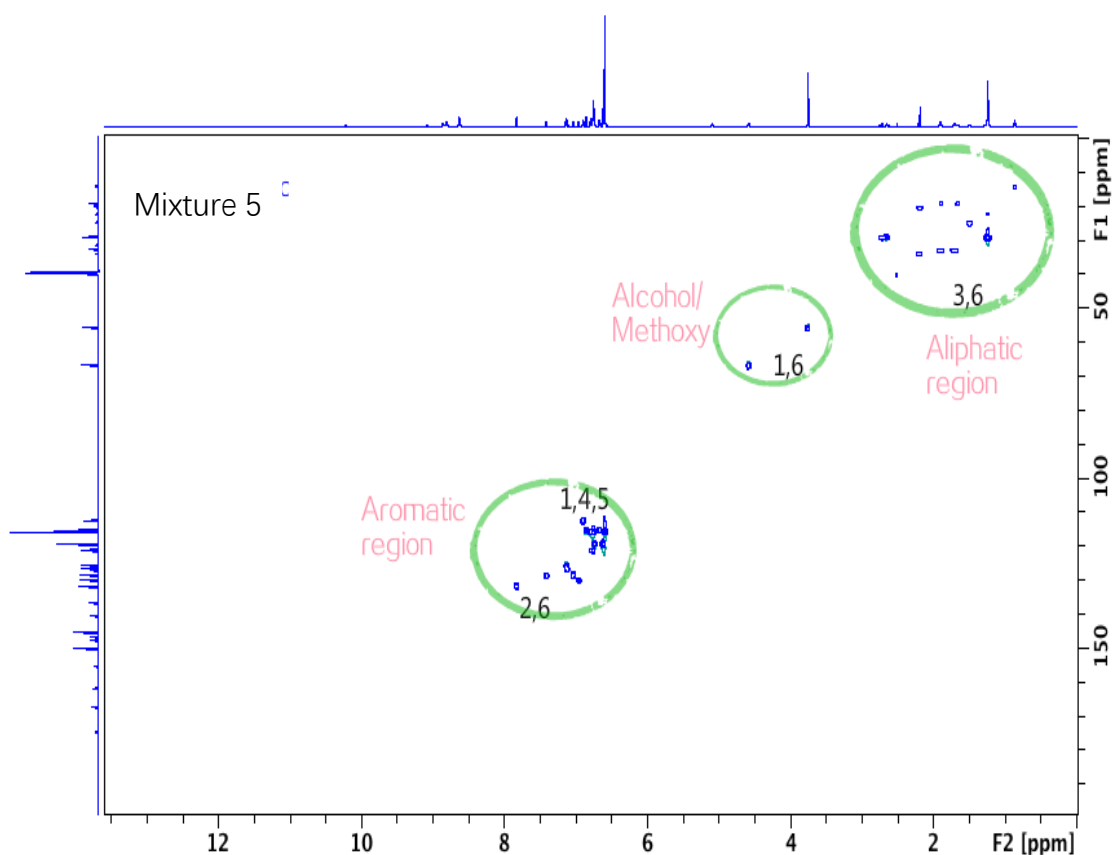


Figure 4-5 Identification of 6 compounds in 2D HSQC NMR spectrum of mixture 5

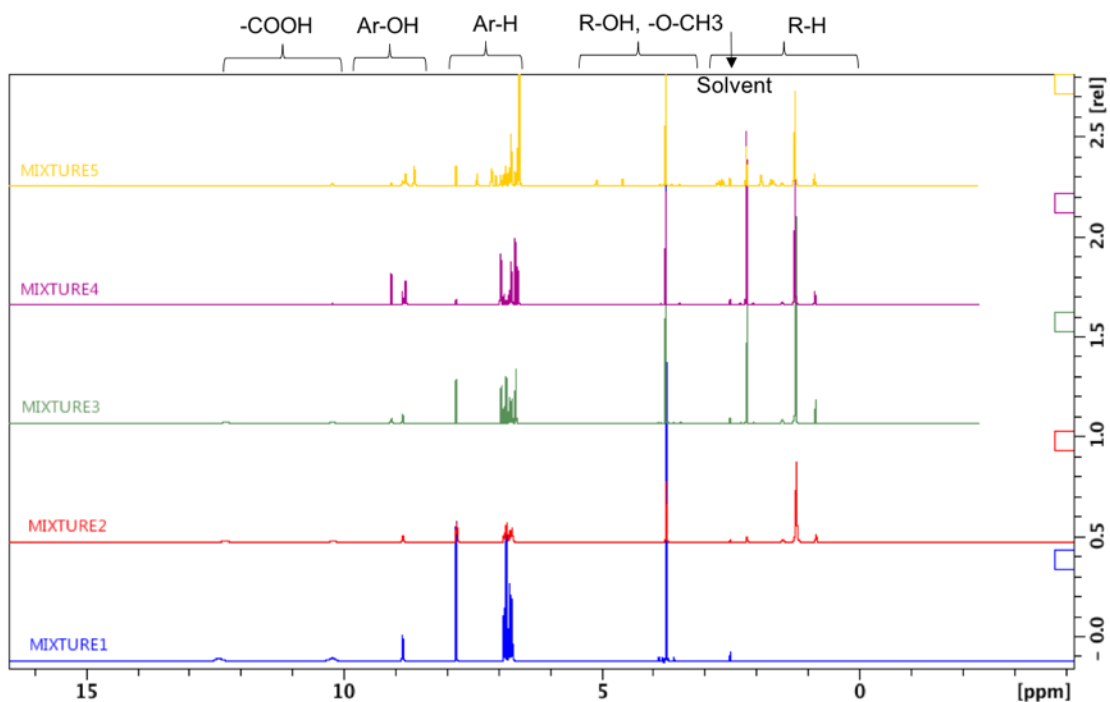


Figure 4-6 ^1H NMR chemical shift assignment of functional groups in Mixtures

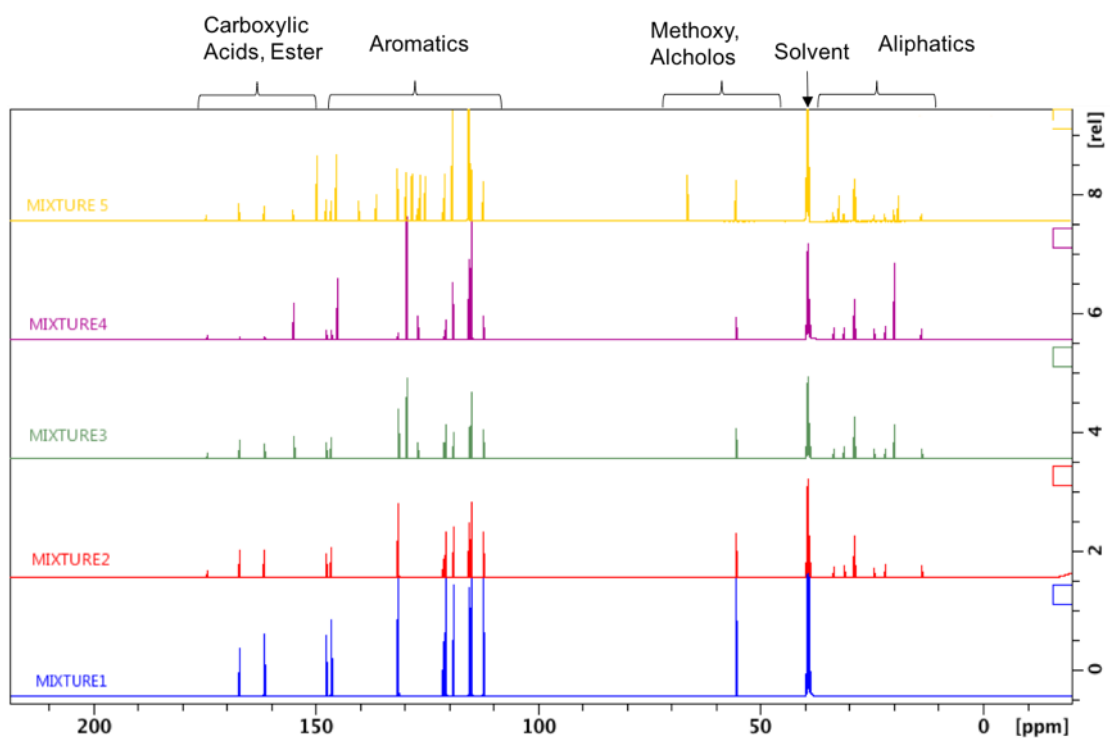


Figure 4-7 ^{13}C NMR chemical shift assignment of functional groups in Mixtures

4.5 LtL-oil NMR experiments

Quantitative ^{13}C NMR spectra of 6 LtL-oil samples in DMSO-d₆ with 0.01 M Cr(acac)₃ were collected on Bruker Advance 850MHz with an inverse-gated

decoupling pulse sequence (pulse program: zgig, sw=238ppm, d1=10, ns=3072). (See spectra in Appendix A25-A30, and integration raw data in Appendix B4-B7)

4.5.1 Two LtL- oils with and without formic acid ^{13}C

Figure 4-8 shows a comparison of ^{13}C NMR spectra of LtL oil-1 and LtL oil-2. Two LtL-oils were converted from same lignin at same condition, while the reaction solvents were standard formic acid and ^{13}C labelled formic acid. The chemical shift of ^{13}C peaks are similar for the two oils. However, it is clear to see two differences in the spectra. The signal from peak at 166.2 ppm in experiment 2 is significant stronger than experiment 1, which is from residual formic acid (One signal at 166.22ppm is acquired in ^{13}C NMR spectra of standard CH_2O_2 [39]). Also, the intensity of the peaks over 170ppm is higher in oil made with ^{13}C -enriched formic acid than the oil made with standard formic acid. This is an evidence demonstrating that the formic acid does not only work as a hydrogen donor but also contributes with carbon to the produced LtL-oil.

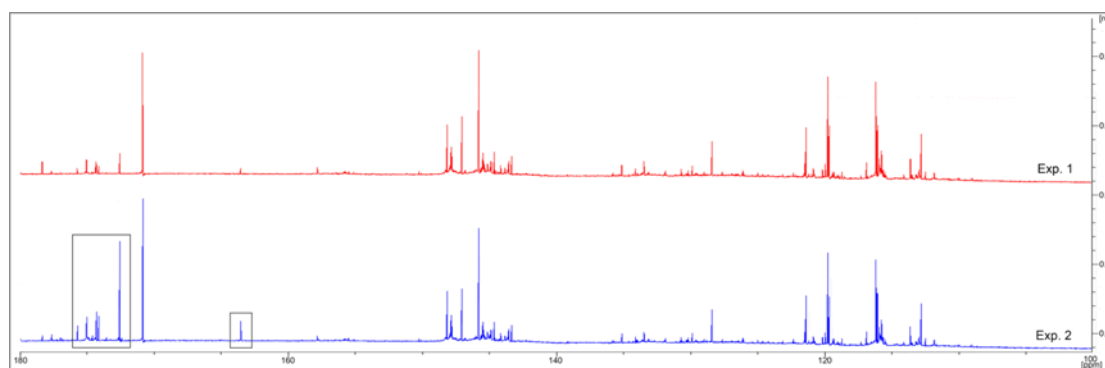


Figure 4-8 ^{13}C NMR spectra of LtL-oil produced on the conditions -360 °C, 0.5 mL, standard formic acid (Exp.1)/ ^{13}C -labeled formic acid (Exp.2)

4.5.2 Four LtL- oils converted under different conditions

Figure 4-9 shows a comparison of quantitative ^{13}C NMR spectra of LtL oil-3,4,5,6.

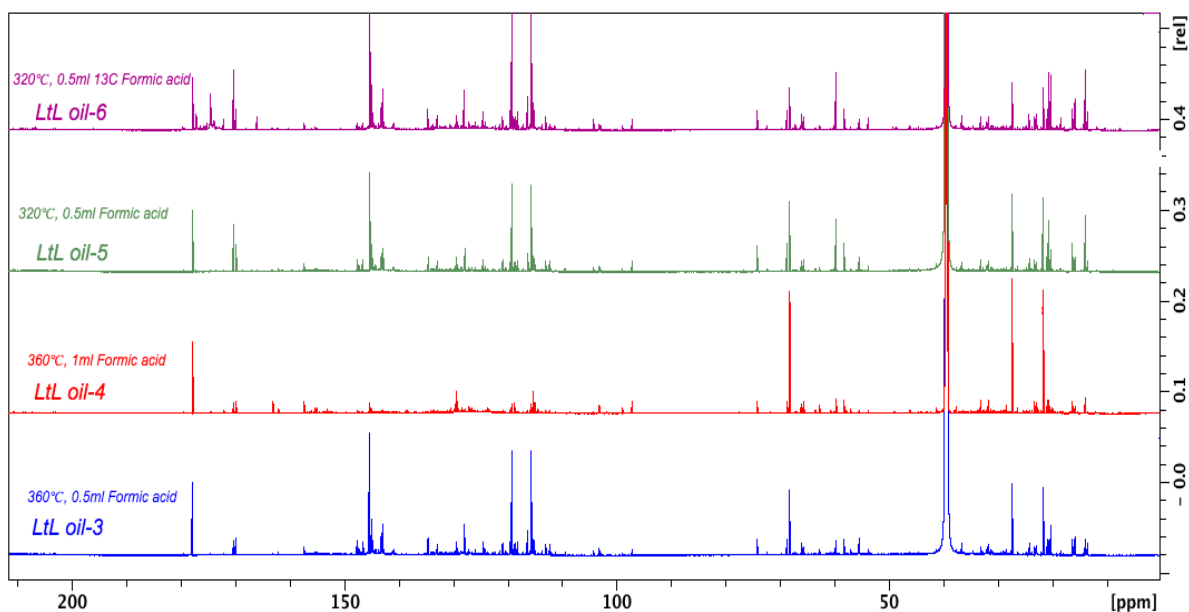


Figure 4-9 Quantitative ¹³C NMR spectra of LtL-oil produced on different conditions

The integration of quantitative ¹³C NMR spectra were manually calculated using the topspin software, a previous developed structural assignment range of bio-oil in Table 4-2 [40] were used in this study.

Table 4-2 ¹³C NMR Chemical Shift Assignment Range of Lignin bio-oil

Functional group	Chemical shift range (ppm)
Carbonyl or Carboxyl bond	215.0-166.5
Aromatic C-O bond	166.5-142.0
Aromatic C-C bond	142.0-125.0
Aromatic C-H bond	125.0-95.8
Aliphatic C-O bond	95.8-60.8
Methoxyl-Aromatic bond	60.8-55.2
aliphatic C-C	55.2-0

The integration results of functional group in four LtL-oil based on ¹³C NMR spectra are summarized in table 4-3.

Table 4-3 Integration Results for LtL-Oils produced under different conditions, detected by quantitative ^{13}C NMR using the assignment range shown in Table 4-2*

Functional group	LtL-Oil 3 360°C, 0.5M FA	LtL-Oil 4 360°C, 1M FA	LtL-Oil 5 320°C, 0.5M FA	LtL-Oil 6 320°C, 0.5M FA- ^{13}C
C=O	4.3%	9.1%	6.8%	23.5%
Aromatic C-O	18.2%	8.3%	14.2%	19.5%
Aromatic C-C	5.4%	9.1%	4.5%	6.7%
Aromatic C-H	31.8%	14.6%	26.2%	24.4%
Aliphatic C-O	7.3%	15.1%	9.2%	4.4%
Methoxyl-	11.5%	3.4%	6.0%	3.1%
Aliphatic C-C	21.5%	40.3%	33.2%	18.3%

* The results were shown as the percentage of carbon, which is the relative amounts of carbon, not the absolute amounts, since this depends on the amount of oil produced to.

On the basis of ^{13}C NMR LtL-oil 3 and LtL-oil 5, the relative number of aromatic carbons increased with increasing temperature from 320°C to 360°C. In the contrast, the number of aliphatic carbons decreased with an increasing temperature. While under the same production temperature 360°C, there are more C=O groups, as well as aliphatic groups and less aromatic groups in LtL-oil 4 than LtL-oil 3 due to double amount of formic acid presented. In LtL-oil 6, significantly higher amount of carbonyl or carboxyl groups were detected, compare with the LtL-oil 5 which was produced under same condition, it's confirmed that some carbon in carbonyl or carboxyl group are transferred from ^{13}C labelled formic acid.

Chapter 5 Discussion

5.1 The choice of deuterated solvent

Before running the samples for NMR experiments, the first step is to choose a suitable solvent. Since deuterium is by far the most popular lock nucleus, the sample is usually dissolved in a deuterated solvent (a deuterated solvent is one in which a large proportion, typically more than 99%, of the hydrogen atoms have been replaced by deuterium). Chloroform-d and DMSO-d6 were compared to be use in this study.

The choice deuterated solvent is decided by solvent solubility, solvent peak overlap, and solvent effect:

a. Solubility

All the samples can be easily dissolved in DMSO-d6. Clearly the more soluble the sample is in the solvent the better. This maximizes the amount of sample within the sensitive volume which increases the sensitivity of the experiment.

b. Interfering peaks

The solvent itself will inevitably produce NMR signals which will obscure regions of the spectrum. These ‘residual solvent peaks’ should not overlap with signals from the sample. The chemical shifts of solvent signals observed for ^1H NMR and ^{13}C NMR spectra are listed in the table 5-1, it shows ^1H chemical shift of DMSO-d6 is less overlap than CDCl_3 in aromatic area:

Table 5-1: Chemical shift of CDCl_3 and DMSO-d6 in ^1H and ^{13}C NMR spectra*:

	^1H NMR chemical shift/ppm	^{13}C NMR chemical shift/ppm
CDCl_3	7.26	77.3
DMSO-d6	2.50	39.5
Aromatics group	7-8	95-166

*Data were acquired from: <https://webspectra.chem.ucla.edu/NotesOnSolvents.html>

^1H chemical shift of DMSO- d_6 is 2.5 ppm which is far away from aromatics group 7-8 ppm, while CDCl_3 is 7.26ppm. It shows the solvent peak of DMSO- d_6 will have less overlap in aromatic area in the ^1H NMR spectra.

c. Solvent effect:

The spectra show that the peak from Ar-OH in guaiacol is more intensive and easier to be identified in high chemical shift range ($>8\text{ppm}$) in DMSO- d_6 . (Spectra in appendix A1&A3). It's assumed that the Aromatics-OH group is more stable in DMSO- d_6 due to hydrogen bond. The difference of ^1H NMR chemical shift in Ar-OH group of Guaiacol can be observed in Figure 5-1. The chemical shifts of OH protons vary over a wide range depending on substrate structure, solvent, temperature and concentration. The shifts are very strongly affected by hydrogen bonding, with large downfield shifts of H-bonded groups compared to free OH groups. Thus OH signals tend to move downfield with increased hydrogen bonding solvents like DMSO.

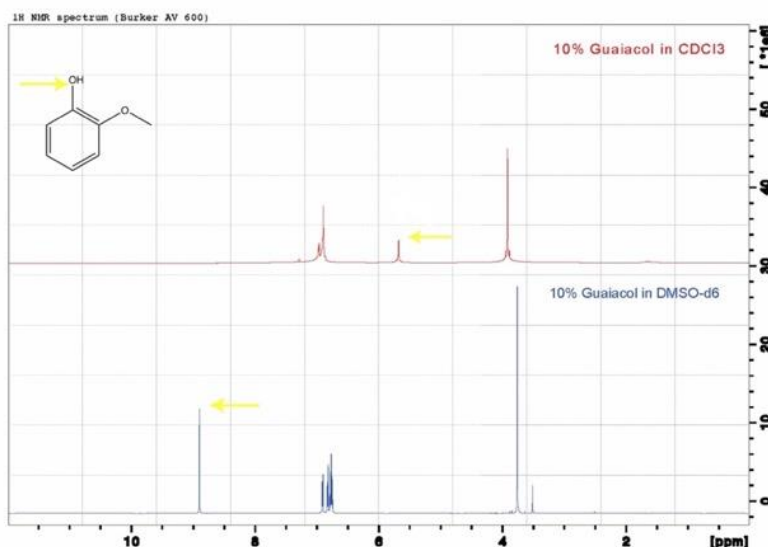


Figure 5-1 A comparison of ^1H NMR spectrum for guaiacol in CDCl_3 and DMSO- d_6

Consequently, DMSO- d_6 was chosen as suitable deuterated solvent for the experiments because of the good solubility, less peak overlap in ^1H NMR spectra and

stable solvent effect.

5.2 The choice of relaxation agent

To get accurate integration of ^{13}C signals, the quantitative ^{13}C NMR were recorded with broad band proton-decoupling only during the acquisition periods (“inverse-gating”), to avoid nonuniform enhancement of carbon signals from proton NOE effects (polarization transfer). As each LtL oil sample was a complex mixture, the concentration of each individual component was very low. These low concentrations, in combination with the intrinsic low sensitivity of the ^{13}C nuclei, required large numbers of transients to achieve visually acceptable signal-to-noise ratios. Such lengthy signal-averaging made the direct determination of accurate carbon T1 relaxation times impractical due to long acquisition times.

It was measured the longest T1 is at least 8,816s in chapter 4.2, which results in that the d1 should be set to 44s (five times the longest T1), which would have allowed the full recovery of all signals and the most accurate measurement of integrated intensities. In that case, a 12 h experiment time is needed (calculated by Topspin with a command “expt” for estimating the experiment time) for one sample with ns=1024.

In order to reduce the experiment time, a relaxation agent Chromium(III) acetylacetonate was considered to add into LtL-oil sample. Relaxation reagents are paramagnetic substances capable of strongly decreasing the relaxation times of substrate nuclei without inducing noticeable shifts. Both an electron nuclear dipolar interaction and an electron-nuclear contact interaction may contribute to the overall relaxation.

Table 5-2: Integral results of Quantitative ^{13}C NMR spectra of three guaiacol samples

Object	Integral [rel]		
	Guaiacol with 0.05M Cr(acac) ₃	Guaiacol with 0.01M Cr(acac) ₃	Guaiacol without relaxation agent
Peak 1	1.0	1.0	0.9
Peak 2	1.0	1.0	1.1
Peak 3	1.0	1.0	1.0
Peak 4	1.0	1.0	1.0
Peak 5	1.0	1.0	1.0
Peak 6	1.0	1.0	0.9
Peak 7	1.0	1.0	1.1

Guaiacol contains 7 carbons and each carbon gives a peak in the ^{13}C NMR spectrum. During the integration process, the sum of integrals is normalized to be 7, then the results of each peak integrals were calculated by topspin in table 5-2. Two samples with 0.01M and 0.05M Cr(acac)₃ got the more accurate carbon integration results than the sample without relaxation agent. Besides, no big difference was observed between 0.01M and 0.05M Cr(acac)₃. (See the integration raw data in Appendix B1-B3)

It is showed that adding Cr(acac)₃ can speeded up the relaxation and the carbons were fully relaxed when d1=10, where d1=55 for without relaxation reagent sample. It reduced the experiment time from 12h to 3h per sample with ns=1024.

Both 0.01M and 0.05M Cr(acac)₃ can reduce the NMR acquisition time to get fully relaxed ^{13}C NMR spectra. However, an excess of this relaxation agent deteriorates the spectral resolution. The spectrum with 0.01M Cr(acac)₃ has better resolution and quality than spectrum with 0.05M Cr(acac)₃. Therefore, 0.01M Cr(acac)₃ was added into LtL oil NMR samples as relaxation agent to achieve more accurate quantitative ^{13}C

NMR spectra of LtL oils.

5.3 Comparison of NMR spectra acquired from different magnets

Quantitative ^{13}C NMR spectra of LtL-oil sample 1 in DMSO- d_6 with 0.01 M Cr(acac) $_3$ were collected on Bruker Advance 850MHz/AV600 MHz/AV500MHz with an inverse-gated decoupling pulse sequence (pulse program: zig 30, sw=238ppm, d1=10, ns=3072). Figure 5-2 shows the quantitative ^{13}C spectra of LtL oil 1 acquired under different magnet field (AV500/AV600/850MHz).

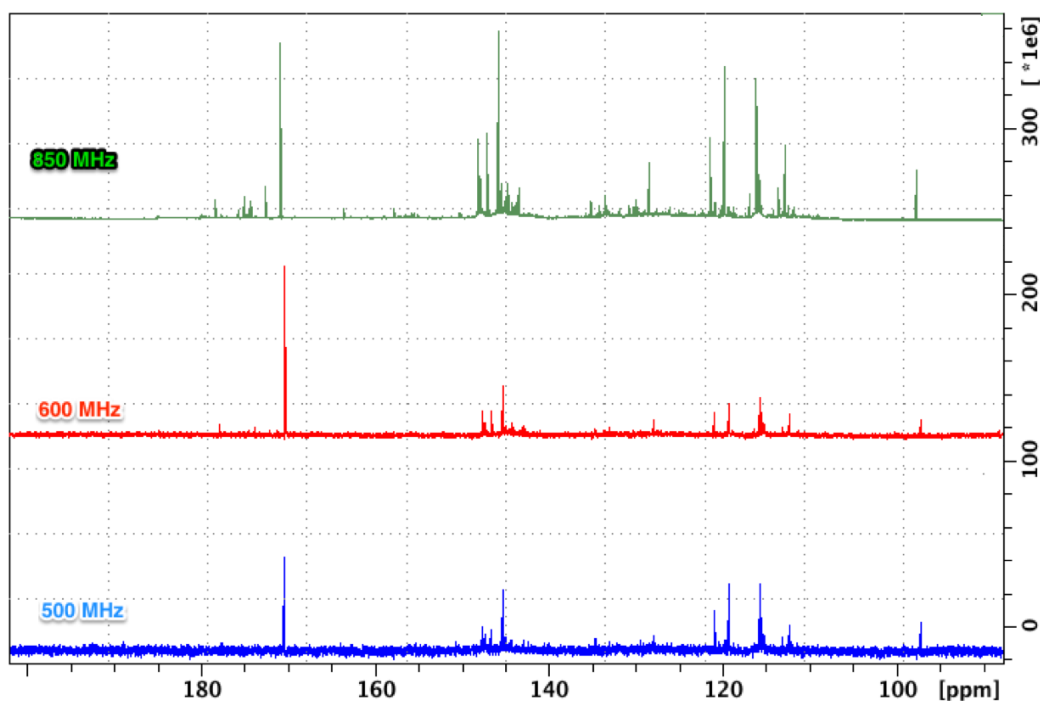


Figure 5-2 A comparison of ^{13}C NMR spectrum for LtL oil 1 in DMSO- d_6 acquired under different magnet field (AV500/AV600/850MHz)

a. Resolution of NMR spectra

The energy separation between the two spin states of a spin $\frac{1}{2}$ nucleus is directly proportional to the strength of the magnetic field ($\Delta E = \mu B_0$). This in turns affects the Boltzmann population differences of the α and β spin states. Thus, stronger magnetic fields result in large increases in the strength of the NMR signal.

It is obvious that more peaks are visible in 850MHz spectrum than 500 MHz and

600MHz spectra. Also, there is a noticeable difference in the spread of the chemical shifts, peaks are better separated in 850MHz spectrum. The larger the separation is, the more chance one has to be able to fully interpret the spectrum with visual inspection.

b. Sensitivity of NMR spectra

The sensitivity of NMR experiments is given by the signal to noise ratio[35]:

$$S/N = \frac{N\gamma_{exc}T_2(\gamma_{det}B_0)^{3/2}\sqrt{ns}}{T}$$

S/N	=	signal to noise ratio
N	=	number of spins in the system (sample concentration)
γ_{exc}	=	gyromagnetic ratio of the excited nucleus
γ_{det}	=	gyromagnetic ratio of the detected nucleus
ns	=	number of scans
B_0	=	external magnetic field
T_2	=	transverse relaxation time (determines the line width)
T	=	sample temperature

with same number of scans, high applied magnetic field is recognized as the most straightforward means to enhance NMR sensitivity. It is clearly seen that 850MHz spectra has higher signal-noise ratio than 500 MHz and 600MHz spectra.

Therefore, LtL-oils were analyzed by 850MHz because of its both highest resolution and sensitivity, which gives better spectra for interpretation, especially for integration of ^{13}C spectra. However, 850 MHz NMR is pricy and should be booked for complex samples. For standard compound samples, 500MHz was efficient to analyze ^1H NMR and ^{13}C NMR since the component was pure and the concentration was high enough to achieve strong signals, it can be used for routine ^1H and ^{13}C experiments where the main goal is identification and characterization of organic compounds.. For mixture samples, 600MHz was chosen to use because its magnet field is higher than 500MHz, and it is allowed to manually shim the sample and try different parameters to achieve preferable spectra.

5.4 Functional groups in LtL-oils

The ^{13}C NMR chemical shift of selected standard compounds was identified in chapter 4.1; however, LtL-oil contains more chemical compounds than used in this study. A complete chemical shift range of different functional groups in LtL-oil cannot be assigned based on only standard compounds, and a previous developed structural assignment range of bio-oil in Table 4-2 [40] were used for integration of quantitative ^{13}C NMR spectra. Several different assignments of ^{13}C NMR chemical shift integration ranges were proposed by researchers, the reason to choose it in this study is that most peaks from standard compounds fits the functional group assigned range in Table 4-2, and the only exceptions came from standard compound 6 - 1,2,3,4-Tetrahydronaphthol. The ^{13}C chemical shift of aromatic C-H in standard compound 6 are 126.0-128.8 ppm, but these peaks would be included into aromatic C-C range (125.0-142.0 ppm) instead of aromatic C-H range (95.8-125.0 ppm) according to Table 4-2. It is considered there would be some overlap in these ranges.

a. Production temperature effect

The integration results of LtL-Oils were calculated in chapter 4.5, Table 5-3 & Table 5-4 show that with an increasing temperature from 320°C to 360°C, the percentage of aromatic carbons increased from 44.9% to 55.4%, while the ratio of C-C: C-O: C-H in aromatic carbons are similar (*0.3: 1: 1.8* and *0.3: 1: 1.7*). Also, Methoxyl C-O increased significantly from 6.0% to 11.5%. In the contrast, the number of aliphatic Carbon with C-C bond decreased from 33.2% to 21.5%, aliphatic carbon with C-O bond decreased from 9.2% to 7.3% and C=O decreased from 6.8% to 4.3%.

Table 5-3 Carbon percentage in functional group of LtL-oils produced from lignin under different temperature (same formic acid amount =0.5ml)

	Temperature	Aromatic carbon	Methoxyl C-O	Aliphatic C-C	Aliphatic C-O	C=O
LtL-Oil 5	320°C	44.9%	6.00%	33.20%	9.20%	6.80%
LtL-Oil 3	360°C	55.4%	11.50%	21.50%	7.30%	4.30%

Table 5-4 The ratio of different carbon bond in Aromatic carbons in LtL-oils produced under different temperature (same formic acid amount =0.5ml)

	Temperature	Aromatic C-C	Aromatic C-O	Aromatic C-H
LtL-Oil 5	320°C	0.3	1	1.8
LtL-Oil 3	360°C	0.3	1	1.7

b. Formic acid amount effect

On the other hand, Table 5-5 & Table 5-6 show that under the same production temperature 360°C, there are more C=O groups, as well as Aliphatic groups and less Aromatic groups in LtL-oil 4 than LtL-oil 3 due to double amount of formic acid presented. With the increasing presented amount of formic acid from 0.5M to 1M, the percentage of aromatic carbons decreased from 55.4% to 32.0%, while the ratio of C-C: C-O: C-H in aromatic carbons are quite different (0.3: 1: 1.7 and 1.1: 1: 1.8) which means significantly more C-C bonds are presented in LtL-oil produced by high amount formic acid. Also, Methoxyl C-O decreased dramatically from 11.5% to 3.4%. In contrast, the number of aliphatic Carbon with C-C bonds increased from 21.5% to 40.3%, aliphatic carbon with C-O bond increased from 7.3% to 15.10% and C=O increased from 4.3% to 9.1%.

Table 5-5 Carbon percentage in functional group of LtL-oils produced from lignin with different amount of formic acid (same production temperature=360°C)

	Formic acid	Aromatic carbon	Methoxyl C-O	Aliphatic C-C	Aliphatic C-O	C=O
LtL-Oil 3	0.5M	55.40%	11.50%	21.50%	7.30%	4.30%
LtL-Oil 4	1M	32.00%	3.40%	40.30%	15.10%	9.10%

Table 5-6 The ratio of different carbon bond in Aromatic carbons in LtL oil produced with different amount of formic acid (same production temperature=360°C)

	Formic acid	Aromatic C-C	Aromatic C-O	Aromatic C-H
LtL-Oil 3	0.5M	0.3	1	1.7
LtL-Oil 4	1M	1.1	1	1.8

c. Carbon incorporation from formic acid

From the results calculated in chapter 4.5, the LtL-oil 6 which contains ^{13}C labelled formic acid, significant higher amount of Carbonyl or Carboxyl group were detected with a carbon percentage of 23%, compare with the LtL-oil 5 which was produced under same condition, it's confirmed that some carbon in Carbonyl or Carboxyl group are transferred from ^{13}C labelled formic acid.

It is also noticed in Table 5-7 that the aliphatic carbon ratio of C-C: C-O: C-H bonds are slightly different between 2 oils ($1 : 1.5 : 5.5$ and $1 : 1.5 : 5.9$), it seems carbon from the formic acid didn't incorporate in the Methoxyl/ Aliphatic C-O structures, but incorporates in aliphatic C-C structure in LtL-oil. In the aromatic composition, the carbon percentage of aromatic carbon is higher in LtL oil 6 which produced under ^{13}C labeled formic acid than LtL oil 5, and Table 5-8 shows the aromatic carbon ratio of C-C: C-O: C-H bonds are slightly varied. It seems some of carbon in aromatic group are also transferred from ^{13}C labelled formic acid.

Table 5-7 The ratio of different carbon bond in Aliphatic carbons in LtL oil produced with with normal and ^{13}C labelled formic acid

	Production condition	Formic acid	Methoxyl- C-O	Aliphatic C-O	Aliphatic C-C
LtL-Oil 5	320°C, 0.5ml FA	<i>Normal</i>	1.0	1.5	5.5
LtL-Oil 6	320°C, 0.5ml FA	^{13}C labelled	1.0	1.5	5.9

Table 5-8 The ratio of different carbon bond in Aromatic carbons in LtL oil produced with normal and ^{13}C labelled formic acid

	Production condition	Formic acid	Aromatic C-C	Aromatic C-O	Aromatic C-H
LtL-Oil 5	320°C, 0.5ml FA	<i>Normal</i>	0.3	1	1.8
LtL-Oil 6	320°C, 0.5ml FA	^{13}C labelled	0.3	1	1.3

In summary, more aromatic carbon and less aliphatic carbon are achieved when the experiment temperature is increasing, while the aromatic compositions are not varied under different temperature. On the other hand, less aromatic carbon and more aliphatic carbon are achieved when higher amount formic acid is presented. Also, the aromatic compositions vary with different amount of formic acid added. When more formic acid is present, then more aromatic C-C bonds are produced compared to aromatic C-O bonds and C-H bonds.

Besides, carbon from the formic acid mostly incorporates in the carbonyl structures, while it seems also partly incorporates in aliphatic C-C structures and aromatic group in LtL oils. It's confirmed that the formic acid is a reactant in the lignin conversion in addition to be a hydrogen donor.

In conclusion, the NMR technologies provide a facile way to analyze LtL-oils. NMR analysis is well-known to be an insensitive research tool, and for bio-oils it is better suited to analyze changes in functional group composition than to identify

individual compounds.

Quantitative measurements of the chemical constituents in whole bio-oil are difficult because of its viscous, acidic, and unstable nature. Further, bio-oil is a complex mixture containing over 300 compounds, mostly present in low concentrations, including ketones, aldehydes, anhydrous sugars, carboxylic acids, and phenolic compounds. The main advantages of the application of NMR to the analysis of bio-oils are (1) the whole bio-oil can be dissolved in an appropriate solvent and information about the whole spectrum of functional groups can be obtained, which does not depend on the volatility of the components in the bio-oils; and (2) the chemical shift ranges for functional groups can be studied, and quantitative analysis of functional groups can be achieved by integration of peaks based on the proposed chemical shift assignment ranges.

Since most LtL research focuses on reducing the oxygen contents in bio-oils through optimizing the solvolysis experiment parameters (e.g., temperature and solvent atmosphere), ^1H , ^{13}C NMR analyses are powerful tools for obtaining structure information about the whole fraction of bio-oils. ^1H - ^{13}C HSQC NMR provides carbon-hydrogen information, which is useful for elucidating possible reactive pathways during solvolysis reactions.

The 1-D NMR techniques are quantitative in nature; however, the spectral overlap problems usually occur because of the complex constitution of bio-oils. The appropriate NMR experiment parameters should carefully select to let the nucleus fully relax. The inaccuracy of the integration results should also be considered. It took days to perform ^{13}C quantitative NMR of LtL oil samples on 850MHz with $n_s=3072$, but it's still not enough to get perfect baseline correction and phase correction of spectra due to very low concentration of each component in the LtL oils. Besides, around 300 peaks for each ^{13}C NMR spectrum of LtL oils were manually picked and integrated, some integration from poor signals could be not accurate enough.

Chapter 6 Conclusion

The aim of this thesis was to characterize LtL-oil by Nuclear Magnetic Resonance (NMR) spectroscopy. An appropriate solvent DMSO-d₆ was selected for NMR sample preparation because of its good solubility, less peak overlap in ¹H NMR spectra and stable solvent effect. The efficiency of quantitative ¹³C NMR experiments was improved by reducing the duration of the experimental time by means of relaxation agents and by the use of greater magnetic field strengths to improve signal sensitivity. The optimum relaxation Cr(acac)₃ concentration was determined to be 0.01M and 850MHz magnet was chosen to use for LtL-oil analysis because of its better resolution and sensitivity.

The ¹H NMR, ¹³C NMR and 2D NMR spectra of 6 standard compounds and mixtures which were predicted to be contained in LtL-oil by previous GC-MS analysis results were achieved and their ¹H chemical shift and ¹³C chemical shift were identified. Then a proposed ¹³C NMR chemical shift assignment range of Lignin bio-oil was selected to use for quantitative analysis of functional groups in LtL-oils, based on the comparison of standard compounds assignments. The integration results of quantitative ¹³C NMR spectra about the functional groups in LtL-oils under different conversion conditions were consequently obtained.

It is discovered that both formic acid and reaction temperature in LtL-solvolysis influences the product composition on a functional group level. It is also confirmed that the formic acid is a reactant in the lignin solvolysis conversion in addition to be a hydrogen donor by comparing ¹³C NMR spectra of normal LtL oil and ¹³C-labeled formic acid LtL oil, which indicated functionality and position of the incorporated carbons from formic acid in the solvolysis products.

Chapter 7 Further work

- **Make the NMR spectra better processed**

When it comes to ^{13}C quantitative analysis, a clear and well processed NMR spectrum is essential. However, the LtL oils are complicated mixture, in addition to small amounts and poor solubility of LtL oil samples, the NMR spectra acquired were not perfect regarding the baseline correction and integration. To reduce the noise and correct the phase in NMR spectra, larger number of ns (scan of numbers) can be settled if time permits. Also, the concentration of NMR samples can be increased if enough sample is available.

- **Make the functional groups integration more accurate.**

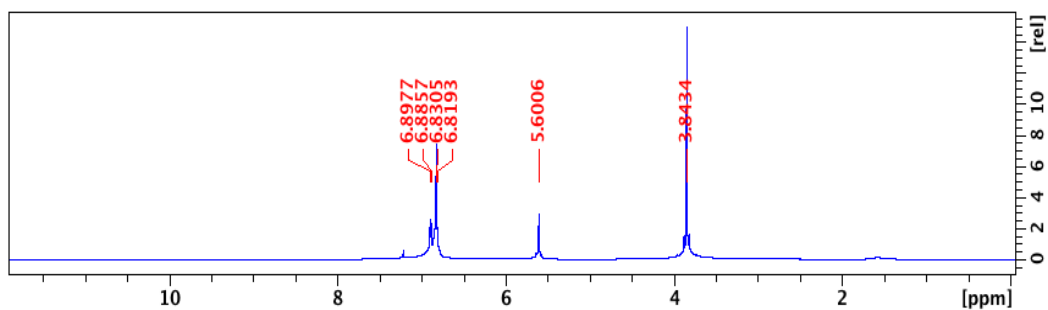
The ^{13}C NMR chemical shift of functional group assignment of Lignin bio-oil is identified based on standard compound NMR spectra assignment and previous published papers. Most peaks from standard compounds fits the functional group assigned range, however, there are a few exceptions and LtL oil contains more chemical compounds than used in this study. In order to find these exceptions to make the functional group integration more accurate, it could be helpful to run NMR experiments of more possible chemical compounds and identify their peaks.

- **Investigation of the role of formic acid**

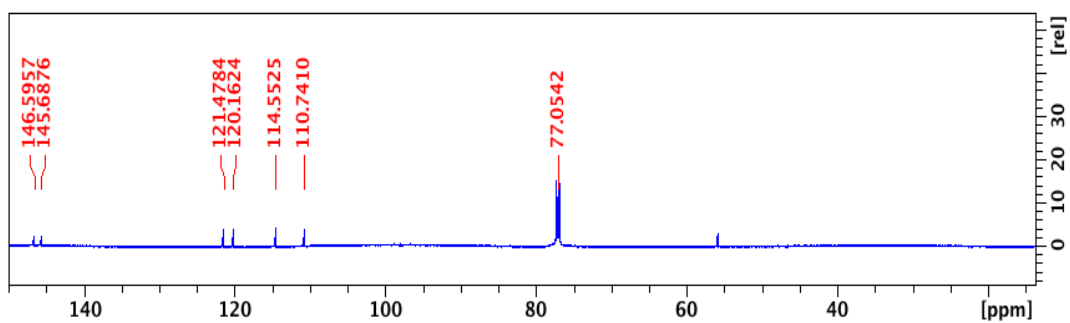
2D NMR, for example HMBC and HSQC can be used further identify relevant functional groups in LtL oil. It will help understand the reaction pathways of producing LtL oils and investigate the role of formic acid.

Appendix A NMR spectra

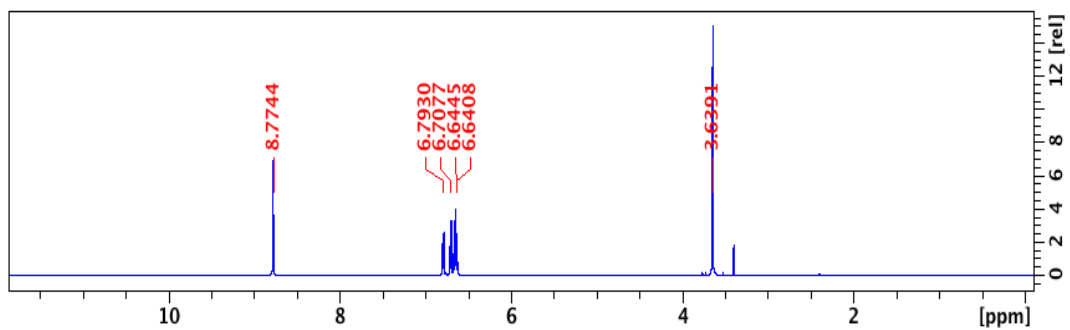
A1. ^1H NMR spectrum of compound 1 in CDCl_3



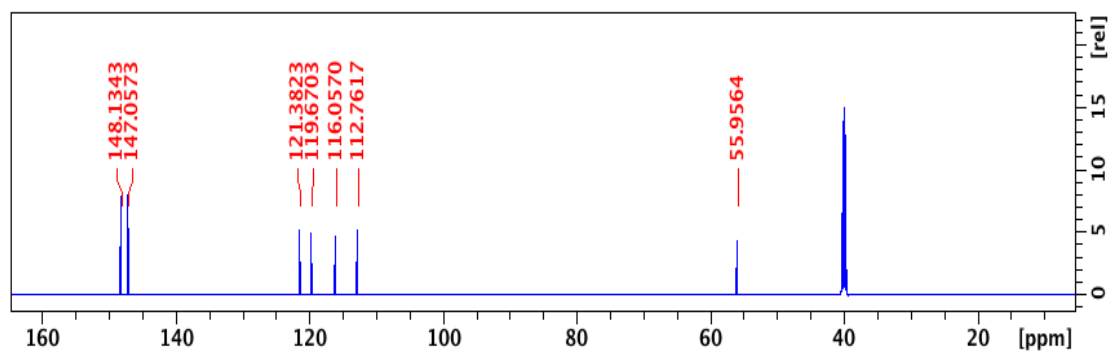
A2. ^{13}C NMR spectrum of compound 1 in CDCl_3



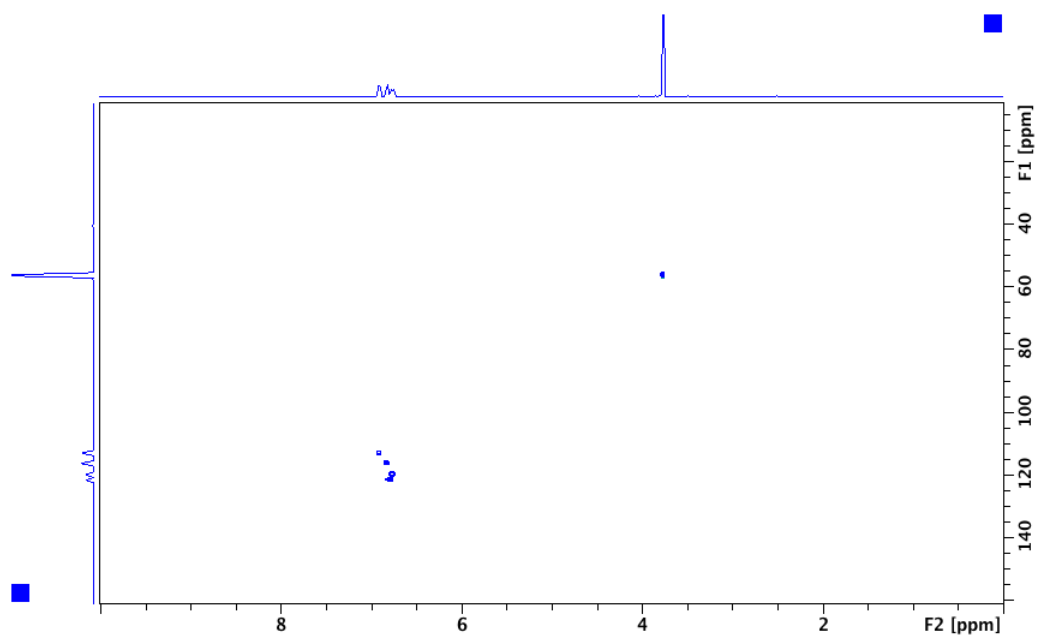
A3. ^1H NMR spectrum of compound 1 in DMSO-d_6



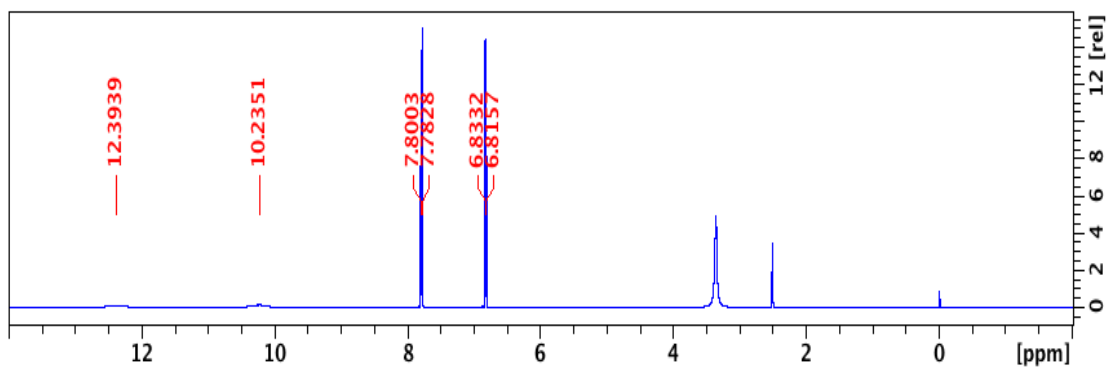
A4. ^{13}C NMR spectrum of compound 1 in DMSO-d₆



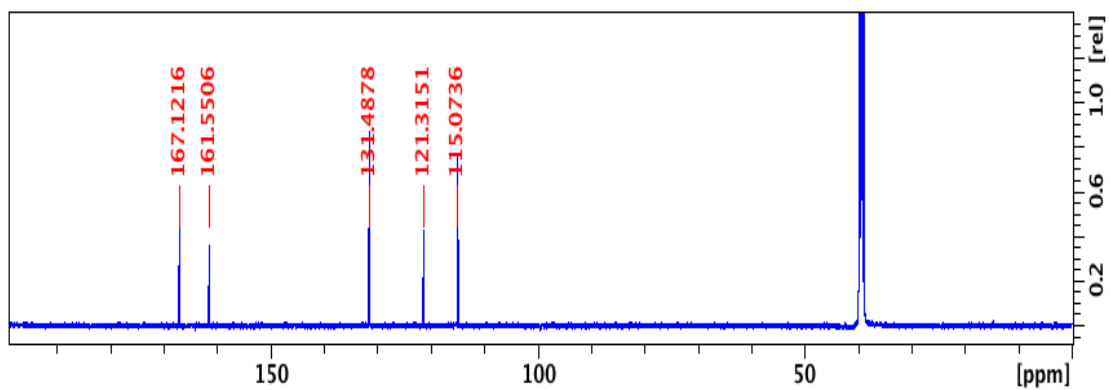
A5. 2D HSQC NMR spectrum of compound 1 in DMSO-d₆



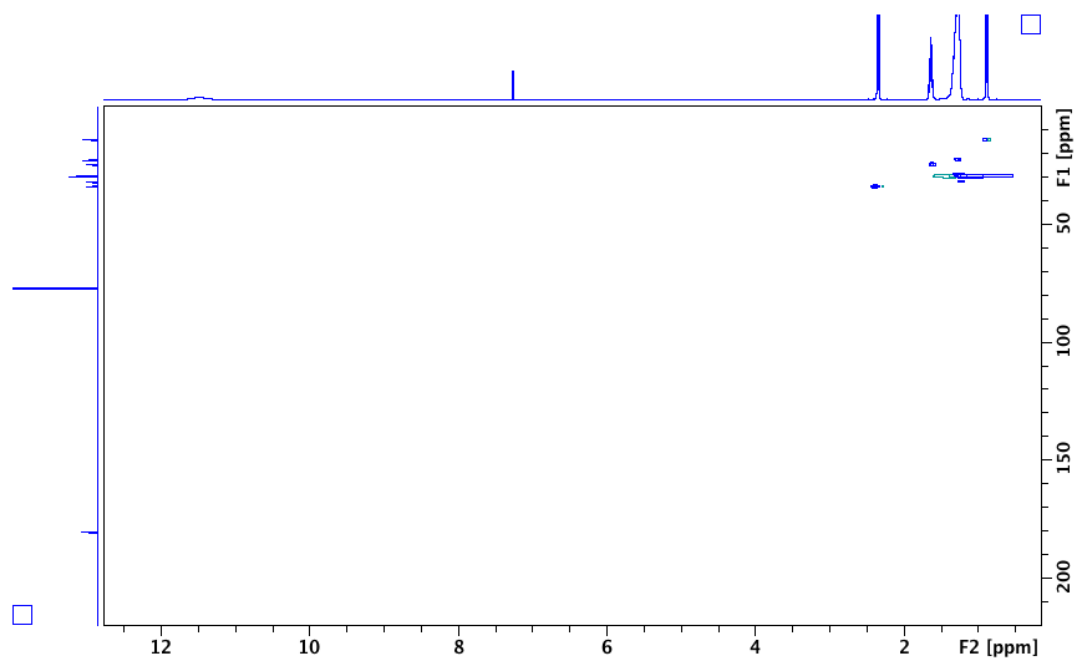
A6. ^1H NMR spectrum of compound 2 in DMSO-d6



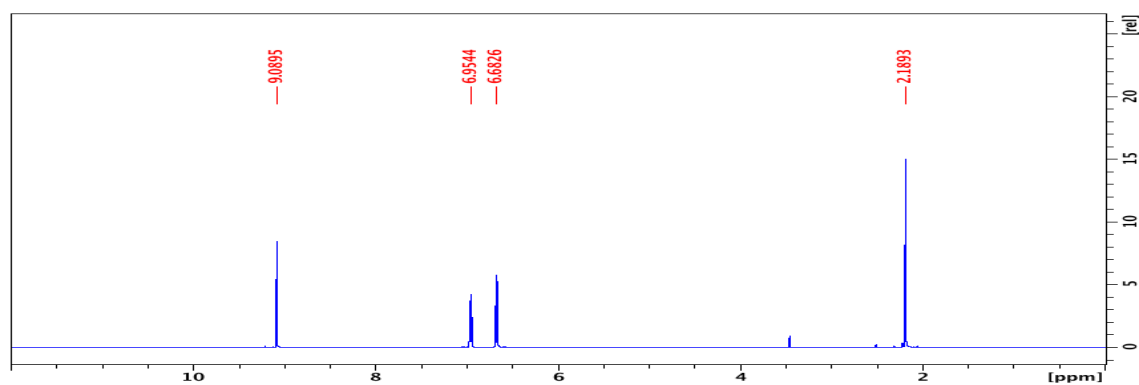
A7. ^{13}C NMR spectrum of compound 2 in DMSO-d6



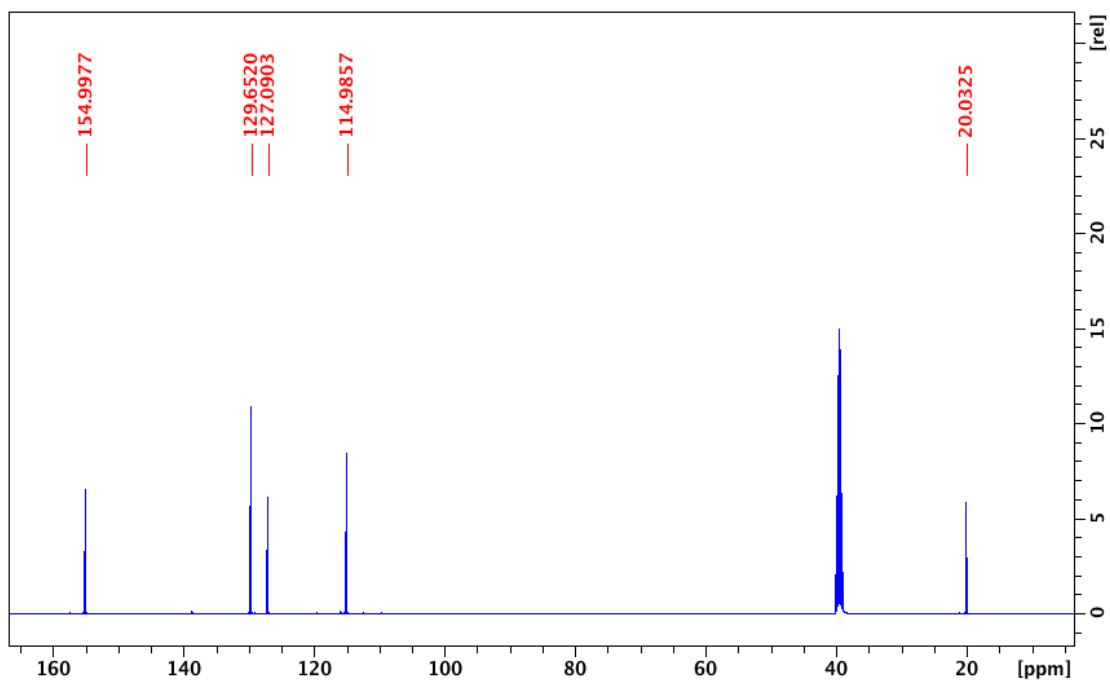
A8. 2D HSQC NMR spectrum of compound 2 in DMSO-d6



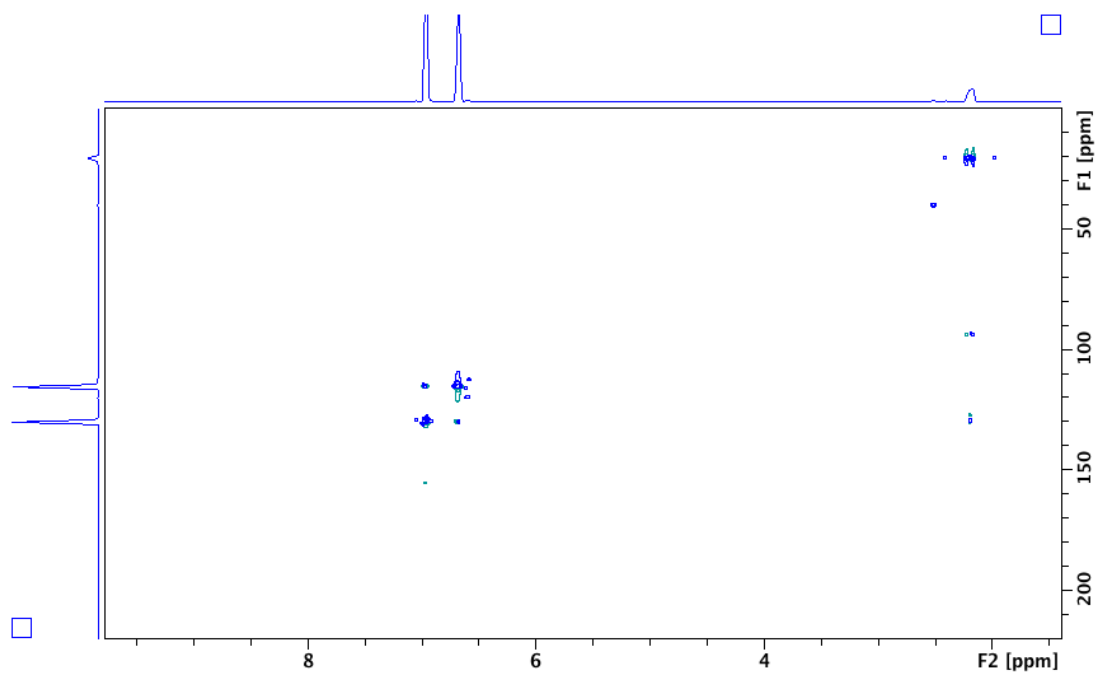
A9. ^1H NMR spectrum of compound 3 in DMSO-d6



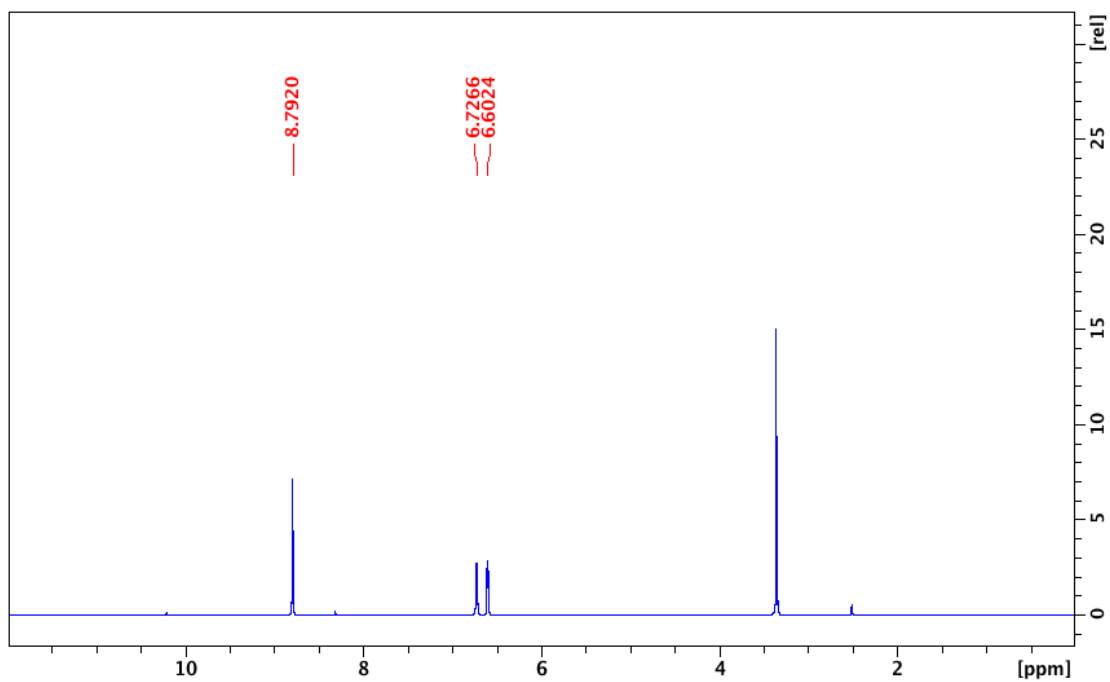
A10. ^{13}C NMR spectrum of compound 3 in DMSO-d6



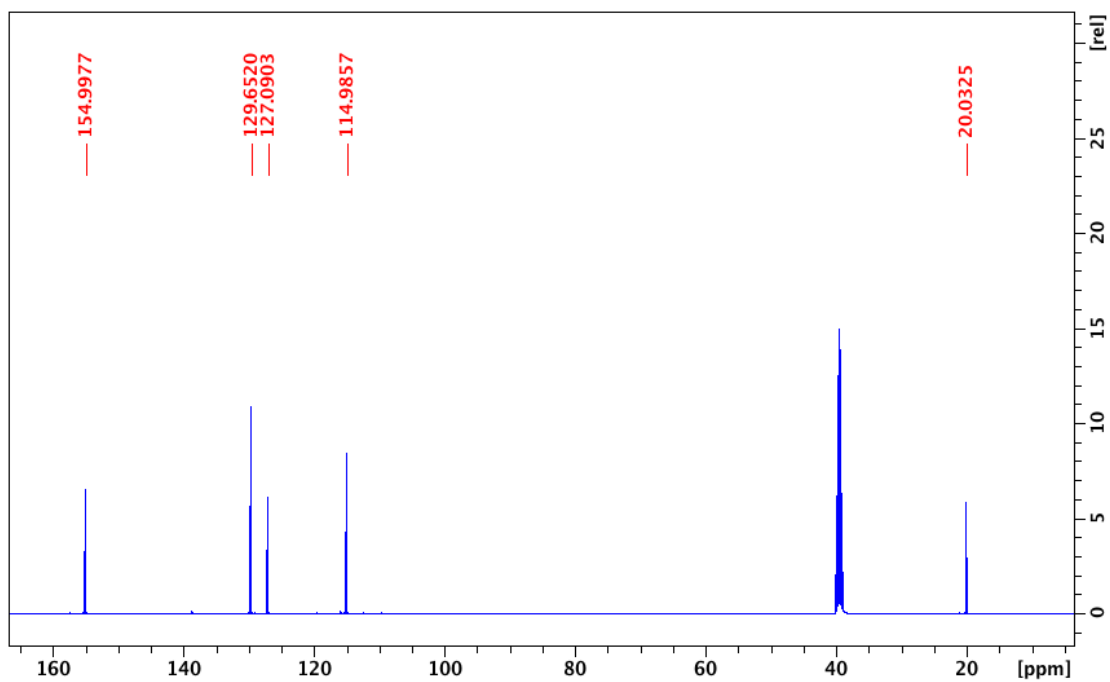
A11. 2D HSQC NMR spectrum of compound 3 in DMSO-d6



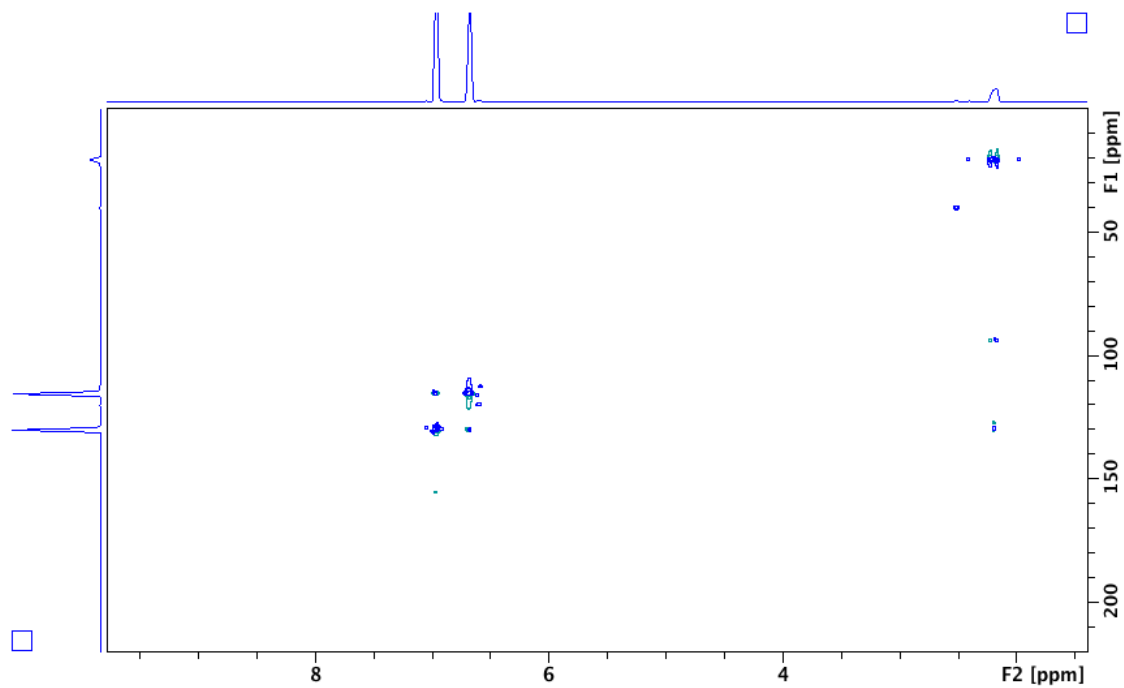
A12. ^1H NMR spectrum of compound 4 in DMSO-d6



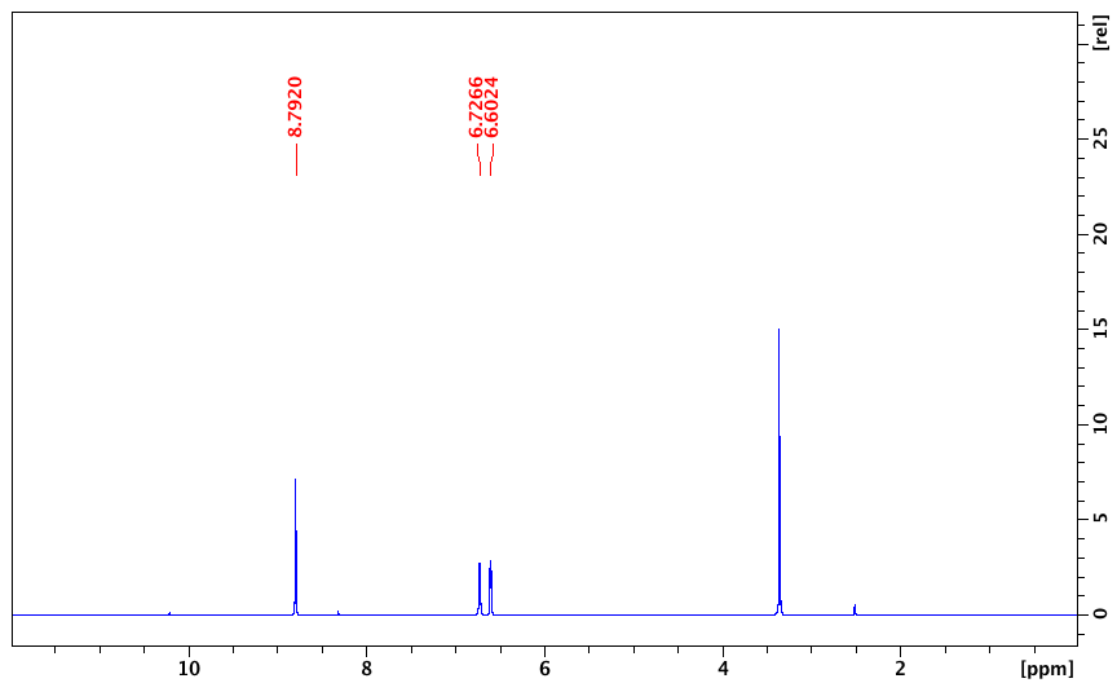
A13. ^{13}C NMR spectrum of compound 4 in DMSO-d6



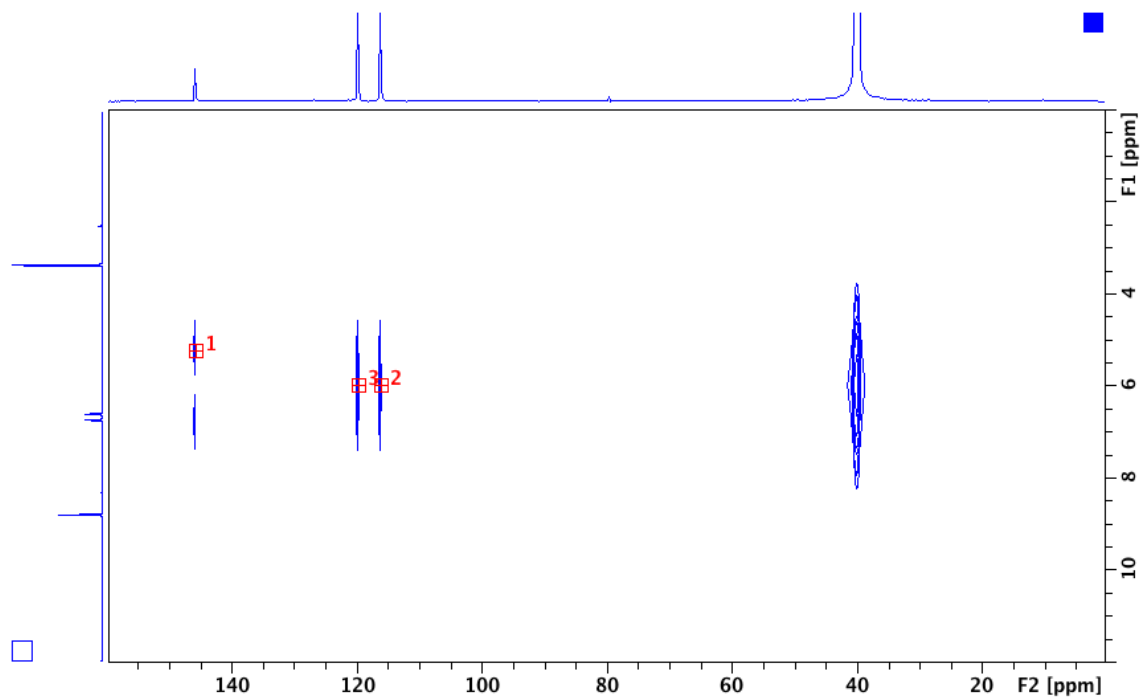
A14. 2D HSQC NMR spectrum of compound 4 in DMSO-d6



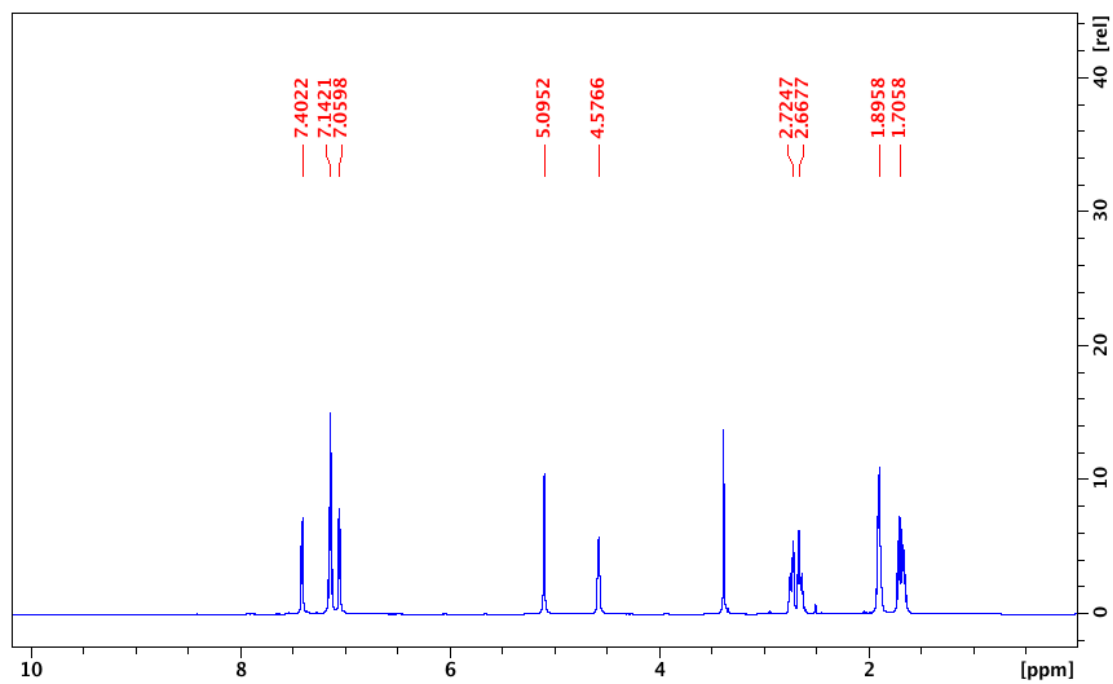
A15. ^1H NMR spectrum of compound 5 in DMSO-d6



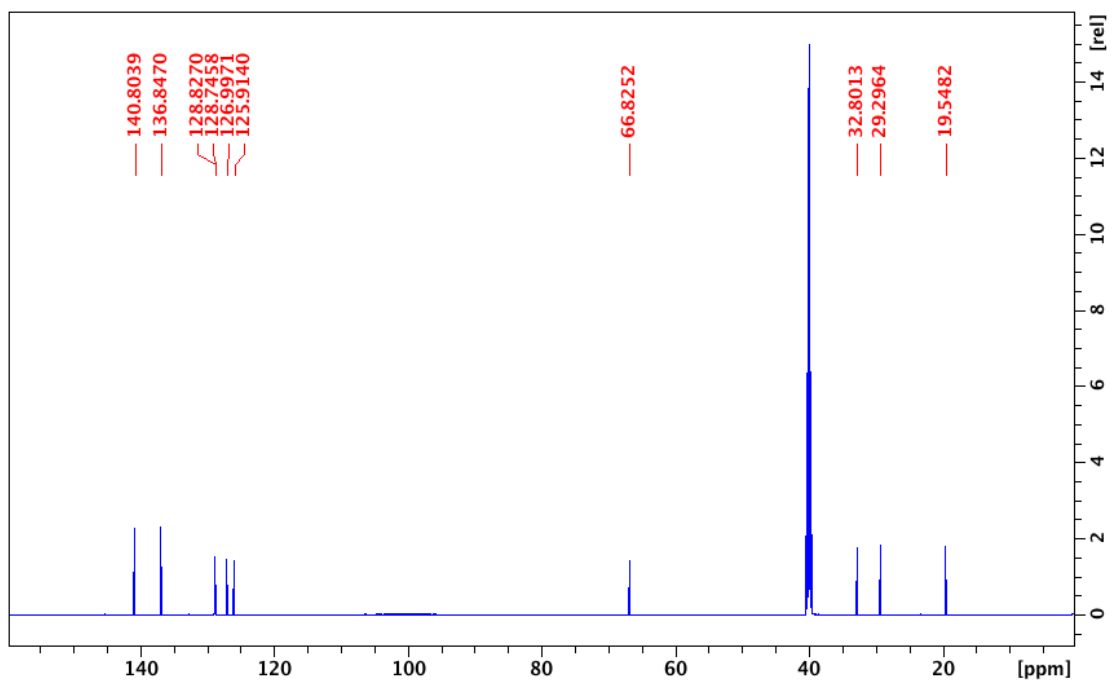
A16. 2D HSQC NMR spectrum of compound 5 in DMSO-d6



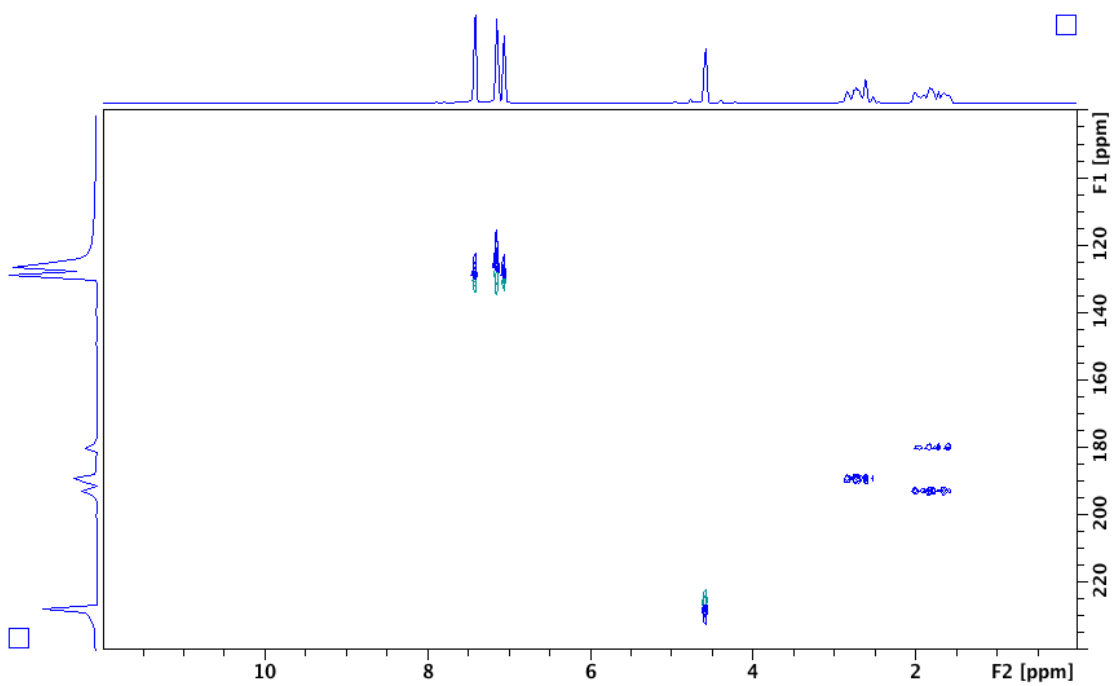
A17. ^1H NMR spectrum of compound 6 in DMSO-d6



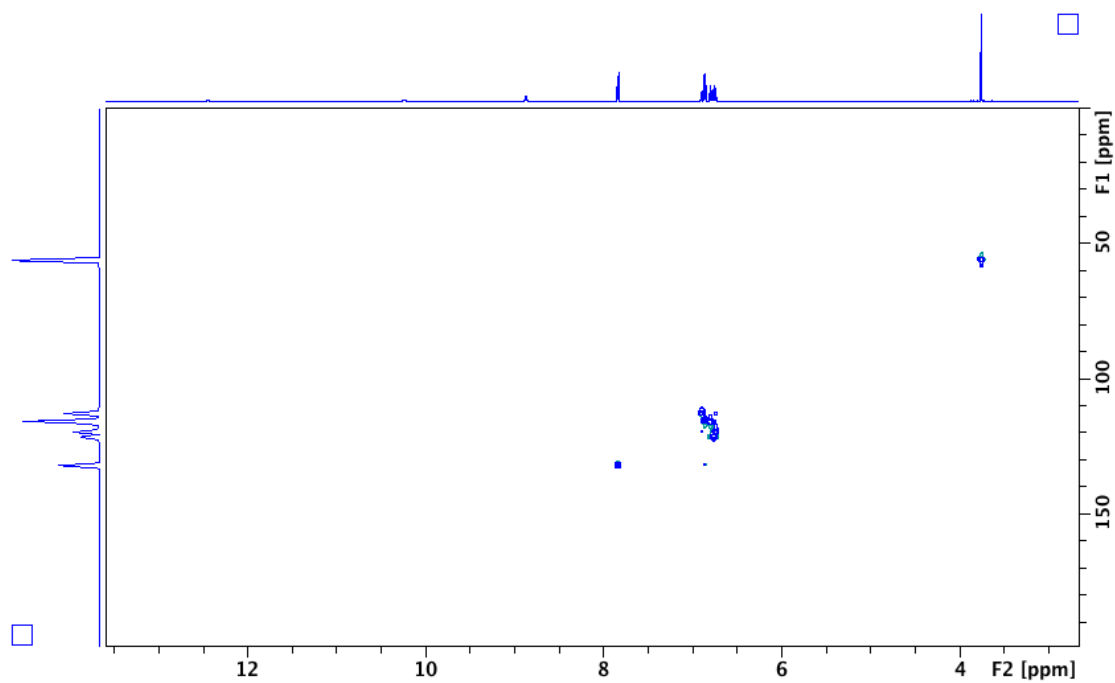
A18. ^{13}C NMR spectrum of compound 6 in DMSO-d₆



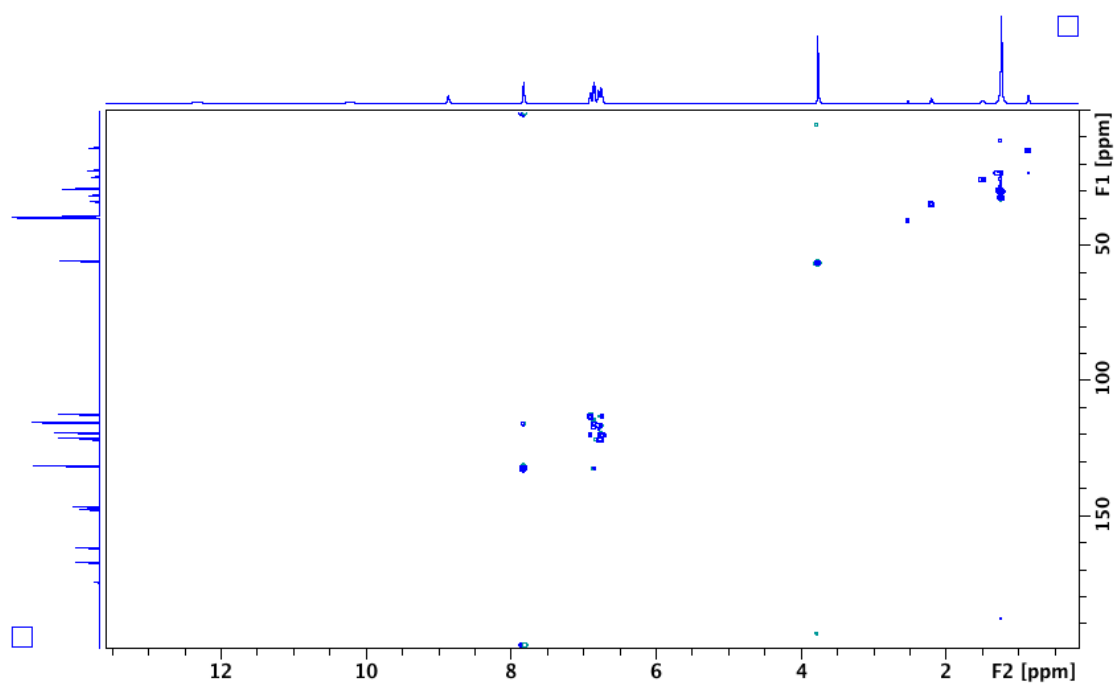
A19. 2D HSQC NMR spectrum of compound 6 in DMSO-d₆



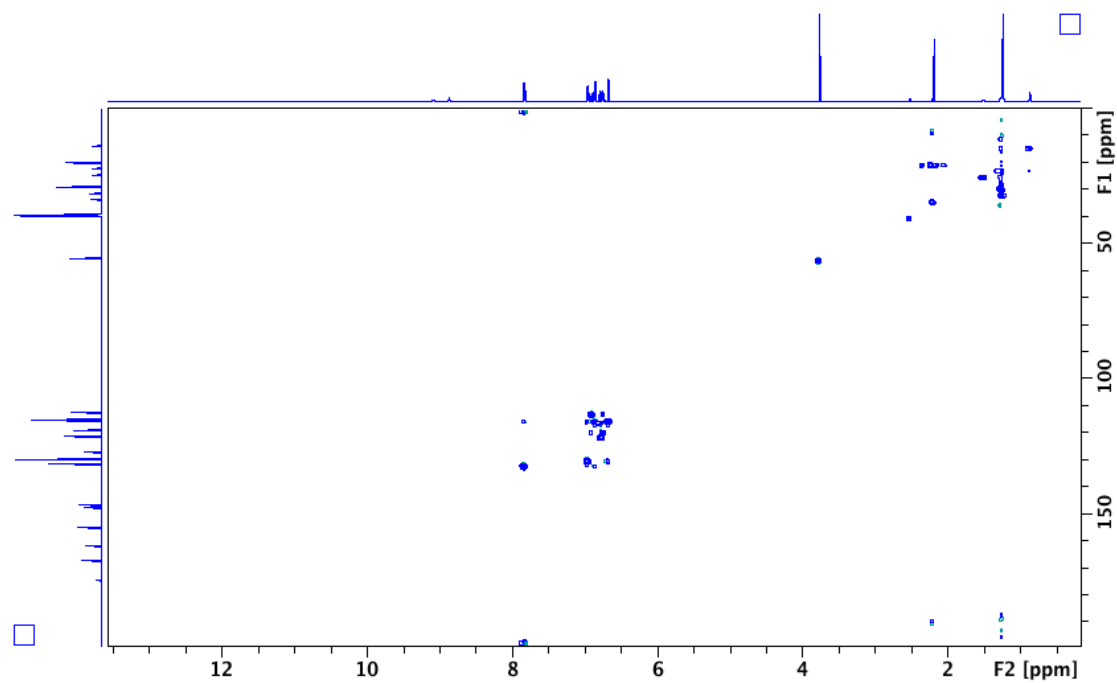
A20. 2D HSQC NMR spectrum of Mixture 1 in DMSO-d6



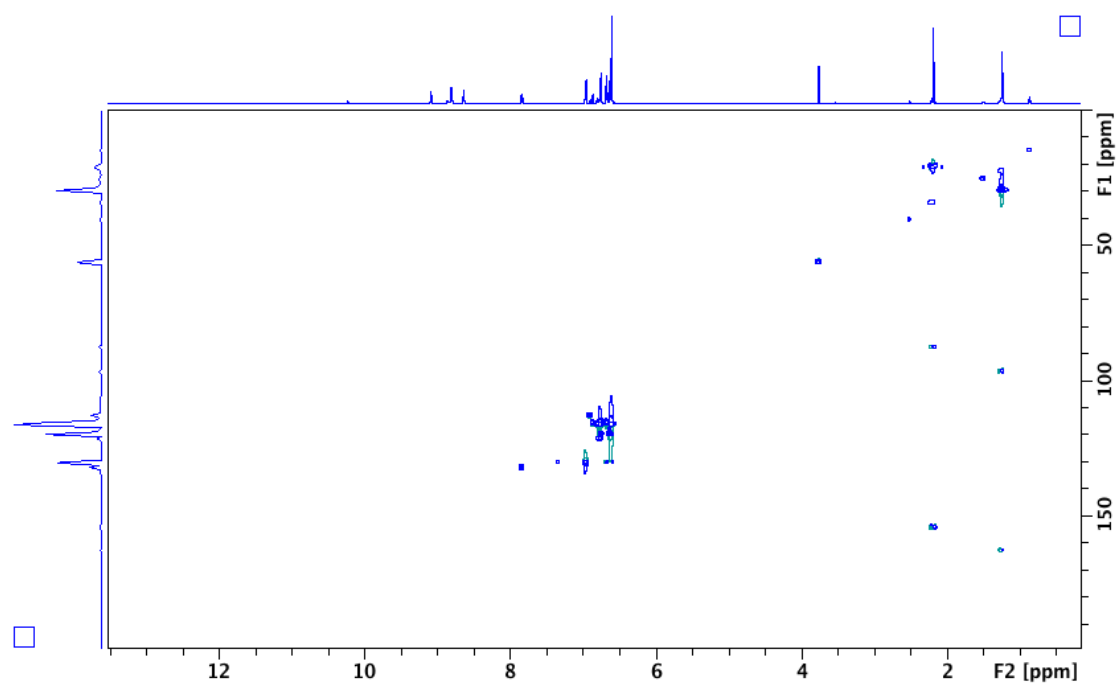
A21. 2D HSQC NMR spectrum of Mixture 2 in DMSO-d6



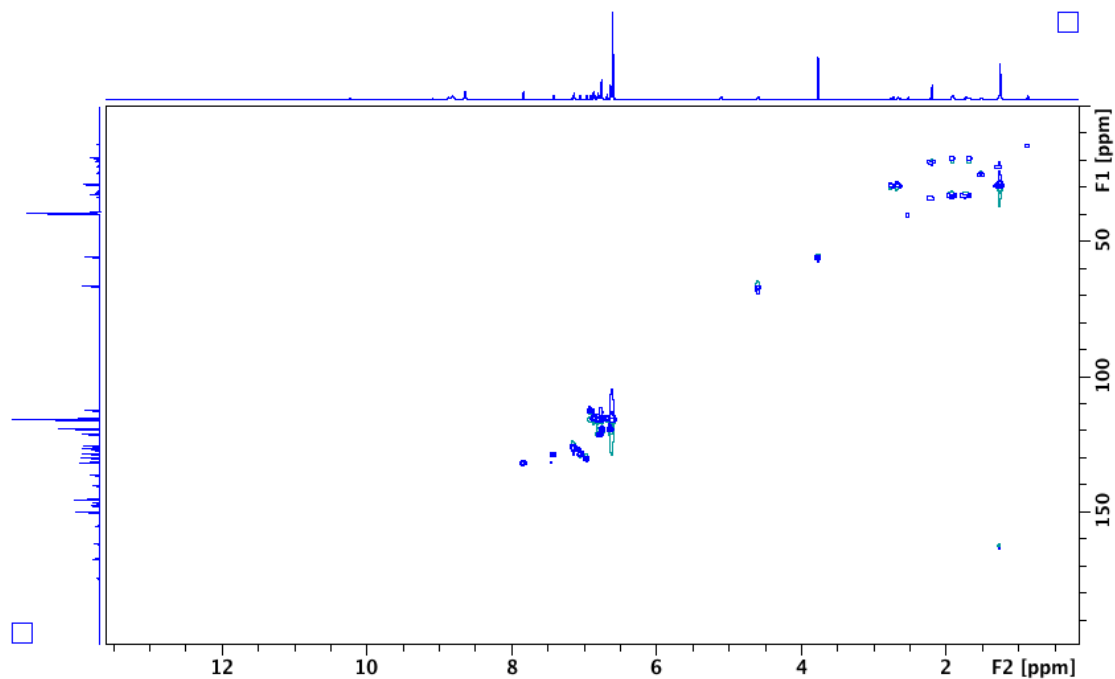
A22. 2D HSQC NMR spectrum of Mixture 3 in DMSO-d6



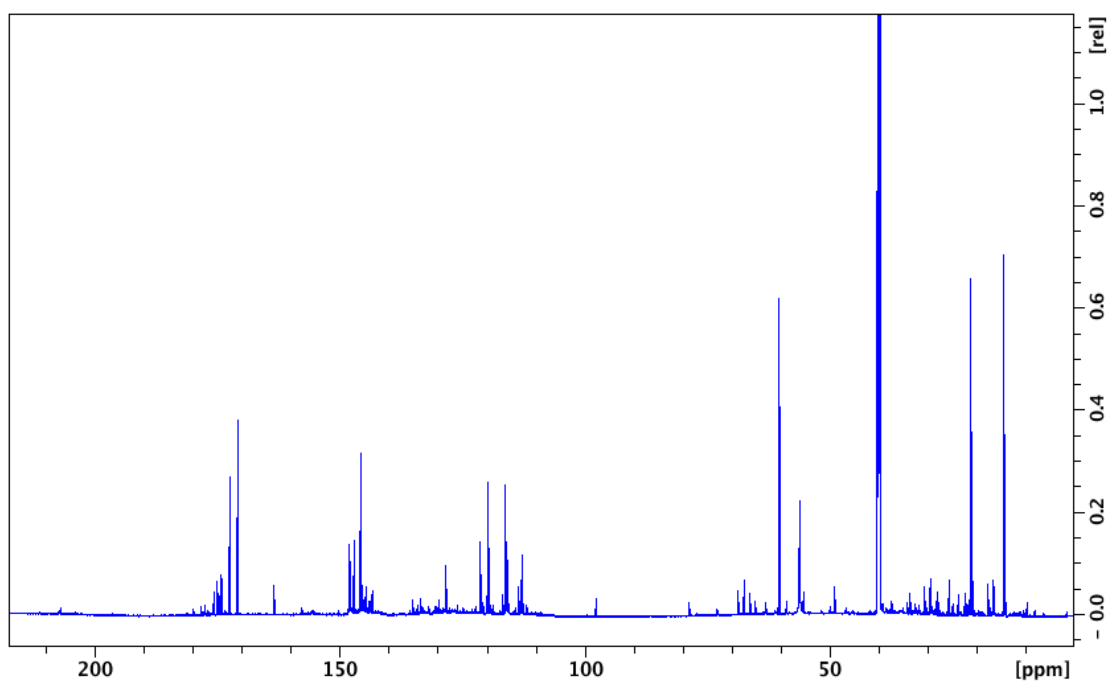
A23. 2D HSQC NMR spectrum of Mixture 4 in DMSO-d6



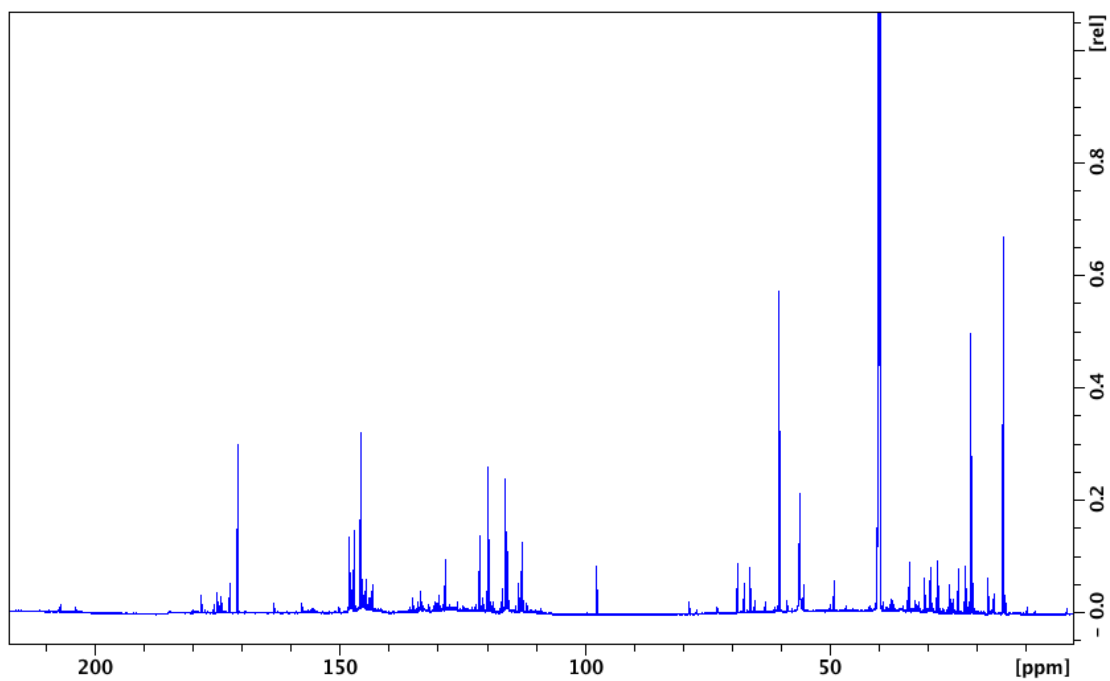
A24. 2D HSQC NMR spectrum of Mixture 5 in DMSO-d6



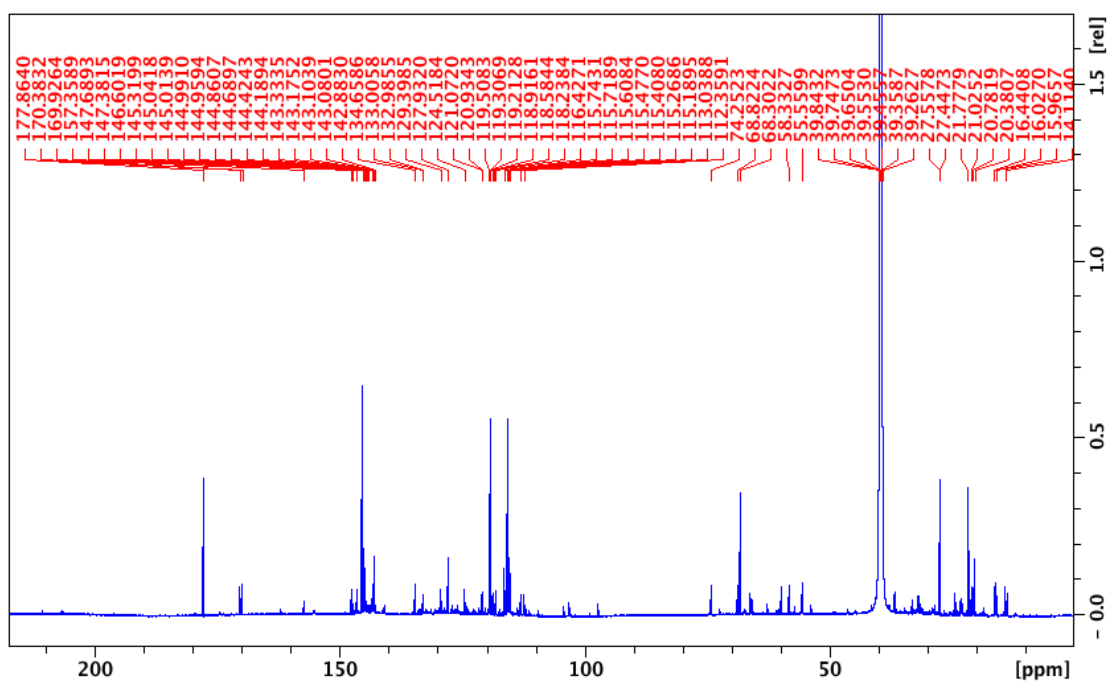
A25. Quantitative ^{13}C NMR spectrum of LtL-oil 1



A26. Quantitative ^{13}C NMR spectrum of LtL-oil 2



A27. Quantitative ^{13}C NMR spectrum of LtL-oil 3



Appendix B NMR spectra integrations

B1. Integrals of Quantitative ^{13}C NMR of guaiacol in DMSO-d6

Object	Integral [abs]	Integral [rel]	Peaks	Range (F1) from	Range (F1) to	v(F1) [ppm]
Integral 7	495338993.5	1.000000267	1	55.53826661	55.4160574	55.477162
Integral 6	488102177	0.985390437	1	112.5584483	111.8340934	112.1962708
Integral 5	455005668	0.918574543	1	115.739311	115.1409308	115.4401209
Integral 4	446036191.5	0.900466784	1	119.2980979	118.9674142	119.1327561
Integral 3	460157866	0.928975903	1	121.0302509	120.6838203	120.8570356
Integral 2	513465909.5	1.036595247	1	146.6916176	146.1853223	146.43847

B2. Integrals of Quantitative ^{13}C NMR of guaiacol with 0.01M Cr(acac)3 in DMSO-d6

Object	Integral [abs]	Integral [rel]	Peaks	Range (F1) from	Range (F1) to	v(F1) [ppm]
Integral 7	231310278.5	1.023627748	1	147.730764	147.5362239	147.6334939
Integral 6	231852651.5	1.026027936	1	146.6458288	146.4662533	146.556041
Integral 5	224416517.5	0.993120478	1	120.9837601	120.8131788	120.8984695
Integral 4	227947103	1.008744537	1	119.3195521	119.0532789	119.1864155
Integral 3	220524783.5	0.975898213	1	115.6470469	115.4999432	115.5734951
Integral 2	219654160.5	0.972045406	1	112.3796118	112.236031	112.3078214

B3. Integrals of Quantitative ^{13}C NMR of guaiacol with 0.01M Cr(acac)3 in DMSO-d6

Object	Integral [abs]	Integral [rel]	Peaks	Range (F1) from	Range (F1) to	v(F1) [ppm]
Integral 7	275862890	1.017773662	1	147.7753862	147.5496977	147.662542
Integral 6	273207266.5	1.007975955	1	146.6800188	146.5010244	146.5905216
Integral 5	269003982.5	0.992468281	1	121.1067041	120.8805622	120.9936331
Integral 4	267485257.5	0.986865069	1	119.4028956	119.1519711	119.2774334
Integral 3	269314169	0.993612689	1	115.7722347	115.505821	115.6390279
Integral 2	273812203	1.010207819	1	112.5195095	112.2654872	112.3924983

B4. Integrals of Quantitative ^{13}C NMR of LtL-oil 3 in DMSO-d6

Object	Integral [abs]	Integral [rel]	Peaks	Range (F1) from	Range (F1) to	v(F1) [ppm]	
Integral 1	1128729,556		1	0	179,6930255	179,6614959	179,6772607
Integral 2	2322358,471	2,057497705	1	179,5176422	179,4585242	179,4880832	

<i>Integral 3</i>	80138756,1	70,99907654	1	177,925398	177,8032209	177,8643095
<i>Integral 4</i>	1120485,974	0,992696584	1	174,6027505	174,533526	174,5681382
<i>Integral 5</i>	15425659,87	13,66639138	1	170,4308182	170,3523637	170,391591
<i>Integral 6</i>	18009852,45	15,95586149	1	169,9785512	169,8908668	169,934709
<i>Integral 7</i>	12537475,78	11,10759944	2	162,240605	162,1075853	162,1740951
<i>Integral 8</i>	8804552,737	7,80040949	1	157,3921725	157,2890143	157,3405934
<i>Integral 9</i>	1352595,15	1,19833413	1	155,407057	155,3724447	155,3897508
<i>Integral 10</i>	466972,2	0,413714869	0	155,3357964	155,3011841	155,3184903
<i>Integral 11</i>	2554809,92	2,263438489	1	155,2726799	155,2278876	155,2502837
<i>Integral 12</i>	2096570,542	1,85746048	1	155,0690783	155,026322	155,0477002
<i>Integral 13</i>	772040,3333	0,683990536	1	148,885219	148,8590951	148,8721571
<i>Integral 14</i>	3487343,813	3,089618585	1	147,8402638	147,7854036	147,8128337
<i>Integral 15</i>	15765109,42	13,96712733	1	147,7436054	147,6391099	147,6913577
<i>Integral 16</i>	12918344,3	11,4450306	2	147,4562427	147,356972	147,4066074
<i>Integral 17</i>	3606734,788	3,195393237	1	147,2812127	147,2237402	147,2524765
<i>Integral 18</i>	1754143,261	1,554086408	1	147,0591598	147,0199739	147,0395668
<i>Integral 19</i>	2881222,032	2,552623893	1	146,8162077	146,7613475	146,7887776
<i>Integral 20</i>	15746137,06	13,95031873	1	146,6516272	146,5680308	146,609829
<i>Integral 21</i>	3521897,88	3,120231824	1	146,3982256	146,3094044	146,353815
<i>Integral 22</i>	1597649,421	1,415440407	1	145,5441092	145,5121746	145,5281419
<i>Integral 23</i>	152073877,3	134,7301278	1	145,3962019	145,2499754	145,3230886
<i>Integral 24</i>	2695860,6	2,388402595	1	145,2062755	145,1709795	145,1886275
<i>Integral 25</i>	2726252,32	2,415328195	1	145,117195	145,0734951	145,0953451
<i>Integral 26</i>	65400345,74	57,9415551	3	145,0701336	144,9726493	145,0213914
<i>Integral 27</i>	9456110,824	8,37765856	1	144,9726493	144,9138225	144,9432359
<i>Integral 28</i>	16206053,17	14,35778225	2	144,8953341	144,8129767	144,8541554
<i>Integral 29</i>	10489711,86	9,293379279	1	144,7171731	144,6398579	144,6785155
<i>Integral 30</i>	3249557,371	2,878951256	1	144,6297733	144,5692658	144,5995196
<i>Integral 31</i>	7687079,636	6,810382167	2	144,4482508	144,389424	144,4188374
<i>Integral 32</i>	3631706,621	3,217517077	1	144,268409	144,2179861	144,2431975
<i>Integral 33</i>	3784467,636	3,352855977	1	144,2095823	144,1726054	144,1910938
<i>Integral 34</i>	4492900	3,980492916	1	143,7053529	143,6381224	143,6717377
<i>Integral 35</i>	21515262,92	19,06148626	1	143,3944115	143,305331	143,3498713
<i>Integral 36</i>	1009617,462	0,894472424	1	143,2952464	143,2733965	143,2843215
<i>Integral 37</i>	6578073,429	5,827856103	1	143,1994429	143,1372546	143,1683487
<i>Integral 38</i>	40352215,62	35,75011873	2	143,1372546	143,0431318	143,0901932
<i>Integral 39</i>	687230	0,608852667	0	142,9674974	142,9439667	142,955732
<i>Integral 40</i>	39778809,21	35,24210828	1	142,9389244	142,8212709	142,8800976
<i>Integral 41</i>	1501029	1,329839369	1	142,7893364	142,7590826	142,7742095
<i>Integral 42</i>	5936649,162	5,259585109	1	142,7540403	142,6884905	142,7212654
<i>Integral 43</i>	1030385,824	0,912872192	1	141,836343	141,8060892	141,8212161
<i>Integral 44</i>	1666768,692	1,476676751	1	141,3421983	141,2968176	141,3195079
<i>Integral 45</i>	3505888,231	3,106048046	1	141,1102528	141,0413414	141,0757971
<i>Integral 46</i>	5126806,2	4,542103265	1	140,8799881	140,8262036	140,8530958

<i>Integral 47</i>	482983,2308	0,427899871	1	135,1284124	135,1065625	135,1174874
<i>Integral 48</i>	24089509,11	21,34214435	1	134,7082213	134,6124177	134,6603195
<i>Integral 49</i>	2398878,364	2,125290644	1	134,1535691	134,1149115	134,1342403
<i>Integral 50</i>	3832182,677	3,395129204	1	133,9031352	133,8493507	133,876243
<i>Integral 51</i>	2824795,125	2,502632372	1	133,6577436	133,6157245	133,636734
<i>Integral 52</i>	18769993,89	16,62931018	2	133,0257762	132,9635879	132,994682
<i>Integral 53</i>	3150875,217	2,791523622	1	132,5753313	132,5366738	132,5560026
<i>Integral 54</i>	2021164,923	1,790654735	1	132,2307746	132,1837132	132,2072439
<i>Integral 55</i>	3134765	2,777250746	1	131,4240078	131,3752656	131,3996367
<i>Integral 56</i>	1488842,25	1,319042496	1	130,6794291	130,6441331	130,6617811
<i>Integral 57</i>	1449480,455	1,284169842	1	130,3752108	130,3365532	130,355882
<i>Integral 58</i>	1273884,063	1,128599899	0	130,2289843	130,2020921	130,2155382
<i>Integral 59</i>	1294089,889	1,146501288	1	129,8726622	129,8407277	129,856695
<i>Integral 60</i>	3575930,189	3,168101847	1	129,7550087	129,6894589	129,7222338
<i>Integral 61</i>	13668081,36	12,10926151	1	129,4306212	129,3533061	129,3919636
<i>Integral 62</i>	1549208,3	1,372523907	1	129,1583374	129,0877453	129,1230413
<i>Integral 63</i>	621078,3871	0,550245525	0	129,0742992	129,0205147	129,0474069
<i>Integral 64</i>	3750338,919	3,322619577	1	128,5650276	128,4994778	128,5322527
<i>Integral 65</i>	1237403,913	1,096280244	1	128,2137478	128,1734095	128,1935786
<i>Integral 66</i>	42811758,95	37,92915561	1	127,9885254	127,8742334	127,9313794
<i>Integral 67</i>	7217101,808	6,394004456	2	127,1935239	127,1010818	127,1473028
<i>Integral 68</i>	1857774,88	1,645899029	1	127,0708281	127,0271282	127,0489781
<i>Integral 69</i>	3539957,529	3,136231803	1	126,6724869	126,6119794	126,6422331
<i>Integral 70</i>	2783385,273	2,465945238	1	126,5245796	126,4859221	126,5052508
<i>Integral 71</i>	14607157,08	12,94123734	1	124,5480007	124,4757279	124,5118643
<i>Integral 72</i>	3367921,143	2,983815854	1	124,2269748	124,2034441	124,2152094
<i>Integral 73</i>	4797696,25	4,250527707	1	124,2034441	124,1614249	124,1824345
<i>Integral 74</i>	1082619	0,95914827	0	124,1614249	124,1278097	124,1446173
<i>Integral 75</i>	3559735,55	3,153754177	1	124,1244481	124,053856	124,0891521
<i>Integral 76</i>	5034735,396	4,460532969	1	123,8168683	123,7311493	123,7740088
<i>Integral 77</i>	1130320,188	1,001409223	0	122,8823633	122,855471	122,8689172
<i>Integral 78</i>	3248874,923	2,878346639	1	122,6924369	122,6470563	122,6697466
<i>Integral 79</i>	3347363,927	2,965603152	0	122,4117492	122,3125841	122,3621667
<i>Integral 80</i>	590121,4286	0,522819152	0	122,2218229	122,1982922	122,2100575
<i>Integral 81</i>	1179417,7	1,044907254	0	121,7091897	121,6738937	121,6915417
<i>Integral 82</i>	4119230,406	3,649439661	1	121,5579209	121,500775	121,529348
<i>Integral 83</i>	15722145,14	13,92906304	1	121,1192415	121,0150341	121,0671378
<i>Integral 84</i>	3808537,714	3,374180906	1	121,0133533	120,9763765	120,9948649
<i>Integral 85</i>	13206575,5	11,70038955	1	120,9595688	120,8973806	120,9284747
<i>Integral 86</i>	4687129,75	4,152571116	1	120,4082781	120,3292822	120,3687802
<i>Integral 87</i>	3492610,296	3,094284436	0	120,2906246	120,2418825	120,2662535
<i>Integral 88</i>	2443994,077	2,165260992	0	119,7208455	119,6754649	119,6981552
<i>Integral 89</i>	33792507,69	29,93853357	1	119,5561306	119,4586463	119,5073884
<i>Integral 90</i>	2206032	1,954438057	0	119,4569655	119,4048618	119,4309136

<i>Integral 91</i>	148065180,2	131,1786153	1	119,3813311	119,2384661	119,3098986
<i>Integral 92</i>	11351425,6	10,05681613	1	119,2384661	119,1846817	119,2115739
<i>Integral 93</i>	2457806,765	2,177498368	0	119,104005	119,0451782	119,0745916
<i>Integral 94</i>	15058542,19	13,34114281	1	118,9846707	118,8737403	118,9292055
<i>Integral 95</i>	9102342,118	8,064236533	2	118,875421	118,8149135	118,8451673
<i>Integral 96</i>	2336040	2,069618881	0	118,8031482	118,7476829	118,7754155
<i>Integral 97</i>	3007927,056	2,664878439	0	118,7426406	118,6787716	118,7107061
<i>Integral 98</i>	964883	0,854839846	0	118,6552409	118,6266679	118,6409544
<i>Integral 99</i>	14760241,04	13,07686236	1	118,6233064	118,5426297	118,582968
<i>Integral 100</i>	488177,3684	0,432501626	0	118,5392682	118,5056529	118,5224605
<i>Integral 101</i>	4289495,629	3,800286444	0	118,4989298	118,4367415	118,4678357
<i>Integral 102</i>	1420858,667	1,258812317	0	118,4249762	118,3728725	118,3989243
<i>Integral 103</i>	1344036,235	1,190751344	0	118,3442995	118,283792	118,3140457
<i>Integral 104</i>	16204469,3	14,35637901	1	118,2770689	118,1963922	118,2367306
<i>Integral 105</i>	349716,8788	0,309832304	0	118,1678192	118,1106732	118,1392462
<i>Integral 106</i>	2600971,854	2,304335738	0	118,1089925	118,0350388	118,0720157
<i>Integral 107</i>	1641227	1,454048042	0	117,4854289	117,4551751	117,470302
<i>Integral 108</i>	4167233,867	3,691968414	0	117,2400373	117,1879336	117,2139855
<i>Integral 109</i>	2966245,733	2,627950795	0	116,8551423	116,8030386	116,8290904
<i>Integral 110</i>	38281243,13	33,91533689	1	116,4937779	116,3542745	116,4240262
<i>Integral 111</i>	1397511,475	1,23812783	0	116,3357861	116,1912403	116,2635132
<i>Integral 112</i>	9241488,397	8,187513432	0	116,1072021	115,9761025	116,0416523
<i>Integral 113</i>	179116202,8	158,6883252	2	115,8181107	115,6533957	115,7357532
<i>Integral 114</i>	16038440,14	14,20928517	1	115,6533957	115,5760806	115,6147382
<i>Integral 115</i>	100717161	89,2305517	2	115,5760806	115,3643043	115,4701924
<i>Integral 116</i>	6698437,765	5,934493105	1	115,3609427	115,3004352	115,330689
<i>Integral 117</i>	16339022,58	14,47558674	1	115,3105198	115,2466508	115,2785853
<i>Integral 118</i>	18859971,36	16,70902588	1	115,2332047	115,1390819	115,1861433
<i>Integral 119</i>	15289053,6	13,54536481	1	115,1189127	114,9609208	115,0399168
<i>Integral 120</i>	1395563,862	1,236402339	0	114,6045988	114,5541759	114,5793873
<i>Integral 121</i>	6370681,547	5,644116889	0	113,754132	113,6600092	113,7070706
<i>Integral 122</i>	6001656,915	5,317178845	0	113,2028413	113,1188031	113,1608222
<i>Integral 123</i>	22797299,64	20,19730902	1	113,0885493	112,9440036	113,0162765
<i>Integral 124</i>	3733411,667	3,307622848	0	112,5372586	112,4733896	112,5053241
<i>Integral 125</i>	2237196,96	1,98204871	0	112,4868357	112,4431358	112,4649858
<i>Integral 126</i>	19091592,21	16,9142308	1	112,4296897	112,3019516	112,3658207
<i>Integral 127</i>	4086696,04	3,62061578	0	112,0700061	112,0263063	112,0481562
<i>Integral 128</i>	2882824,514	2,554043614	0	112,0094986	111,9456296	111,9775641
<i>Integral 129</i>	3390829,375	3,004111444	0	111,3909773	111,3338313	111,3624043
<i>Integral 130</i>	2409975,277	2,135121974	0	109,5522211	109,4681829	109,510202
<i>Integral 131</i>	1719497,067	1,52339155	0	109,1353915	109,0816071	109,1084993
<i>Integral 132</i>	5555557,061	4,921955869	0	104,2981517	104,2410057	104,2695787
<i>Integral 133</i>	10411495,73	9,224083556	1	103,2628009	103,1350628	103,1989318
<i>Integral 134</i>	3315676,4	2,937529529	0	103,1417858	103,0711937	103,1064898

<i>Integral 135</i>	1961432,156	1,737734382	0	102,9165634	102,8594174	102,8879904
<i>Integral 136</i>	6921887,875	6,132459136	0	97,2624721	97,20532611	97,2338991
<i>Integral 137</i>	20117713,23	17,82332457	1	74,33012314	74,12843142	74,22927728
<i>Integral 138</i>	8737009,892	7,740569784	0	72,51489763	72,38043648	72,44766706
<i>Integral 139</i>	19156482,51	16,97172048	1	68,91133884	68,74998546	68,83066215
<i>Integral 140</i>	85582728,84	75,82217407	1	68,39366342	68,21214086	68,30290214
<i>Integral 141</i>	34081741,47	30,19478076	0	66,2389235	66,03050872	66,13471611
<i>Integral 142</i>	13425716,57	11,89453798	0	65,83890159	65,70444044	65,77167101
<i>Integral 143</i>	756338,5263	0,670079491	0	63,68080015	63,64718486	63,6639925
<i>Integral 144</i>	1644072,079	1,456568645	0	63,56650817	63,45221619	63,50936218
<i>Integral 145</i>	6821725,381	6,043719992	0	62,84041796	62,76646433	62,80344115
<i>Integral 146</i>	8899080,93	7,884156915	0	60,87728519	60,72265487	60,79997003
<i>Integral 147</i>	2306913,368	2,043814089	0	60,27893308	60,2117025	60,24531779
<i>Integral 148</i>	18510099,28	16,39905608	1	59,87554963	59,72091931	59,79823447
<i>Integral 149</i>	19948377,31	17,67330111	1	58,41664616	58,28218501	58,34941559
<i>Integral 150</i>	1612247,947	1,428373998	0	58,07377023	58,00653966	58,04015495
<i>Integral 151</i>	1497124,022	1,326379747	0	57,91913991	57,83846322	57,87880157
<i>Integral 152</i>	6783105,538	6,009504673	0	57,19304971	57,0989269	57,14598831
<i>Integral 153</i>	15889613,19	14,07743166	0	55,7879307	55,61313121	55,70053095
<i>Integral 154</i>	40418317,94	35,8086822	1	55,64002344	55,35093197	55,4954777
<i>Integral 155</i>	4860385,652	4,306067497	0	53,88530544	53,80462875	53,8449671
<i>Integral 156</i>	1427753,538	1,264920841	0	48,69285584	48,64638764	48,66962174
<i>Integral 157</i>	2172897,074	1,925082109	0	46,45076561	46,32878661	46,38977611
<i>Integral 158</i>	7158900,471	6,342440876	0	46,31136104	46,18938204	46,25037154
<i>Integral 159</i>	3191510,909	2,827524887	0	44,76629368	44,70820844	44,73725106
<i>Integral 160</i>	3187797,917	2,824235355	0	37,00051078	36,95886879	36,97968978
<i>Integral 161</i>	10811722,77	9,578665426	1	36,76347175	36,71862653	36,74104914
<i>Integral 162</i>	2532371	2,243558687	0	36,47197781	36,42392936	36,44795359
<i>Integral 163</i>	3833077,732	3,395922179	0	34,67496573	34,60129144	34,63812859
<i>Integral 164</i>	6227965	5,517676904	0	33,24312188	33,16624436	33,20468312
<i>Integral 165</i>	3624679,548	3,21129143	0	32,95162794	32,89717303	32,92440049
<i>Integral 166</i>	2294711,452	2,033003779	0	32,82349874	32,76904383	32,79627128
<i>Integral 167</i>	12131382,07	10,74781998	0	32,17324303	32,0739429	32,12359297
<i>Integral 168</i>	3062677,606	2,713384788	0	32,0419106	31,98425246	32,01308153
<i>Integral 169</i>	28653882,17	25,38595896	1	31,95542339	31,83049741	31,8929604
<i>Integral 170</i>	6735899,385	5,967682295	0	31,81768449	31,77283927	31,79526188
<i>Integral 171</i>	3624439,667	3,211078906	0	31,49736148	31,44931303	31,47333726
<i>Integral 172</i>	13033935,45	11,54743879	0	31,35641936	31,26993215	31,31317575
<i>Integral 173</i>	2584090,259	2,289379459	0	31,1482094	31,10016095	31,12418518
<i>Integral 174</i>	7363326,938	6,523552876	0	31,07773834	30,96242206	31,0200802
<i>Integral 175</i>	1543080,094	1,367094612	0	30,7205193	30,66282583	30,69167257
<i>Integral 176</i>	1267190,789	1,122669982	0	30,26649677	30,19876965	30,23263321
<i>Integral 177</i>	1810576,267	1,604083332	0	29,57919193	29,52651529	29,55285361
<i>Integral 178</i>	1263421,133	1,119330248	0	28,9972404	28,97215628	28,98469834

<i>Integral 179</i>	7833544,147	6,940142666	0	28,6586048	28,53569263	28,59714871
<i>Integral 180</i>	2775440,156	2,458906248	0	27,83082891	27,77564385	27,80323638
<i>Integral 181</i>	2544474,478	2,254281786	0	27,63266438	27,59252979	27,61259708
<i>Integral 182</i>	16538964	14,6527252	1	27,58751297	27,52480267	27,55615782
<i>Integral 183</i>	82967305,74	73,50503523	1	27,50724379	27,3818232	27,44453349
<i>Integral 184</i>	2038035,436	1,805601196	0	27,37931479	27,31158767	27,34545123
<i>Integral 185</i>	2816129,385	2,494954944	0	26,5114043	26,46625289	26,4888286
<i>Integral 186</i>	670702,1818	0,594209816	0	25,47543023	25,43780405	25,45661714
<i>Integral 187</i>	1114930,741	0,987774915	0	25,19448811	25,14682828	25,1706582
<i>Integral 188</i>	875962	0,776060125	0	25,00635722	24,97625628	24,99130675
<i>Integral 189</i>	2592109,348	2,296483985	0	24,83829363	24,79815904	24,81822634
<i>Integral 190</i>	4503588,813	3,989962689	0	24,45450663	24,39681316	24,42565989
<i>Integral 191</i>	14819425,84	13,12929724	1	24,39932157	24,31152716	24,35542436
<i>Integral 192</i>	1485291,75	1,315896924	0	24,27390098	24,23878321	24,2563421
<i>Integral 193</i>	3512688,333	3,112072609	0	23,96787474	23,92523174	23,94655324
<i>Integral 194</i>	6111314,917	5,414330551	0	23,35833067	23,31568767	23,33700917
<i>Integral 195</i>	6298220,63	5,579920007	0	23,21284279	23,16518296	23,18901288
<i>Integral 196</i>	7148282,435	6,333033807	0	23,1300652	23,08993061	23,1099979
<i>Integral 197</i>	4417755,75	3,913918731	0	22,99711937	22,95447637	22,97579787
<i>Integral 198</i>	1988857	1,762031472	0	22,83908943	22,8114969	22,82529316
<i>Integral 199</i>	1286381,96	1,139672434	0	22,75380343	22,70865202	22,73122772
<i>Integral 200</i>	3377484,424	2,992288461	0	22,15930983	22,10161636	22,1304631
<i>Integral 201</i>	86842837,74	76,93856984	1	21,85328359	21,70277888	21,77803124
<i>Integral 202</i>	1779277,963	1,576354543	0	21,18102923	21,13336941	21,15719932
<i>Integral 203</i>	2270631,2	2,011669836	0	21,11581052	21,06313388	21,0894722
<i>Integral 204</i>	16064130,37	14,23204548	1	21,05811705	20,99540676	21,02676191
<i>Integral 205</i>	14595523,38	12,93093045	1	20,8122927	20,74707399	20,77968334
<i>Integral 206</i>	3780500,667	3,349341433	0	20,62917863	20,59406087	20,61161975
<i>Integral 207</i>	35427751,48	31,38728077	1	20,43352251	20,35827016	20,39589634
<i>Integral 208</i>	2005230,12	1,776537267	0	20,15007198	20,10742898	20,12875048
<i>Integral 209</i>	2938817,75	2,603650924	0	18,64753331	18,58983984	18,61868658
<i>Integral 210</i>	18863524,63	16,71217391	1	16,47023187	16,37491222	16,42257205
<i>Integral 211</i>	9602565,389	8,507410249	0	16,14413834	16,08142804	16,11278319
<i>Integral 212</i>	46107312,75	40,84885748	2	16,06888598	15,94848222	16,0086841
<i>Integral 213</i>	2662488,636	2,358836644	0	15,93844857	15,90082239	15,91963548
<i>Integral 214</i>	2547748,353	2,257182281	0	15,87573828	15,81553639	15,84563733
<i>Integral 215</i>	15209155	13,4745785	1	14,14242572	14,0746986	14,10856216
<i>Integral 216</i>	2964573,815	2,626469556	0	14,01449672	13,9668369	13,99066681
<i>Integral 217</i>	15565880,28	13,79061991	1	13,71097889	13,64074336	13,67586113

B5. Integrals of Quantitative ¹³C NMR of LtL-oil 4 in DMSO-d6

<i>Object</i>	<i>Integral [abs]</i>	<i>Integral [rel]</i>	<i>Peak</i>	<i>Range (F1) from</i>	<i>Range (F1) to</i>	<i>ν(F1) [ppm]</i>
<i>Integral 1</i>	833745,0909	0,207403118	0	179,5897375	179,5509832	179,5703604
<i>Integral 2</i>	4019925,538	1	1	179,4917119	179,4461186	179,4689153
<i>Integral 3</i>	124378017,9	30,94037855	6	177,8480734	177,7432088	177,7956411
<i>Integral 4</i>	1510538,087	0,375762703	0	175,2378568	175,1968228	175,2173398
<i>Integral 5</i>	1086316,571	0,270233008	0	174,6269066	174,5904319	174,6086692
<i>Integral 6</i>	11153136,94	2,774463564	1	174,5813133	174,4354147	174,508364
<i>Integral 7</i>	2033367,871	0,505822272	1	174,3305501	174,2758381	174,3031941
<i>Integral 8</i>	4662618,943	1,159876943	1	172,078241	171,982495	172,030368
<i>Integral 9</i>	17844418,51	4,438992299	1	170,3912888	170,2864242	170,3388565
<i>Integral 10</i>	1415604,571	0,352146963	1	170,1861189	170,1359663	170,1610426
<i>Integral 11</i>	16556006,24	4,118485799	1	169,9125591	169,8396098	169,8760845
<i>Integral 12</i>	24127912,8	6,002079533	1	163,0879458	163,016719	163,0523324
<i>Integral 13</i>	29181094,34	7,259113152	3	162,2100335	162,0013223	162,1056779
<i>Integral 14</i>	19383844	4,821941057	1	157,4129882	157,3384485	157,3757184
<i>Integral 15</i>	4918524,833	1,223536303	1	157,3334792	157,2920682	157,3127737
<i>Integral 16</i>	774142,4615	0,192576319	1	157,2771603	157,2556266	157,2663934
<i>Integral 17</i>	2648581,953	0,658863436	1	156,2286348	156,1524387	156,1905368
<i>Integral 18</i>	3429981,714	0,853245087	2	156,1027455	156,0530524	156,0778989
<i>Integral 19</i>	3122451,467	0,776743608	1	156,0033592	155,9768562	155,9901077
<i>Integral 20</i>	7531448,632	1,873529387	1	155,4335445	155,3672869	155,4004157
<i>Integral 21</i>	6494967,442	1,61569347	2	155,3142809	155,2380847	155,2761828
<i>Integral 22</i>	7457952,098	1,855246329	1	155,095631	155,0227478	155,0591894
<i>Integral 23</i>	1004681,778	0,24992547	1	153,8662493	153,8359219	153,8510856
<i>Integral 24</i>	1213150,222	0,301784252	1	153,5872369	153,5569095	153,5720732
<i>Integral 25</i>	1336477,818	0,332463327	1	153,2869953	153,2687988	153,2778971
<i>Integral 26</i>	4070120,889	1,012486637	1	153,0838015	153,0524631	153,0681323
<i>Integral 27</i>	1237179,294	0,307761744	1	152,2902038	152,2611697	152,2756868
<i>Integral 28</i>	1351660	0,336240059	1	152,2227699	152,1918626	152,2073162
<i>Integral 29</i>	1090773,778	0,271341787	1	151,5400011	151,5090939	151,5245475
<i>Integral 30</i>	1016045,846	0,252752405	1	151,4510258	151,4294844	151,4402551
<i>Integral 31</i>	2321838,971	0,577582582	1	149,6386971	149,5766554	149,6076763
<i>Integral 32</i>	3199502	0,795910762	1	147,7213129	147,6563168	147,6888148
<i>Integral 33</i>	4415008,276	1,098281108	1	147,4613285	147,3579257	147,4096271
<i>Integral 34</i>	3877725,905	0,964626302	1	146,7847785	146,7109193	146,7478489
<i>Integral 35</i>	2731159,871	0,679405587	1	146,6547863	146,6016077	146,628197
<i>Integral 36</i>	17037156,24	4,238177069	1	145,3560784	145,2950205	145,3255494
<i>Integral 37</i>	2151931,556	0,535316273	1	145,1829692	145,1514337	145,1672014
<i>Integral 38</i>	8283974,788	2,060728416	2	145,0715888	145,0125438	145,0420663
<i>Integral 39</i>	1455642,4	0,362106807	1	144,9816793	144,9662471	144,9739632
<i>Integral 40</i>	3600085,333	0,895560204	1	144,9273311	144,8910989	144,909215
<i>Integral 41</i>	1689977,143	0,420400111	1	144,7146348	144,6777317	144,6961833

<i>Integral 42</i>	1363132	0,339093843	1	144,4663774	144,4435645	144,454971
<i>Integral 43</i>	1189940,364	0,296010549	1	144,4422226	144,4241065	144,4331646
<i>Integral 44</i>	1404900,615	0,349484238	1	144,3892163	144,3664034	144,3778099
<i>Integral 45</i>	2203520,952	0,548149694	1	143,3680059	143,3331157	143,3505608
<i>Integral 46</i>	1953829,25	0,486036179	1	143,1304839	143,1029743	143,1167291
<i>Integral 47</i>	1906032	0,474146096	1	143,1036453	143,0794905	143,0915679
<i>Integral 48</i>	2856435,111	0,710569159	1	142,9198006	142,8875942	142,9036974
<i>Integral 49</i>	4118155,636	1,0244358	1	138,7088175	138,6699014	138,6893594
<i>Integral 50</i>	2252380,6	0,56030406	1	138,3760183	138,341128	138,3585732
<i>Integral 51</i>	2660359,667	0,66179327	1	136,9320993	136,8891575	136,9106284
<i>Integral 52</i>	2592776,077	0,64498112	1	134,9461906	134,8544944	134,9003425
<i>Integral 53</i>	5132882,75	1,276860156	1	134,6683234	134,5543979	134,6113606
<i>Integral 54</i>	3794335,478	0,94388203	1	134,1431545	134,062573	134,1028637
<i>Integral 55</i>	1949290,909	0,484907218	0	133,9958849	133,9375328	133,9667088
<i>Integral 56</i>	14238623,34	3,542011712	1	133,8875167	133,6680016	133,7777591
<i>Integral 57</i>	6744064,93	1,677659167	1	133,6402149	133,4846093	133,5624121
<i>Integral 58</i>	7393912,727	1,839315842	1	133,4762733	133,3567904	133,4165318
<i>Integral 59</i>	6616431,563	1,645908985	1	133,0344645	132,9205389	132,9775017
<i>Integral 60</i>	2320849,872	0,577336533	1	132,4870661	132,403706	132,4453861
<i>Integral 61</i>	3061367,915	0,761548413	0	132,2230923	132,1397322	132,1814122
<i>Integral 62</i>	5160377,514	1,283699776	1	131,1714366	131,1065158	131,1389762
<i>Integral 63</i>	7315083,059	1,819706109	1	130,574584	130,5138517	130,5442179
<i>Integral 64</i>	18210765,52	4,530125086	2	129,7452731	129,6698812	129,7075772
<i>Integral 65</i>	35751058,81	8,893462944	2	129,4353287	129,3389946	129,3871616
<i>Integral 66</i>	2350958,4	0,584826355	1	129,190305	129,1567975	129,1735512
<i>Integral 67</i>	8183840,818	2,035819007	2	129,1484206	129,0709345	129,1096775
<i>Integral 68</i>	7915630,737	1,969098846	1	128,5369085	128,4698935	128,503401
<i>Integral 69</i>	6170836,706	1,53506244	1	128,2248698	128,1641374	128,1945036
<i>Integral 70</i>	4648948,457	1,156476261	1	127,9442444	127,8814178	127,9128311
<i>Integral 71</i>	9618941,917	2,392815943	2	127,1610063	127,0751433	127,1180748
<i>Integral 72</i>	4263599,833	1,060616619	1	127,0269763	126,9850919	127,0060341
<i>Integral 73</i>	5452333,625	1,35632702	1	126,8782867	126,8217428	126,8500147
<i>Integral 74</i>	7332102,872	1,823939971	1	126,6353572	126,566248	126,6008026
<i>Integral 75</i>	4660567,882	1,159366719	1	126,5138925	126,3924278	126,4531601
<i>Integral 76</i>	4507367,322	1,121256421	1	126,2919052	126,1871943	126,2395497
<i>Integral 77</i>	13128186,68	3,265778572	2	126,0259394	125,8835324	125,9547359
<i>Integral 78</i>	4245900,211	1,056213646	1	125,6573567	125,5903417	125,6238492
<i>Integral 79</i>	6848811,676	1,703716054	1	124,5495146	124,4845938	124,5170542
<i>Integral 80</i>	13682340,62	3,403630364	1	123,8039725	123,6427176	123,723345
<i>Integral 81</i>	6670736,286	1,659417873	1	123,47518	123,3746575	123,4249188
<i>Integral 82</i>	4481986,621	1,114942697	0	119,6280989	119,5757434	119,6019211
<i>Integral 83</i>	4670106,2	1,161739479	0	119,5212937	119,4500902	119,4856919
<i>Integral 84</i>	17745246,61	4,414322215	1	119,343285	119,2616105	119,3024477
<i>Integral 85</i>	2327601,231	0,579016006	0	119,2113492	119,1673706	119,1893599

<i>Integral 86</i>	1946682,074	0,484258242	0	118,9181585	118,8720856	118,8951221
<i>Integral 87</i>	15394201,85	3,829474377	1	118,8343897	118,7904111	118,8124004
<i>Integral 88</i>	6562303,484	1,632444039	1	118,7778458	118,7233961	118,7506209
<i>Integral 89</i>	1865801,625	0,464138355	0	118,5830834	118,5265394	118,5548114
<i>Integral 90</i>	524300,75	0,130425488	0	118,4511475	118,4239227	118,4375351
<i>Integral 91</i>	2282531,84	0,567804507	0	118,398792	118,3548134	118,3768027
<i>Integral 92</i>	2286410,167	0,568769283	0	118,2480082	118,2061238	118,227066
<i>Integral 93</i>	2244444,167	0,558329786	0	118,0804707	118,0385863	118,0595285
<i>Integral 94</i>	6743810,359	1,67759584	0	116,4700159	116,4009067	116,4354613
<i>Integral 95</i>	4263107,179	1,060494066	0	115,9255188	115,8564096	115,8909642
<i>Integral 96</i>	20057041,92	4,989406328	1	115,7663582	115,6784009	115,7223796
<i>Integral 97</i>	3608958,545	0,897767511	0	115,6260455	115,5883495	115,6071975
<i>Integral 98</i>	3179711,125	0,790987568	0	115,487827	115,431283	115,459555
<i>Integral 99</i>	2569864,593	0,63928164	0	115,4166235	115,3705507	115,3935871
<i>Integral 100</i>	1938903,4	0,482323213	0	115,3454201	115,3119125	115,3286663
<i>Integral 101</i>	32559438,38	8,099512808	1	115,297253	115,2051073	115,2511802
<i>Integral 102</i>	2992639,833	0,744451559	0	115,2009189	115,1590345	115,1799767
<i>Integral 103</i>	24054549,6	5,983829643	1	115,0815484	115,0019681	115,0417582
<i>Integral 104</i>	3104751,833	0,772340633	0	115,0019681	114,9600837	114,9810259
<i>Integral 105</i>	2054612,131	0,511107012	0	114,8176767	114,7087773	114,763227
<i>Integral 106</i>	3719835,059	0,925349244	0	114,6857409	114,6250086	114,6553747
<i>Integral 107</i>	8302232	2,065270095	1	114,6061606	114,5454282	114,5757944
<i>Integral 108</i>	8802163,525	2,189633475	1	114,5119207	114,4072097	114,4595652
<i>Integral 109</i>	1148581,143	0,285721995	0	114,4009271	114,3653253	114,3831262
<i>Integral 110</i>	3128242,615	0,778184219	0	113,9360103	113,8438647	113,8899375
<i>Integral 111</i>	3302524	0,8215386	0	113,7265884	113,6595734	113,6930809
<i>Integral 112</i>	12008396,69	2,987218687	0	113,1255474	113,0208364	113,0731919
<i>Integral 113</i>	11312211,23	2,814035018	0	112,4218897	112,3381209	112,3800053
<i>Integral 114</i>	2289958,526	0,569651976	0	111,2072424	111,1737349	111,1904886
<i>Integral 115</i>	1582133,143	0,393572748	0	104,2754991	104,2641811	104,2698401
<i>Integral 116</i>	20880178,48	5,194170458	1	103,2345488	103,1612512	103,1979
<i>Integral 117</i>	4988172	1,24086179	0	103,1361205	103,0900477	103,1130841
<i>Integral 118</i>	7681000	1,910731909	1	98,96263098	98,93589164	98,94926131
<i>Integral 119</i>	15218021,63	3,78564764	1	97,25572489	97,20825009	97,23198749
<i>Integral 120</i>	1859027,2	0,462453143	0	74,60855267	74,55574647	74,58214957
<i>Integral 121</i>	19639615,1	4,885566888	1	74,28841511	74,19930465	74,24385988
<i>Integral 122</i>	2044398,667	0,508566302	0	71,82962666	71,77682046	71,80322356
<i>Integral 123</i>	18353199,17	4,565556998	1	68,83446462	68,78284147	68,80865304
<i>Integral 124</i>	196771151,4	48,94895427	2	68,30667568	68,23964263	68,27315916
<i>Integral 125</i>	7083066,261	1,761989418	0	66,32650404	66,28566781	66,30608593
<i>Integral 126</i>	12391355,11	3,082483741	1	66,18781497	66,15776568	66,17279033
<i>Integral 127</i>	2064232	0,513500059	0	66,15006073	66,13388034	66,14197053
<i>Integral 128</i>	2209631,333	0,549669717	0	66,11847044	66,09843758	66,10845401
<i>Integral 129</i>	1424801,5	0,354434799	0	66,08456867	66,07069976	66,07763422

<i>Integral 130</i>	6478509,091	1,611599277	0	66,0653063	66,02755206	66,04642918
<i>Integral 131</i>	16607616	4,131324285	1	65,75864937	65,71781315	65,73823126
<i>Integral 132</i>	2213957,067	0,55074579	0	65,61071437	65,58528804	65,5980012
<i>Integral 133</i>	4166113,787	1,03636591	0	60,87423354	60,79022695	60,83223024
<i>Integral 134</i>	8690423,149	2,161836846	0	60,76349758	60,67949098	60,72149428
<i>Integral 135</i>	3256932,242	0,810197157	0	60,12581115	60,06853393	60,09717254
<i>Integral 136</i>	1810547,273	0,450393236	0	60,06089696	60,02271215	60,04180456
<i>Integral 137</i>	2068433	0,514545103	0	59,99598278	59,96161644	59,97879961
<i>Integral 138</i>	21387397,23	5,320346615	1	59,80887718	59,7630554	59,78596629
<i>Integral 139</i>	24019071,53	5,975004089	1	58,3731281	58,25093669	58,31203239
<i>Integral 140</i>	3245272,485	0,807296666	0	58,07910502	58,02182779	58,0504664
<i>Integral 141</i>	3796808,444	0,944497207	0	57,91109183	57,86527005	57,88818094
<i>Integral 142</i>	6500588,941	1,617091879	0	57,15885095	57,09775525	57,1283031
<i>Integral 143</i>	2208557,231	0,549402522	0	55,76510517	55,6963725	55,73073884
<i>Integral 144</i>	6836157,647	1,700568227	0	55,57418109	55,51308539	55,54363324
<i>Integral 145</i>	3108570,4	0,773290542	0	53,8663652	53,77615357	53,82125939
<i>Integral 146</i>	779098,087	0,193809084	0	49,28561922	49,24552516	49,26557219
<i>Integral 147</i>	3175364	0,789906174	0	46,40636477	46,37880011	46,39258244
<i>Integral 148</i>	3485415,6	0,867034866	0	46,2284474	46,1933651	46,21090625
<i>Integral 149</i>	1793460,48	0,446142712	0	44,73494378	44,69234385	44,71364382
<i>Integral 150</i>	1564101,333	0,389087141	0	44,37660315	44,34653261	44,36156788
<i>Integral 151</i>	968654,0909	0,240963192	0	44,18114462	44,10346239	44,14230351
<i>Integral 152</i>	1246337,333	0,310039905	0	43,14120503	43,10862861	43,12491682
<i>Integral 153</i>	856913,4884	0,213166508	0	42,48215897	42,40698261	42,44457079
<i>Integral 154</i>	6263975,368	1,558231691	0	41,37205478	41,33947835	41,35576657
<i>Integral 155</i>	3027160,69	0,753038996	0	38,22688239	38,1761752	38,2015288
<i>Integral 156</i>	3068114	0,763226575	0	38,11912962	38,07159163	38,09536062
<i>Integral 157</i>	3900528,941	0,970298804	0	38,00503844	37,94482365	37,97493105
<i>Integral 158</i>	11062682,67	2,751962085	1	37,82439408	37,76734849	37,79587128
<i>Integral 159</i>	3769606,244	0,937730365	0	37,04160184	36,96871026	37,00515605
<i>Integral 160</i>	2886979	0,718167283	0	36,7753891	36,73418951	36,7547893
<i>Integral 161</i>	2977489,418	0,740682729	0	36,64228272	36,49966875	36,57097574
<i>Integral 162</i>	4154751,158	1,033539333	0	36,47748436	36,41093117	36,44420777
<i>Integral 163</i>	3852927,364	0,958457396	0	36,32536279	36,24613281	36,2857478
<i>Integral 164</i>	1422030,452	0,353745471	0	35,87216729	35,8182909	35,84522909
<i>Integral 165</i>	2906584,69	0,72304441	0	35,43798698	35,3334034	35,38569519
<i>Integral 166</i>	4768327,636	1,186173125	0	35,07669825	35,01965266	35,04817546
<i>Integral 167</i>	803569,44	0,199896598	0	34,87703869	34,8326699	34,8548543
<i>Integral 168</i>	869610,4286	0,216325009	0	34,78513191	34,73759392	34,76136292
<i>Integral 169</i>	1485773,6	0,369602269	0	34,25587562	34,22101443	34,23844503
<i>Integral 170</i>	12573201,74	3,127720058	0	33,79317252	33,65689695	33,72503474
<i>Integral 171</i>	1387690,769	0,345203103	0	33,50160618	33,45723739	33,47942179
<i>Integral 172</i>	21993475,6	5,471115171	1	33,25123943	33,14348666	33,19736305
<i>Integral 173</i>	1400068,2	0,348282123	0	33,11496386	33,08010267	33,09753327

<i>Integral 174</i>	5330943,111	1,326129815	0	32,9406579	32,89311991	32,9168889
<i>Integral 175</i>	5670182,146	1,410519198	0	32,82656672	32,75367514	32,79012093
<i>Integral 176</i>	3088351,04	0,768260758	0	32,74416754	32,65542996	32,69979875
<i>Integral 177</i>	2203697,455	0,5481936	0	32,57619998	32,53816958	32,55718478
<i>Integral 178</i>	2990870,323	0,744011374	0	32,53500038	32,481124	32,50806219
<i>Integral 179</i>	3771959,75	0,938315825	0	32,25928004	32,17371166	32,21649585
<i>Integral 180</i>	2661345,744	0,662038567	0	32,16420406	32,09448168	32,12934287
<i>Integral 181</i>	1419068	0,353008529	0	32,05328209	32,0089133	32,03109769
<i>Integral 182</i>	36257231,46	9,019378869	1	31,92651411	31,82193053	31,87422232
<i>Integral 183</i>	11826022,44	2,941851119	1	31,81876133	31,75537735	31,78706934
<i>Integral 184</i>	5104861,419	1,269889547	0	31,65396297	31,54304099	31,59850198
<i>Integral 185</i>	6836825,63	1,700734395	0	31,5208566	31,47331861	31,4970876
<i>Integral 186</i>	3005502,815	0,747651365	0	31,45113421	31,40359622	31,42736522
<i>Integral 187</i>	3566158,37	0,887120504	0	31,37507343	31,32753544	31,35130443
<i>Integral 188</i>	1931366,316	0,480448281	0	31,30218184	31,27048985	31,28633585
<i>Integral 189</i>	2027225,333	0,50429425	0	31,07083029	31,0454767	31,05815349
<i>Integral 190</i>	3011903,2	0,74924353	0	31,0106155	30,97575431	30,99318491
<i>Integral 191</i>	3867242,074	0,962018335	0	30,84898634	30,80144835	30,82521734
<i>Integral 192</i>	3459789,867	0,860660187	0	30,5384048	30,48452841	30,51146661
<i>Integral 193</i>	2805282,326	0,69784435	0	30,39896003	30,32289925	30,36092964
<i>Integral 194</i>	1998903,31	0,49724884	0	30,25000766	30,19930047	30,22465407
<i>Integral 195</i>	470911,4667	0,117144326	0	30,12007049	30,0947169	30,10739369
<i>Integral 196</i>	-160383,75	-0,039897194	0	30,0883785	30,0598557	30,0741171
<i>Integral 197</i>	623871,8333	0,155194873	0	30,03133291	29,99013332	30,01073311
<i>Integral 198</i>	1701437,867	0,42325109	0	29,98062572	29,92674933	29,95368753
<i>Integral 199</i>	816174,8	0,203032318	0	29,87287294	29,83801175	29,85544235
<i>Integral 200</i>	1562891,636	0,388786215	0	29,78730456	29,74927417	29,76828936
<i>Integral 201</i>	3046662,824	0,757890362	0	29,74293577	29,68272098	29,71282838
<i>Integral 202</i>	3212526,4	0,799150723	0	29,5749682	29,53059941	29,55278381
<i>Integral 203</i>	2377192,833	0,591352454	0	29,49890742	29,45770783	29,47830763
<i>Integral 204</i>	3605674,947	0,896950681	0	28,81436036	28,74780718	28,78108377
<i>Integral 205</i>	12773804,67	3,177622208	1	28,57983961	28,51645563	28,54814762
<i>Integral 206</i>	1874958,065	0,466416118	0	28,22805849	28,1741821	28,20112029
<i>Integral 207</i>	205558323,9	51,13485858	3	27,46998927	27,39697826	27,43348376
<i>Integral 208</i>	6107772,615	1,519374565	0	26,49813152	26,47703945	26,48758549
<i>Integral 209</i>	17214331,17	4,282251251	1	23,35541289	23,31485121	23,33513205
<i>Integral 210</i>	3345940,923	0,83233903	0	23,18343138	23,16071684	23,17207411
<i>Integral 211</i>	2606128,8	0,648302755	0	23,16071684	23,13637984	23,14854834
<i>Integral 212</i>	10049900,92	2,500021661	1	23,12015517	23,0990631	23,10960913
<i>Integral 213</i>	3826373,412	0,951851813	0	23,09744063	23,06823622	23,08283843
<i>Integral 214</i>	7221259,529	1,79636649	1	23,03254195	23,00171507	23,01712851
<i>Integral 215</i>	2905063,733	0,722666056	0	22,99360274	22,96764327	22,980623
<i>Integral 216</i>	6533114,857	1,625183052	1	22,75347762	22,72914061	22,74130912
<i>Integral 217</i>	2182885,684	0,543016447	0	22,2375331	22,20508376	22,22130843

<i>Integral 218</i>	3728886	0,927600764	0	22,16127715	22,12720535	22,14424125
<i>Integral 219</i>	206024538,3	51,25083446	3	21,82218154	21,71185379	21,76701767
<i>Integral 220</i>	7781841,875	1,935817418	0	21,09856125	21,0417749	21,07016808
<i>Integral 221</i>	20914810,32	5,202785503	1	21,03852997	20,97038635	21,00445816
<i>Integral 222</i>	18880920,47	4,696833384	1	20,78704758	20,7270163	20,75703194
<i>Integral 223</i>	-117347	-0,029191337	0	20,52096299	20,50798325	20,51447312
<i>Integral 224</i>	3411465,037	0,848638863	0	20,39116562	20,34411408	20,36763985
<i>Integral 225</i>	12183984	3,030897932	1	20,12832597	20,08127442	20,10480019
<i>Integral 226</i>	3270517,143	0,813576548	0	20,06018235	20,02286561	20,04152398
<i>Integral 227</i>	2450154,286	0,609502406	0	19,59777925	19,56046251	19,57912088
<i>Integral 228</i>	2712656,444	0,67480266	0	19,07638735	19,04483005	19,0606087
<i>Integral 229</i>	1764928	0,439044948	0	18,66877223	18,64510426	18,65693825
<i>Integral 230</i>	988967,6364	0,246016407	0	18,62143629	18,60302786	18,61223207
<i>Integral 231</i>	1729142,435	0,430142901	0	18,41894362	18,37949699	18,3992203
<i>Integral 232</i>	2394418,947	0,595637637	0	18,19804252	18,16648522	18,18226387
<i>Integral 233</i>	1171344,2	0,291384551	0	18,13492792	18,10074085	18,11783439
<i>Integral 234</i>	2342654,957	0,582760784	0	16,91997191	16,88052528	16,90024859
<i>Integral 235</i>	418209,8462	0,104034227	0	16,78585338	16,74114721	16,7635003
<i>Integral 236</i>	554937,3333	0,13804667	0	16,71484946	16,68329216	16,69907081
<i>Integral 237</i>	1687459,5	0,41977382	0	16,62543711	16,58336071	16,60439891
<i>Integral 238</i>	19357632,56	4,815420676	1	16,46502084	16,40190624	16,43346354
<i>Integral 239</i>	3535489,412	0,879491269	0	16,40190624	16,34142142	16,37166383
<i>Integral 240</i>	1821679,333	0,453162457	0	16,1231501	16,0810737	16,1021119
<i>Integral 241</i>	5056669,2	1,25790121	0	16,06266528	16,0284782	16,04557174
<i>Integral 242</i>	1202260,75	0,299075378	0	16,02584843	15,99955068	16,01269955
<i>Integral 243</i>	7414420,889	1,844417469	1	15,99955068	15,96799338	15,98377203
<i>Integral 244</i>	6157343,294	1,531705808	0	15,96799338	15,93906586	15,95352962
<i>Integral 245</i>	8071578,225	2,007892472	0	15,83650463	15,53145074	15,68397769
<i>Integral 246</i>	24312093,03	6,047896359	1	14,12452115	14,06140656	14,09296385
<i>Integral 247</i>	5865010,667	1,458984902	0	13,99829196	13,94043691	13,96936443
<i>Integral 248</i>	6870986,595	1,709232305	0	13,70638694	13,64064257	13,67351475
<i>Integral 249</i>	2103405,905	0,523244992	0	13,64064257	13,60382572	13,62223414
<i>Integral 250</i>	3178610,462	0,790713766	0	13,59330662	13,52493247	13,55911954
<i>Integral 251</i>	1154732	0,287252087	0	12,15481973	12,12589221	12,14035597
<i>Integral 252</i>	4241828,7	1,055200814	0	10,48491265	10,41390873	10,44941069
<i>Integral 253</i>	940987,2727	0,234080772	0	9,13583811	9,117429685	9,126633898
<i>Integral 254</i>	331838,4	0,082548395	0	9,046425762	8,993830264	9,020128013

B6. Integrals of Quantitative ¹³C NMR of LtL-oil 5 in DMSO-d6

<i>Object</i>	<i>Integral [abs]</i>	<i>Integral [rel]</i>	<i>Peak</i>	<i>Range (F1) from</i>	<i>Range (F1) to</i>	<i>ν(F1) [ppm]</i>
<i>Integral 1</i>	3179721,846	0,173213134	1	179,549799	179,361522	179,4556605
<i>Integral 2</i>	55765692,81	3,037797289	1	177,9437914	177,5943491	177,7690702
<i>Integral 3</i>	3930884,5	0,214132196	1	174,6102967	174,4663202	174,5383085
<i>Integral 4</i>	1074013,25	0,058506124	1	173,691062	173,591386	173,641224
<i>Integral 5</i>	4691190,757	0,25554935	1	172,1183954	171,985494	172,0519447
<i>Integral 6</i>	33776957,13	1,839976222	3	170,4501951	170,2172336	170,3337143
<i>Integral 7</i>	21718494,68	1,183099876	1	170,0813393	169,7513105	169,9163249
<i>Integral 8</i>	11694124,5	0,63702929	2	162,2383007	161,9082719	162,0732863
<i>Integral 9</i>	10258938,03	0,558848507	2	157,4625891	157,249041	157,3558151
<i>Integral 10</i>	661065,7778	0,036011098	1	155,3957184	155,380865	155,3882917
<i>Integral 11</i>	847983,1667	0,046193293	1	155,2555028	155,2358964	155,2456996
<i>Integral 12</i>	1027489,333	0,055971766	1	155,0546855	155,0356733	155,0451794
<i>Integral 13</i>	988534,2222	0,053849713	1	147,8264028	147,7963629	147,8113829
<i>Integral 14</i>	7816154	0,425779547	1	147,7062432	147,6681926	147,6872179
<i>Integral 15</i>	2928308,087	0,159517544	1	147,4499026	147,4098494	147,429876
<i>Integral 16</i>	7375589,077	0,40178008	1	147,4038414	147,3577802	147,3808108
<i>Integral 17</i>	2305084,667	0,125567882	1	147,2636551	147,2336152	147,2486352
<i>Integral 18</i>	2141549,4	0,116659413	1	147,0794104	147,0093172	147,0443638
<i>Integral 19</i>	1734408,087	0,094480673	1	146,8070485	146,7669953	146,7870219
<i>Integral 20</i>	8051793,778	0,43861586	1	146,6248064	146,5927638	146,6087851
<i>Integral 21</i>	86420397,1	4,707690962	5	145,3766821	145,2489007	145,3127914
<i>Integral 22</i>	1454981,412	0,079259099	1	145,1934484	145,1645167	145,1789825
<i>Integral 23</i>	840070,8571	0,045762275	0	145,1633112	145,140407	145,1518591
<i>Integral 24</i>	1301317,818	0,070888382	1	145,0970095	145,0777218	145,0873657
<i>Integral 25</i>	10180568,92	0,554579404	1	145,0753108	145,0295024	145,0524066
<i>Integral 26</i>	13940766,27	0,759413536	1	145,0295024	145,0041872	145,0168448
<i>Integral 27</i>	691038,5714	0,037643845	1	145,0005707	144,9897214	144,9951461
<i>Integral 28</i>	545734	0,02972848	0	144,9897214	144,976461	144,9830912
<i>Integral 29</i>	7244159,111	0,394620524	1	144,9752555	144,9439129	144,9595842
<i>Integral 30</i>	928590,2857	0,050584309	0	144,941502	144,9185977	144,9300499
<i>Integral 31</i>	7118257,9	0,387762143	1	144,8776113	144,8438577	144,8607345
<i>Integral 32</i>	657076,8333	0,035793803	1	144,8330083	144,8125151	144,8227617
<i>Integral 33</i>	357384	0,019468245	1	144,7908163	144,7715286	144,7811724
<i>Integral 34</i>	5836398,229	0,31793373	1	144,7136653	144,6521855	144,6829254
<i>Integral 35</i>	-7228,222222	-0,000393752	0	144,6497746	144,6341033	144,6419389
<i>Integral 36</i>	556072,1667	0,030291644	0	144,6232539	144,6039661	144,61361
<i>Integral 37</i>	1018410,4	0,055477197	1	144,5907058	144,573829	144,5822674
<i>Integral 38</i>	2218610,933	0,120857286	1	144,4460476	144,4219379	144,4339927
<i>Integral 39</i>	1099996,857	0,059921563	1	144,4219379	144,409883	144,4159105
<i>Integral 40</i>	221261,1111	0,012053045	0	144,4050611	144,3905953	144,3978282
<i>Integral 41</i>	1153439,778	0,062832829	1	144,2495535	144,2362932	144,2429233

<i>Integral 42</i>	2202908,941	0,120001931	1	144,210978	144,1832518	144,1971149
<i>Integral 43</i>	1109827,778	0,060457095	1	144,1217721	144,091635	144,1067035
<i>Integral 44</i>	744798,5714	0,040572384	1	143,7010578	143,6757426	143,6884002
<i>Integral 45</i>	802147,4118	0,043696422	1	143,4720156	143,4442895	143,4581525
<i>Integral 46</i>	1397782,571	0,076143232	1	143,3707549	143,3466452	143,3587
<i>Integral 47</i>	7420473,565	0,404225131	1	143,3478507	143,3080696	143,3279601
<i>Integral 48</i>	1096662,154	0,059739907	1	143,2972203	143,276727	143,2869737
<i>Integral 49</i>	3203720	0,174520416	1	143,2080144	143,1622059	143,1851102
<i>Integral 50</i>	506280	0,027579251	0	143,1634114	143,1453292	143,1543703
<i>Integral 51</i>	300054	0,016345233	0	143,1321039	143,1207806	143,1264422
<i>Integral 52</i>	4945024	0,269376739	1	143,1205172	143,1012938	143,1109055
<i>Integral 53</i>	9243404,933	0,503528048	1	143,0970805	143,0712737	143,0841771
<i>Integral 54</i>	236470,4	0,01288156	0	143,0736438	143,0578437	143,0657437
<i>Integral 55</i>	503046,5	0,027403108	0	142,9651501	142,9509301	142,9580401
<i>Integral 56</i>	16527223,33	0,900308984	1	142,9019499	142,8661365	142,8840432
<i>Integral 57</i>	1187117,231	0,064667385	1	142,7934562	142,7713362	142,7823962
<i>Integral 58</i>	1602005,167	0,087268116	1	142,7428961	142,7218294	142,7323627
<i>Integral 59</i>	504049,5	0,027457746	0	142,7202494	142,7076093	142,7139293
<i>Integral 60</i>	722744,8	0,03937102	0	142,276268	142,2499346	142,2631013
<i>Integral 61</i>	475474,1818	0,025901125	1	142,0582274	142,0392673	142,0487473
<i>Integral 62</i>	310909,8462	0,016936597	1	141,8338667	141,8117466	141,8228067
<i>Integral 63</i>	1214236,8	0,066144704	1	141,0757712	141,0590201	141,0673956
<i>Integral 64</i>	1983938,375	0,108073662	1	140,8704626	140,8438326	140,8571476
<i>Integral 65</i>	15002237,33	0,817236433	1	134,6681042	134,5993121	134,6337081
<i>Integral 66</i>	2227816,667	0,121358762	1	134,1491721	134,1192625	134,1342173
<i>Integral 67</i>	489339,6667	0,026656438	1	133,8635352	133,8425984	133,8530668
<i>Integral 68</i>	95528,90909	0,005203871	0	133,8470849	133,8291391	133,838112
<i>Integral 69</i>	438896,3636	0,023908574	1	133,8052114	133,7872656	133,7962385
<i>Integral 70</i>	1320437,059	0,071929889	1	133,6347265	133,6048169	133,6197717
<i>Integral 71</i>	4024515,333	0,219232671	1	133,0081198	132,9767147	132,9924172
<i>Integral 72</i>	5187227,294	0,282570595	1	132,9767147	132,9483005	132,9625076
<i>Integral 73</i>	1561387,04	0,085055472	1	132,5579799	132,5146109	132,5362954
<i>Integral 74</i>	2090182,483	0,113861235	1	132,2185056	132,1676592	132,1930824
<i>Integral 75</i>	862366,8	0,046976831	1	131,3929998	131,3660812	131,3795405
<i>Integral 76</i>	828699,4286	0,045142824	1	131,1537228	131,1297951	131,1417589
<i>Integral 77</i>	711166,6667	0,038740309	1	130,6721778	130,651241	130,6617094
<i>Integral 78</i>	1208081,9	0,06580942	1	130,573476	130,5375844	130,5555302
<i>Integral 79</i>	621101,75	0,033834085	1	130,2145605	130,1876418	130,2011011
<i>Integral 80</i>	471149,5556	0,025665544	0	130,1502548	130,1352999	130,1427773
<i>Integral 81</i>	2907853,091	0,158403273	1	129,7494657	129,7120787	129,7307722
<i>Integral 82</i>	383803,2	0,020907412	0	129,7120787	129,6956284	129,7038535
<i>Integral 83</i>	9365177,067	0,5101615	1	129,4114869	129,3591451	129,385316
<i>Integral 84</i>	446745,6364	0,024336157	1	129,1303364	129,1123906	129,1213635
<i>Integral 85</i>	1197591,143	0,065237944	1	129,0435985	129,0077069	129,0256527

<i>Integral 86</i>	1960631,852	0,106804055	1	128,5500896	128,5022342	128,5261619
<i>Integral 87</i>	18746749,04	1,021216101	1	127,9399332	127,8920778	127,9160055
<i>Integral 88</i>	1605198,842	0,087442089	1	127,1667693	127,1323732	127,1495712
<i>Integral 89</i>	1690003	0,092061737	1	127,1323732	127,1054545	127,1189139
<i>Integral 90</i>	1544631,405	0,08414272	2	127,0800313	127,0142302	127,0471308
<i>Integral 91</i>	1815653,263	0,098906447	1	126,6463417	126,6134411	126,6298914
<i>Integral 92</i>	2640287,84	0,143827842	1	126,5281987	126,4848297	126,5065142
<i>Integral 93</i>	2022591,429	0,11017926	1	126,0182395	125,9943118	126,0062757
<i>Integral 94</i>	312116,9091	0,017002351	1	125,5980092	125,5606222	125,5793157
<i>Integral 95</i>	743960,5714	0,040526734	0	124,5556587	124,5302355	124,5429471
<i>Integral 96</i>	6153044,421	0,335182811	1	124,5227581	124,4898575	124,5063078
<i>Integral 97</i>	88944,8	0,004845206	0	124,4898575	124,4734072	124,4816323
<i>Integral 98</i>	163127,6364	0,008886264	0	124,2356256	124,2176799	124,2266527
<i>Integral 99</i>	2057273,077	0,112068518	1	124,2206708	124,1997341	124,2102024
<i>Integral 100</i>	3477978,444	0,189460454	1	124,1982386	124,1668335	124,182536
<i>Integral 101</i>	927266,4167	0,050512192	0	124,1638425	124,1204736	124,142158
<i>Integral 102</i>	1374558,833	0,074878135	1	124,0860775	124,0666362	124,0763569
<i>Integral 103</i>	1132084,737	0,061669528	0	123,8138999	123,7809993	123,7974496
<i>Integral 104</i>	1263379,273	0,068821706	1	123,7735219	123,7555761	123,764549
<i>Integral 105</i>	1272143,6	0,069299137	1	123,7555761	123,7301529	123,7428645
<i>Integral 106</i>	386266,4615	0,021041597	0	123,4594708	123,438534	123,4490024
<i>Integral 107</i>	478090,5455	0,02604365	0	122,9839077	122,9644664	122,974187
<i>Integral 108</i>	401802	0,021887884	0	122,8687556	122,8523053	122,8605304
<i>Integral 109</i>	2419401,556	0,131795215	1	122,6758385	122,6444334	122,6601359
<i>Integral 110</i>	899698,6667	0,049010458	0	122,3931925	122,3737512	122,3834719
<i>Integral 111</i>	738980,8889	0,040255469	0	122,2167257	122,1853206	122,2010231
<i>Integral 112</i>	567998,2667	0,03094131	0	121,6948027	121,6693795	121,6820911
<i>Integral 113</i>	2146853,217	0,116948334	1	121,5273087	121,4869307	121,5071197
<i>Integral 114</i>	7643304,786	0,416363707	1	121,0786643	121,0293134	121,0539889
<i>Integral 115</i>	2369784,444	0,129092358	1	121,0008993	120,9694942	120,9851967
<i>Integral 116</i>	7777173,273	0,423656099	1	120,9500529	120,9111704	120,9306116
<i>Integral 117</i>	435473,7647	0,02372213	0	120,6404882	120,6105786	120,6255334
<i>Integral 118</i>	2942932,261	0,160314186	1	120,3997157	120,3593377	120,3795267
<i>Integral 119</i>	2444358,686	0,133154738	0	120,2725998	120,2097896	120,2411947
<i>Integral 120</i>	3014423,391	0,164208615	1	119,7117943	119,6714163	119,6916053
<i>Integral 121</i>	14952991,85	0,81455382	1	119,5188772	119,4725172	119,4956972
<i>Integral 122</i>	601214,6667	0,03275075	0	119,4695263	119,4426076	119,4560669
<i>Integral 123</i>	206438,6667	0,011245603	0	119,4366257	119,4216709	119,4291483
<i>Integral 124</i>	1133444,375	0,061743593	0	119,4216709	119,3947522	119,4082115
<i>Integral 125</i>	-515644,0952	-0,028089353	0	119,3902657	119,3543742	119,37232
<i>Integral 126</i>	80550045,05	4,387907621	1	119,3274555	119,2586634	119,2930595
<i>Integral 127</i>	476370,8421	0,02594997	1	119,251186	119,2182854	119,2347357
<i>Integral 128</i>	8696237,692	0,473721494	1	119,2227718	119,1779074	119,2003396
<i>Integral 129</i>	689914,125	0,037582592	0	119,0627553	119,0358366	119,049296

<i>Integral 130</i>	6184895	0,336917849	1	118,9326484	118,8907749	118,9117117
<i>Integral 131</i>	1028863,75	0,056046636	1	118,8937659	118,8653518	118,8795588
<i>Integral 132</i>	1228158,588	0,066903084	1	118,8653518	118,8369376	118,8511447
<i>Integral 133</i>	5281303,083	0,287695308	1	118,8354421	118,7935687	118,8145054
<i>Integral 134</i>	1184321,667	0,064515098	0	118,7771184	118,7576771	118,7673977
<i>Integral 135</i>	603931,3333	0,032898739	0	118,7158036	118,6843985	118,7001011
<i>Integral 136</i>	7638766,593	0,416116492	1	118,5856967	118,5378413	118,561769
<i>Integral 137</i>	2306166	0,125626787	0	118,4630673	118,4316621	118,4473647
<i>Integral 138</i>	10008994,1	0,545232985	1	118,2596818	118,1893941	118,224538
<i>Integral 139</i>	1258153,813	0,068537053	0	118,0757376	118,0189093	118,0473234
<i>Integral 140</i>	2322776,786	0,126531648	0	117,2397634	117,1904125	117,215088
<i>Integral 141</i>	19531047,75	1,063940228	1	116,4665995	116,3798616	116,4232306
<i>Integral 142</i>	866402,8	0,047196689	0	116,3275197	116,2931237	116,3103217
<i>Integral 143</i>	816460,4	0,044476112	0	116,2318089	116,2048903	116,2183496
<i>Integral 144</i>	1680143,5	0,091524647	0	116,0628195	116,0284235	116,0456215
<i>Integral 145</i>	673352	0,036680381	0	116,0284235	116,0059912	116,0172073
<i>Integral 146</i>	2221765,913	0,121029151	0	115,9013075	115,8190561	115,8601818
<i>Integral 147</i>	108763717,9	5,924827806	2	115,7906419	115,6500667	115,7203543
<i>Integral 148</i>	7939196,667	0,432482211	1	115,6336164	115,5857609	115,6096887
<i>Integral 149</i>	31413220,08	1,711213293	2	115,5199598	115,4332218	115,4765908
<i>Integral 150</i>	9017879,6	0,491242712	1	115,4332218	115,3539613	115,3935916
<i>Integral 151</i>	5088165,143	0,277174254	1	115,3539613	115,3046104	115,3292859
<i>Integral 152</i>	13148640,8	0,716263053	1	115,2866647	115,2328273	115,259746
<i>Integral 153</i>	-273226,7692	-0,014883838	0	115,2253499	115,2044132	115,2148816
<i>Integral 154</i>	8853192,615	0,482271504	1	115,2044132	115,1580533	115,1812332
<i>Integral 155</i>	695600	0,037892326	0	115,1550623	115,1281436	115,141603
<i>Integral 156</i>	6916313,733	0,376761375	1	115,0743063	115,0219644	115,0481354
<i>Integral 157</i>	3385227,212	0,184407895	0	115,0219644	114,9636407	114,9928026
<i>Integral 158</i>	1170902,444	0,063784095	0	114,5987432	114,5673381	114,5830406
<i>Integral 159</i>	688466,7692	0,037503748	0	114,4581679	114,4357357	114,4469518
<i>Integral 160</i>	1207962,105	0,065802895	0	113,728373	113,6969679	113,7126704
<i>Integral 161</i>	2406792	0,131108318	0	113,1780358	113,1451352	113,1615855
<i>Integral 162</i>	824179,8824	0,044896625	0	113,1286849	113,0987753	113,1137301
<i>Integral 163</i>	8764890,471	0,477461306	1	113,0733521	113,0135329	113,0434425
<i>Integral 164</i>	4162880,909	0,226770039	0	112,5409607	112,5020782	112,5215195
<i>Integral 165</i>	1250516,143	0,068120996	0	112,4975918	112,4482409	112,4729163
<i>Integral 166</i>	937831,0909	0,051087696	0	112,4273041	112,3884216	112,4078629
<i>Integral 167</i>	9018654,714	0,491284936	1	112,3899171	112,3405662	112,3652417
<i>Integral 168</i>	3555470,933	0,193681804	0	112,0803524	112,0265151	112,0534338
<i>Integral 169</i>	577797,125	0,031475096	0	111,3729897	111,3460711	111,3595304
<i>Integral 170</i>	3384268,177	0,184355653	0	110,3067114	110,1631452	110,2349283
<i>Integral 171</i>	1707747	0,093028329	0	109,5514933	109,5170972	109,5342953
<i>Integral 172</i>	2815360,727	0,153364815	0	104,2758704	104,2585053	104,2671878
<i>Integral 173</i>	3147873,429	0,171478213	0	103,2246195	103,1999077	103,2122636

<i>Integral 174</i>	593228,6667	0,032315719	0	103,1999077	103,1915592	103,1957335
<i>Integral 175</i>	1147492,875	0,062508875	0	103,1842125	103,1561613	103,1701869
<i>Integral 176</i>	161626,5	0,008804491	0	103,1387963	103,1267744	103,1327853
<i>Integral 177</i>	2212751,636	0,120538105	0	103,1214313	103,0840297	103,1027305
<i>Integral 178</i>	1071866,667	0,05838919	0	102,9117154	102,885	102,8983577
<i>Integral 179</i>	1837802,222	0,100112996	0	98,96585131	98,93379284	98,94982208
<i>Integral 180</i>	7873377,385	0,428896751	1	97,24270823	97,19729205	97,22000014
<i>Integral 181</i>	22443365,57	1,222586713	1	74,28630546	74,22056673	74,2534361
<i>Integral 182</i>	20081083,68	1,093903052	1	68,82921793	68,79518847	68,8122032
<i>Integral 183</i>	61691194,61	3,360584874	1	68,33965778	68,25303735	68,29634756
<i>Integral 184</i>	807258,4	0,043974839	0	67,73408812	67,70779263	67,72094038
<i>Integral 185</i>	1307549,059	0,071227825	0	67,6196254	67,59023632	67,60493086
<i>Integral 186</i>	1754156,643	0,095556462	0	67,34893653	67,30098594	67,32496124
<i>Integral 187</i>	1613220,313	0,087879053	0	67,25612892	67,20044436	67,22828664
<i>Integral 188</i>	852114,3529	0,046418336	0	67,0705137	67,04267142	67,05659256
<i>Integral 189</i>	1229160,421	0,066957658	0	66,89263245	66,86014978	66,87639112
<i>Integral 190</i>	2968635,368	0,161714345	0	66,3311464	66,29866374	66,31490507
<i>Integral 191</i>	6417383,905	0,34958252	0	66,20121574	66,1640927	66,18265422
<i>Integral 192</i>	647448,6667	0,035269315	0	66,1640927	66,13315683	66,14862477
<i>Integral 193</i>	5567539,263	0,303287825	0	66,13006324	66,09758058	66,11382191
<i>Integral 194</i>	1828844,615	0,099625037	0	66,0929402	66,04808319	66,07051169
<i>Integral 195</i>	168412	0,009174126	0	66,04962998	66,03416204	66,04189601
<i>Integral 196</i>	296856,4211	0,016171047	0	66,02952166	65,997039	66,01328033
<i>Integral 197</i>	1097539,52	0,059787701	0	66,00013259	65,95682237	65,97847748
<i>Integral 198</i>	929289,0769	0,050622375	0	65,95527558	65,90887177	65,93207367
<i>Integral 199</i>	1995584	0,108708049	0	65,90732498	65,854734	65,88102949
<i>Integral 200</i>	112133,1667	0,006108376	0	65,85009362	65,82843851	65,83926606
<i>Integral 201</i>	291687,1667	0,015889455	0	65,82379813	65,80368981	65,81374397
<i>Integral 202</i>	2410276,727	0,131298147	0	65,8052366	65,76811356	65,78667508
<i>Integral 203</i>	8283208,769	0,451222031	0	65,76811356	65,72170975	65,74491166
<i>Integral 204</i>	25329,81818	0,001379824	0	65,72016296	65,70160144	65,7108822
<i>Integral 205</i>	907341,7037	0,049426808	0	65,70160144	65,65519763	65,67839954
<i>Integral 206</i>	275270,3077	0,014995158	0	65,65674443	65,63508932	65,64591687
<i>Integral 207</i>	1953537,6	0,106417601	0	65,63508932	65,58249834	65,60879383
<i>Integral 208</i>	4516690,645	0,246043578	0	65,56084323	65,45102089	65,50593206
<i>Integral 209</i>	1512493,647	0,082392038	0	63,69076987	63,63199171	63,66138079
<i>Integral 210</i>	3039520,909	0,165575785	0	63,57940073	63,52062258	63,55001166
<i>Integral 211</i>	2132798,813	0,11618273	0	62,81528474	62,75960017	62,78744245
<i>Integral 212</i>	1253301,543	0,068272729	0	62,7286643	62,66679256	62,69772843
<i>Integral 213</i>	2901183,708	0,158039963	0	60,88798001	60,80135958	60,84466979
<i>Integral 214</i>	3486960,564	0,189949749	0	60,77042371	60,70236479	60,73639425
<i>Integral 215</i>	1243100,3	0,067717023	0	60,24703684	60,21208381	60,22956032
<i>Integral 216</i>	926719,1111	0,050482378	0	60,04430927	60,01335087	60,02883007
<i>Integral 217</i>	400335,6667	0,021808007	0	59,98239248	59,96142066	59,97190657

<i>Integral 218</i>	46165690,63	2,514843854	1	59,84757365	59,73372665	59,79065015
<i>Integral 219</i>	1984562,4	0,108107655	0	59,73472531	59,69078436	59,71275483
<i>Integral 220</i>	23044904,81	1,255355145	1	58,41749547	58,2457263	58,33161089
<i>Integral 221</i>	1722342,233	0,093823394	0	58,07395713	57,99805913	58,03600813
<i>Integral 222</i>	2228434,31	0,121392407	0	57,93015039	57,80232217	57,86623628
<i>Integral 223</i>	2749864,681	0,149796964	0	57,16318108	57,07929382	57,12123745
<i>Integral 224</i>	2967665,103	0,16166149	0	55,78902774	55,73709753	55,76306264
<i>Integral 225</i>	2053419,2	0,111858581	0	55,74708411	55,70314316	55,72511364
<i>Integral 226</i>	708655	0,038603488	0	55,70514048	55,67717805	55,69115927
<i>Integral 227</i>	19498707,81	1,062178532	0	55,59728542	55,52338473	55,56033507
<i>Integral 228</i>	2351054	0,12807203	0	55,52538205	55,48543573	55,50540889
<i>Integral 229</i>	4056007,826	0,220948202	0	53,87559911	53,83565279	53,85562595
<i>Integral 230</i>	1258163,636	0,068537588	0	49,4074677	49,35022424	49,37884597
<i>Integral 231</i>	3846025,385	0,209509555	0	49,10852962	48,96860116	49,03856539
<i>Integral 232</i>	3391529,238	0,184751194	0	48,69510462	48,5806177	48,63786116
<i>Integral 233</i>	467973,8298	0,025492549	0	46,54529462	46,46260962	46,50395212
<i>Integral 234</i>	2314656,48	0,1260893	0	46,44352847	46,35448309	46,39900578
<i>Integral 235</i>	3790201,93	0,206468611	0	46,30360001	46,20183385	46,25271693
<i>Integral 236</i>	2275362,233	0,123948773	0	44,78254182	44,70646421	44,74450301
<i>Integral 237</i>	5445284,435	0,296628078	1	37,84391835	37,76185812	37,80288824
<i>Integral 238</i>	3014994,133	0,164239706	1	36,99501933	36,94125573	36,96813753
<i>Integral 239</i>	6633885,789	0,36137631	1	36,75166828	36,72054198	36,73610513
<i>Integral 240</i>	1433834,286	0,078107124	1	36,48002059	36,44323497	36,46162778
<i>Integral 241</i>	2348593,556	0,127937999	1	36,33853742	36,27628483	36,30741113
<i>Integral 242</i>	990992,7586	0,053983641	0	35,87730229	35,82636835	35,85183532
<i>Integral 243</i>	3044528,25	0,165848557	0	35,45568244	35,38494086	35,42031165
<i>Integral 244</i>	1768346,526	0,096329446	0	35,09914485	35,03123293	35,06518889
<i>Integral 245</i>	3082300,824	0,167906191	0	34,67469534	34,61527241	34,64498388
<i>Integral 246</i>	9663770,769	0,526427184	1	33,24571532	33,15233643	33,19902588
<i>Integral 247</i>	2915088,211	0,158797401	0	32,95425999	32,88634807	32,92030403
<i>Integral 248</i>	358221,5385	0,019513869	0	32,86654043	32,82126581	32,84390312
<i>Integral 249</i>	1868675,833	0,101794815	0	32,80994716	32,76750221	32,78872468
<i>Integral 250</i>	3713173,481	0,20227254	0	32,74769457	32,65148601	32,69959029
<i>Integral 251</i>	778004,8	0,04238127	0	32,6288487	32,44775024	32,53829947
<i>Integral 252</i>	1075526,5	0,058588557	0	32,17893222	32,12233895	32,15063559
<i>Integral 253</i>	789430,8485	0,043003696	0	32,12233895	32,06291602	32,09262749
<i>Integral 254</i>	1674001,838	0,091190085	0	32,04310838	31,97802612	32,01056725
<i>Integral 255</i>	2564152,615	0,139680428	0	31,95538881	31,9101142	31,93275151
<i>Integral 256</i>	13449528,09	0,732653679	1	31,9101142	31,83371329	31,87191374
<i>Integral 257</i>	8505985,489	0,463357637	0	31,83088362	31,74599372	31,78843867
<i>Integral 258</i>	2326149,241	0,12671536	0	31,51396132	31,46302738	31,48849435
<i>Integral 259</i>	2510322,914	0,136748093	0	31,37530781	31,31305522	31,34418152
<i>Integral 260</i>	2268440,35	0,123571709	0	31,31305522	31,24231363	31,27768443
<i>Integral 261</i>	1742079,739	0,094898581	0	31,23665431	31,15459407	31,19562419

<i>Integral 262</i>	1500472,154	0,081737176	0	31,14610508	31,10083046	31,12346777
<i>Integral 263</i>	3423800,507	0,186509149	0	31,08668215	30,96783628	31,02725922
<i>Integral 264</i>	1821341,9	0,099216332	0	30,73014456	30,65940297	30,69477377
<i>Integral 265</i>	4288962,03	0,23363822	0	30,59149105	30,47264519	30,53206812
<i>Integral 266</i>	1908018,353	0,103937971	0	30,2858874	30,22646447	30,25617594
<i>Integral 267</i>	1465665,467	0,079841106	0	30,22646447	30,17270087	30,19958267
<i>Integral 268</i>	5480258,17	0,298533248	0	30,07649231	29,90671251	29,99160241
<i>Integral 269</i>	2834348,077	0,154399138	0	29,60393852	29,51055963	29,55724908
<i>Integral 270</i>	5653574,159	0,307974516	0	29,14545064	29,03165971	29,08855517
<i>Integral 271</i>	2378665,784	0,129576162	0	28,99872023	28,93284127	28,96578075
<i>Integral 272</i>	3270771,646	0,178173008	0	28,83402283	28,546551	28,69028691
<i>Integral 273</i>	4783654,239	0,260586234	0	27,9027339	27,78295397	27,84284393
<i>Integral 274</i>	1986268,588	0,108200598	0	27,63622356	27,5763336	27,60627858
<i>Integral 275</i>	10447705,42	0,569131478	1	27,5733391	27,53141613	27,55237761
<i>Integral 276</i>	59513662,74	3,241965341	1	27,48050966	27,41762519	27,44906743
<i>Integral 277</i>	1545234,667	0,084175583	0	27,37570222	27,34575724	27,36072973
<i>Integral 278</i>	1418748,688	0,077285347	0	26,87262653	26,81573106	26,8441788
<i>Integral 279</i>	3714198,56	0,202328381	0	26,52825924	26,43842429	26,48334177
<i>Integral 280</i>	2244318,7	0,122257699	0	25,48317938	25,37537745	25,42927841
<i>Integral 281</i>	4727470,693	0,257525674	0	25,24062503	25,05796064	25,14929284
<i>Integral 282</i>	4684259,586	0,25517178	0	24,88427975	24,77947231	24,83187603
<i>Integral 283</i>	1334898,513	0,072717667	0	24,72557134	24,65669789	24,69113462
<i>Integral 284</i>	3391430	0,184745788	0	24,45007751	24,39617655	24,42312703
<i>Integral 285</i>	11331850,29	0,617294655	1	24,39617655	24,3093361	24,35275632
<i>Integral 286</i>	609793,4667	0,033218075	0	24,27939112	24,22549015	24,25244063
<i>Integral 287</i>	3172692,767	0,172830229	0	23,9859303	23,87812836	23,93202933
<i>Integral 288</i>	1185242,194	0,064565243	0	23,77631543	23,72241446	23,74936494
<i>Integral 289</i>	9303874,192	0,506822069	1	23,37804717	23,28521773	23,33163245
<i>Integral 290</i>	12671864,81	0,690291013	0	23,24030026	23,12650932	23,18340479
<i>Integral 291</i>	4437258,645	0,241716575	1	23,12650932	23,07260836	23,09955884
<i>Integral 292</i>	737626,25	0,040181676	0	23,04266338	23,0007404	23,02170189
<i>Integral 293</i>	5458108,56	0,297326664	0	23,0097239	22,91988895	22,96480642
<i>Integral 294</i>	752037,4118	0,040966714	0	22,91389996	22,85400999	22,88395497
<i>Integral 295</i>	1131807,692	0,061654436	0	22,848021	22,80310352	22,82556226
<i>Integral 296</i>	505957,3793	0,027561676	0	22,76716955	22,71626308	22,74171631
<i>Integral 297</i>	483420,8889	0,026334016	0	22,47071423	22,42280226	22,44675824
<i>Integral 298</i>	806909,0435	0,043955808	0	22,3449453	22,30601683	22,32548107
<i>Integral 299</i>	1271045,619	0,069239325	0	22,23714337	22,16228092	22,19971214
<i>Integral 300</i>	3240248,679	0,176510291	0	22,16228092	22,06346248	22,1128717
<i>Integral 301</i>	65252700,72	3,554595439	1	21,83288612	21,72508418	21,77898515
<i>Integral 302</i>	3020765,306	0,164554087	0	21,2160195	21,12917905	21,17259927
<i>Integral 303</i>	2321485,905	0,126461328	0	21,12917905	21,0543166	21,09174782
<i>Integral 304</i>	22981573,8	1,251905232	1	21,06629459	20,99442663	21,03036061
<i>Integral 305</i>	43794170,21	2,385656931	1	20,82374024	20,7219273	20,77283377

<i>Integral 306</i>	2684212	0,14622058	0	20,62310886	20,56920789	20,59615838
<i>Integral 307</i>	20813262,08	1,13378796	1	20,43744997	20,34162603	20,389538
<i>Integral 308</i>	2856577,724	0,155610083	0	20,16195614	20,11104967	20,13650291
<i>Integral 309</i>	1411864,621	0,076910342	0	18,68566855	18,63476208	18,66021531
<i>Integral 310</i>	2353205,636	0,128189239	0	18,63775658	18,5988281	18,61829234
<i>Integral 311</i>	2316190	0,126172838	0	18,22451583	18,15564237	18,1900791
<i>Integral 312</i>	24160824,65	1,316144101	1	16,49369589	16,38289946	16,43829767
<i>Integral 313</i>	7378432	0,401934946	0	16,17029009	16,06847715	16,11938362
<i>Integral 314</i>	582037,4286	0,031706084	0	16,07147165	16,04751566	16,05949366
<i>Integral 315</i>	8521478,609	0,464201614	1	16,04751566	16,00858719	16,02805143
<i>Integral 316</i>	3758195,2	0,204725067	0	16,00607602	15,98068457	15,9933803
<i>Integral 317</i>	9939427,7	0,541443404	1	15,97919096	15,94433994	15,96176545
<i>Integral 318</i>	1669004,842	0,090917877	0	15,93936122	15,90650169	15,92293146
<i>Integral 319</i>	250794,1176	0,013661835	0	15,9010251	15,87214855	15,88658683
<i>Integral 320</i>	484526	0,026394216	0	15,87015706	15,85522091	15,86268899
<i>Integral 321</i>	2997837,692	0,163305121	0	15,8537273	15,80742523	15,83057626
<i>Integral 322</i>	1615151,742	0,087984267	0	14,90030307	14,84653293	14,873418
<i>Integral 323</i>	5713902,822	0,311260878	0	14,53984398	14,41039735	14,47512067
<i>Integral 324</i>	1125178,4	0,06129331	0	14,40840587	14,38251654	14,3954612
<i>Integral 325</i>	502443,2308	0,027370245	0	14,23713801	14,21523166	14,22618484
<i>Integral 326</i>	3013037,85	0,164133139	0	14,21124869	14,13955517	14,17540193
<i>Integral 327</i>	43756966,44	2,383630281	1	14,13955517	14,0598957	14,09972544
<i>Integral 328</i>	3180099,28	0,173233694	0	14,03599786	13,94638096	13,99118941
<i>Integral 329</i>	499497,1	0,027209757	0	13,94638096	13,91252569	13,92945333
<i>Integral 330</i>	15545475,14	0,846828935	1	13,72134297	13,62176864	13,67155581
<i>Integral 331</i>	2347274,316	0,127866134	0	13,59986229	13,53215174	13,56600702
<i>Integral 332</i>	2009310,188	0,109455773	0	12,02641383	11,96902808	11,99772095
<i>Integral 333</i>	6180998,022	0,336705564	0	10,52618633	10,35812806	10,4421572
<i>Integral 334</i>	1384250,746	0,075406096	0	9,103839505	8,997265967	9,050552736

B7. Integrals of Quantitative ¹³C NMR of LtL-oil 6 in DMSO-d6

<i>Object</i>	<i>Integral [abs]</i>	<i>Integral [rel]</i>	<i>Peak</i>	<i>Range (F1) from</i>	<i>Range (F1) to</i>	<i>v(F1) [ppm]</i>
<i>Integral 1</i>	5938010,531	1	0	210,9180267	210,6521808	210,7851037
<i>Integral 2</i>	816159,2632	0,137446584	0	209,5683476	209,3638508	209,4660992
<i>Integral 3</i>	1529111,934	0,2575125	0	209,0980049	208,9344074	209,0162062
<i>Integral 4</i>	5311435,717	0,894480683	0	208,6481119	208,4436151	208,5458635
<i>Integral 5</i>	1704645,77	0,287073551	0	207,0734863	206,8689895	206,9712379
<i>Integral 6</i>	9393441,774	1,581917332	1	206,7258417	206,4395462	206,5826939
<i>Integral 7</i>	2667269,059	0,44918564	0	203,4743422	203,290295	203,3823186
<i>Integral 8</i>	2310798,789	0,389153703	0	203,1471473	202,8813014	203,0142243
<i>Integral 9</i>	6498050,198	1,094314361	0	196,2556042	195,9079596	196,0817819

<i>Integral 10</i>	1139617,913	0,191919147	0	183,1064583	182,9837602	183,0451092
<i>Integral 11</i>	1389837,912	0,234057839	0	180,5706976	180,4479995	180,5093486
<i>Integral 12</i>	1606865,516	0,270606714	0	180,0594556	179,8958581	179,9776568
<i>Integral 13</i>	27695153,81	4,664045924	3	179,77316	179,384616	179,578888
<i>Integral 14</i>	2745839,496	0,462417418	0	179,3641664	179,1187702	179,2414683
<i>Integral 15</i>	98761927,22	16,63215764	2	177,9725456	177,7879028	177,8802242
<i>Integral 16</i>	23380486,71	3,937427627	2	177,6417273	177,2032007	177,422464
<i>Integral 17</i>	63655630,38	10,72002652	3	177,2032007	176,8800759	177,0416383
<i>Integral 18</i>	14557175,37	2,451524007	7	176,8569956	176,1722786	176,5146371
<i>Integral 19</i>	21320967,59	3,590591072	1	175,9030079	175,6183503	175,7606791
<i>Integral 20</i>	39522533,37	6,655854376	2	175,5183355	175,2413714	175,3798534
<i>Integral 21</i>	644434430	108,5269934	5	175,1413565	174,1181279	174,6297422
<i>Integral 22</i>	188607802,3	31,76279351	6	174,1335148	173,5795865	173,8565506
<i>Integral 23</i>	23679873,67	3,987846359	1	173,5411192	173,2641551	173,4026372
<i>Integral 24</i>	47427226,34	7,987056623	1	173,2487682	172,6255989	172,9371835
<i>Integral 25</i>	1981770,87	0,333743239	0	172,3871019	172,2640068	172,3255544
<i>Integral 26</i>	40996439,41	6,904069839	1	172,1793788	172,0332033	172,1062911
<i>Integral 27</i>	5626091,459	0,947470778	0	171,4092398	171,1024946	171,2558672
<i>Integral 28</i>	119711738,3	20,16024352	4	170,6117022	170,2436079	170,4276551
<i>Integral 29</i>	45268513,71	7,623515229	1	170,0391111	169,8141646	169,9266378
<i>Integral 30</i>	35419707,99	5,964911616	0	169,0575263	168,0145925	168,5360594
<i>Integral 31</i>	4026003,137	0,678005389	0	167,4829008	167,2988536	167,3908772
<i>Integral 32</i>	126140113,8	21,24282421	2	166,3581682	165,6833287	166,0207484
<i>Integral 33</i>	18194027,96	3,063993887	1	163,3725145	163,0657693	163,2191419
<i>Integral 34</i>	11428776,65	1,924681101	1	162,5136279	162,2682317	162,3909298
<i>Integral 35</i>	23755556,22	4,000591797	2	162,2273323	161,9001374	162,0637349
<i>Integral 36</i>	32653398,54	5,499046923	1	160,6527068	159,8756188	160,2641628
<i>Integral 37</i>	15324830,51	2,580802178	2	157,4830059	157,2989588	157,3909824
<i>Integral 38</i>	16143466,3	2,718665825	4	155,4584874	154,967695	155,2130912
<i>Integral 39</i>	746318,4425	0,125684931	0	152,3296859	152,1251891	152,2274375
<i>Integral 40</i>	659216,1053	0,111016325	0	150,7959597	150,5914629	150,6937113
<i>Integral 41</i>	39466067,44	6,646345141	6	147,9534538	147,1354665	147,5444602
<i>Integral 42</i>	25100058,76	4,227014862	2	147,1354665	146,4401773	146,7878219
<i>Integral 43</i>	9723300,266	1,63746767	3	146,4294065	146,198603	146,3140048
<i>Integral 44</i>	34238852,68	5,766047821	4	145,8447044	145,4215647	145,6331346
<i>Integral 45</i>	342407178,2	57,66361923	3	145,460032	145,2830827	145,3715573
<i>Integral 46</i>	3663575,315	0,616970161	2	145,2600023	145,1292137	145,194608
<i>Integral 47</i>	231388189,3	38,96729184	4	145,1292137	144,929184	145,0291989
<i>Integral 48</i>	96556740,04	16,26078963	5	144,929184	144,5675919	144,748388
<i>Integral 49</i>	29960092,02	5,045476404	6	144,5137378	144,0829047	144,2983212
<i>Integral 50</i>	17764930,96	2,991731131	3	144,0213571	143,6136043	143,8174807
<i>Integral 51</i>	39935786,35	6,725448892	3	143,5828305	143,2904795	143,436655
<i>Integral 52</i>	143558930,7	24,17626745	4	143,2673991	143,0058219	143,1366105
<i>Integral 53</i>	96050211,2	16,17548684	2	142,9981284	142,8442595	142,921194

<i>Integral 54</i>	19859274,34	3,344432321	3	142,8211791	142,7057774	142,7634783
<i>Integral 55</i>	287167,8636	0,048360956	0	142,6596167	142,5826822	142,6211495
<i>Integral 56</i>	13798971,46	2,323837485	2	142,5596019	142,3595722	142,459587
<i>Integral 57</i>	2562501,657	0,431542121	1	142,3057181	142,2441705	142,2749443
<i>Integral 58</i>	6248880,225	1,0523525	2	142,1749294	142,013367	142,0941482
<i>Integral 59</i>	1954296,458	0,329116368	1	141,8825784	141,7979505	141,8402644
<i>Integral 60</i>	1403416,933	0,236344635	0	141,6902422	141,5286798	141,609461
<i>Integral 61</i>	7410892,681	1,248043034	1	141,4209715	141,2517156	141,3363435
<i>Integral 62</i>	40482012,63	6,817436989	3	141,2594091	140,6977873	140,9785982
<i>Integral 63</i>	4211513,533	0,709246559	1	140,6900939	140,5823856	140,6362397
<i>Integral 64</i>	857237,9574	0,144364506	1	140,451597	140,366969	140,409283
<i>Integral 65</i>	3858227,778	0,649750915	0	140,1515525	139,9899901	140,0707713
<i>Integral 66</i>	2196352,512	0,369880198	0	139,8976687	139,8207342	139,8592014
<i>Integral 67</i>	2287639,436	0,385253516	0	139,5206897	139,4514487	139,4860692
<i>Integral 68</i>	2380259,318	0,400851313	0	139,0744697	138,9975352	139,0360024
<i>Integral 69</i>	2216028,567	0,373193775	0	138,7667317	138,6590234	138,7128776
<i>Integral 70</i>	1065176,35	0,179382698	0	138,1435624	138,0743214	138,1089419
<i>Integral 71</i>	2598227,771	0,437558633	1	135,1815847	135,1200371	135,1508109
<i>Integral 72</i>	3456944,385	0,582172155	0	134,9430878	134,8507664	134,8969271
<i>Integral 73</i>	52109611,97	8,775601137	1	134,7892188	134,5814957	134,6853573
<i>Integral 74</i>	6666088,567	1,122613126	1	134,1737429	134,0660347	134,1198888
<i>Integral 75</i>	24929212,98	4,198243309	4	134,0121805	133,4120915	133,712136
<i>Integral 76</i>	10378271,83	1,747769186	2	133,3890112	133,0889667	133,2389889
<i>Integral 77</i>	43156261,08	7,267798004	2	133,112047	132,9581781	133,0351125
<i>Integral 78</i>	22061723,67	3,715339263	2	132,9274043	132,4888777	132,708141
<i>Integral 79</i>	19517288,43	3,286839645	2	132,4657974	132,0734315	132,2696144
<i>Integral 80</i>	18831610,39	3,171366958	2	132,0888184	131,7272263	131,9080223
<i>Integral 81</i>	2158184,9	0,363452521	0	131,7041459	131,5964377	131,6502918
<i>Integral 82</i>	4264749,674	0,718211875	1	131,5887442	131,5118097	131,550277
<i>Integral 83</i>	15636239,31	2,633245466	1	131,5195032	131,3848678	131,4521855
<i>Integral 84</i>	1549737,077	0,260985909	0	131,1194439	131,0271225	131,0732832
<i>Integral 85</i>	1810470,545	0,304895139	0	130,9848085	130,9271076	130,9559581
<i>Integral 86</i>	16867789,18	2,840646558	4	130,7232313	130,3539457	130,5385885
<i>Integral 87</i>	6843416,139	1,152476255	1	130,015434	129,8077109	129,9115724
<i>Integral 88</i>	11083412,5	1,866519509	2	129,7692436	129,6115279	129,6903858
<i>Integral 89</i>	27626929,46	4,652556495	1	129,4499655	129,3499507	129,3999581
<i>Integral 90</i>	11833798,62	1,992889463	2	129,1576145	129,0306726	129,0941435
<i>Integral 91</i>	1849262,97	0,311428038	1	128,7075477	128,6498469	128,6786973
<i>Integral 92</i>	6798124,26	1,144848805	1	128,592146	128,4536639	128,522905
<i>Integral 93</i>	9884181,113	1,664561062	1	128,3998098	128,1766998	128,2882548
<i>Integral 94</i>	5576296,436	0,939084969	1	128,1228456	128,0228308	128,0728382
<i>Integral 95</i>	86070608,54	14,49485616	1	128,0112906	127,9112758	127,9612832
<i>Integral 96</i>	1473434,171	0,248135998	1	127,7727937	127,7112461	127,7420199
<i>Integral 97</i>	1415661,31	0,238406669	1	127,3034934	127,253486	127,2784897

<i>Integral 98</i>	12960582,84	2,182647331	1	127,211172	127,1342375	127,1727047
<i>Integral 99</i>	4204645,243	0,708089893	1	127,1226973	127,057303	127,0900002
<i>Integral 100</i>	2700661,742	0,454809187	1	126,9649816	126,9111275	126,9380546
<i>Integral 101</i>	6133073,333	1,032849858	2	126,7110978	126,6149297	126,6630138
<i>Integral 102</i>	3776659,455	0,636014274	1	126,5264551	126,4687542	126,4976046
<i>Integral 103</i>	14095354,5	2,373750337	1	126,0840818	125,9840669	126,0340744
<i>Integral 104</i>	10417678,06	1,754405453	1	125,6917159	125,5032264	125,5974712
<i>Integral 105</i>	2290568,308	0,385746757	1	125,3147369	125,2685762	125,2916566
<i>Integral 106</i>	35048179,17	5,90234372	1	124,5800126	124,4723043	124,5261584
<i>Integral 107</i>	1428339,73	0,240541798	0	124,3569026	124,2915083	124,3242054
<i>Integral 108</i>	27176225,36	4,576654963	4	124,2568877	124,0914786	124,1741832
<i>Integral 109</i>	5374439,742	0,905090975	1	124,0683982	124,0145441	124,0414712
<i>Integral 110</i>	3546963,286	0,597331929	1	123,8145144	123,764507	123,7895107
<i>Integral 111</i>	12910618,67	2,174233036	1	123,7452734	123,5606306	123,652952
<i>Integral 112</i>	3506199,419	0,590467026	0	122,9028408	122,8489867	122,8759137
<i>Integral 113</i>	5364334,103	0,903389119	0	122,7989792	122,7297382	122,7643587
<i>Integral 114</i>	3926965,234	0,661326755	0	122,7105046	122,6258766	122,6681906
<i>Integral 115</i>	7143002,103	1,2029285	1	122,4220003	122,3527592	122,3873797
<i>Integral 116</i>	1983909,636	0,334103422	0	122,2412042	122,202737	122,2219706
<i>Integral 117</i>	9452479,268	1,591859634	1	121,5795676	121,5064799	121,5430238
<i>Integral 118</i>	1039165,622	0,17500232	0	121,2795231	121,2141288	121,246826
<i>Integral 119</i>	26881976,34	4,527101493	1	121,1333476	121,0487197	121,0910337
<i>Integral 120</i>	2429940,897	0,409218017	0	121,0217926	120,9717852	120,9967889
<i>Integral 121</i>	10801873,17	1,819106436	1	120,9563983	120,9063909	120,9313946
<i>Integral 122</i>	1917501,091	0,322919786	0	120,6794342	120,6409669	120,6602005
<i>Integral 123</i>	3676271,892	0,619108348	0	120,4178569	120,3524626	120,3851598
<i>Integral 124</i>	10808532,61	1,820227929	1	120,3216888	120,2332141	120,2774515
<i>Integral 125</i>	1565131,304	0,263578398	0	119,9524033	119,913936	119,9331696
<i>Integral 126</i>	3080139,478	0,518715732	0	119,7485269	119,6677457	119,7081363
<i>Integral 127</i>	2830312,387	0,476643208	0	119,6485121	119,5946579	119,621585
<i>Integral 128</i>	73686464,78	12,40928496	1	119,5850411	119,4946431	119,5398421
<i>Integral 129</i>	2280228,2	0,384005415	1	119,4869496	119,4523291	119,4696394
<i>Integral 130</i>	332757583,8	56,03856411	2	119,3907815	119,2426826	119,3167321
<i>Integral 131</i>	8101212,632	1,364297451	1	119,2388359	119,2061388	119,2224873
<i>Integral 132</i>	820089,8333	0,138108518	0	119,1926752	119,1715182	119,1820967
<i>Integral 133</i>	5522833,684	0,93008149	1	119,0868903	119,0541931	119,0705417
<i>Integral 134</i>	26573411,19	4,475137094	2	118,9695652	118,9137877	118,9416765
<i>Integral 135</i>	3340500	0,562562155	1	118,9060943	118,8830139	118,8945541
<i>Integral 136</i>	10547957,57	1,77634538	1	118,8541635	118,8041561	118,8291598
<i>Integral 137</i>	3843236,32	0,647226255	0	118,8022327	118,7579954	118,780114
<i>Integral 138</i>	3200387,778	0,538966336	0	118,7541487	118,7214515	118,7378001
<i>Integral 139</i>	1200754,533	0,202214955	0	118,7060646	118,6810609	118,6935627
<i>Integral 140</i>	26436744,65	4,452121551	1	118,6310535	118,554119	118,5925862
<i>Integral 141</i>	3508077,286	0,590783271	0	118,4887247	118,4406406	118,4646827

<i>Integral 142</i>	3704077,846	0,623791054	0	118,3983267	118,352166	118,3752463
<i>Integral 143</i>	40741936,65	6,861209902	1	118,3040819	118,2156073	118,2598446
<i>Integral 144</i>	793126,75	0,133567757	0	118,1540597	118,1271326	118,1405961
<i>Integral 145</i>	1109479,333	0,186843612	0	118,0944354	118,0694317	118,0819336
<i>Integral 146</i>	5652349,636	0,951892828	1	118,0636616	118,0251944	118,044428
<i>Integral 147</i>	3621107,818	0,609818356	0	117,5135801	117,4751128	117,4943465
<i>Integral 148</i>	10571897,05	1,780376946	1	117,2520028	117,1846851	117,218344
<i>Integral 149</i>	1775714,941	0,299042067	0	116,8634837	116,8346332	116,8490585
<i>Integral 150</i>	3608924,957	0,607766682	0	116,7653922	116,7250016	116,7451969
<i>Integral 151</i>	80784503,05	13,60464126	1	116,5153551	116,4018768	116,4586159
<i>Integral 152</i>	16572370,67	2,79089614	2	116,0883687	116,0248978	116,0566333
<i>Integral 153</i>	386491060,4	65,08763473	2	115,8267915	115,6786926	115,752742
<i>Integral 154</i>	1118780,941	0,188410064	1	115,6652291	115,6363786	115,6508038
<i>Integral 155</i>	8440149,333	1,421376619	1	115,6344553	115,6036815	115,6190684
<i>Integral 156</i>	1645910,571	0,277182158	1	115,5921413	115,569061	115,5806011
<i>Integral 157</i>	132282620,9	22,27726276	1	115,5498273	115,4498125	115,4998199
<i>Integral 158</i>	23345383,63	3,931516036	1	115,4498125	115,394035	115,4219237
<i>Integral 159</i>	5703663,75	0,960534462	1	115,3709546	115,3151771	115,3430659
<i>Integral 160</i>	21630563,28	3,64272902	1	115,2997903	115,2555529	115,2776716
<i>Integral 161</i>	38408127,82	6,468181156	1	115,2324726	115,1747717	115,2036221
<i>Integral 162</i>	11676710,53	1,966434797	1	115,1016839	115,0497532	115,0757186
<i>Integral 163</i>	777925,4545	0,131007759	0	115,0497532	115,0305196	115,0401364
<i>Integral 164</i>	3392351,5	0,571294288	0	115,0228261	114,9728187	114,9978224
<i>Integral 165</i>	3979457,886	0,670166862	0	114,6266135	114,5650659	114,5958397
<i>Integral 166</i>	862936,3158	0,145324147	0	114,4862081	114,4535109	114,4698595
<i>Integral 167</i>	3759701,714	0,633158479	0	114,3727297	114,3361858	114,3544578
<i>Integral 168</i>	1674720,5	0,282033939	0	113,8438051	113,816878	113,8303416
<i>Integral 169</i>	7279953,724	1,225992054	1	113,7649473	113,7149398	113,7399436
<i>Integral 170</i>	2646591,765	0,445703447	0	113,1456247	113,1148509	113,1302378
<i>Integral 171</i>	29085094,52	4,898121074	1	113,0879238	113,0340696	113,0609967
<i>Integral 172</i>	4047141,474	0,681565223	0	113,0340696	113,0013725	113,0177211
<i>Integral 173</i>	7148357,083	1,203830314	0	112,5089918	112,4666778	112,4878348
<i>Integral 174</i>	10683375,18	1,799150595	1	112,3724331	112,3128088	112,342621
<i>Integral 175</i>	5015486,182	0,844640837	0	112,0339213	111,9954541	112,0146877
<i>Integral 176</i>	1653900	0,27852763	0	111,591548	111,5607742	111,5761611
<i>Integral 177</i>	8527890,6	1,436152825	1	111,410752	111,3395876	111,3751698
<i>Integral 178</i>	2354065,789	0,396440151	0	109,5143169	109,4816198	109,4979684
<i>Integral 179</i>	1796972,286	0,302621943	0	107,023563	106,9985593	107,0110611
<i>Integral 180</i>	17968310	3,025981498	1	104,309699	104,2500748	104,2798869
<i>Integral 181</i>	18665724,04	3,143430605	1	103,3037806	103,2018424	103,2528115
<i>Integral 182</i>	1619746,737	0,272775996	0	103,2018424	103,1691452	103,1854938
<i>Integral 183</i>	11907337,96	2,005273971	0	103,1691452	103,0902874	103,1297163
<i>Integral 184</i>	8579421,613	1,444830987	0	102,9364184	102,8825643	102,9094913
<i>Integral 185</i>	1445145,067	0,243371927	0	101,9266533	101,8997262	101,9131898

<i>Integral 186</i>	6887780,952	1,159947581	0	98,97621576	98,93967188	98,95794382
<i>Integral 187</i>	20597783,56	3,468802128	1	97,28173372	97,22595621	97,25384496
<i>Integral 188</i>	2036002,444	0,342876193	0	74,61106416	74,58029037	74,59567727
<i>Integral 189</i>	36045072,76	6,070227154	1	74,29370941	74,21869829	74,25620385
<i>Integral 190</i>	9874641,829	1,662954583	0	72,49536581	72,43381822	72,46459201
<i>Integral 191</i>	35855162,38	6,038244997	1	68,86213472	68,78904696	68,82559084
<i>Integral 192</i>	88139616,03	14,84329062	1	68,34475031	68,23511867	68,28993449
<i>Integral 193</i>	12088798,23	2,035833073	0	67,38306924	67,30421139	67,34364031
<i>Integral 194</i>	5867407,1	0,988109918	0	67,26766751	67,19650311	67,23208531
<i>Integral 195</i>	3172626,583	0,534291168	0	67,08879483	67,04840423	67,06859953
<i>Integral 196</i>	2757269,846	0,464342364	0	66,90607543	66,85991474	66,88299508
<i>Integral 197</i>	2020170,333	0,340209961	0	66,64065145	66,60026085	66,62045615
<i>Integral 198</i>	2134080,783	0,35939323	0	66,44062179	66,40023118	66,42042649
<i>Integral 199</i>	1026280,375	0,17283236	0	66,3790742	66,35214713	66,36561067
<i>Integral 200</i>	6447637,091	1,085824462	0	66,34637704	66,3079098	66,32714342
<i>Integral 201</i>	21330885,93	3,592261384	1	66,21558842	66,16750437	66,19154639
<i>Integral 202</i>	1964500,769	0,330834841	0	66,165581	66,14442402	66,15500251
<i>Integral 203</i>	9705342,64	1,634443487	0	66,14250066	66,09826333	66,12038199
<i>Integral 204</i>	754028,4	0,126983338	0	66,10018669	66,08479979	66,09249324
<i>Integral 205</i>	1303482,364	0,219514997	0	66,08672316	66,04825591	66,06748953
<i>Integral 206</i>	27180738,55	4,577415012	1	65,82899263	65,75013478	65,7895637
<i>Integral 207</i>	3373215,333	0,568071632	0	65,6270396	65,57510883	65,60107421
<i>Integral 208</i>	1168874,857	0,196846208	0	65,53279486	65,47124727	65,50202106
<i>Integral 209</i>	1865620,182	0,3141827	0	63,70944755	63,65174669	63,68059712
<i>Integral 210</i>	1395855,333	0,235071212	0	63,57096548	63,50749452	63,53923
<i>Integral 211</i>	8097971,072	1,363751551	0	62,82470096	62,70160579	62,76315338
<i>Integral 212</i>	3929272,75	0,661715356	0	60,88787529	60,81671089	60,85229309
<i>Integral 213</i>	4875150,5	0,821007385	0	60,79363054	60,70900261	60,75131658
<i>Integral 214</i>	7454871,5	1,255449357	0	60,32625354	60,2473957	60,28682462
<i>Integral 215</i>	2162499,167	0,364179072	0	60,09545009	60,05313612	60,0742931
<i>Integral 216</i>	1055595,1	0,177769153	0	60,03197914	59,99735862	60,01466888
<i>Integral 217</i>	112231160,8	18,9004651	1	59,85887654	59,74732154	59,80309904
<i>Integral 218</i>	2385546	0,401741625	0	59,74539818	59,6934674	59,71943279
<i>Integral 219</i>	33663168,5	5,669098821	1	58,3625008	58,28364295	58,32307187
<i>Integral 220</i>	1996266,923	0,33618447	0	58,12208053	58,07591984	58,09900019
<i>Integral 221</i>	1416019,2	0,23846694	0	57,96821156	57,91628078	57,94224617
<i>Integral 222</i>	5910589,484	0,995382116	0	57,19117325	57,13539575	57,1632845
<i>Integral 223</i>	11984256,88	2,018227624	0	55,7563451	55,66210035	55,70922273
<i>Integral 224</i>	18942024,9	3,189961486	1	55,56208552	55,51207811	55,53708181
<i>Integral 225</i>	2813520,5	0,473815344	0	55,51207811	55,48515104	55,49861457
<i>Integral 226</i>	996535,3333	0,1678231	0	55,39667638	55,36590258	55,38128948
<i>Integral 227</i>	17870963,1	3,00958764	1	53,86760348	53,81374934	53,84067641
<i>Integral 228</i>	4614567,459	0,777123489	0	49,42656029	49,36116598	49,39386314
<i>Integral 229</i>	3508846,732	0,590912851	0	49,0553514	48,98226364	49,01880752

<i>Integral 230</i>	5425573,143	0,913702176	0	48,70145277	48,60143793	48,65144535
<i>Integral 231</i>	4614363,907	0,777089209	0	46,42803872	46,35110423	46,38957147
<i>Integral 232</i>	10231987,23	1,723133897	0	46,29340337	46,19146517	46,24243427
<i>Integral 233</i>	4910490,37	0,826958852	0	44,76048374	44,71432305	44,73740339
<i>Integral 234</i>	722355,7778	0,121649461	0	43,1467829	43,11600911	43,13139601
<i>Integral 235</i>	4272597,077	0,719533429	0	41,37151965	41,32728232	41,34940098
<i>Integral 236</i>	5201692,87	0,875999267	0	37,87100055	37,83060995	37,85080525
<i>Integral 237</i>	5395208,75	0,908588613	0	37,82676322	37,78444925	37,80560624
<i>Integral 238</i>	2661956,842	0,448291028	0	37,02279785	36,95548017	36,98913901
<i>Integral 239</i>	1477252,643	0,248779054	0	36,8573887	36,80930465	36,83334668
<i>Integral 240</i>	25285823,89	4,258298931	1	36,78814767	36,72082999	36,75448883
<i>Integral 241</i>	5421711,643	0,913051874	0	36,45155929	36,40347524	36,42751726
<i>Integral 242</i>	6206479,463	1,045211933	0	36,35923791	36,28615015	36,32269403
<i>Integral 243</i>	3183612,72	0,536141306	0	35,42256055	35,38024658	35,40140356
<i>Integral 244</i>	1845367,077	0,310771944	0	35,12828614	35,08212545	35,10520579
<i>Integral 245</i>	1450623,467	0,244294526	0	35,07827872	35,02634794	35,05231333
<i>Integral 246</i>	2369706	0,399074065	0	34,88786587	34,8455519	34,86670889
<i>Integral 247</i>	9540076,154	1,606611525	0	34,72053336	34,62821198	34,67437267
<i>Integral 248</i>	1497668,417	0,252217205	0	34,42241223	34,38009827	34,40125525
<i>Integral 249</i>	2188258,267	0,368517074	0	33,77423919	33,72038505	33,74731212
<i>Integral 250</i>	901318,1818	0,151787906	0	33,49150496	33,47227134	33,48188815
<i>Integral 251</i>	21321721,56	3,590718044	1	33,23954452	33,17607357	33,20780904
<i>Integral 252</i>	7611624,375	1,281847571	0	32,95104019	32,89333933	32,92218976
<i>Integral 253</i>	750704	0,126423487	0	32,82217493	32,79140114	32,80678803
<i>Integral 254</i>	1468862,3	0,247366065	0	32,77024415	32,73562364	32,75293389
<i>Integral 255</i>	912211,8333	0,153622468	0	32,51059026	32,4682763	32,48943328
<i>Integral 256</i>	34558621,02	5,81989891	2	32,17977198	32,07398706	32,12687952
<i>Integral 257</i>	8848088,923	1,490076327	1	32,04898335	32,00282266	32,02590301
<i>Integral 258</i>	5637103,412	0,949325264	1	31,94896852	31,92011809	31,9345433
<i>Integral 259</i>	18576174,6	3,128349892	1	31,91627136	31,83933688	31,87780412
<i>Integral 260</i>	28096698,63	4,731668711	2	31,82394998	31,75663231	31,79029114
<i>Integral 261</i>	4067725,6	0,685031726	0	31,51044195	31,45658781	31,48351488
<i>Integral 262</i>	16271068,58	2,740154888	0	31,38734677	31,2642516	31,32579919
<i>Integral 263</i>	2898170,609	0,488070978	0	31,23732453	31,19693392	31,21712923
<i>Integral 264</i>	5677689,826	0,956160282	0	31,17193021	31,13153961	31,15173491
<i>Integral 265</i>	6756347,692	1,137813356	0	31,08153219	31,03729486	31,05941353
<i>Integral 266</i>	2482190,267	0,418017155	0	30,6699327	30,64492899	30,65743084
<i>Integral 267</i>	2003186,889	0,337349838	0	30,54106743	30,51029364	30,52568054
<i>Integral 268</i>	2523554,5	0,424983163	0	30,23332949	30,17755199	30,20544074
<i>Integral 269</i>	1201182,8	0,202287078	0	29,66016757	29,61593024	29,63804891
<i>Integral 270</i>	5001178,615	0,842231348	0	29,59092654	29,54476584	29,56784619
<i>Integral 271</i>	6431531,25	1,083112133	0	29,46975472	29,38512679	29,42744075
<i>Integral 272</i>	8993894,541	1,514630952	0	29,11777945	29,05238514	29,08508229
<i>Integral 273</i>	4033210,889	0,679219221	0	28,96391048	28,91774979	28,94083013

<i>Integral 274</i>	1563566,435	0,263314864	0	28,85235547	28,81388823	28,83312185
<i>Integral 275</i>	1131622,545	0,190572674	0	28,79273125	28,77542099	28,78407612
<i>Integral 276</i>	5176740,533	0,871797129	0	28,59270158	28,54077081	28,5667362
<i>Integral 277</i>	1555951,636	0,262032482	0	28,44652606	28,40805882	28,42729244
<i>Integral 278</i>	3373725,882	0,568157612	0	28,2311095	28,16956191	28,20033571
<i>Integral 279</i>	3334066,571	0,561478723	0	27,85220716	27,81566328	27,83393522
<i>Integral 280</i>	47448355,47	7,990614908	1	27,66179431	27,56754956	27,61467194
<i>Integral 281</i>	86416551,39	14,55311521	1	27,46368801	27,40021706	27,43195253
<i>Integral 282</i>	1523408,516	0,256552006	0	26,88283264	26,8289785	26,85590557
<i>Integral 283</i>	5059014,476	0,851971287	0	26,51739383	26,48084995	26,49912189
<i>Integral 284</i>	2488619,846	0,419099938	0	26,20580917	26,15964848	26,18272882
<i>Integral 285</i>	7355649,042	1,238739642	0	25,49608854	25,40953724	25,45281289
<i>Integral 286</i>	4893937,302	0,824171206	0	25,23258792	25,1575768	25,19508236
<i>Integral 287</i>	3772780,767	0,635361077	0	25,09987594	24,99216766	25,0460218
<i>Integral 288</i>	7989318,667	1,345453772	0	24,81329498	24,74405394	24,77867446
<i>Integral 289</i>	1904049,375	0,320654429	0	24,44977953	24,42285246	24,436316
<i>Integral 290</i>	24795794,87	4,175774823	1	24,39400203	24,35361143	24,37380673
<i>Integral 291</i>	774156	0,130372958	0	24,35361143	24,33822453	24,34591798
<i>Integral 292</i>	3932610,6	0,662277472	0	24,25551996	24,22089944	24,2382097
<i>Integral 293</i>	3282170,667	0,552739112	0	23,95162874	23,92085494	23,93624184
<i>Integral 294</i>	4397139,059	0,740507117	0	23,92085494	23,89200451	23,90642973
<i>Integral 295</i>	2436471,538	0,41031782	0	23,43809105	23,4150107	23,42655087
<i>Integral 296</i>	22832957,31	3,845220078	1	23,37846682	23,31499587	23,34673135
<i>Integral 297</i>	1463136	0,246401719	0	23,25921837	23,22844457	23,24383147
<i>Integral 298</i>	4520166,706	0,761225781	1	23,22459785	23,19574742	23,21017263
<i>Integral 299</i>	2086762,4	0,351424503	1	23,19574742	23,17843716	23,18709229
<i>Integral 300</i>	1439956,923	0,24249821	0	23,17074371	23,14766336	23,15920354
<i>Integral 301</i>	435926	0,073412803	0	23,14189328	23,12650638	23,13419983
<i>Integral 302</i>	19845350,22	3,342087408	1	23,13227647	23,10150267	23,11688957
<i>Integral 303</i>	992691,3333	0,167175745	0	23,10150267	23,08034569	23,09092418
<i>Integral 304</i>	7207760	1,213834156	1	23,04187845	23,00918129	23,02552987
<i>Integral 305</i>	15592142,9	2,625819342	1	23,00672819	22,95219848	22,97946334
<i>Integral 306</i>	3416988,8	0,575443372	0	22,90729165	22,87361154	22,8904516
<i>Integral 307</i>	3731782,615	0,628456719	0	22,86398865	22,84153523	22,85276194
<i>Integral 308</i>	6631415,492	1,116773953	0	22,24972745	22,14547946	22,19760345
<i>Integral 309</i>	84285286,29	14,19419616	1	21,80386684	21,73009134	21,76697909
<i>Integral 310</i>	4659271,111	0,784651877	0	21,19441708	21,14790645	21,17116176
<i>Integral 311</i>	32732980,5	5,512449049	1	21,02441268	20,96827915	20,99634591
<i>Integral 312</i>	2361448,5	0,397683448	0	20,85921972	20,83195486	20,84558729
<i>Integral 313</i>	99379888,16	16,73622633	1	20,78223659	20,71648017	20,74935838
<i>Integral 314</i>	2967318,706	0,499715972	0	20,60742074	20,57855207	20,59298641
<i>Integral 315</i>	4725418,526	0,795791537	0	20,54807958	20,51600328	20,53204143
<i>Integral 316</i>	1623461,273	0,273401548	0	20,47270027	20,4550583	20,46387929
<i>Integral 317</i>	11732730,3	1,975868894	1	20,43581252	20,40052859	20,41817056

<i>Integral 318</i>	90627434,65	15,26225563	1	20,39892477	20,34439506	20,37165992
<i>Integral 319</i>	4502952,4	0,758326779	0	20,12948383	20,08618082	20,10783233
<i>Integral 320</i>	3927691,833	0,661449119	0	19,29389614	19,25219695	19,27304655
<i>Integral 321</i>	3508175,926	0,590799883	0	19,04690861	19,00039797	19,02365329
<i>Integral 322</i>	15908406,7	2,679080243	1	18,63152049	18,59142511	18,6114728
<i>Integral 323</i>	6173548,5	1,039666142	0	18,02207074	17,95791813	17,98999443
<i>Integral 324</i>	29633187,11	4,990423469	1	16,43589756	16,40542507	16,42066132
<i>Integral 325</i>	3093444,588	0,5209564	0	16,38938692	16,36051825	16,37495259
<i>Integral 326</i>	13575346,32	2,286177541	1	16,12315361	16,08947349	16,10631355
<i>Integral 327</i>	23548629,45	3,96574397	1	16,05739719	16,01890563	16,03815141
<i>Integral 328</i>	5346781,111	0,90043308	1	16,02211326	16,00767892	16,01489609
<i>Integral 329</i>	6871402,167	1,157189286	1	16,0060751	15,98682932	15,99645221
<i>Integral 330</i>	43392332,93	7,307554054	1	15,98362169	15,93390342	15,95876256
<i>Integral 331</i>	2149508,267	0,361991319	0	15,92748816	15,90182712	15,91465764
<i>Integral 332</i>	371097,4286	0,062495246	0	15,85852411	15,84729741	15,85291076
<i>Integral 333</i>	1668083,5	0,280916225	0	15,84408977	15,82484399	15,83446688
<i>Integral 334</i>	1646195,091	0,277230073	0	15,82484399	15,80559821	15,8152211
<i>Integral 335</i>	1723751,333	0,290291054	0	14,87698924	14,85774346	14,86736635
<i>Integral 336</i>	3315393,467	0,55833405	0	14,50490413	14,47924309	14,49207361
<i>Integral 337</i>	8177634,727	1,377167434	1	14,41990193	14,38301418	14,40145806
<i>Integral 338</i>	92018925,64	15,49659186	1	14,11196942	14,07508167	14,09352555
<i>Integral 339</i>	647632	0,109065485	0	13,69978893	13,67412789	13,68695841
<i>Integral 340</i>	18958766,74	3,192780922	1	13,67092026	13,63724014	13,6540802
<i>Integral 341</i>	5613324,783	0,945320786	0	11,99814107	11,95804569	11,97809338

References

1. Schobert, H., *Chemistry of fossil fuels and biofuels*. Cambridge Series in Chemical Engineering. 2013, Cambridge: Cambridge University Press.
2. Crocker, M. and C. Royal Society of, *Thermochemical conversion of biomass to liquid fuels and chemicals*. RSC energy and environment series. Vol. No 1. 2010, Cambridge: RSC Publishing.
3. Ragauskas, A., et al., *The path forward for biofuels and biomaterials*. Science, 2006. **311**(5760): p. 484-489.
4. Petrou, E.C. and C.P. Pappis, *Biofuels: A Survey on Pros and Cons*. Energy & Fuels, 2009. **23**(2): p. 1055-1066.
5. McKendry, P., *Energy production from biomass (part 1): overview of biomass*, in *Bioresour. Technol.* 2002. p. 37-46.
6. Yunpu, W., et al., *Review of microwave-assisted lignin conversion for renewable fuels and chemicals*. Journal of Analytical and Applied Pyrolysis, 2016. **119**: p. 104-113.
7. Collard, F.-X. and J. Blin, *A review on pyrolysis of biomass constituents: Mechanisms and composition of the products obtained from the conversion of cellulose, hemicelluloses and lignin*. Renewable and Sustainable Energy Reviews, 2014. **38**: p. 594.
8. Wang, Y., et al., *A supramolecular structure insight for conversion property of cellulose in hot compressed water: Polymorphs and hydrogen bonds changes*. Carbohydrate Polymers, 2015. **133**: p. 94-103.

9. Effendi, A., H. Gerhauser, and A.V. Bridgwater, *Production of renewable phenolic resins by thermochemical conversion of biomass: A review*. Renewable and Sustainable Energy Reviews, 2008. **12**(8): p. 2092-2116.
10. Kai, D., et al., *Towards lignin-based functional materials in a sustainable world*. Green Chem., 2016. **18**(5): p. 1175-1200.
11. Ragauskas, A., et al., *Lignin Valorization: Improving Lignin Processing in the Biorefinery*. Science, 2014. **344**(6185): p. 709-+.
12. Azadi, P., et al., *Liquid fuels, hydrogen and chemicals from lignin: A critical review*. Liquid fuels, hydrogen and chemicals from lignin: A critical review, 2013. **21**: p. 506-523.
13. Mohan, D., C.U. Pittman, and P. Steele, *Pyrolysis of wood/biomass for bio-oil: A critical review*. Energy Fuels, 2006. **20**(3): p. 848-889.
14. Thakur, V., et al., *Progress in Green Polymer Composites from Lignin for Multifunctional Applications: A Review*. ACS Sustain. Chem. Eng., 2014. **2**(5): p. 1072-1092.
15. El Hage, R., et al., *Characterization of milled wood lignin and ethanol organosolv lignin from miscanthus*. Polymer Degradation and Stability, 2009. **94**(10): p. 1632-1638.
16. Dutta, S., K.C.W. Wu, and B. Saha, *Emerging strategies for breaking the 3D amorphous network of lignin*. Catal. Sci. Technol., 2014. **4**(11): p. 3785-3799.
17. Van Parijs, F., et al., *Modeling lignin polymerization, I: simulation model of*

- dehydrogenation polymers*. PLANT PHYSIOLOGY, 2010. **153**(3): p. 110.154468.
18. Vanholme, R., et al., *Lignin biosynthesis and structure*. Plant physiology, 2010. **153**(3): p. 895.
 19. Bonawitz, N.D. and C. Chapple, *The genetics of lignin biosynthesis: connecting genotype to phenotype*. Annual review of genetics, 2010. **44**(1): p. 337.
 20. Wertz, J.-L. and O. Bédué, *Lignocellulosic biorefineries*. 2013, Lausanne: EPFL.
 21. Brennan, L. and P. Owende, *Biofuels from microalgae -- A review of technologies for production, processing, and extractions of biofuels and co-products.(Report)*. Renewable and Sustainable Energy Reviews, 2010. **14**(2): p. 557.
 22. Gasson, J.R., *Solvolytic lignin degradation in an alcohol / formic acid medium - Analysis of the reaction system and products by means of kinetic modelling and chemometrics*. 2012, The University of Bergen.
 23. Yu, J. and P. Savage, *Decomposition of formic acid under hydrothermal conditions*. Ind. Eng. Chem. Res., 1998. **37**(1): p. 2-10.
 24. Løhre, C., T. Barth, and M. Kleinert, *The effect of solvent and input material pretreatment on product yield and composition of bio-oils from lignin solvolysis*. Journal of Analytical and Applied Pyrolysis, 2016. **119**: p. 208-216.

25. Oregui Bengoetxea, M., *Formic acid aided catalytic lignin conversion in ethanol and water media*. 2016, The University of Bergen.
26. Kanaujia, P.K., et al., *Review of analytical strategies in the production and upgrading of bio-oils derived from lignocellulosic biomass*. Journal of Analytical and Applied Pyrolysis, 2014. **105**: p. 55-74.
27. Mu, W., et al., *Lignin Pyrolysis Components and Upgrading—Technology Review*. BioEnergy Research, 2013. **6**(4): p. 1183-1204.
28. Løhre, C., H.V. Halleraker, and T. Barth, *Composition of Lignin-to-Liquid Solvolysis Oils from Lignin Extracted in a Semi-Continuous Organosolv Process*. 2017. **18**(1).
29. Hao, N., et al., *Review of NMR characterization of pyrolysis oils*. Energy and Fuels, 2016. **30**(9).
30. Vaz Jr, S., *Analytical techniques for the chemical analysis of plant biomass and biomass products*. Anal. Methods, 2014. **6**(20): p. 8094-8105.
31. Alwehaibi, A.S., D.J. Macquarrie, and M.S. Stark, *Effect of spruce-derived phenolics extracted using microwave enhanced pyrolysis on the oxidative stability of biodiesel*. Green Chem., 2016. **18**(9): p. 2762-2774.
32. Ben, H. and A.J. Ragauskas, *NMR Characterization of Pyrolysis Oils from Kraft Lignin*. Energy & Fuels, 2011. **25**(5): p. 2322-2332.
33. Neuhaus, D. and M.P. Williamson, *The nuclear Overhauser effect in structural and conformational analysis*. 1989, New York: VCH Publishers.
34. Reich, H.J., *Structure Determination Using NMR*.

<https://www2.chem.wisc.edu/areas/reich/Check>.

35. Friebolin, H., *Basic one- and two-dimensional NMR spectroscopy*. 5th completely rev. and enl. ed. ed. Ein- und zweidimensionale NMR-Spektroskopie. 2011, Weinheim: Wiley-VCH.
36. Keeler, J., *Understanding NMR spectroscopy*. 2nd ed. ed. 2010, Chichester: Wiley.
37. Løhre, C., *The effect of input material pretreatment on product yield and composition of bio-oils from LtL-solvolytic : a continuous process for organosolv fractionation of lignocellulosic biomass and solvolytic conversion of lignin*, B. Universitetet i, Editor. 2017, University of Bergen: Bergen.
38. Løhre, C., M. Kleinert, and T. Barth, *Organosolv extraction of softwood combined with lignin-to-liquid-solvolytic as a semi-continuous percolation reactor*. *Biomass and Bioenergy*, 2017. **99**: p. 147-155.
39. *Spectral Database for Organic Compounds SDBS*.
http://sdb.sdb.aist.go.jp/sdb/cgi-bin/direct_frame_top.cgi.
40. Mullen, C.A., G.D. Strahan, and A.A. Boateng, *Characterization of Various Fast-Pyrolysis Bio-Oils by NMR Spectroscopy*. *Energy & Fuels*, 2009. **23**(5): p. 2707-2718.