

Reconstructing the time history of the air-sea CO₂ disequilibrium and its rate of change in the eastern subpolar North Atlantic, 1972–1989

Abdirahman M. Omar^{1,2} and Are Olsen^{1,2}

Received 8 December 2005; revised 9 January 2006; accepted 13 January 2006; published 18 February 2006.

[1] This study determines the temporal changes of wintertime surface ocean partial pressure of CO₂ (pCO₂^{SW}) in the eastern subpolar North Atlantic (esNA) (50–64°N; 32–10°W) by using data of carbon-system parameters and chlorofluorocarbon-12 acquired in 1993. Wintertime pCO₂^{SW} and its temporal trend from early 1970s through to the late 1980s were reconstructed through the application of a back-calculation method that isolates surface variations which have been transmitted to the ocean interior during the formation of Subpolar Mode Water. Our computations suggest a pCO₂^{SW} growth rate (3 μatm/yr) which is twice as large as that of atmospheric pCO₂, 1.47 μatm/yr. The sensitivity of the estimated pCO₂^{SW} growth rate to remineralization ratios as well as to the CFC-12-derived ages is discussed. Cooling and northward advection of surface water equilibrated with the increasing atmospheric CO₂ is suggested as the process responsible for the excessive pCO₂^{SW} growth rate. **Citation:** Omar, A. M., and A. Olsen (2006), Reconstructing the time history of the air-sea CO₂ disequilibrium and its rate of change in the eastern subpolar North Atlantic, 1972–1989, *Geophys. Res. Lett.*, 33, L04602, doi:10.1029/2005GL025425.

1. Introduction

[2] It is generally accepted that the North Atlantic (NA) is an important sink region for atmospheric carbon dioxide (CO₂). Assuming a constant wind field, the strength of the sink is proportional to the CO₂ disequilibrium across the air-sea interface $\Delta p\text{CO}_2 = p\text{CO}_2^{\text{atm}} - p\text{CO}_2^{\text{SW}}$, i.e., the difference between sea surface and atmospheric CO₂ partial pressures, pCO₂^{SW} and pCO₂^{atm}, respectively. In the subpolar NA, extensive heat loss and primary production lower pCO₂^{SW} so that a disequilibrium exists throughout most of the year and the flux of CO₂ is directed into the ocean. Additionally the pCO₂^{atm} is currently increasing so that an increasing air-sea CO₂ disequilibrium is expected, and this is often indicated by ocean carbon models [e.g., *Wetzel et al.*, 2005], and has been an underlying assumption employed when compiling the global air-sea CO₂ flux climatology [*Takahashi et al.*, 2002]. However, the contrary has been suggested in a number of recent studies. *Anderson and Olsen* [2002] employed a conceptual advective model to calculate the change of the air-sea CO₂ flux in the NA since pre-industrial times. For latitudes north of ~50°N, their computation suggested a slightly decreasing $\Delta p\text{CO}_2$.

Lefèvre et al. [2004] analyzed pCO₂^{SW} data acquired in the region and showed that the disequilibrium decreased at an annual mean rate of $-0.3 \mu\text{atm yr}^{-1}$ over the period 1982–1998. And *Friis et al.* [2005] provided measurement-based estimates of anthropogenic carbon (C_{ant}) accumulated in the subpolar NA over the period 1981–1999. For the surface water, they found that the increase of the C_{ant} was only reconcilable with a decreasing $\Delta p\text{CO}_2$.

[3] In this paper, we reconstruct the time history of wintertime $\Delta p\text{CO}_2$ in the eastern subpolar NA (50–64°N; 32–10°W) (Figure 1) over the time period 1972–1989.

[4] Our approach takes advantage by the fact that mode water characteristics in the interior of the ocean reflect variations of wintertime sea surface properties in the formation region [e.g., *Hanawa and Talley*, 2001]. The winter mixed layer in the eastern subpolar NA (esNA) is dominated by a type of Subpolar Mode Water (SPMW) and during summer this is isolated from the surface by a seasonal pycnocline [*McCartney and Talley*, 1982]. We assume that climatological wintertime (January–March) surface seawater properties can be used to define the SPMW that forms in the esNA during winter. This water type is henceforth referred to as eastern Subpolar Mode Water (eSPMW) in order to differentiate it from the broader term SPMW.

[5] By using data acquired during the third leg of the 1993 Ocean Atmosphere Carbon Exchange Study (OACES) cruise of the R/V *Malcolm Baldrige* (18–29 August 1993) (Figure 1) we identify eSPMW that formed in the esNA at times determined from their apparent CFC-12 ages. We then extend the back-calculation approach of *Brewer* [1978] to estimate preformed (i.e., wintertime) pCO₂^{SW} values for the eSPMW encountered during the cruise. The results are combined with data for the atmospheric mole fraction of CO₂ (xCO₂) to compute $\Delta p\text{CO}_2$ and the temporal variability is analyzed.

2. Data

[6] The *Baldrige* data have been described in detail by others [*Körtzinger et al.*, 2003, and references therein] and were made available by the Global Ocean Data Analysis Project (GLODAP) [*Key et al.*, 2004]. Data for salinity (S), temperature (T), total dissolved inorganic carbon (C_T), total alkalinity (A_T), phosphate (P), silicate (Si), oxygen (O₂) and CFC-12 ages (see *Key et al.* [2004] for the computation of these ages) were retrieved from the database (<http://cdiac.esd.ornl.gov/oceans/home.html>, accessed on March 4, 2005).

[7] The accuracy of the C_T, A_T, and O₂ data have been determined to 1.5, 2.5, and 1 μmol kg⁻¹, respectively [*Körtzinger et al.*, 2003, and references therein]. A value of 7.5 μmol kg⁻¹ were subtracted from the O₂ data because

¹Bjerknes Centre for Climate Research, University of Bergen, Bergen, Norway.

²Geophysical Institute, University of Bergen, Bergen, Norway.

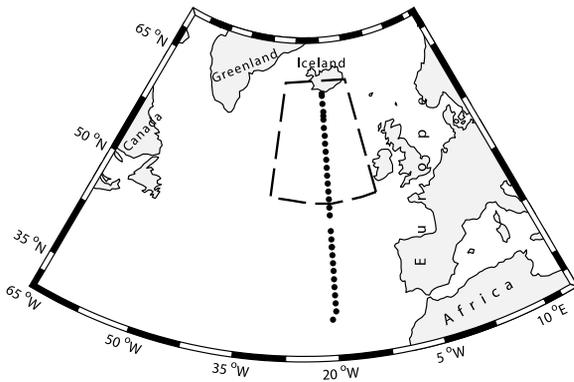


Figure 1. Map of the northern North Atlantic. The box show the eastern subpolar North Atlantic and the dots shown the sampling stations during the third leg of *Malcolm Baldrige* cruise (leg 2B, 18–29 August 1993).

these data have been double corrected (first after *Wanninkhof et al.* [1999] and then after *Gouretski and Jancke* [2001]) (R. Key, personal communication, 2005). After this re-correction, apparent oxygen utilization (AOU = O₂ saturation – O₂ measured) was recomputed.

3. Methods

[8] Wintertime climatological ranges of surface density (σ), salinity, and temperature for the esNA were determined using data from the World Ocean Atlas 2001 [*Conkright et al.*, 2002]. The resulting ranges (Figure 2) of temperature (5.5–11.2°C), salinity (34.93–35.49), and density (27.06–27.65 kg m⁻³) agree well with those reported by *McCartney and Talley* [1982], for the winter mixed layer of the esNA and were used to define eSPMW. All water samples with properties falling within the climatological ranges were identified in the cruise data and assumed to be remnants of eSPMW that formed in the esNA at times given by their apparent CFC-12 ages.

[9] The preformed pCO₂^{SW} values for the eSPMW were determined from preformed A_T (A_T⁰) and C_T (C_T⁰). The former was computed from the empirical relationship:

$$A_T^0 = 49.87 \times S + 566 \quad (1)$$

which was determined by regressing surface (depth < 200 m) salinity with alkalinity. The coefficient of determination (r^2) was 0.79 and the standard error of estimate was $\pm 3 \mu\text{mol kg}^{-1}$. C_T⁰ was computed according to *Körtzinger et al.* [2003]:

$$C_T^0 = C_T - R_{C:O} \times \text{AOU} - \frac{1}{2} (A_T - A_T^0 + R_{N:O} \times \text{AOU}) \quad (2)$$

where C_T and A_T are the measured concentrations, R_{C:O} and R_{N:O} are the remineralization ratios between C_T and O₂, and between NO₃ and O₂. We used the R_{C:O} and R_{N:O} values which were estimated by *Körtzinger et al.* [2001] for the North Atlantic. Alternative values will affect our results as discussed in section 5.

[10] Preformed pCO₂^{SW} was then computed from C_T⁰, A_T⁰, P, Si, S, and potential temperature (θ), using the constants of

Merbach et al. [1973] refitted by *Dickson and Millero* [1987]. These constants have been recommended by *Lueker et al.* [2000] for the computation of pCO₂^{SW} from C_T and A_T. By propagating the uncertainties in C_T⁰ (= 3.3 $\mu\text{mol kg}^{-1}$) and A_T⁰ (= $\pm 3 \mu\text{mol kg}^{-1}$) the maximum uncertainty in the computed pCO₂^{SW} values was obtained to be $\pm 7 \mu\text{atm}$ (or 2 %).

[11] The *Baldrige* cruise was carried out during August and, in order to limit seasonal biasing effects, only data from deeper than 200 m were used for the temporal analysis.

[12] Values of pCO₂^{atm} were calculated from xCO₂ data measured at Niwot Ridge, Colorado, USA which have been made available by the Carbon Cycle Greenhouse Gases Group at the National Oceanic and Atmospheric Administration (NOAA), USA. Monthly xCO₂ values were retrieved (<ftp://ftp.cmdl.noaa.gov/ccg/co2/flask/month/>), averaged over January to March, and converted to pCO₂^{atm} according to:

$$pCO_2^{\text{atm}} = xCO_2(\text{SLP} - VP_{H_2O}) \quad (3)$$

where SLP is the monthly sea level pressure averaged over January to March for each year, VP_{H₂O} is the wintertime water vapor pressure and was computed from θ according to *Cooper et al.* [1998]. The SLP data originate from the NCEP/NCAR reanalysis project [*Kalnay et al.*, 1996] and were obtained from the IRI/LDEO Climate Data Library (<http://ingrid.ldeo.columbia.edu/>).

[13] Finally, ΔpCO_2 was computed for each sample as the difference between pCO₂^{atm} and pCO₂^{SW}.

4. Results

[14] Figure 3 depicts the spatial distribution of eSPMW during the *Baldrige* cruise. From 46°N and northward, eSPMW was found at each station throughout the 200–1000 m layer. South of 45°N the presence of eSPMW was sporadic both with respect to station and depth, indicating that eSPMW indeed forms in the subpolar region.

[15] There were no trends in wind speed (not shown) so over the study period, the waters embodying the winter mixed layer of the esNA (i.e., eSPMW) changed from being strong sink to weak sink of atmospheric CO₂. This can be appreciated from Figure 4 which show wintertime pCO₂^{atm}

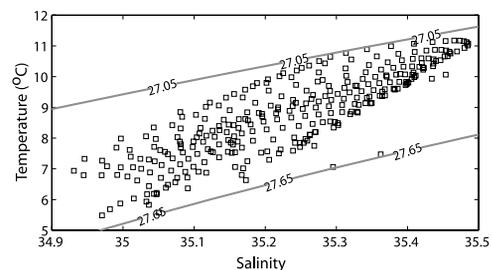


Figure 2. Wintertime (January–March) climatological temperature-salinity relationship for surface seawater in the eastern subpolar North Atlantic. Maximum and minimum density values are also shown (lines). These data were obtained from the World Ocean Atlas 2001 [*Conkright et al.*, 2002].

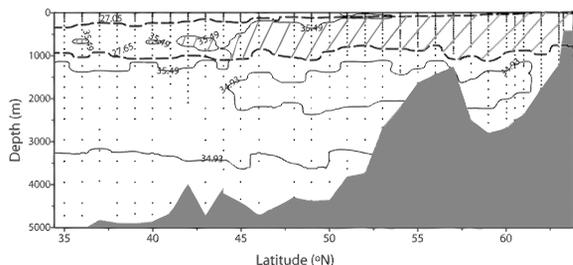


Figure 3. The distribution of 34.93 and 35.49 isohalines (solid contours) and 27.05 and 27.65 isopycnals (dashed contours) during the *Baldrige* 93 cruise. The hatched area shows the spatial distribution of eastern Subpolar Mode Water (eSPMW) as defined by salinity and potential density ranges of 34.93–35.49 and 27.05–27.65.

and the estimated values of $p\text{CO}_2^{\text{SW}}$ and $\Delta p\text{CO}_2$ as a function of formation year (= 1993 – CFC-12 age) of eSPMW. Evidently, the eSPMW was undersaturated with respect to atmospheric CO₂ throughout the study period (Figure 4a). However, this disequilibrium decreased consistently over the years and nearly vanished at the end of the study period (Figure 4b). Straight lines fitted to the data (Figures 4a and 4b) suggest linear trends of 1.47, 3.00, and $-1.53 \mu\text{atm yr}^{-1}$ for $p\text{CO}_2^{\text{atm}}$, $p\text{CO}_2^{\text{SW}}$ and $\Delta p\text{CO}_2$, respectively.

5. Discussion

[16] Our computations suggest that the wintertime $p\text{CO}_2^{\text{SW}}$ in the eastern subpolar North Atlantic increased at a greater rate than the atmospheric $p\text{CO}_2$ so that the disequilibrium decreased between 1972 and 1989. This is in qualitative agreement with *Lefèvre et al.* [2004]. Quantitatively, however, the $p\text{CO}_2^{\text{SW}}$ growth rate estimated in this study ($3.00 \mu\text{atm yr}^{-1}$) is larger than the $\sim 1.8 \mu\text{atm yr}^{-1}$ estimated by *Lefèvre et al.* [2004] for the months January–March. The difference in growth rates can be reconciled as our estimate, being based on reconstructions rather than observations, is sensitive to the choice of $R_{\text{C:O}}$ and $R_{\text{N:O}}$ values as well as to the CFC-12 ages.

[17] If we were to use a lower value of $R_{\text{C:O}}$ then the estimated $p\text{CO}_2^{\text{SW}}$ growth rate would decrease because the correction for remineralized carbon (= $-R_{\text{C:O}} \times \text{AOU}$, see equation (2)) would be reduced more for “older” eSPMW than for “younger” eSPMW. Changing $R_{\text{N:O}}$ has a similar effect on the estimated $p\text{CO}_2^{\text{SW}}$ growth rate, but the effect is much smaller than that of $R_{\text{C:O}}$ because $R_{\text{N:O}} \times \text{AOU}$ is an order of magnitude smaller than $R_{\text{C:O}} \times \text{AOU}$. There are several alternative remineralization quotients available [e.g., *Redfield et al.*, 1963; *Takahashi et al.*, 1985; *Anderson and Sarmiento*, 1994], but only those of *Redfield et al.* [1963] are applicable to our study area and at the same time free from known biases. These latter quotients, however, are higher than those of *Körtzinger et al.* [2001] and give a $p\text{CO}_2^{\text{SW}}$ growth rate of $3.2 \mu\text{atm yr}^{-1}$.

[18] The use of apparent CFC-12 ages implicitly assumes that the surface-to-interior spreading of water can be estimated from a single transit time. This assumption is strictly only valid for the limiting case of no mixing and, in order to account for mixing, *Waugh et al.* [2004] applied the concept of transit time distribution (TTD) in the subpolar North

Atlantic. They found that CFC-12 ages are smaller than the mean transit times (mean ages, henceforth) for much of the gyre. Being unable to constrain by how much CFC-12 ages underestimate the mean ages they provided several possible relationships between CFC-12 ages and mean ages [*Waugh et al.*, 2004, Figure 6b] by choosing the ratio Δ/Γ (which denotes the width of the TTD divided by the mean age) equal to either 0, 0.75, 1, and 1.25. Thus, the only certain consequence for our results is that since the CFC-12 ages are a lower limit of a range of possible ages the estimated $p\text{CO}_2^{\text{SW}}$ growth rate is an upper limit. This is so because increasing the ages implies that the total $p\text{CO}_2^{\text{SW}}$ increase of $\approx 50 \mu\text{atm}$ (Figure 4a) would be distributed over a longer time period and, thus, the growth rate would be reduced.

[19] In order to evaluate the combined effect of the loosely constrained remineralization quotients and the possible underestimation of water mass ages on the $p\text{CO}_2^{\text{SW}}$ growth rate we (i) utilized the remineralization quotients of both *Körtzinger et al.* [2001] and *Redfield et al.* [1963] for the determination of $p\text{CO}_2^{\text{SW}}$, (ii) replaced the CFC-12 ages with mean ages obtained through employing the relationships of *Waugh et al.* [2004] and (iii) re-estimated the $p\text{CO}_2^{\text{SW}}$ growth rate. Important to mention is that, when the mean ages obtained by setting Δ/Γ equal to 0.75, 1, or 1.25 were used, the temporal trend of $p\text{CO}_2^{\text{SW}}$ became quadric so that the data best fitted to equations of the form:

$$p\text{CO}_2^{\text{SW}} = A \times (\text{year})^2 + B \times (\text{year}) + C \quad (4)$$

where year = 1993 – mean age and A, B, and C are constants. For these cases, therefore, the $p\text{CO}_2^{\text{SW}}$ growth rate was determined as the mean value of the time derivative of equation (4) evaluated for the years 1972 through to 1989. The re-estimated growth rates are depicted in Figure 5 and fall between $1.36 \mu\text{atm yr}^{-1}$ and $3.2 \mu\text{atm yr}^{-1}$. This suggests that during 1972–1989 $p\text{CO}_2^{\text{SW}}$ in the eSNA increased at a rate around or above the atmospheric $p\text{CO}_2$ growth rate of $1.47 \mu\text{atm/yr}$. Figure 5 also indicate that the $1.8 \mu\text{atm yr}^{-1}$ growth rate estimated by *Lefèvre et al.* [2004] would be obtained by using a Δ/Γ value between 0.75 and 1.

6. Conclusions and Further Remarks

[20] Using an extended back calculation approach we have shown that wintertime $p\text{CO}_2^{\text{SW}}$ in the surface waters of

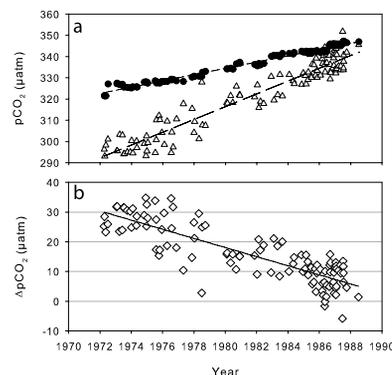


Figure 4. (a) Temporal changes of wintertime $p\text{CO}_2^{\text{atm}}$ (dots) and $p\text{CO}_2^{\text{SW}}$ (triangles), (b) Temporal changes of wintertime $\Delta p\text{CO}_2 (= p\text{CO}_2^{\text{atm}} - p\text{CO}_2^{\text{SW}})$. Linear regressions of the data are also shown (lines).

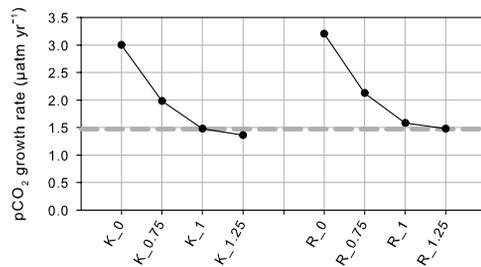


Figure 5. Growth rates of reconstructed pCO₂^{SW} for the eastern subpolar North Atlantic during 1972–1989 for different combinations of remineralization quotients and transit times. On the x axis K_{0.75} denotes that the quotients of Körtzinger *et al.* [2001] were used and CFC-12 ages were converted to mean transit times according to Waugh *et al.* [2004] by choosing $\Delta/\Gamma = 0.75$ (see sections 5 of the main text), R_{1.25} denotes quotients of Redfield *et al.* [1963] and $\Delta/\Gamma = 1.25$, etc. The horizontal dashed line indicates the mean atmospheric pCO₂ growth rate during 1972–1989.

the eastern subpolar North Atlantic increased between 1972 and 1989. Our approach resulted at first hand in a pCO₂^{SW} growth rate estimate of 3 μatm yr⁻¹. After taking the caveats of the analysis into account we have shown that this is an upper limit, and a lower limit is close to the atmospheric growth rate of 1.47 μatm yr⁻¹.

[21] With respect to the cause of the trend of increasing pCO₂^{SW}, no single process can be pointed out. The estimated pCO₂^{SW} depends mainly on preformed concentrations of total dissolved inorganic carbon (C_T⁰) and alkalinity (A_T⁰), potential temperature (θ), and salinity (S). Of these four parameters only C_T⁰ exhibited a significant temporal trend (not shown) which, in turn, have caused the trend of increasing pCO₂^{SW}. There are several processes that may have caused the increase in C_T⁰. Analysis of NCEP/NCAR reanalysis data reveal no increase in the mean wind speed during the winter season so it is unlikely that enhanced air-sea CO₂ flux into the study area increased C_T⁰ during the study period. Similarly, a long term decrease in the amount of carbon fixed into organic matter and/or increased remineralization is also unlikely because the preformed phosphate ($P^0 = P - R_{C:O} \times AOU$) did not show any significant trend (not shown). We suggest that uptake of anthropogenic carbon is the main cause of the estimated pCO₂^{SW} increase. Especially, since this process can produce excessive pCO₂^{SW} growth rates in the North Atlantic as shown by Wallace [2001] and Anderson and Olsen [2002]. The reason for this is that the surface water feeding this region originates mainly further south where temperatures are higher and buffer capacities larger. As water with any given C_T change are moved northward the corresponding pCO₂^{SW} change will be amplified since the buffer capacity decreases due to cooling induced C_T uptake. Therefore if the pCO₂^{SW} in the south changed at the same rate as the atmosphere, greater than atmospheric increases can take place in the north.

[22] **Acknowledgments.** This is a contribution to the EU IP CARBOOCEAN (contract 511176-2) and publication A120 of the Bjerknes Centre for Climate Research. Comments from Karsten Friis and a second anonymous referee helped improve the manuscript.

References

- Anderson, L. A., and J. L. Sarmiento (1994), Redfield ratios of remineralization determined by nutrient data analysis, *Global Biogeochem. Cycles*, 8(1), 65–80.
- Anderson, L. G., and A. Olsen (2002), Air-sea flux of anthropogenic carbon dioxide in the North Atlantic, *Geophys. Res. Lett.*, 29(17), 1835, doi:10.1029/2002GL014820.
- Brewer, P. (1978), Direct observation of the oceanic CO₂ increase, *Geophys. Res. Lett.*, 5, 997–1000.
- Conkright, M. E., et al. (2002), *World Ocean Atlas 2001: Objective Analyses, Data Statistics, and Figures, CD-ROM Documentation*, 17 pp., Natl. Oceanogr. Data Cent., Silver Spring, Md.
- Cooper, D. J., A. J. Watson, and R. D. Ling (1998), Variation of pCO₂ along a North Atlantic shipping route (UK to Caribbean): a year of automated observations. *Mar. Chem.*, 60, 147–164.
- Dickson, A. G., and F. J. Millero (1987), A comparison of the equilibrium constants for the dissociation of carbonic acid in seawater media, *Deep Sea Res.*, 34, 1733–1743.
- Friis, K., et al. (2005), On the temporal increase of anthropogenic CO₂ in the subpolar North Atlantic, *Deep Sea Res., Part I*, 52, 681–698.
- Gouretski, V. V., and K. Jancke (2001), Systematic errors as the cause for an apparent deep water property variability: Global analysis of the WOCE and historical hydrographic data, *Prog. Oceanogr.*, 48, 337–402.
- Hanawa, K., and L. D. Talley (2001), Mode waters, in *Ocean Circulation and Climate: Observing and Modelling the Global Ocean*, edited by G. Siedler, J. Church, and J. Gould, Elsevier, New York.
- Kalnay, E., et al. (1996), The NCEP/NCAR 40-year reanalysis project, *Bull. Am. Meteorol. Soc.*, 77, 438–471.
- Key, R. M., et al. (2004), A global ocean carbon climatology: Results from GLODAP, *Global Biogeochem. Cycles*, 18, GB4031, doi:10.1029/2004GB002247.
- Körtzinger, A., J. I. Hedges, and P. D. Quay (2001), Redfield ratios revisited: Removing the biasing effect of anthropogenic CO₂, *Limnol. Oceanogr.*, 46, 964–970.
- Körtzinger, A., P. D. Quay, and R. E. Sonnerup (2003), Relationship between anthropogenic CO₂ and the ¹³C Suess effect in the North Atlantic Ocean, *Global Biogeochem. Cycles*, 17(1), 1005, doi:10.1029/2001GB001427.
- Lefèvre, N., et al. (2004), A decrease in the sink for atmospheric CO₂ in the North Atlantic, *Geophys. Res. Lett.*, 31, L07306, doi:10.1029/2003GL018957.
- Lueker, T. J., A. G. Dickson, and C. D. Keeling (2000), Ocean pCO₂ calculated from dissolved inorganic carbon, alkalinity, and equations for K₁ and K₂: Validation based on laboratory measurements of CO₂ in gas and seawater at equilibrium, *Mar. Chem.*, 70, 105–119.
- McCartney, M. S., and L. D. Talley (1982), The subpolar mode water of the North Atlantic Ocean, *J. Phys. Oceanogr.*, 51, 118–1169.
- Merbach, C., et al. (1973), Measurement of apparent dissociation constants of carbonic acid in seawater at atmospheric pressure, *Limnol. Oceanogr.*, 18, 533–541.
- Redfield, A. C., B. H. Ketchum, and F. A. Richards (1963), The influence of organisms on the composition of sea water, in *The Sea*, vol. 2, edited by M. N. Hill, pp. 26–77, Wiley-Interscience, Hoboken, N. J.
- Takahashi, T., W. Broecker, and S. Langer (1985), Redfield ratio based on chemical data from isopycnal surfaces, *J. Geophys. Res.*, 90, 6907–6924.
- Takahashi, T., et al. (2002), Global sea-air CO₂ flux based on climatological surface ocean pCO₂, and seasonal biological and temperature effects, *Deep Sea Res., Part II*, 49, 1601–1622.
- Wallace, D. W. R. (2001), Storage and transport of excess CO₂ in the oceans: The JGOFS/WOCE global CO₂ survey, in *Ocean Circulation and Climate: Observing and Modelling the Global Ocean*, edited by G. Siedler, J. Church, and J. Gould, Elsevier, New York.
- Wanninkhof, R., et al. (1999), Comparison of methods to determine the anthropogenic CO₂ invasion into the Atlantic Ocean, *Tellus, Ser. B*, 51, 511–530.
- Waugh, D. W., T. W. N. Haine, and T. M. Hall (2004), Transport times and anthropogenic carbon in the subpolar North Atlantic, *Deep Sea Res., Part I*, 51, 1475–1491.
- Wetzel, P., A. Winguth, and E. Maier-Reimer (2005), Sea-to-air CO₂ flux from 1948 to 2003: A model study, *Global Biogeochem. Cycles*, 19, GB2005, doi:10.1029/2004GB002339.

A. Olsen, Geophysical Institute, University of Bergen, Allégaten 70, N-5007 Bergen, Norway.

A. M. Omar, Bjerknes Centre for Climate Research, University of Bergen, Allégaten 55, N-5007 Bergen, Norway. (abdirahman.omar@bjerknes.uib.no)