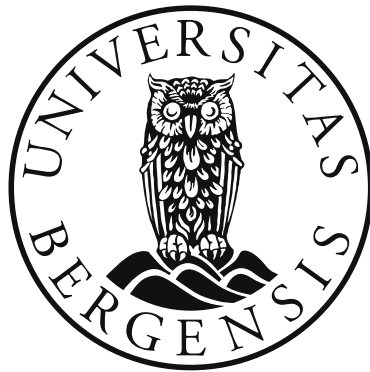


# The inorganic carbon cycle of the Red Sea

**Elsheikh Bashir Ali**



Thesis for the degree of philosophiae doctor (PhD)  
at the University of Bergen

2017

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## Abstract

The aim of this thesis has been to improve the general scientific understanding of the marine inorganic carbon cycle of the Red Sea. The region houses an exceptional range of ecosystems and it is considered the most important repository of biodiversity in the world with vast calcification areas rich in coral reefs. Therefore, it is important to understand the interaction between biogeochemical processes and the Red Sea ecosystem. In spite of this fact, the area is poorly studied, and this is particularly true for the marine inorganic carbon cycle.

Prior to this work, no systematic sampling of biogeochemical variables had been done off the coast of Sudan. As part of this work, a new biogeochemical time series from the Sudanese coastal waters has been established, which represents the very first data on hydrography and inorganic carbon from this part of the Red Sea. The sampling site has been operated since 2007, and here is presented data over a period of 8 years, from 2007 to 2015. Consequently, there now exists a comprehensive dataset allowing the quantification of interannual to seasonal variations in biogeochemical variables, and identify important processes that control these changes. The establishment of such a baseline is important to compare future changes caused by climate and environmental changes when approaching the high CO<sub>2</sub> world.

The three papers in this thesis improve and renew our understanding of the Red Sea hydrography, the inorganic carbon cycle, how the variation goes, and what drives the observed changes.

A common assumption has been that the temperature and salinity variations in the Red Sea is solely a function of local heat and mass flux. However, results from the current work evidence that advection of temperature and salinity gradients also plays a role for establishing the temperature and salinity maxima in the area (**Paper I**). The finding of this paper also show that the annual temperature range is approximately 6°C, with

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highest temperatures during summer and autumn. Changes in salinity lag that of temperature with about 3 months.

The inorganic carbonate measurements (**Paper II**) document for the very first time the seasonal as well as interannual variability of dissolved inorganic carbon (DIC) and total alkalinity ( $A_T$ ) in this part of the ocean. This pristine dataset has been used to construct a baseline for the inorganic carbon cycle and constitutes an important reference for years to come. Higher  $A_T$  and DIC is measured during winter compared to the summer with an annual change of approximately  $40 \mu\text{mol kg}^{-1}$  for  $A_T$  and  $32 \mu\text{mol kg}^{-1}$  for DIC.  $A_T$  is mainly driven by physical processes such as advection and local evaporation (through salinity changes) as well as calcification, while changes in DIC are mainly a result of air-sea exchange and likely biological activity, and to a smaller degree along shore advection.

The Red Sea, which is situated in the sub-tropical and tropical area, has previously been regarded as a net annual source for atmospheric  $\text{CO}_2$ , but this view has to be revised. **Paper III** shows that  $p\text{CO}_2$  is high during summer and autumn and low during spring and winter, with a seasonal amplitude of about  $60 \mu\text{atm}$ . Consequently, the Sudanese coastal area acts as a source for atmospheric  $\text{CO}_2$  during summer and autumn, while during winter and spring, the area is a sink for atmospheric  $\text{CO}_2$ . Over an annual cycle, the area is a net sink of atmospheric  $\text{CO}_2$  of size  $24.4 \text{ mmol CO}_2 \text{ m}^{-2} \text{ y}^{-1}$ . The change from being a net annual source for atmospheric  $\text{CO}_2$  to becoming a net sink likely occurred in the 2000s.

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## List of publications

### Paper I

Ali, E. B., J. H. Churchill, K. Barthel, I. Skjelvan, A. M. Omar, T. E. de Lange, and E. B. A. Eltaib, 2017. Seasonal and interannual variations of hydrographic parameters in the Sudanese coast of the Red Sea, 2009-2015, *submitted to Regional Studies in Marine Sciences journal*.

### Paper II

Ali, E. B., A. M. Omar, I. Skjelvan, and T. Johannessen, 2017. Dissolved inorganic carbon and total alkalinity at the Sudanese coastal Red Sea, 2009-2013, *submitted to Regional Studies in Marine Sciences journal*.

### Paper III

Ali, E. B., I. Skjelvan, A. M. Omar, A. Olsen, T. E. de Lange, T. Johannessen, S. Elageed, 2017, Sea surface pCO<sub>2</sub> variability and air sea gas exchange in the coastal Sudanese Red Sea, *Manuscript in preparation*.

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## 1. Aim of the study

The main goal of this thesis has been to achieve a better understanding of the inorganic carbon cycle at the western Red Sea through unravelling the biogeochemical setting and examining the spatiotemporal variation of hydrography and carbon parameters. The environmental settings for the Red Sea is extreme in terms of biogeochemistry. The sea is considered an important calcification area with numerous coral reefs and thus its coastline is very vulnerable, but despite of this fact, there is a limited amount of scientific surveys in the area, which has resulted in a poorly understanding of the marine inorganic carbon cycle, in particular. The lack of data hinders adequate analyses of the interactions between the biogeochemical processes and the inorganic carbon cycle in the area. This is critical in light with the ongoing global warming, exemplified in a temperature increase in the Red Sea of 0.7 °C from 1985 to 2007 (Raitsos et al., 2011), expanding oxygen–minimum zones in the tropical regions (Stramma et al., 2008), and the increasing atmospheric CO<sub>2</sub> (Le Quéré et al., 2016).

With help of a pristine time series of hydrography and inorganic carbon data, the aim is to develop new and improved knowledge regarding the inorganic carbon cycle in this area.

## 2. The marine carbon system

### 2.1 Introduction

Due to the carbon emissions produced by the combustion of fossil fuels, production of cement, and land-use change, the atmospheric carbon dioxide ( $\text{CO}_2$ ) concentration has increased exponentially from approximately 277 parts per million (ppm) at the beginning of industrial revolution in 1750 to 402 ppm in 2016. According to Le Quéré et al. (2016), combustion of fossil fuels globally emitted an amount of  $9.3 \pm 0.5 \text{ GtC y}^{-1}$  into the atmosphere while land-use change contributed  $1.0 \pm 0.5 \text{ GtC y}^{-1}$  during the period between 2006 and 2015. Ocean takes up about a quarter of the annual carbon emissions ( $2.6 \pm 0.5 \text{ GtC y}^{-1}$ ) and the global residual terrestrial  $\text{CO}_2$  sink is  $3.1 \pm 0.9 \text{ GtC y}^{-1}$ . The growth rate of atmospheric  $\text{CO}_2$  concentration for the same period is  $4.5 \pm 0.1 \text{ GtC y}^{-1}$  (Le Quéré et al., 2016) (see Fig.2.1).

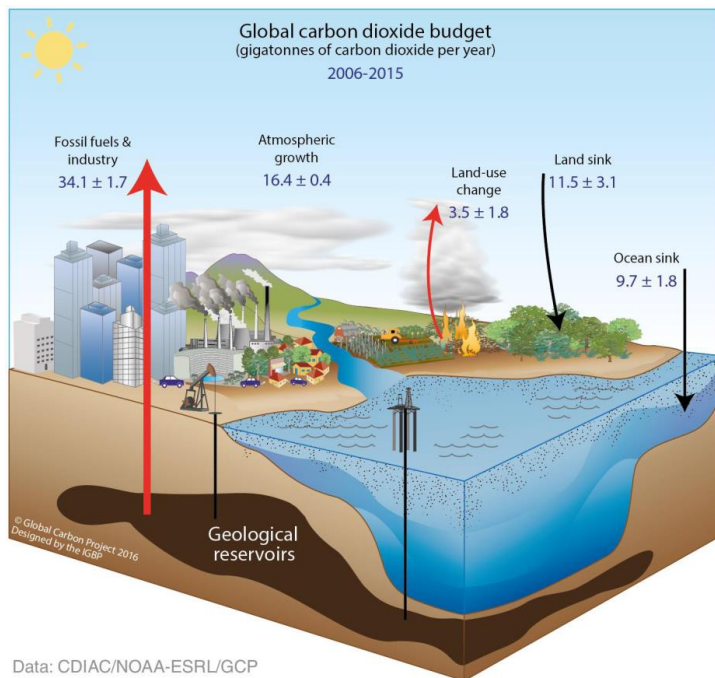


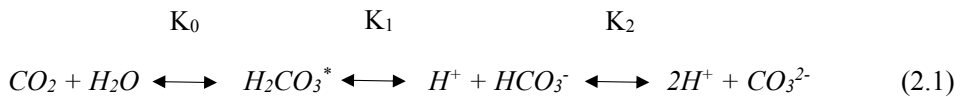
Fig. 2.1 The global carbon dioxide budget with reservoirs (in  $\text{GtC y}^{-1}$ ) for the period between 2006 and 2015. 1  $\text{GtC}$  equals  $10^{15}$  g C. Figure from Le Quéré et al. (2016).

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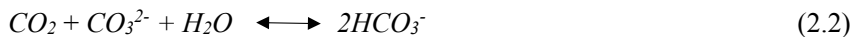
The CO<sub>2</sub> continuously cycles between the atmosphere, land, and ocean, but the amount of CO<sub>2</sub> is not evenly distributed, and e.g. the ocean stores about 50 times more CO<sub>2</sub> than the atmosphere (Field and Raupach, 2004; Zeebe and Wolf-Gladrow, 2001). The CO<sub>2</sub> exchange between the surface ocean and atmosphere is faster than the exchange between the surface layer and deep layer of the ocean. The latter one takes from several hundreds to thousands of years because it is driven by slower processes such as deep-water formation and the marine biological production (Emerson and Hedges, 2008).

## 2.2 The carbon chemistry

When CO<sub>2</sub> is dissolved in seawater, it is converted to aqueous CO<sub>2</sub> and aqueous carbonic acid (H<sub>2</sub>CO<sub>3</sub>). This weak acid is dissociated in two steps producing one proton (H<sup>+</sup>) and bicarbonate ion (HCO<sub>3</sub><sup>-</sup>) in the first step and two protons and carbonate (CO<sub>3</sub><sup>2-</sup>) ion in the second step (see Eq. 2.1):



where K<sub>0</sub> is Henry's constant for CO<sub>2</sub>, and K<sub>1</sub> and K<sub>2</sub> are the first and the second dissociation constants of carbonic acid. All constants depend on temperature, salinity, and pressure of seawater (Lueker et al., 2000; Mehrbach et al., 1973). The star (\*) indicate aquatic solution. Additional CO<sub>2</sub> dissolved in seawater generates more H<sup>+</sup> ions which drive more CO<sub>3</sub><sup>2-</sup> to react with H<sup>+</sup> and produce HCO<sub>3</sub><sup>-</sup>; known as the buffer reaction:



CO<sub>3</sub><sup>2-</sup> is also used to form calcium carbonate (CaCO<sub>3</sub>), which is a building block for skeleton and shells of some marine organisms. The CaCO<sub>3</sub> saturation state describes if the water is supersaturated or under saturated with respect to CO<sub>3</sub><sup>2-</sup>.

---

The marine carbon system is described by four carbon variables; total dissolved inorganic carbon (DIC), total alkalinity ( $A_T$ ), pH, and fugacity of carbon ( $fCO_2$ ). If two of the carbon variables are known in addition to the equilibrium constants, the other two variables can be calculated. The total dissolved inorganic carbon (DIC), sometimes called  $TCO_2$ ,  $C_T$ , or  $\sum CO_2$ , is defined as the sum of inorganic carbon components (Dickson and Goyet, 1994):

$$DIC = [CO_2]^* + [HCO_3^-] + [CO_3^{2-}] \quad (2.3)$$

where  $[CO_2]^*$  is the sum of aquatic  $CO_2$  and  $H_2CO_3$ .

The  $A_T$  is defined by Dickson (1981) as "the number of moles of hydrogen ion equivalent to the excess of proton acceptors over proton donors in one kilogram of sample":

$$A_T = [HCO_3^-] + 2[CO_3^{2-}] + [B(OH)_4^-] + [OH^-] + [HPO_4^{2-}] + 2[PO_4^{3-}] + [H_3SiO_4^-] + [NH_3] + [HS^-] - [H^+]_F - [HSO_4^-] - [HF] - [H_3PO_4] \quad (2.4)$$

where  $[H^+]_F$  is the free concentration of  $H^+$  ions. The major parts of total alkalinity in seawater is:

$$A_T = [HCO_3^-] + 2[CO_3^{2-}] + [B(OH)_4^-] + [OH^-] - [H^+]_F \quad (2.5)$$

The thermodynamic state of the acid-base system is described by the pH, which is defined as the negative logarithm of hydrogen ions concentration in gram atoms per liter:

$$pH = -\log[H^+] \quad (2.6)$$

The ocean can be considered as a natural buffer system and seawater at normal conditions is slightly basic ( $\approx 8.1$ ) (Zeebe, 2012). At this pH,  $HCO_3^-$  is the major component of DIC (88.6%) followed by  $CO_3^{2-}$  (10.9%) while  $[CO_2]^*$  only takes about 0.5% (Fig. 2.2).

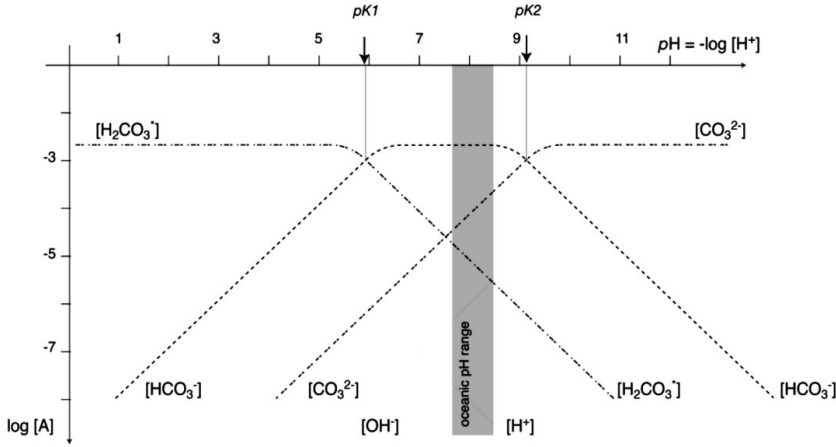


Fig. 2.2 Concentrations of the different inorganic carbon components as a function of pH. The dissociation constants of Eq. 2.1 are presented in the figure.  $[\text{H}_2\text{CO}_3]^*$  equals  $[\text{CO}_2]^*$ , which is mentioned in the text. Figure from Sarmiento and Gruber (2006).

The partial pressure of  $\text{CO}_2$  ( $p\text{CO}_2$ ) is describing the amount of  $\text{CO}_2$  in gas phase that is in equilibrium with seawater. According to Sarmiento and Gruber (2006), Eq. 2.1 and its dissociation constants can be reformulated to express the  $p\text{CO}_2$  as:

$$p\text{CO}_2 = \frac{K_2}{K_0 \cdot K_1} \frac{[\text{HCO}_3^-]^2}{[\text{CO}_3^{2-}]} \quad (2.7)$$

The fugacity of  $\text{CO}_2$  ( $f\text{CO}_2$ ) differs from  $p\text{CO}_2$  by taking into account the non-ideal behavior of  $\text{CO}_2$  and the difference between them is less than 0.5% (Dickson and Goyet, 1994). Zeebe and Wolf-Gladrow (2001) has described the relation between  $p\text{CO}_2$  and  $f\text{CO}_2$  as:

$$f\text{CO}_2 = p\text{CO}_2 \exp\left(P \frac{B + 2\delta}{RT}\right) \quad (2.8)$$

---

where  $P$  is the total atmospheric pressure,  $B$  is the first virial coefficient of  $\text{CO}_2$ ,  $\delta$  is the cross virial coefficient,  $R$  is the gas constant, and  $T$  is absolute temperature. The unit for these parameters are:  $f\text{CO}_2$  and  $p\text{CO}_2$  in  $\mu\text{atm}$ ,  $P$  in Pa (1 atm = 101325 Pa),  $B$  and  $\delta$  are in  $\text{m}^3 \text{mol}^{-1}$ ,  $R = 8.314 \text{ J K}^{-1} \text{mol}^{-1}$ , and  $T$  in Kelvin. According to Weiss (1974),  $B$  and  $\delta$  can be determined, respectively, as:

$$B = (-1636.75 + 12.0408T - 3.27957 \cdot 10^{-2}T^2 + 3.16528 \cdot 10^{-5}T^3)10^{-6} \quad (2.9)$$

$$\delta = (57.7 - 0.118T)10^{-6} \quad (2.10)$$

The buffer capacity reflects the capacity of seawater to buffer changes in pH occurring because  $\text{CO}_2$  is absorbed in the sea, and it can be quantified through the Revelle factor ( $\gamma$ ).  $\gamma$  describes how  $f\text{CO}_2$  changes for a given change in DIC when  $A_T$  is constant:

$$\gamma = \frac{\Delta f\text{CO}_2}{\Delta \text{DIC}} \frac{\text{DIC}}{f\text{CO}_2} \quad (2.11)$$

According to Takahashi et al. (1993), the Revelle factor is high (approximately 14) in cold polar and subpolar surface water and low (ca. 8) in warm surface water in tropical and subtropical areas. A global Revelle factor of 10 indicates that 1% change in DIC will drive the surface  $f\text{CO}_2$  to change by about 10%. The current  $\gamma$  values are higher by one unit compared to the values prior to the industrial revolution (Sabine et al., 2004). Increasing surface  $f\text{CO}_2$  drives the  $\gamma$  values to increase, thus the surface ocean has become less able to absorb additional  $\text{CO}_2$ .

### 2.3 The main processes controlling the marine carbon cycle

In general, the processes described below have been assessed and make up the theoretical basis in the general carbon cycle for the three papers presented in this thesis.



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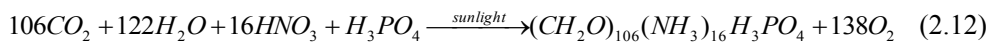
### 2.3.1 Solubility pump

The solubility of any gas in seawater depends on temperature and salinity (Gordon and Jones, 1973). Therefore, cold water at high latitudes contain more CO<sub>2</sub> in equilibrium with the atmosphere than warm water at lower latitudes. By help of the global thermohaline circulation (Broecker, 1991), dissolved CO<sub>2</sub> sinks towards deep layers through deep-water formation at high latitudes, while at low latitudes, upwelling brings carbon rich deep water to the surface, which is warm and has low gas solubility, and thus, CO<sub>2</sub> is emitted into the atmosphere. This physical process (called solubility pump) takes long time (hundreds of years) and constantly exchange CO<sub>2</sub> between ocean and atmosphere.

### 2.3.2 Biological pump

The biological pump can be split into two parts: organic carbon pump and calcium carbonate counter pump.

The organic carbon pump is described through the primary production taking place in the surface water and remineralization of organic matter which occurs in sub-surface and deep waters. During primary production, the phytoplankton use aquatic CO<sub>2</sub> from the surface water and transform it into organic matter through photosynthesis:



A minor part of the sinking organic particles is trapped in the sediment, while most of the organic matter is remineralized by bacteria and regenerated into CO<sub>2</sub> and nutrients. Upwelling of deep water brings water rich in carbon and nutrients into the surface (Fig. 2.3).

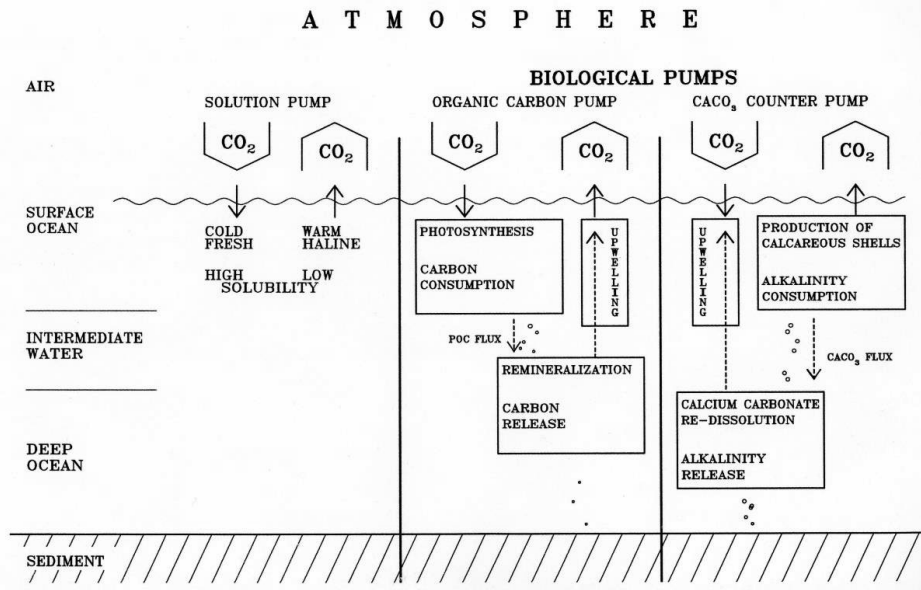
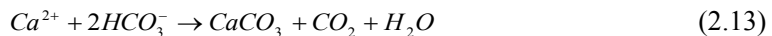
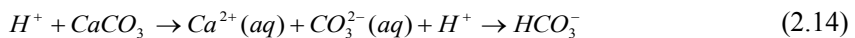


Fig. 2.3 The solubility and biological CO<sub>2</sub> pumps in the ocean (Heinze et al., 1991).

The calcium carbonate counter pump is described through the production of calcium carbonate in the surface layer and dissolution in the deep waters. The coral reefs and many planktonic organisms such as Coccolithophorids uses CaCO<sub>3</sub> to form their shells and skeletons according to



In the deep water, where the conditions are more acidic, the CaCO<sub>3</sub> shells are dissolved:



Photosynthesis will decrease DIC, and  $fCO_2$ , while pH increases.  $A_T$  is only affected to a minor degree, by a small increase due to nutrient consumption (Fig.2.4). On the other hand, calcium carbonate formation decreases  $A_T$  by two units and DIC by one

unit, and only to a minor degree decreases pH and increases  $f\text{CO}_2$  (Fig.2.4) (Zeebe and Wolf-Gladrow, 2001).

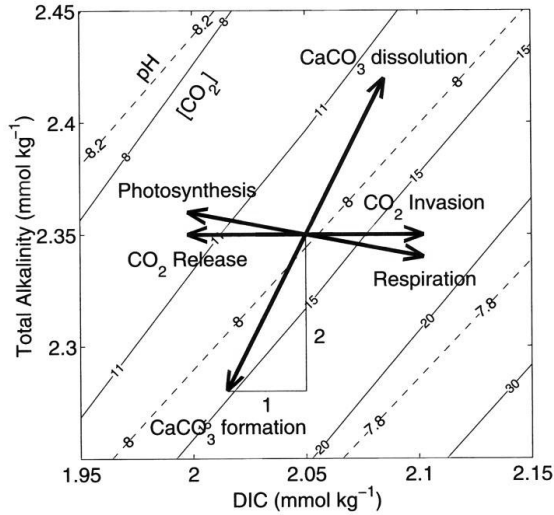


Fig. 2.4 Effects of the different processes on DIC,  $A_T$ , pH, and  $[\text{CO}_2]$  (from Zeebe and Wolf-Gladrow, 2001).

### 2.3.3 Air-sea gas exchange

Air-sea gas exchange affects the DIC and pH (Fig. 2.4), and the direction of  $\text{CO}_2$  fluxes between air and sea depends on differences in  $\text{CO}_2$  concentration ( $f\text{CO}_2$  or  $p\text{CO}_2$ ) between atmosphere and surface ocean, as well as the wind at sea surface, the temperature and to a small degree the salinity. The  $\text{CO}_2$  flux is determined as

$$F = SK(f\text{CO}_2^{\text{seawater}} - f\text{CO}_2^{\text{atmosphere}}) \quad (2.15)$$

where  $S$  is solubility of  $\text{CO}_2$  in seawater ( $\text{mol kg}^{-1}\text{atm}^{-1}$ ), which depends on temperature and salinity of the surface water (Weiss, 1974).  $K$  is the gas transfer velocity, which depends on the molecular diffusivity, kinematic viscosity, and turbulence at the air-

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water interface.  $K$  is commonly parameterized as a function of wind speed because the wind speed is important for the turbulence.

There are numerous relationships of  $K$  in the literature (Liss and Merlivat, 1986; Wanninkhof, 1992; Wanninkhof and McGillis, 1999; Nightingale et al., 2000; Sweeney et al., 2007), and in this thesis, the one from Nightingale et al. (2000) is used:

$$K = \left(0.222U_{10}^2 + 0.333U_{10}\right) \left(\frac{Sc}{660}\right)^{-1/2} \quad (2.16)$$

where  $U_{10}$  is wind speed at 10 m above the sea surface, and  $Sc$  is the Schmidt number, which show the ratio between kinematic viscosity and molecular diffusivity.

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### 3. Description of the study area

The Red Sea separates the northeastern Africa from the Arabian Peninsula, and linking between the tropical and sub-tropical parts of the ocean. It is about 1930 km long, on average 200 km wide, and represent an area of approximately  $0.46 \times 10^6 \text{ km}^2$ . The Red Sea is connected with the Gulf of Aden and the Indian Ocean through the narrow strait of Bab Al Mandab (BAM) (Fig. 3.1). In the northern part, the Red Sea is divided into two main branches; Gulf of Aqaba and Gulf of Suez, the latter is linked with the Mediterranean Sea via the Suez Canal. The bottom topography of the Red Sea is naturally wedge shaped with relatively large maximum depth (2920 m) in the central of the basin. The average depth of Bab Al Mandab strait is about 300 m while the Gulf of Suez has a relatively flat bottom with depth about 60 m. The Gulf of Aqaba is a deep basin with narrow shelves and a mean depression depth of about 1200 m (Morcos, 1970; Patzert, 1974; Edwards, 1987; Maillard and Soliman, 1986).

The climate of the Red Sea is influenced by two wind regimes (Fig. 3.1). North of  $19^\circ \text{N}$  the north-northwesterly wind (NNW), which is controlled by eastern Mediterranean weather systems, is active throughout the year. During the summer, the NNW wind extends as far south as the BAM Strait. The area south of  $14^\circ \text{N}$  is influenced by Indian Monsoon system, which switches between south-southeasterly wind (SSE) during winter and NNW during summer. The area between  $14^\circ \text{N}$  and  $19^\circ \text{N}$  is characterized as a convergent zone for the wind field for most of the year, see Fig. 3.1 (Morcos, 1970; Pedgley, 1974; Patzert, 1974).

The average sea surface temperature of the Red Sea is about  $26^\circ \text{C}$  in the north and  $30^\circ \text{C}$  in the south during summer, while during winter, the temperature is  $2\text{-}4^\circ \text{C}$  lower. The highest surface temperatures ( $28^\circ \text{C}$  -  $34^\circ \text{C}$ ) are found in the south-central parts of the Red Sea where the wind field is convergent and thus weak for most of the year (Sofianos and Johns, 2003). Because of high evaporation, low precipitation, and supply of relative fresh water from the south, the surface salinity of the Red Sea is increasing from approximately 36.5 psu in the south to more than 41.0 psu in the north. Deeper

than about 300 m, the temperature is about 21.5°C and salinity about 40.5 psu all the way to the bottom (Edwards, 1987).

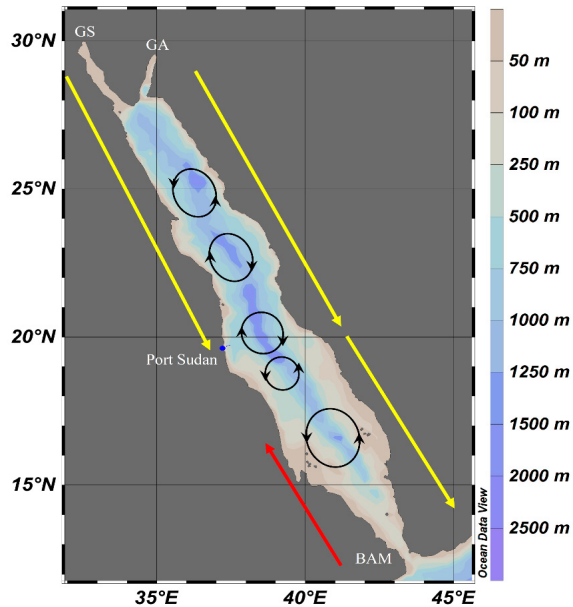


Fig. 3.1 Map showing the location and bathymetry of the Red Sea. Bab Al Mandab Strait (BAM), Gulf of Suez (GS), and Gulf of Aqaba (GA) are indicated in the figure. Arrows refer to wind directions: NNW wind = yellow arrows; SSE wind = red arrows. Arrows to the right in the figure indicate summer situation, while those to the left are winter situation. Locations of cyclonic and anticyclonic gyres are also indicated.

There are four dominant water masses in the Red Sea, all affected by two major masses in the Gulf of Aden. During wintertime, the Gulf of Aden Surface Water (GASW) enters the Red Sea through the BAM Strait as a result of southeasterly winds (Morcos, 1970; Pedgley, 1974; Patzert, 1974; Quadfasel and Baudner, 1993). At deeper layers, the Red Sea Overflow Water (RSOW), which is a mixture of Red Sea Intermediate Water (RSIW) and Red Sea Deep Water (RSDW), is observed to flow out of the Red Sea throughout the year (Sofianos and Johns, 2003; Yao et al., 2014 a; b). During

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summer, when northwesterly winds extend as far south as the BAM Strait, an outflow of Red Sea Surface Water (RSSW) is induced while south of the BAM Strait, southwesterly winds provoke an upwelling of Gulf of Aden Intermediate Water (GAIW), as it moves towards the Red Sea (Morcos, 1970; Patzert, 1974; Smeed, 1997).

The overall circulation of the Red Sea is influenced by the monsoon winds. Surface currents during winter flow from the Indian Ocean northwards throughout the Red Sea while the direction of currents during summer, is reversed i.e. flowing southwards to the Indian Ocean as deep currents. The surface circulation in the south (around 15°N) is featured by anticyclonic rotation during winter and cyclonic rotation during summer. Several cyclonic and anticyclonic gyres are distributed along the north-south axis of the Red Sea, and the strength, size and location of these gyres vary with time (Fig.3.1) (Morcos and Soliman, 1974; Quadfasel and Bauner, 1993; Yao et al., 2014 a; b).

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## 4. Objectives

The main objective of this thesis is to explore the inorganic carbon cycle of the Red Sea, which is poorly known with respect to seasonal and interannual variability. The main objectives has been to:

- establish and maintain a time series to document seasonal and interannual changes.
- understand the coastal physical oceanographic setting.
- determine the drivers of biogeochemical variability at multiple time scales.

The thesis is divided into three topics, which are covered in separate papers. The three papers are based upon the new time series of discrete and continuous data collected in the Sudanese coastal waters off Port Sudan during the period 2007-2015.

Generally, there are few studies focusing on temporal variations of temperature or salinity within the Red Sea, and more specific, there is no study, which fully has dealt with the dynamics responsible for the observed temperature and salinity variations.

**Paper I** aims towards getting a better understanding on the temporal variability of ocean physics based on temperature and salinity as well as pointing at the mechanisms responsible for these variations. **Paper II** aims to unravel the seasonality of the marine inorganic carbon cycle, represented by DIC and  $A_T$ , the relationship with the hydrography, and the drivers causing the observed variability of DIC and  $A_T$ .

**Paper III** focus on how to determine the air-sea flux of  $CO_2$  over an annual cycle using continuous atmospheric and oceanic  $pCO_2$  measurements of moored autonomous sensors deployed in our study area. Further, the seasonal variations and drivers of the oceanic  $pCO_2$  variability have been identified.



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## 5. Summary

Combined, the three papers in this thesis improve and renew our understanding of the Red Sea inorganic carbon cycle, the hydrography, how it varies, and what drives the observed changes.

A new time-series of discrete and continuous data from the Sudanese coast shows that, in contradiction to previous knowledge, advection in addition to local heat and mass flux drive the temperature and salinity variability (**Paper I**). The surface temperature (SST) is at the highest value during summer and autumn and at lowest during winter, with a seasonal amplitude of approximately 6°C. The salinity lagged temperature with about 3 months, and the seasonal amplitude was about 1.1 psu. Validated satellite-derived SST data confirmed the above temperature findings, which was also confirmed by computed T arrived from the local heat flux when taking into account the adjustment for advection and mixing. The impact of alongshore advection on seasonal distribution of temperature and salinity has been estimated using a simple model based on gradient features of temperature, salinity and geostrophic surface velocity. The geostrophic surface velocity is computed from sea level anomaly field. SSS increases from south to north and the maximum SST zone is located south of the area of study, in the central Red Sea. The geostrophic current directs from south to north during spring and summer introducing relatively warm and fresh water from the south. During autumn and winter, the geostrophic current reverses and brings waters, which is cold and more saline towards the study area. The close match between estimated and observed seasonal temperature and salinity supports the conclusion that the observed seasonal T and S signals off Port Sudan are largely the product of local heat and mass flux and alongshore advection.

A study of the seasonal variations of  $A_T$  and DIC off the Sudanese coast (**Paper II**) shows that  $A_T$  and DIC are high during winter and low during summer with an average annual variability of 40  $\mu\text{mol kg}^{-1}$  for  $A_T$  and 32  $\mu\text{mol kg}^{-1}$  for DIC. The seasonal signals are associated with the maxima and minima of salinity described in **Paper I**. Advection of water, and thus changing salinity is an important factor controlling  $A_T$  variations,

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while the observed change in DIC is primarily controlled by air-sea gas exchange, through change in temperature, and, very likely, by biological production. The remaining changes of DIC are caused by alongshore advection during autumn and winter (**Paper I**). Seasonality for DIC and  $A_T$  have been reconstructed using the observed DIC-SST and  $A_T$ -SSS relationships together with SST satellite data and SSS-advection from **Paper I**. The calculated results fit the observed variability of DIC and  $A_T$ . The interannual change of the surface  $A_T$  were twice as high as the seasonal variation whereas for surface DIC, the interannual changes were found to be less than the seasonal changes.

An annual cycle of oceanic  $p\text{CO}_2$  between October 2014 and October 2015 is presented in **Paper III**. The highest values occur during summer-autumn and lowest during winter-spring, with a seasonal amplitude of approximately 60  $\mu\text{atm}$  overlaid a high frequency signal of about 10  $\mu\text{atm}$ . More than half of the variability of oceanic  $p\text{CO}_2$  is driven by SST changes, which is in line with findings in **Paper II** that half of the observed change in DIC is due to temperature driven air-sea  $\text{CO}_2$  exchange. The  $p\text{CO}_2$ -SST relationship throughout a year has an elliptical shape, which confirms that beside the temperature influence, other processes also contribute in controlling  $p\text{CO}_2$  changes, e.g. along-coast advection described in **Paper I**. Based on oceanic  $p\text{CO}_2$  and atmospheric data, the area is a net annual sink for atmospheric  $\text{CO}_2$  of size 24.4  $\text{mmol CO}_2 \text{ m}^{-2} \text{ y}^{-1}$ . During summer and autumn, the area act as a source for atmospheric  $\text{CO}_2$ , while during winter and spring the area is a sink for atmospheric  $\text{CO}_2$ . The air-sea gas exchange was examined for the period between 1977-2015. It shows that the area most likely switched from being a net annual source area for atmospheric  $\text{CO}_2$  to becoming a net annual sink during the 2000s.

Most of the work done is pristine and for this reason important, and a useful baseline is made for a region with extreme condition in term of physical, chemical, and ecological properties. Through this baseline, future changes in climate and environment and their predicted trends and impacts can be documented and assessed. In this regard, sustaining the coastal time series initiated by this study will be of extreme importance.

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## 6. Future plans

In light of the vulnerable environment of the Red Sea as well as its natural extreme environmental settings, there are several topics, which are important and interesting to explore in the future. Multiple stressors, e.g. increasing temperature, deoxygenation, and rising atmospheric CO<sub>2</sub> concentration, affect the ocean and its ecosystems. Warming of the ocean will e.g. decrease the gas solubility, and thus, reduce oceanic oxygen concentration, which has been observed at several locations (Shepherd et al., 2017, and references therein). However, the current understanding and implications of such a deoxygenation is not yet resolved.

Further, exploration of precipitation and dissolution of calcium carbonate, CaCO<sub>3</sub>, has not been sufficiently discussed in this thesis due to lack of time and data scarceness. The Red Sea is a region with high CaCO<sub>3</sub> production, and recent studies by Takahashi et al. (2014), Steiner et al. (2014), and Elageed et al. (“Oxygen and alkalinity utilization rates in the Red Sea”, manuscript in preparation) show that such a production affects A<sub>T</sub> through calcification within the pelagic layer and by corals.

At present, ocean acidification is not an imminent threat for the Red Sea. The area is super saturated with respect to calcium carbonate, and thus has relatively high resistance to Ocean Acidification (Elageed, 2010; Omer, 2010). However, this might change in the future, and continuous monitoring of the marine carbon cycle is important.

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## 7. Bibliography

- Broecker, W.S., 1991. The Great Ocean Conveyor. *Oceanography*, 4, 79-89.
- Dickson, A. G., 1981. An exact definition of total alkalinity and a procedure for the estimation of alkalinity and total inorganic carbon from titration data, *Deep-Sea Res.*, 28A:609-623.
- Dickson, A., Goyet, C., 1994. Handbook of methods for the analysis of the various parameters of the carbon dioxide system in sea water, ORNL/CDIAC-74.
- Edwards, F.J., 1987. Climate and oceanography, in: Edwards, A. J., Head, S. M. (Eds), *Red Sea*, Pergamon press, Oxford, pp. 45-70.
- Elageed, S., 2010. Factors controlling the total alkalinity in the Arabian Sea and Red Sea, MS thesis, Univ. of Bergen, Bergen, Norway.
- Emerson, S., Hedges, J., 2008. *Chemical Oceanography and the Marine Carbon Cycle*, 462 pp., Cambridge University Press, Cambridge.
- Field C. B., Raupach, M. R. (Eds), 2004. *The Global Carbon Cycle: Integrating Humans, Climate, and the Natural World*. Island Press, Washington, DC.
- Gordon, L., Jones, L., 1973. The effect of temperature on carbon dioxide partial pressures in seawater, *Marine Chemistry*, 1 (4), 317-322.
- Heinze, C., Maier-Reimer, E., Winn, K., 1991. Glacial pCO<sub>2</sub> reduction by the World Ocean - experiments with the Hamburg Carbon Cycle Model, *Paleoceanography*, 6, 395-430.
- Le Quéré, C., Andrew, R. M., Canadell, J. G., Sitch, S., Korsbakken, J. I., Peters, G. P., Manning, A. C., Boden, T. A., Tans, P. P., Houghton, R. A., Keeling, R. F., Alin, S., Andrews, O. D., Anthoni, P., Barbero, L., Bopp, L., Chevallier, F., Chini, L.P., Ciais, P., Currie, K., Delire, C., Doney, S. C., Friedlingstein, P., Gkritzalis, T., Harris, I., Hauck, J., Haverd, V., Hoppema, M., Klein Goldewijk, K., Jain, A.K., Kato, E., Körtzinger, A., Landschützer, P., Lefèvre, N., Lenton, A., Lienert, S., Lombardozzi, D., Melton, J. R., Metzl, N., Millero, F., Monteiro, P.M.S., Munro, D.R., Nabel, J. E. M. N., Nakaoka, S.-i., O'Brien, K., Olsen, A., Omar, A. M., Ono, T., Pierrot, D., Poulter, B., Rödenbeck, C., Salisbury, J.E., Schuster, U., Schwinger, J., Séférian, R., Skjelvan, I., Stocker, B. D., Sutton, A.J., Takahashi, T., Tian, H., Tilbrook, B., van der Laan-Luijks, I. T., van der Werf, G.R., Viovy, N., Walker, A. P., Wiltshire, A.J., Zaehle, S., **2016**. Global carbon budget 2016, *Earth Syst. Sci. Data*, 8, 605–649,
- Liss, P. S., Merlivat, L., 1986. Air- Sea gas exchange rates: introduction and synthesis .In: *The*

- 
- Role of Air-Sea Exchange in Geochemical Cycling, (ed .P.Buat-Menard).NATO ASI Series C: Mathematical and Physical Science 185,D.Reidel Publishing Company .pp.113-127 .
- Lueker, T.J., Dickson, A. G., Keeling. C. D., 2000. Ocean pCO<sub>2</sub> calculated from dissolved inorganic carbon, alkalinity, and equations for K<sub>1</sub> and K<sub>2</sub>: validation based on laboratory measurements of CO<sub>2</sub> in gas and seawater at equilibrium. *Mar. Chem.* 70: 105– 119.
- Maillard, C., Soliman, G., 1986. Hydrography of the Red Sea and exchanges with the Gulf of Aden in summer. *Oceanologica Acta*, 9 - N°3, 249–269.
- Mehrbach, C., Culberson, C. H., Hawley, J. E., Pytkowicz, R. M., 1973. Measurement of the apparent dissociation constants of carbonic acid in seawater at atmospheric pressure. *Limnology and Oceanography* 18:897-907.
- Morcos, S.A., 1970. Physical and chemical oceanography of the Red Sea, *Oceanography and Marine Biology Annual Review*, 8, 73-202.
- Morcos, S., Soliman, G.F., 1974. Circulation and deep water formation in the Northern Red Sea in winter (Based on R/V Mabahiss sections, January–February, 1935). In *L'Océanographie physique de la Mer Rouge*, pp. 91–103, Centre National pour l'Exploration des Océans (CNEXO), Paris, 91–103.
- Nightingale, P.D., Malin, G., Law, C. S., Watson, A. J., Liss, P. S., Liddicoat, M. I., Boutin, J., Upstill-Goddard, R. C., 2000. In-situ evaluation of air-sea gas exchange parameterizations using novel conservative and volatile tracers, *Global Biogeochem. Cycles*, 14, 373-387.
- Omer, W. M. M., 2010. Ocean acidification in the Arabian Sea and the Red Sea - factors controlling pH MS thesis, Univ. of Bergen, Bergen, Norway.
- Patzert, W. C., 1974. Wind-induced reversal in Red Sea circulation, *Deep Sea Res.*, 21, 109–121.
- Pedgley, D. E., 1974. An outline of the weather and climate of the Red Sea, in *L'Océanographie Physique de la Mer Rouge*, pp. 9 – 27, Cent. Natl. pour l'Exploitation des Océans, Paris.
- Quadfasel, D., Baudner, H., 1993. Gyre-scale circulation cells in the Red Sea, *Oceanol. Acta*, 16, 221-229.

- 
- Raitsos, D. E., Hoteit, I., Prihartato, P. K., Chronis, T., Triantafyllou, G., Abualnaja, Y., 2011. Abrupt warming of the Red Sea, *Geophys. Res. Lett.*, 38, L14601, doi: 10.1029/2011GL047984.
- Sabine, C.L., Feely, R., Gruber, N., Key, R., Lee, K., Bullister, J., Wanninkhof, R., Wong, C. S., Wallace, D. W., Tilbrook, B., Millero, F. J., Peng, T., Kozyr, A., Ono, T., Rios, A., 2004. The oceanic sink for anthropogenic CO<sub>2</sub>, *Science* 305, 367-371.
- Sarmiento, J. L., Gruber, N., 2006. *Ocean Biogeochemical Dynamics*. Princeton University Press, Princeton, NJ, USA.
- Shepherd, J.G., Brewer, P.G., Oschlies, A. Watson, A. J., 2017. Ocean ventilation and deoxygenation in a warming world: introduction and overview, *Phil. Trans. R. Soc A* 375: 20170240, <http://dx.doi.org/10.1098/rsta.2017.0240>.
- Smeed, D. A., 1997. Seasonal Variation of the flow in the strait of Bab el Mandab, *Oceanologica Acta*, 20 - N°6, 773–781.
- Sofianos, S. S., Johns, W. E., 2003. An Oceanic General Circulation Model (OGCM) investigation of the Red Sea circulation: 2. Three-dimensional circulation in the Red Sea, *J. Geophys. Res.*, 108, NO. C3, 3066, doi:10.1029/2001JC001185.
- Steiner, Z., Erez, J., Shemesh, A., Yam, R., Lazar, B., 2014. Basin-scale estimates of pelagic and coral reef calcification in the Red Sea and Western Indian Ocean, *PANS*, Vol. 111, No. 46, 16303-16308.
- Stramma, L., Johnson, G.C., Sprintall, J., and Mohrholz, V., 2008. Expanding Oxygen-Minimum Zones in the Tropical Oceans, *Science*, 320.
- Sweeney, C., Gloor, E., Jacobson, A. R., Key, R. M., McKinley, G., Sarmiento, J. L., Wanninkhof, R., 2007. Constraining global air-sea gas exchange for CO<sub>2</sub> with recent bomb 14C measurements, *Global Biogeochem. Cycles*, 21, GB2015, doi:10.1029/2006GB002784.
- Takahashi, T., Olafsson, J., Goddard, J., Chipman, D.W., Sutherland, S.C., 1993. Seasonal variation of CO<sub>2</sub> and nutrients in the high-latitude surface oceans: A comparative study. *Glob. Biogeochem. Cycles* 7, 843–878.
- Takahashi, T., Sutherland, S. C., Chipman, D. W., Goddard, J. G., Ho, C., Newberger, T., and others. 2014. Climatological distributions of pH, pCO<sub>2</sub>, total alkalinity, and CaCO<sub>3</sub> saturation in the global surface ocean, and temporal changes at selected locations. *Mar. Chem.* 164:95–125.

- 
- Wanninkhof, R., 1992. Relationship between gas exchange and wind speed over the ocean, *J. Geophys. Res.*, 97, 7373-7381.
- Wanninkhof, R., McGillis, W. M., 1999. A cubic relationship between gas transfer and wind speed, *Geophys. Res. Lett.*, 26, 1889-1893.
- Weiss, R. F., 1974. Carbon dioxide in water and sea water: the solubility of a non-ideal gas, *Mar. Chem.*, 2:203-215.
- Yao, F. C., Hoteit, I., Pratt, L. J., Bower, A. S., Zhai, P., Kohl, A., Gopalakrishnan, G., 2014a. Seasonal overturning circulation in the Red Sea: 1. Model validation and summer circulation. *Journal of Geophysical Research-Oceans*. 119:2238-2262.
- Yao, F., Hoteit, I., Pratt, L. J., Bower, A. S., Kohl, A., Gopalakrishnan, G., Rivas, D., 2014b. Seasonal overturning circulation in the Red Sea: 2. Winter circulation, *J. Geophys. Res. Oceans*, 119, 2263–2289, doi:10.1002/2013JC009331.
- Zeebe, R.E., Wolf-Gladrow, D., 2001. *CO<sub>2</sub> in Seawater: Equilibrium, Kinetics, Isotopes*, Elsevier Oceanography Series: 2001.
- Zeebe, R. E., 2012. History of seawater carbonate chemistry, atmospheric CO<sub>2</sub>, and ocean acidification, *Annu. Rev. Earth Planet. Sci.*, 40(1).





# Paper I

## **Seasonal and interannual variations of hydrographic parameters in the Sudanese coast of the Red Sea, 2009-2015**

Ali, E. B., J. H. Churchill, K. Barthel, I. Skjelvan, A. M. Omar, T. E. de Lange, and E. B. A. Eltaib

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1     **Seasonal and interannual variations of hydrographic parameters in**  
2     **the Sudanese coast of the Red Sea, 2009-2015**  
3  
4

5     Elsheikh B. Ali<sup>1,2</sup>, James H. Churchill<sup>3</sup>, Knut Barthel<sup>1</sup>, Ingunn Skjelvan<sup>4, 1</sup>, Abdirahman M.  
6     Omar<sup>4, 1</sup>, Tor E. de Lange<sup>1</sup>, and Elfatih B. A. Eltaib<sup>2</sup>

7     <sup>1</sup>Geophysical Institute (GFI), University of Bergen (UiB), Bergen, Norway

8     <sup>2</sup>Institute of Marine Research (IMR), Red Sea University (RSU), Port Sudan, Sudan

9     <sup>3</sup>Department of Physical Oceanography, Woods Hole Oceanographic Institution, Woods Hole, Massachusetts  
10    02543, USA

11    <sup>4</sup>Uni Research Climate, Bergen, Norway  
12  
13

14    **Abstract:**

15    The temporal variations of temperature and salinity in the Sudanese coast of the Red Sea have  
16    been studied based on new time series acquired over 2009-2015 from a mooring and from survey  
17    cruises. The observations show that temporal variations in temperature and salinity above the  
18    main pycnocline are dominated by seasonal signals. Highest temperature of approximately 32°C  
19    occur during summer and early autumn and lowest temperature of roughly 26°C are seen in  
20    winter. The seasonal salinity signal lags that of temperature by roughly 3 months, and varies from  
21    approximately 38.5 psu in late spring and early summer to 39-40 psu in late autumn and early  
22    winter. Using estimates of heat flux, circulation and horizontal temperature/salinity gradients  
23    derived from a number of sources, we determined that the observed seasonal temperature and  
24    salinity could not be the product of local heat and mass flux alone, but also due to advection of  
25    alongshore temperature and salinity gradients.  
26

27    **Keywords**

28    Coastal Red Sea; temperature; salinity; time series; seasonality; alongshore advection  
29

30    **1. Introduction**

31    Encompassing a highly diverse ecosystem, the Red Sea ranks as one of the warmest and most  
32    saline of the world's seas. Fluctuations in near-surface temperature and salinity within Red Sea  
33    coastal waters can have profound ecological consequences, particularly for the numerous reef  
34    systems that fringe the Red Sea basin. It is well established that the growth rate and overall health

35 of coral communities are highly sensitive to changes in temperature and affected to a lesser  
36 degree by variations in salinity (e.g., Ferrier-Pages et al., 1999; Furby et al., 2013; Kuanui et al.,  
37 2015). It is also well established that changes in temperature and salinity affect the inorganic  
38 carbon cycle; e.g. warmer water reduces the gas solubility and thus increases the flux of carbon  
39 dioxide from the surface water into the atmosphere, and changes in alkalinity are tightly  
40 connected to changes in salinity (e.g. Sarmiento and Gruber, 2006, and references therein).

41 There are currently few published studies of the temporal variations of temperature or  
42 salinity within the Red Sea. Churchill et al. (2014) examined temperature fluctuations waters of  
43 the coastal zone of the central Red Sea off the Saudi Arabian coast using data acquired from  
44 moorings deployed at ~50 m depth. They showed that near-surface (upper 15 m) temperature  
45 variations span a range of order 8°C and are predominately due to a seasonal signal with a range  
46 of order 6°C. A seasonal near-surface temperature signal with a similar range was reported by  
47 Sultan and Ahmad (1991) based on sea surface measurements acquired off of Jeddah Saudi  
48 Arabia, by Berman et al. (2003) based on hydrographic data acquired in summer and winter, and  
49 by Davis et al. (2011) based on temperature sensors placed on platform reef tops. Published  
50 reports of temporal salinity variations in the Red Sea are very rare. Churchill et al. (2014) briefly  
51 discussed salinity records from the moorings references above but showed no time series. Sultan  
52 and Ahmad (1991) presented an 8-year record of monthly-averaged surface salinities acquired off  
53 of Jeddah Saudi Arabia that span a range of order 1 psu, but noted that the salinities may have  
54 been affected by discharge from the Jeddah desalination facility.

55 It is notable that all of the studies reviewed above utilized data from the coastal zone of  
56 the central Red Sea west of the Saudi Arabian coast and so may not be representative of  
57 temperature and salinity variations over the full Red Sea. Furthermore, no study has dealt fully  
58 with the dynamics responsible for the observed temperature and salinity variations. While some  
59 studies have related temperature variations to surface heat flux (Sultan and Ahmad, 1991;  
60 Berman et al., 2003; Churchill et al., 2014), the manner in which temperature and salinity  
61 variations are influenced by water mass transport has not yet been assessed.

62 The work reported here is aimed at furthering the understanding of temperature and  
63 salinity variations in the coastal Red Sea. Using data from moored instruments and hydrographic  
64 surveys, we describe the temporal variations in temperature and salinity within near-surface  
65 coastal waters off of Port Sudan, Sudan (Fig. 1). With the aid of reanalysis and satellite-derived

66 data, we examine the mechanisms responsible for these variations, concentrating on the relative  
67 importance of local processes (surface heat and mass flux) and water mass advection. Our focus  
68 is on the seasonal signal of temperature and salinity. As noted above, previous observations have  
69 shown that the range of near-surface temperature variation in the central Red Sea is primarily due  
70 to the seasonal signal.

71 In the following sections, we first describe the data sets and methodology employed  
72 (Section 2). We then detail the temperature and salinity variations in the coastal zone off of Port  
73 Sudan (Section 3.1) and examine the dominant mechanisms responsible for these variations  
74 (Section 3.2). We conclude with a summary of our findings and a discussion of how they relate to  
75 prior work on the dynamics of the Red Sea system (Section 4).

76

## 77 **2. Data and Methods**

78 Our analysis employed six datasets. Three were used to describe the seasonal variation of  
79 temperature, salinity and potential density ( $\sigma_\theta$ ) off of Port Sudan, and three were employed in  
80 assessing the mechanisms responsible for the seasonal variation of these properties. Below, we  
81 detail these data sets and present our methods for estimating near-surface temperature and salinity  
82 change.

83

### 84 **2.1 Measurements of coastal temperature and salinity**

#### 85 *2.1.1 Cruise data*

86 Our study employed cruise measurements of temperature and salinity acquired at two set of  
87 stations (Fig. 1): near-shore stations in the area of Port Sudan Harbour (PSH), and stations  
88 situated along a transect between Port Sudan and the Sanganeb atoll reef (30 km northeast of Port  
89 Sudan) (the SPS stations). The data acquired at all stations were from a SAIV A/S model 204  
90 CTD, deployed using a hand winch and equipped with an inductive cell conductivity sensor  
91 (resolution  $0.01 \text{ mS cm}^{-1}$ , accuracy  $\pm 0.02 \text{ mS cm}^{-1}$ ), a temperature sensor (resolution  $0.001$ ,  
92 accuracy  $\pm 0.01^\circ\text{C}$ ), and a pressure sensor (resolution  $0.01 \text{ mbar}$ , accuracy  $\pm 0.02\%$ ).

93 The salinity was computed to a resolution of  $0.01 \text{ psu}$  and to an accuracy of  $\pm 0.02 \text{ psu}$   
94 (Operating Manual for SAIV CTD model 204, 2006). Calibration of the salinity measurements  
95 was done using water samples acquired during a subset of the CTD casts by a Hydrobios 2-liter

96 water sampler. The bottle salinity was determined using the Guildline Portasal Salinometer  
97 (model 8410A) to an accuracy of  $\pm 0.003$  psu (Datasheet for Guildline 8410A Portasal, 2002).

98 Measurements at the PSH stations encompassed the period of 2010-2013 and extended  
99 vertically from near the surface to roughly 150 m depth. Measurements from the SPS transect  
100 were from 2009-2013 and vertically extended to roughly 200 m. Both sets of data resolved the  
101 seasonal signal of temperature and salinity (Fig. 2). The SPS data were taken at a quarterly  
102 interval (collected roughly during October, February, April, and July), whereas the PSH data  
103 interval was shorter but varied considerably.

104

### 105 ***2.1.2 Mooring data***

106 Additional temperature and salinity time series data of our study were acquired from a Sea Bird  
107 CTD (SBE 37-SM MICROCAT, SN 3939) affixed to a mooring deployed on 1 October 2014 at  
108 station 4 (37.395°E, 19.720°N; Fig. 1) of SPS transect. The bottom depth at the mooring location  
109 was  $\sim 800$  m, and the CTD, which was positioned at a nominal depth of 37 m, was set to record at  
110 hourly intervals. The mooring was recovered on 15 October 2015. The CTD was equipped with  
111 conductivity (resolution 0.0001, accuracy  $\pm 0.003$  mS  $\text{cm}^{-1}$ ), temperature (resolution 0.0001,  
112 accuracy  $\pm 0.002^\circ\text{C}$ ), and pressure (resolution 0.002%, accuracy  $\pm 0.1\%$ ) sensors. The salinity  
113 measurements were calibrated using bottle salinity measurements determined as described above.

114

### 115 ***2.1.3 Satellite-derived sea surface temperature***

116 The sea surface temperature (SST) data used in this study were from the NOAA High-resolution  
117 Blended Analysis of Daily SST (Version 2) dataset. Formulated by combining observations from  
118 different platforms (satellites, ships, buoys) (Reynolds et al., 2007), the dataset is comprised of  
119 daily temperatures specified on a  $0.25^\circ$  latitude by  $0.25^\circ$  longitude global grid.

120

## 121 **2.2 Data used to compute heat and salt fluxes**

### 122 ***2.2.1 The Simple Ocean Data Assimilation (SODA) dataset***

123 The SODA dataset is comprised of data produced by an ocean general circulation model with an  
124 average resolution of  $0.25^\circ$  latitude by  $0.4^\circ$  longitude, and 40 vertical levels. Direct  
125 contemporaneous observations are continuously used to correct the model error of the generated  
126 ocean variables (Carton et. al., 2008). We employed SODA data to estimate horizontal salinity

127 and temperature gradients, which were used in the computation of heat and salt fluxes due to  
128 horizontal advection.

129

### 130 **2.2.2 NCEP datasets**

131 We extracted data from the National Centers for Environmental Prediction (NCEP) / National  
132 Center for Atmospheric Research (NCAR) dataset to estimate the surface wind stress as well as  
133 surface heat and mass flux in our study region. All data were from a 2° latitude by 2° longitude  
134 cell centered at 37.50°E, 20.00°N. The wind stress was computed from 10 m NCEP Reanalysis-2  
135 wind velocities according to the formulae of Large and Pond (1981).

136 To compute the surface heat flux, we downloaded individual surface heat flux  
137 components: net shortwave radiation ( $Q_{sw}$ ), net longwave radiation ( $Q_{lw}$ ), latent heat flux ( $Q_l$ )  
138 and sensible heat flux ( $Q_s$ ). Net surface heat flux ( $Q_{net}$ ) was determined according to (Wallcraft  
139 et al., 2008):

$$140 \quad Q_{net} = Q_{sw} - Q_{lw} + Q_l + Q_s \quad (1)$$

141

142 Net surface mass flux ( $m\ s^{-1}$ ) was computed as:

$$143 \quad M = E - P \quad (2)$$

144

145 where  $E$  is the evaporation rate and  $P$  is the precipitation rate.  $P$  was computed from the NCEP  
146 Reanalysis-2 precipitation rate ( $P_R$ ,  $kg\ m^{-2}\ s^{-1}$ ) according to:

$$147 \quad P = \frac{P_R}{\rho_0} \quad (3)$$

148 where  $\rho_0$  is the fresh water density.  $E$  was computed from the latent heat flux according to  
149 (Sumner and Belaineh, 2005):

$$150 \quad E = \frac{-Q_l}{\rho_0 \lambda} \quad (4)$$

151 where  $\lambda$  is latent heat of vaporization of water ( $2.3 \times 10^6\ J\ kg^{-1}$ ).

152

153 To assess the NCEP heat flux estimates, we compared the net heat flux series determined using  
154 measurement from an air–sea interaction buoy maintained in the central Red Sea (at 38° 30.1'E;  
155 22° 9.6'N) over 2008-2010 (Farrar et al., 2009; Bower and Farrar, 2015) with a net heat flux time  
156 series determined from NCEP data for the cell which included the buoy location. The two series

157 were highly correlated ( $R^2=0.89$ ) and exhibited closely matched seasonal cycles, with net heat  
 158 flux into the ocean over March through September and net heat loss over the rest of the year. The  
 159 net heat transfer over the two years of the comparison (2009-2010) was negative for both series,  
 160 but was slightly larger in magnitude for the buoy-derived series ( $-1.2 \times 10^9$  versus  $-0.9 \times 10^9$  J m<sup>2</sup>).  
 161

162

### 163 **2.2.3 Sea Level Anomaly (SLA)**

164 Estimation of near-surface velocities was done using altimeter-derived sea level anomaly (SLA)  
 165 fields computed with respect to the CLS01 (Centre de Localisation des Satellites) long-term mean  
 166 sea surface height. Downloaded from AVISO (<http://www.aviso.oceanobs.com/>), the SLA fields  
 167 were determined by combining sea level data from all available satellites and objectively  
 168 mapping SLA on a  $0.25^\circ$  latitude by  $0.25^\circ$  longitude grid. We computed geostrophic surface  
 169 velocity from the gridded SLA field through:

$$170 \quad u_G = -\frac{g}{f} \frac{\partial \delta}{\partial y}; v_G = \frac{g}{f} \frac{\partial \delta}{\partial x} \quad (5)$$

171 where  $\delta$  is SLA,  $u_G$  and  $v_G$  are the east and north components of geostrophic velocity,  
 172 respectively,  $g$  is the gravitational acceleration and  $f$  is local Coriolis parameter.  
 173

### 174 **2.3 Estimation of near-surface temperature and salinity changes**

175 With the data described above, we sought to roughly assess the contributions of surface heat and  
 176 mass fluxes and horizontal advection in driving the observed seasonal signal of near-surface  
 177 temperature and salinity in our study region. We did not consider the effects of mixing or vertical  
 178 advection as these were not well suited for study with the available data. Our focus was on  
 179 temperature and salinity changes in the layer above the permanent pycnocline. For simplicity, we  
 180 assumed that this layer extended to a constant depth,  $h$ . We also assumed that the advective  
 181 changes in temperature and salinity were principally due to fluxes in the alongshore (roughly N-S  
 182 in the area of Port Sudan) direction. With these assumptions, the changes in temperature ( $T$ ) and  
 183 salinity ( $S$ ) averaged over the surface layer may be approximated as:

184

$$185 \quad \frac{\partial T}{\partial t} = \frac{Q_{net}}{h\rho_s C_p} - V \frac{\partial T}{\partial y} \quad (6)$$

186



187 
$$\frac{\partial S}{\partial t} = \frac{M}{h} S - V \frac{\partial S}{\partial y} \quad (7)$$

188 where  $C_p$  is the specific heat capacity of water ( $4.2 \times 10^3 \text{ J kg}^{-1}$ ),  $y$  is the alongshore coordinate,  $\rho_s$   
 189 is the near-surface density and  $V$  is the vertically-averaged alongshore velocity in the layer above  
 190 the main pycnocline.

191 Our approach was to determine the near-surface temperature and salinity signal from  
 192 observed starting values of  $T_o$  and  $S_o$ , respectively. For simplicity, we took the horizontal  
 193 temperature and salinity gradients as constants (determined from the SODA data as explained in  
 194 Section 2.2). With these assumptions, the seasonal temperature and salinity signals were  
 195 estimated from

196  
 197 
$$T(t) = T_0 + \int_0^t \left[ \frac{Q_{net}(\tau)}{h\rho_s c_p} - V(\tau) \frac{\partial T}{\partial y} \right] d\tau \quad (8)$$

198 
$$S(t) = S_0 + \int_0^t \left[ \frac{M(\tau)}{h} S(\tau) - V(\tau) \frac{\partial S}{\partial y} \right] d\tau. \quad (9)$$

199

## 200 **3. Results**

### 201 **3.1 Seasonal variation**

#### 202 **3.1.1 SPS and PSH**

203 The measurements from the coastal hydrographic surveys (SPS and PSH) clearly show a seasonal  
 204 signal in near-surface values of temperature, salinity and  $\sigma_\theta$  (Fig. 2). The near-surface  
 205 temperature signal is marked by maxima of close to  $32^\circ\text{C}$  during summer and early autumn and  
 206 minima of roughly  $26^\circ\text{C}$  in winter. The timing and range of this signal closely match those of the  
 207 seasonal temperature signal shown by Churchill et al. (2014) based on moored measurements  
 208 acquired in the coastal zone of the central Red Sea. In particular, Churchill et al. (2014) show  
 209 highest near-surface temperatures, of roughly  $32^\circ\text{C}$ , over July-October contrasting with the  
 210 lowest near-surface temperatures, of roughly  $26^\circ\text{C}$ , over January-March (their Fig. 2). The  
 211 seasonal near-surface salinity signal lags the temperature signal by roughly 3 months with  
 212 maxima (39.0-39.5 psu at SPS and 39-40 psu at PSH) in late autumn/early winter and minima  
 213 (approximately 38.5 psu in both areas) in late spring/early summer. The  $\sigma_\theta$  seasonal signal is  
 214 roughly the inverse of the temperature signal, with the densest water ( $\sigma_\theta = 26\text{-}27 \text{ kg m}^{-3}$ )

215 appearing in winter whereas the lowest density water ( $24\text{-}25 \text{ kg m}^{-3}$ ) is seen in late summer/early  
216 autumn. The  $\sigma_\theta$  data also reveal a seasonal variation in density vertical stratification over the  
217 upper 50 m characterized by nearly uniform  $\sigma_\theta$  in winter and stronger stratification over the rest  
218 of the year. During all seasons, the top of the main pycnocline, roughly marked by the 26  $\sigma_\theta$   
219 contour, appears at about 50 m depth (Fig. 2).

220 Seasonal  $\theta$ -S diagrams (Fig. 3a-d) clearly show a yearly progression in vertical  
221 stratification in the survey region. The  $\sigma_\theta$  range over the vertical extent of the surveys is greatest  
222 during summer, extending from roughly  $23.5 \text{ kg m}^{-3}$  at the surface to  $28.45 \text{ kg m}^{-3}$  at 200 m.  
223 During summer, the density stratification in the upper 50 m, above  $\sigma_\theta = 26 \text{ kg m}^{-3}$ , is largely due  
224 to temperature stratification. The  $\sigma_\theta$  range observed during autumn is slightly smaller,  $23.9\text{-}28.5$   
225  $\text{kg m}^{-3}$ . The higher near-surface densities of autumn relative to summer are principally due to the  
226 higher salinities observed in autumn. At  $\sigma_\theta < 26 \text{ kg m}^{-3}$ , salinities of the autumn surveys are  
227 roughly 0.5 psu greater than those of the summer surveys. The near-surface layer of strong  
228 temperature stratification seen in the summer and autumn surveys is absent during the winter  
229 surveys. The weak vertical temperature stratification observed in winter results in weak vertical  
230 density stratification, with  $\sigma_\theta$  varying by about  $0.5 \text{ kg m}^{-3}$  in the upper 50 m. Near-surface  
231 vertical temperature and density gradients are slightly greater during the spring surveys, signaling  
232 the formation of the seasonal thermocline. It's noteworthy that during all seasons the largest  
233 temporal variations in  $\theta$ , S and  $\sigma_\theta$  occur above the permanent pycnocline (roughly at  $\sigma_\theta < 26 \text{ kg}$   
234  $\text{m}^{-3}$ ), with much smaller variations of these properties seen in the permanent pycnocline and  
235 below.

236

### 237 **3.1.2 Mooring and SST data**

238 The temperature and salinity records from the mooring (Fig. 4) show seasonal signals in close  
239 agreement with those exhibited by the cruise data. In particular, the mooring data show the order  
240 3-month shift of the seasonal signal of salinity relative to temperature. The shift is particularly  
241 well defined by the temperature and salinity maxima, which occur, respectively, in late  
242 September 2015 and mid-December 2014. However, the shift is not as well defined by the  
243 minima of temperature and salinity. The seasonal decline in temperature occurs over the autumn  
244 and early winter and terminates in a clear minimum in mid-January 2015. By contrast, the  
245 seasonal salinity decline occurs over December 2014 - March 2015 and is not followed by a

246 clear salinity minimum over the ensuing four months. Nevertheless, both the salinity and  
247 temperature records show a dominant seasonal signal relative to higher frequency variations. The  
248 seasonal temperature signal extends over 6°C, roughly from 25 to 31°C, while the higher  
249 frequency variations are of magnitude 2°C or less. The seasonal salinity signal is approximately  
250 1.1 psu in magnitude, roughly from 38.7 to 39.8 psu, upon which fluctuations of order 0.4 psu  
251 are superimposed.

252 The distribution of  $\theta$ -S derived from the mooring data (Fig. 3f) show a seasonal  
253 progression of near-surface density related to the seasonal variation of temperature and salinity.  
254 The increase in near-surface density from its minimum in October to its maximum in January is  
255 largely the product of declining temperatures, while the subsequent decline in near-surface  
256 density over January-April is principally due to a decrease in near-surface salinity. Completing  
257 the cycle, the May-October decrease in density is primarily the result of rising near-surface  
258 temperatures.

259 Comparing the moored temperature record with the satellite-derived SST record from the  
260 0.25° latitude by 0.25° longitude cell encompassing the mooring indicates predominately well-  
261 mixed or weakly stratified conditions above the moored CTD (Fig. 4a). Significant temperature  
262 stratification above the mooring (indicated by a difference between the SST and mooring  
263 temperature) is confined to the period of early-May through mid-August. However, the near-  
264 surface stratification is eroded on a number of occasions during this period (i.e. in mid-May,  
265 early-June, and mid-July 2015). These mixing events appear to be at least partly due to the action  
266 of the surface wind stress and they correspond to peaks in the surface wind stress record derived  
267 from the NCEP winds (Fig. 4c).

268 The long-term (6-year) record of SST from the cell encompassing the mooring (Fig. 5a)  
269 very clearly shows the dominance of the seasonal signal over higher frequency variations. The  
270 seasonal SST signal spans a range of roughly 6°C, between minima and maxima of 25 and 31°C,  
271 whereas the magnitude of the higher frequency SST variations is at most 1°C.

272

### 273 **3.2 Factors controlling the seasonal signals**

274 The observations reviewed above clearly show that the temporal variations of temperature and  
275 salinity above the permanent pycnocline in the near-shore region off of Port Sudan are  
276 dominated by seasonal signals that differ in phase by roughly 3 months (with the seasonal

277 temperature signal leading). We now consider the extent to which these seasonal signals may be  
278 due to surface mass and heat flux (local processes) and to the effect of alongshore advection. We  
279 focus first on the impact of local processes on the seasonal signals.

280

### 281 **3.2.1 Local processes**

282 The change in mean temperature and salinity above the main pycnocline was estimated by  
283 evaluating Eq. 8 and 9 with inclusion of only the first term in the integral of each equation. The  
284 depth of the layer above the main pycnocline,  $h$ , was approximated as 50 m based on the  
285 contoured temperature and salinity fields derived from survey data (Fig. 2).

286 The computed temperature driven by local heat flux exhibits a seasonal signal similar to  
287 observed temperature fields but with the minima and maxima occurring 1-2 months later (Figs.  
288 4a and 5a). In addition the computed temperature series shows a long-term decline (of roughly  
289  $0.8^{\circ}\text{C yr}^{-1}$ ) that is not matched by the observations (Fig. 5a). This trend in the computed  
290 temperature cannot be attributed to the omission of vertical mixing, as mixing with the cooler  
291 water below the main pycnocline (Fig. 2) would tend to reduce near-surface temperatures even  
292 further.

293 Reflecting the dominance of evaporation over precipitation, the computed salinity series  
294 driven only by surface mass flux shows a steady increase with time and no vestige of a seasonal  
295 signal (Figs. 4b and 5b). Altering this trend to match the salinity observations requires both a  
296 long-term delivery of lower salinity near-surface water to the region as well as a mechanism to  
297 produce the observed seasonal oscillations.

298

### 299 **3.2.2 The effect of alongshore advection**

300 Evaluating the impact of alongshore advection on the seasonal temperature and salinity signals  
301 from Eq. 8 and 9 required estimates of  $V(t)$ ,  $dS/dy$ , and  $dT/dy$ . In estimating the latter two  
302 properties, we used SODA temperature and salinity data from 2009-2010. These data show that  
303 near-surface temperatures (averaged over the upper 50 m) tend to decline going northward over  
304 the central and northern Red Sea (Fig. 6a). A similar trend has been observed based on analysis  
305 of SST data (Raitzos et al., 2013) and hydrographic survey measurements (Neumann and McGill,  
306 1962; Maillard and Soliman, 1962; Sofianos and Johns, 2007). Also shown by the SODA data is  
307 a tendency for the near-surface salinity to increase going northward over the entire Red Sea (Fig.

308 6b), consistent with trends observed by hydrographic survey data (Neumann and McGill, 1962;  
 309 Maillard and Soliman, 1962; Sofianos and Johns, 2007). To estimate  $dS/dy$  and  $dT/dy$  in our  
 310 study area, we used temperature and salinity data from SODA grid cells extending between 18  
 311 and 22.5°N and arranged in roughly the alongshore direction (Figs. 6a and b). Averages (over the  
 312 upper 50 m) of near-surface temperature and salinity from these cells show alongshore gradients  
 313 that do not appear to vary appreciably with season (Figs. 6c and d). Based on these averages, we  
 314 assigned values to  $dT/dy$  and  $dS/dy$  of  $-0.2/111^\circ\text{C km}^{-1}$  and  $0.35/111 \text{ psu km}^{-1}$ , respectively (i.e.,  
 315  $0.2^\circ\text{C}$  decrease and a  $0.35 \text{ psu}$  increase over a degree of latitude).

316 In specifying  $V(t)$ , we assumed that the velocity signal impacting the seasonal  
 317 temperature and salinity signals varied on a similar time scale as these signals and can be  
 318 expressed as

$$319 \quad V(t) = a \sin \left[ \frac{2\pi t}{P} + \phi \right] + V_o \quad (10)$$

320  
 321 where  $a$  is the sinusoidal flow amplitude,  $P$  is the period of 1 year,  $V_o$  is the long-term mean flow  
 322 and  $\phi$  is a phase relative to the beginning of each year. Surface geostrophic velocities determined  
 323 from the SLA data tend to support this form of a seasonal velocity signal in that they show a  
 324 tendency for alongshore velocity to be directed northward over winter-spring and southward  
 325 over summer-autumn (Figs. 4d and 5c). To assign the required parameters in Eq. 10, we used a  
 326 6-year (2009-2014) series of alongshore (N-S) geostrophic velocity determined from SLA data in  
 327 the vicinity of our study area (19-20.5°N; 37.5-38°E). Applying a nonlinear least-squares  
 328 regression technique (with MATLAB function *nlinfit*) to fit Eq. 10 to this time series gave  
 329 estimates of  $a = 0.047 \text{ m s}^{-1}$ ,  $V_o = 0.017 \text{ m s}^{-1}$  and  $\phi = 0.136$  (zero crossings at June 23 and  
 330 December 23).

331 Inclusion of the advection terms in Eqs. 8 and 9 has two principal effects on the  
 332 computed temperature and salinity series. One is that the mean influx of warmer and less saline  
 333 water from the south, carried by the steady northward flow, counteracts the long-term trend of  
 334 declining temperatures and rising salinities seen in the series computed with only the local  
 335 surface flux terms (Figs. 4 and 5). These trends are not apparent in the series computed with the  
 336 addition of the advection term. The second effect is produced by the yearly oscillation in  
 337 alongshore velocity and most profoundly affects the computed salinity signal, which acquires a  
 338 seasonal variation with the inclusion of the advection term (Figs. 4b and 4d).

339           It is unrealistic to expect a close match between the observed and computed temperature  
340 and salinity series as our crude calculation omits many factors that may influence near-surface  
341 temperature and salinity. These include, but are not limited to, mixing of water across the main  
342 pycnocline, temporal variations in the alongshore temperature and salinity gradients, differences  
343 between the actual flows and our representation of the alongshore velocity signal and the effect  
344 of across-shore advection. Nevertheless, the seasonal signals of the temperature and salinity  
345 series computed with the inclusion of the advection terms exhibit many of the features of the  
346 observed seasonal signals. In close agreement with observed temperatures, the computed  
347 temperature series has a seasonal signal spanning a range of roughly 6°C, with minima close to  
348 26°C in late-winter/early spring and maxima near 32°C predominately occurring in late autumn  
349 (Figs. 4a and 5a). The seasonal signal of the computed salinity series resembles observed  
350 salinity signal in that it extends over a range of roughly 1.2 psu, with maxima winter and minima  
351 in early summer (Figs. 4b and 5b). Notably, the observed 3-month phase difference between the  
352 observed seasonal temperature and salinity series (with temperature leading) is reproduced by  
353 the computed series (Fig. 5).

354           However there are differences between the computed and observed seasonal temperature  
355 and salinity signals worth noting. The seasonal variation of the computed temperature series is  
356 somewhat smaller (by  $\sim 1^\circ\text{C}$ ) than the range of the observed temperatures (Figs. 4a and 5a). The  
357 seasonal signal of the computed salinity series roughly matches that observed at the mooring, but  
358 does not show the abrupt decline in salinity observed in February 2015 (Fig. 4b). The long-term  
359 salinity signal determined from the SPS survey data is roughly in phase with the computed  
360 salinity signal, but varies over a much smaller range (Fig. 5b). We can offer no clear reason as to  
361 why the range of the survey-averaged salinities is smaller than the range of the salinities derived  
362 from the computations and the mooring data. One possibility is that the survey-derived salinities,  
363 which are essentially point measurements, do not capture the full range of the seasonal signal as  
364 they are aliased by salinity variations on time scales shorter than that of the seasonal signal.

365

#### 366 **4. Discussion and Conclusions**

367 As noted in Section 3.1.1, our observation is not the first of a dominant seasonal signal in near-  
368 surface temperature records from the Red Sea, as this was previously reported by Churchill et al.  
369 (2014). The similarity of the seasonal oscillations of near-surface temperature observed in our

370 study with those detected further to the north and on the opposite side of the Red Sea by  
371 Churchill et al. (2014) suggests that such oscillations may be a ubiquitous over the central Red  
372 Sea. Our analysis has revealed that these oscillations cannot be solely ascribed to local surface  
373 heat exchange, as this would produce a long-term heat loss and a multiyear decline in near-  
374 surface temperature. According to the results of our simple heat-flux model, the long-term trend  
375 of heat loss through surface exchange near Port Sudan is largely balanced by the advection of  
376 warmer water from the south. In our model, this advection is the product of a long-term mean  
377 northward flow, inferred from analysis of SLA fields, and a tendency for temperatures to increase  
378 going southward from Port Sudan, as deduced from SODA data. Presently, there are no published  
379 long-term velocity records to verify the existence of a long-term mean northward flow off of Port  
380 Sudan. It is noteworthy, however, that such a flow often appears in the results of hydrodynamic  
381 models of the Red Sea, taking the form of a western boundary current flowing northward over the  
382 southern and central Red Sea (Sofianos and Johns, 2003; Yao et al., 2014b; Zhai et al., 2015).  
383 The alongshore temperature gradient inferred from the SODA data is associated with a near-  
384 surface temperature maximum in the central/southern Red Sea south of Port Sudan. As noted in a  
385 review by Morcos (1970), this feature of the Red Sea surface temperature field has been  
386 recognized since the early twentieth century. More recently, it has appeared in large-scale survey  
387 data (Maillard and Soliman, 1986; Sofianos and Johns, 2007), SST fields derived from satellite  
388 measurements (Raitso et al., 2013) and hydrodynamic model results (Sofianos and Johns, 2003).  
389 In interpreting their model results, Sofianos and Johns (2003) ascribe the surface temperature  
390 maximum to relatively weak winds in the central Red Sea area of wind convergence.

391 Our observations show that near-surface salinity variations off of Port Sudan are also  
392 dominated by a seasonal signal. Because of the dominance of evaporation over precipitation in  
393 the Red Sea, the observed oscillations in near-surface salinity cannot be attributed to local mass  
394 flux at the surface, as this would produce a nearly steady increase in near-surface salinity. Our  
395 simple salt-flux model indicates that this tendency for salinity to increase due to local evaporation  
396 is largely balanced by a northward flux of less saline water from the south. The trend of  
397 increasing near-surface salinity going northward over the Red Sea is commonly seen in both  
398 observations and model results (Morcos, 1970; Clifford et al., 1997; Sofianos and Johns, 2003,  
399 2007; Yao et al., 2014a,b). This northward salinity increase has largely been attributed to the  
400 influx of relatively low salinity water into the southern Red Sea through the Strait of Bab al-

401 Mandeb. It is well documented that this influx takes two forms. During summer and early  
402 autumn, low-salinity water, commonly referred to as Gulf of Aden Intermediate Water, enters the  
403 Red Sea in sub-surface (30-120 m) layer (Patzert, 1974; Murray and Johns, 1997). During the rest  
404 of the year, low salinity water, commonly referred to as Gulf of Aden Surface Water, enters the  
405 Red Sea over a surface layer of order 50 m depth (Murray and Johns, 1997; Smeed, 2004).

406 Our analysis indicates that the yearly oscillations in near-surface salinity off of Port Sudan  
407 may be largely due to the advection of the alongshore salinity gradient by yearly oscillations of  
408 the alongshore velocity. As noted by Sofianos and Johns (2003), seasonal variations in Red Sea  
409 flow patterns are likely as in the Red Sea ‘both wind and thermohaline forcing are highly variable  
410 at the seasonal timescales’. Seasonal averages of their modeled flows in the Port Sudan area are  
411 consistent with the alongshore velocity component derived from the SLA data (Fig. 5), directed  
412 northward over winter (September-May) and southward over the summer (June-August) (Figs. 4  
413 and 5 of Sofianos and Johns, 2003).

414 Prominent among the flux terms not included in our analysis are those associated with  
415 vertical mixing and across-shore advection. Although we cannot estimate vertical mixing with the  
416 data used in our study, we can assert that vertical mixing through the main pycnocline would not  
417 counteract the long-term trend of declining temperatures and rising salinity associated with local  
418 heat and mass exchange through the surface. Because cooler and more saline water is found  
419 below the pycnocline, vertical mixing through the pycnocline would tend to further reduce  
420 temperature and increase salinity in near surface waters.

421 Given the prevalence of basin-scale eddies within the central Red Sea (Zhan et al., 2014),  
422 it is likely that eddies may frequently cause exchange of near-shore and basin water within the  
423 central Red Sea. However, because the eddy lifespan is typically 6 weeks (Zhan et al., 2014), this  
424 exchange is likely to produce temperature and salinity changes at intervals relatively short  
425 compared with the observed seasonal signals of temperature and salinity.

426 Although our study has provided new insight into the character and dynamics of seasonal  
427 temperature and salinity changes in the Red Sea, it has been based on limited data from a small  
428 region and has not specifically dealt with the full suite of dynamics that may influence seasonal  
429 temperature and salinity changes. For example, it remains unclear to what extent vertical mixing  
430 may influence the observed yearly oscillations of temperature and salinity. It is uncertain to what  
431 degree our findings are applicable to other areas of the Red Sea, which may experience different



432 seasonal currents and conditions of heat and mass exchange than in the region near Port Sudan.  
433 Perhaps most importantly, further research is required to understand how seasonal variations in  
434 temperature and salinity may be influenced by a changing climate, and how this may in turn  
435 affect the flora and fauna of the coastal Red Sea.

436

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450

## 451 **5. References**

- 452 Berman, T., Paldor, N., Brenner, S., 2003. Annual SST cycle in the Eastern Mediterranean, Red Sea  
453 and Gulf of Elat, *Geophys. Res. Lett.*, 30, 1261, doi:10.1029/2002GL015860, 5.
- 454 Bower, A. S., Farrar, J. T., 2015. Air–sea interaction and horizontal circulation in the Red Sea.  
455 In: Rasul, N.M.A., Stewart, I.C.F. (Eds.), *The Red Sea*. Springer-Verlag, Berlin, pp. 329–342.
- 456 Carton, J. A., Giese, B., 2008. A Reanalysis of Ocean Climate Using Simple Ocean Data  
457 Assimilation (SODA). *Mon. Weath. Rev.*, 136, 2999–3017.
- 458 Churchill, J. H., Lentz, S. J., Farrar, J. T., Abualnaja, Y., 2014. Properties of Red Sea coastal  
459 currents. *Cont. Shelf Res.*, 78, 51–61, doi:10.1016/j.csr.2014.01.025
- 460 Clifford, M., Horton, C., Schmitz, J., Kantha, L. H., 1997. An oceanographic nowcast/forecast  
461 system for the Red Sea, *J. Geophys. Res.*, 102, 25,101– 25,122.
- 462 Davis, K. A., Lentz, S. J., Pineda, J., Farrar, J. T., Starczak, V. R., Churchill, J. H., 2011.  
463 Observations of the thermal environment on Red Sea platform reefs: a heat budget analysis.  
464 *Coral Reefs* 30:25–36.

465 Farrar, J. T., Lentz, S., Churchill, J., Bouchard, P., Smith, J., Kemp, J., Lord, J., Allsup, G.,  
466 Hosom, D., 2009. King Abdullah University of Science and Technology (KAUST) mooring  
467 deployment cruise and fieldwork report, technical report. Woods Hole Oceanographic  
468 Institute, Woods Hole, Mass, 88 pp.

469 Ferrier-Pages, C., Gattuso, J., Jaubert, J., 1999. Effect of small variations in salinity on the rates of  
470 photosynthesis and respiration of the zooxanthellate coral *Stylophora pistillata*. *Marine*  
471 *EcologyProgress Series* 181, 309–314.

472 Furby, K. A., Bouwmeester, J., Berumen, M. L., 2013. Susceptibility of central Red Sea corals  
473 during a major bleaching event. *Coral Reefs* 32. doi: 10.1007/s00338-012- 0998-5.

474 Guildline, 2002. Portasal Salinometer Datasheet, Model 8410A, Guildline Instrument limited,  
475 Ontario, Canada.

476 Kuanui, P., Chavanich, S., Viyakarn, V., Omori, M., Lin, C., 2015. Effects of temperature and  
477 salinity on survival rate of cultured corals and photosynthetic efficiency of zooxanthellae in  
478 coral tissues. *Ocean Sci. J.* 50 (2), 263–268.

479 Large, W. G., Pond, S., 1981. Open ocean momentum flux measurements in moderate to strong  
480 winds, *J. Phys. Oceanogr.*, 11, 324–336.

481 Maillard, C., Soliman, G., 1986. Hydrography of the Red Sea and exchanges with the Gulf of  
482 Aden in summer. *Oceanologica Acta*, 9 - N°3, 249–269.

483 Morcos, S.A., 1970. Physical and chemical oceanography of the Red Sea, *Oceanography and*  
484 *Marine Biology Annual Review*, 8, 73-202.

485 Murray, S. P., Johns, W., 1997. Direct observations of seasonal exchange through the Bab el  
486 Mandeb Strait. *Geophys. Res. Lett.*, 24, 2557–2560.

487 Neumann, A. C., McGill, D. A., 1962., Circulation of the Red Sea in early summer, *Deep Sea Res.*,  
488 8, 223 – 235.

489 Patzert, W. C., 1974. Wind-induced reversal in Red Sea circulation, *Deep Sea Res.*, 21, 109– 121.

490 Raitos, D. E., Pradhan, Y., Brewin, R. J. W., Stenchikov, G., Hoteit, I., 2013. Remote Sensing the  
491 Phytoplankton Seasonal Succession of the Red Sea. *PLoS ONE* 8(6): e64909.  
492 doi:10.1371/journal.pone.0064909.

493 Reynolds, R. W., Smith, T. M., Liu, C., Chelton, D. B., Casey, K. S., Schlax, M. G., 2007. Daily  
494 High-Resolution-Blended Analyses for Sea Surface Temperature. *J. Climate*, 20, 5473-5496.

495 SAIV, 2006. Operating manual for SAIV CTD model 204 with sound velocity and optional  
496 sensors, SAIV Company, Bergen, Norway.

497 Sarmiento, J. L., Gruber, N., 2006. Ocean Biogeochemical Dynamics. Princeton University Press,  
498 Princeton, NJ, USA.

499 Smeed, D. A., 2004. Exchange through the Bab el Mandab, Deep Sea Res., Part I, 51, 455–474.

500 Sofianos, S. S., Johns, W. E., 2003. An Oceanic General Circulation Model (OGCM)  
501 investigation of the Red Sea circulation: 2. Three-dimensional circulation in the Red Sea, J.  
502 Geophys. Res., 108, NO. C3, 3066, doi:10.1029/2001JC001185.

503 Sofianos, S. S., Johns, W. E., 2007. Observations of the summer Red Sea circulation, J. Geophys.  
504 Res., 112, C06025, doi:10.1029/2006JC003886.

505 Sultan, S. A. R., Ahmad, F., 1991. Long-Term Temperature and Salinity Variations near Jeddah  
506 in Relation to Certain Meteorological Factors over the Red Sea, J. K.A.U. :Mar. Sci., 2, 19-  
507 29.

508 Sumner, D., Belaineh, G., 2005. Evaporation, Precipitation, and Associated Salinity Changes at a  
509 Humid, Subtropical Estuary. Estuaries 28 (6), 844–855.

510 Wallcraft, A. J., Kara, A. B., Hurlburt, H. E., Chassignet, E. P., Halliwell, G. H., 2008. Value of  
511 bulk heat flux parametrizations for ocean SST prediction. *J. Marine Syst.* 74, 241–258.

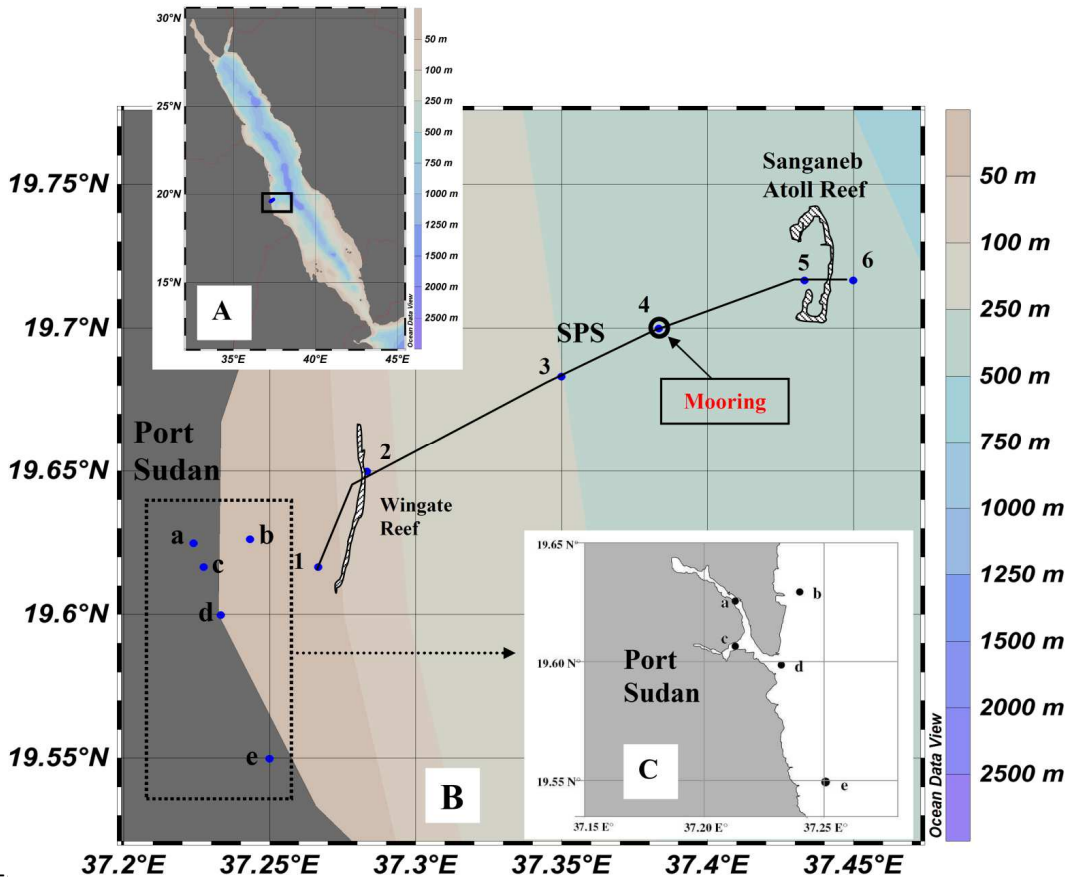
512 Yao, F. C., Hoteit, I., Pratt, L. J., Bower, A. S., Zhai, P., Kohl, A., Gopalakrishnan,  
513 G., 2014. Seasonal overturning circulation in the Red Sea: 1. Model validation and summer  
514 circulation. *Journal of Geophysical Research-Oceans.* 119:2238-2262.

515 Yao, F., Hoteit, I., Pratt, L. J., Bower, A. S., Kohl, A., Gopalakrishnan, G., Rivas, D., 2014.  
516 Seasonal overturning circulation in the Red Sea: 2. Winter circulation, J. Geophys. Res.  
517 Oceans, 119, 2263–2289, doi:10.1002/2013JC009331.

518 Zhai, P., Bower, A. S., Smethie, W. M., Pratt, L. J., 2015. Formation and spreading of Red Sea  
519 Outflow Water in the Red Sea. J. Geophys. Res. Oceans 120 (9), 6542–6563,  
520 <http://dx.doi.org/10.1002/2015JC01075>.

521 Zhan, P., Subramanian, A. C., Yao, F., Hoteit, I., 2014. Eddies in the Red Sea: A statistical and  
522 dynamical study, J. Geophys. Res. Oceans, 119, 3909–3925, doi:10.1002/2013JC009563.

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528 Figure 1. (A) The Red Sea with our study area enclosed in the black-bordered box. (B)  
 529 Locations of our observations: the Sanganeb section (SPS; black solid line connecting six CTD  
 530 stations) and Port Sudan Harbour stations (PSH; five CTD stations labelled a-e), and a CTD  
 531 mooring (at the same position of station 4 in SPS). (C) PSH stations in detail: (a) IMR jetty, (b)  
 532 Abu Hashish lagoon, (c) Khorkilab, (d) Harbour Inlet, and (e) Refinery stations.

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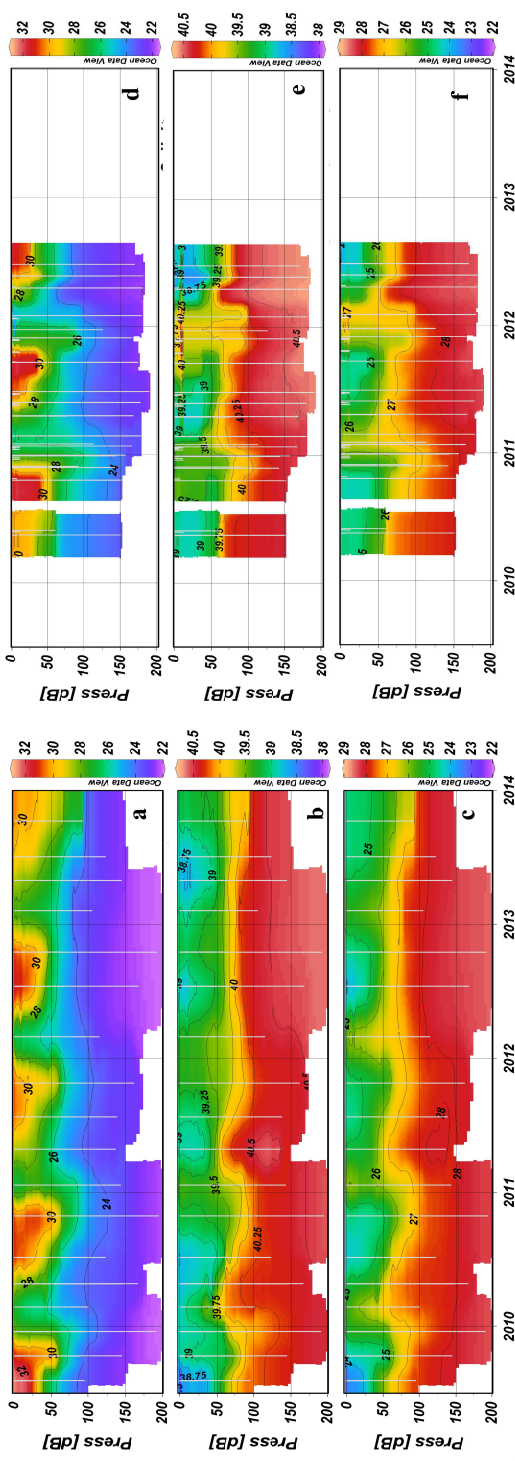
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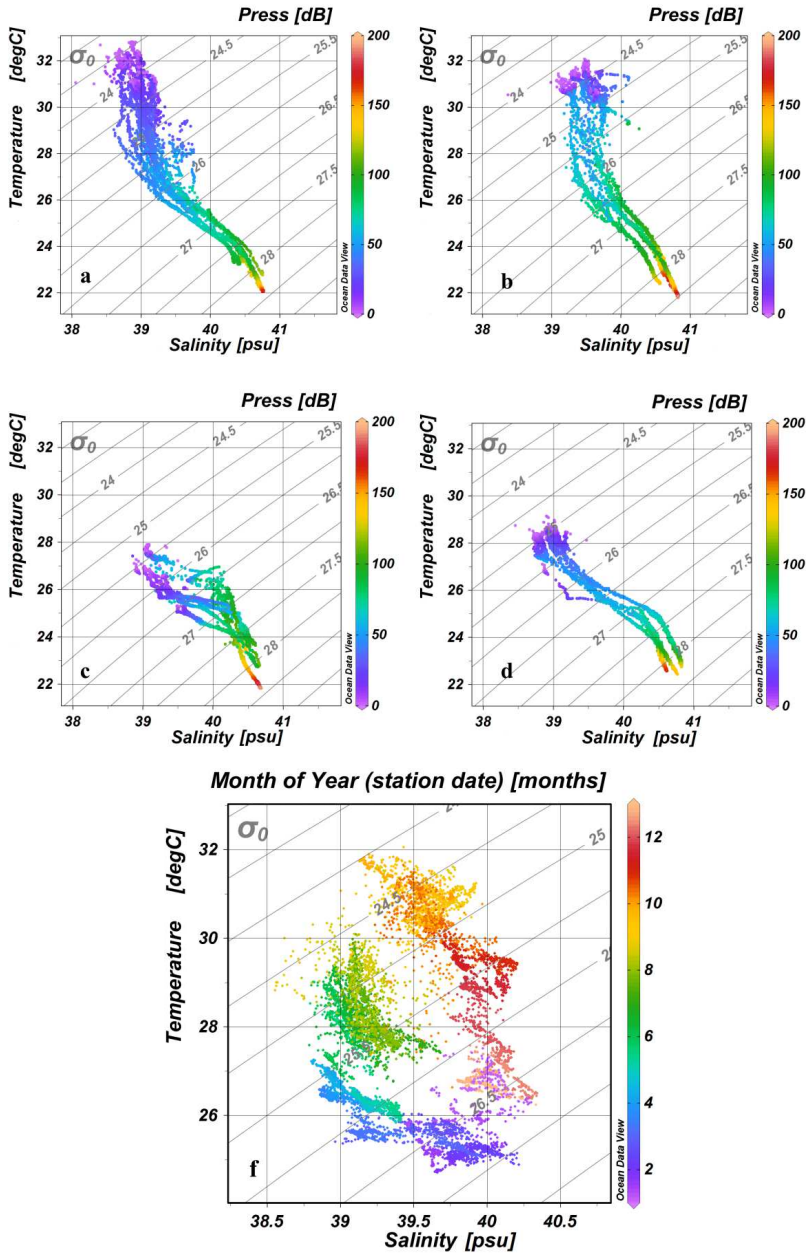
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5. Figure 2. Depth/time contours of temperature [°C], (a) and (d), salinity [psu], (b) and (e), and potential density anomaly  $[\sigma_\theta, \text{kg m}^{-3}]$ , (c) and (f). The left panels are averages values from the Port Sudan-Sanganeb transect (SPS; Station 1-6) and the right panels are averages values from Port Sudan harbor stations (PSH; IMR jetty, Abu Hashish lagoon, Khorkilab, Harbour Inlet, and Refinery stations). The vertical white lines indicate the measurements times and depth.

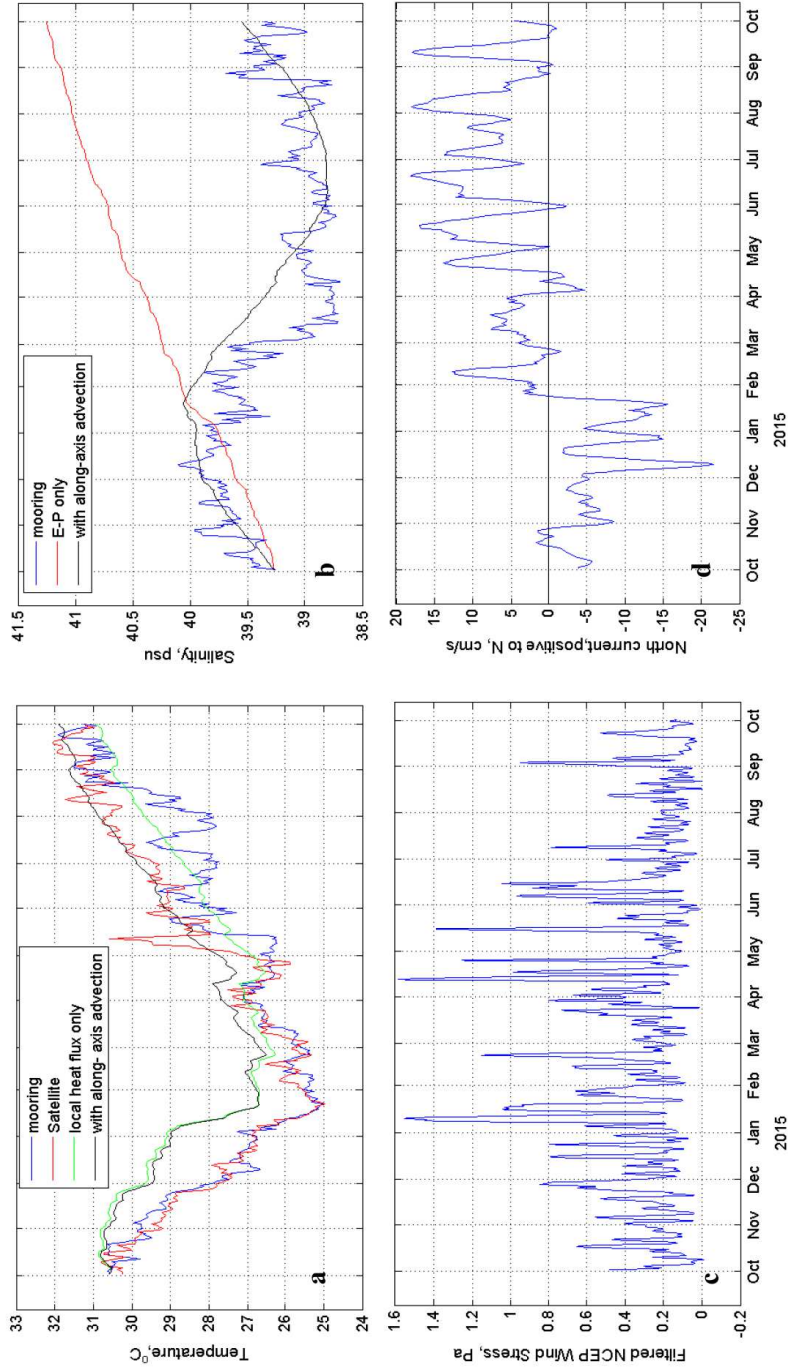
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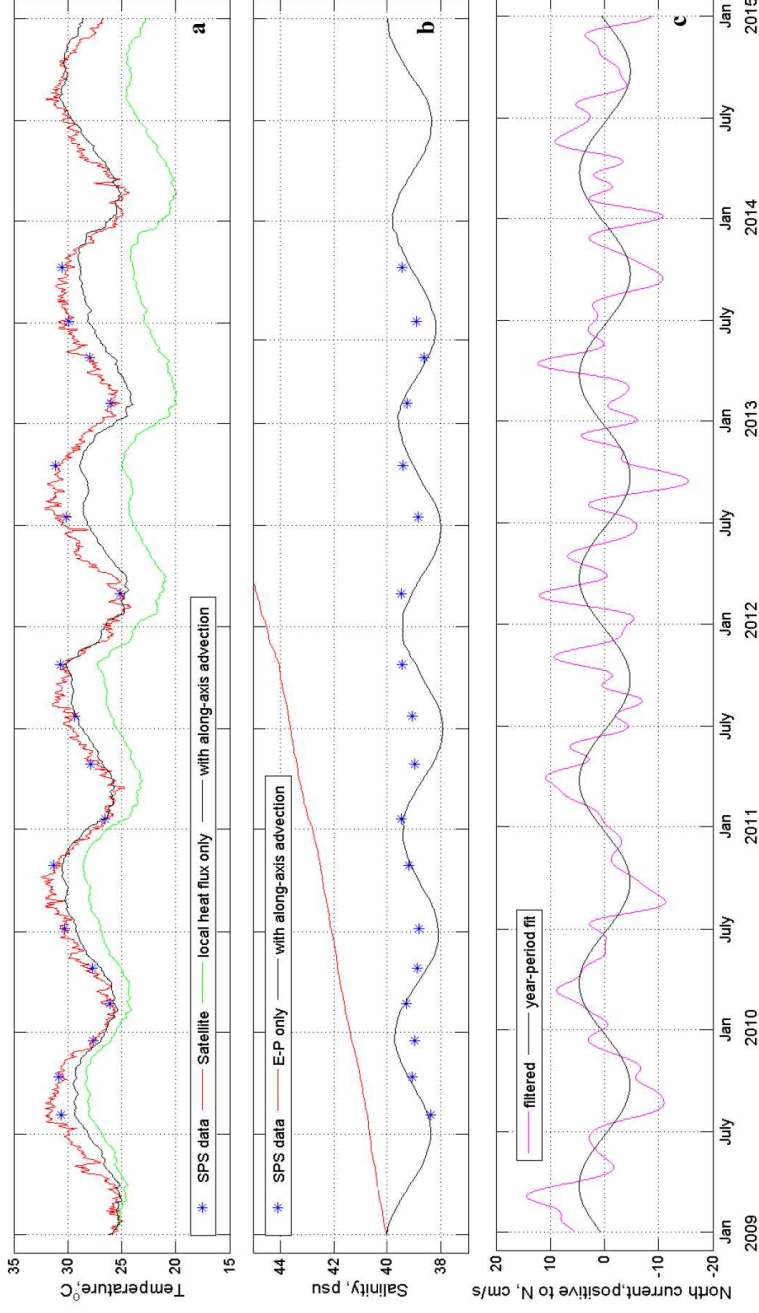
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552 Figure 3. (a-d) T-S diagrams with seasons: (a) summer, (b) autumn, (c) winter, and (d) spring based on  
 553 data from SPS. The colors indicating the depth of sampling. (f) a T-S diagram based on mooring data in  
 554 which the colors indicate the month of sampling.



555

556 Figure 4. The seasonal variation of temperature (a), salinity (b), wind stress (c), and velocity of northward geostrophic current (d). The  
 557 temperature values are from the mooring (blue line), satellite-derived (red line), computed from NCEP heat flux (green line), and computed  
 558 with alongshore advection at a station close to the mooring (black line). The salinity data are from the mooring (blue line), computed based on  
 559 local evaporation/precipitation (red line), and computed with alongshore advection (black line). The wind stress is calculated based on NCEP  
 560 wind and filtered to 50-hour half-power period.



561

562 Figure 5. The seasonality of near-surface temperature (a), salinity (b), and the northward geostrophic current (c). The different temperatures  
 563 are; average temperature (over 50 m) from SPS data (blue stars), SST from satellite-derived data (red line), computed from NCEP heat flux  
 564 (green line), and calculated with alongshore advection included (black line). The different salinities are; average salinity (over 50 m) from  
 565 SPS transect (blue stars), calculated salinities based on local evaporation/precipitation (red line), and computed salinities with alongshore  
 566 advection (black line) included. Shown in (c) are the northward geostrophic current off of Port Sudan, filtered with a 50-days half-power  
 567 period filter (magenta line) and yearly-period velocity signal (black line).



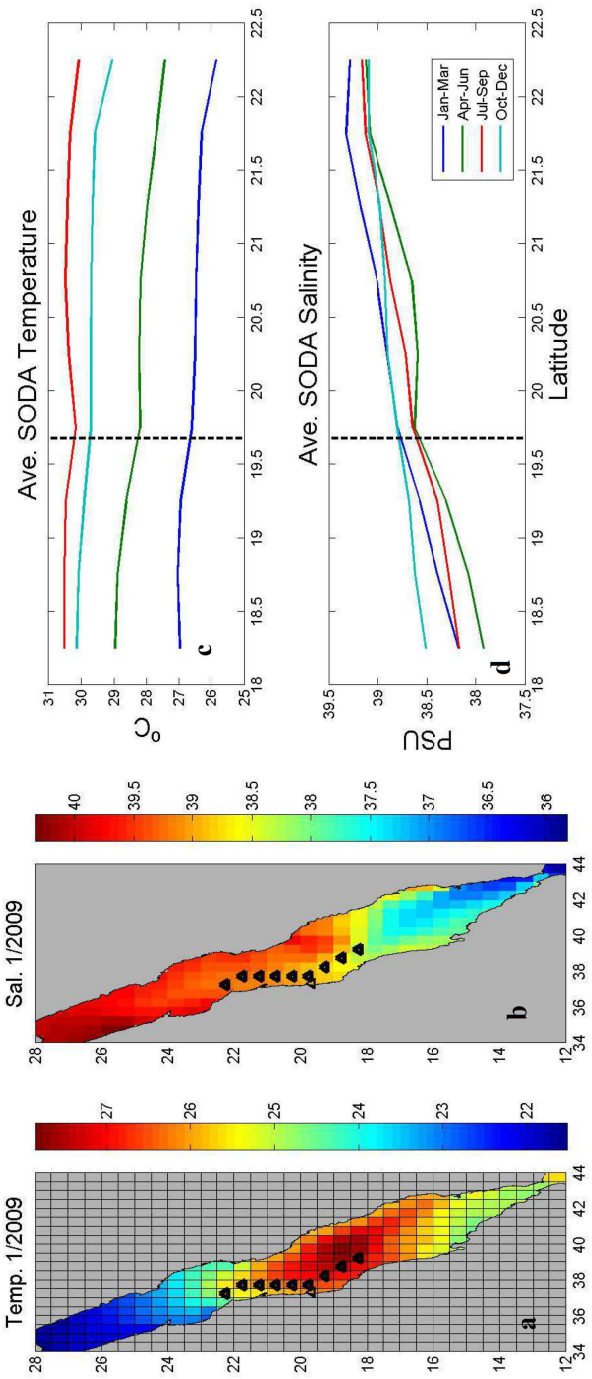


Figure 6. Properties of mean temperature and salinity, averaged over the upper 50-m, as determined from SODA data. (a) Average temperature field of Jan. 2009 with SODA grid cell indicated. (b) average salinity field of Jan 2009. The open triangle in (a) and (b) marks the location of our study's mooring. (c) and (d) Seasonally-averaged temperature and salinity (determined from 2009-2010 SODA data) at the locations shown by the closed triangles in (a) and (b).

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# Paper II

**Dissolved inorganic carbon and total alkalinity at the Sudanese coastal Red Sea, 2009-2013**

Ali, E. B., A. M. Omar, I. Skjelvan, and T. Johannessen

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# Dissolved inorganic carbon and total alkalinity at the Sudanese coastal Red Sea, 2009-2013

Elsheikh B. Ali<sup>1,2</sup>, Abdirahman M. Omar<sup>3,1</sup>, Ingunn Skjelvan<sup>3,1</sup>, and Truls Johannessen<sup>1,3</sup>

<sup>1</sup>Geophysical Institute (GFI), University of Bergen (UiB), Bergen, Norway

<sup>2</sup>Institute of Marine Research (IMR), Red Sea University (RSU), Port Sudan, Sudan

<sup>3</sup>Uni Research Climate, Bjerknes Centre for Climate Research, Bergen, Norway

## Abstract

For the first time, the seasonal variations of total alkalinity ( $A_T$ ) and dissolved inorganic carbon (DIC) in the Sudanese coastal Red Sea have been studied based on a new time series collected between Port Sudan and Sanganeb reef in 2007-2013. The average surface (4 m)  $A_T$  is at its maximum during autumn/winter ( $2460 \mu\text{mol kg}^{-1}$ ) and at minimum during spring/summer ( $2420 \mu\text{mol kg}^{-1}$ ). The sub-surface ( $>10$  m)  $A_T$  varies between  $2452$  and  $2471 \mu\text{mol kg}^{-1}$ , and for both surface and sub-surface,  $A_T$  correlate with salinity, which is driven by alongshore advection and local evaporation. The average DIC concentration at the surface decreases from  $2080 \mu\text{mol kg}^{-1}$  during winter to  $2048 \mu\text{mol kg}^{-1}$  during summer, while in the sub-surface water, the change from winter to summer is much smaller (from  $2091$  to  $2088 \mu\text{mol kg}^{-1}$ ). The DIC change is primarily driven by air-sea gas exchange caused by temperature changes, and, most likely, biological activity. The remaining change in DIC was controlled by alongshore advection and, thus, salinity variation. For surface  $A_T$ , the interannual changes were twice as much as the seasonal variations whereas for surface DIC the interannual changes were found to be less than the seasonal changes.

## Keywords

Red Sea; total alkalinity; dissolved inorganic carbon; time series; seasonal variation; Sudanese coast.

## 32 **1. Introduction**

33 The ocean covers 71 % of the earth surface and thus plays a significant role in the global carbon  
34 cycle by, for instance, taking up about a quarter of the annual carbon emissions produced by the  
35 combustion of fossil fuels and production of cement (Canadell et al., 2007). The oceanic uptake  
36 of CO<sub>2</sub> from the atmosphere occurs primarily at high latitudes whereas in low latitude regions  
37 the ocean releases CO<sub>2</sub> into the atmosphere (e.g. Takahashi et al., 2009). On a regional scale, the  
38 picture is more diverse, and it is important to understand the marine carbon cycle and the  
39 processes controlling its variations, to be able to understand the diversity.

40 The Red Sea is located at tropical and subtropical latitudes between northeast of Africa and  
41 the Arab Peninsula. The area is not very well studied in terms of biogeochemistry and carbon  
42 cycling, and consequently not well understood. According to Souvermezoglou et al. (1989), the  
43 Gulf of Aden surface and intermediate waters are the main source of dissolved inorganic carbon  
44 (DIC), total alkalinity ( $A_T$ ) and nutrients in the Red Sea. The carbon content is influenced by air-  
45 sea gas exchange, which is primarily directed from sea to air (Souvermezoglou et al., 1989; Ali,  
46 2008) and sedimentation of calcium carbonate (Metzl et al., 1989; Souvermezoglou et al., 1989).  
47 Additionally, evaporation, primary production/remineralization of organic matter, and the  
48 formation of coral reefs are assumed to play a role in the Red Sea carbon cycle (Souvermezoglou  
49 et al., 1989), but to what extent is still poorly understood.

50 In the south, at 12.5°N, the Red Sea links to the Gulf of Aden and Indian Ocean through  
51 Bab Al Mandab Strait. In the north, at 28°N, the sea is divided into two branches; the Gulf of  
52 Aqaba and the Gulf of Suez, which connects the Red Sea to the Mediterranean Sea through the  
53 Suez Canal. The Red Sea is featured by high temperature, high biodiversity, and a well-developed  
54 coral reef system. Due to the high evaporation and lack of fresh water runoff, the Red Sea is  
55 considered one of the saltiest seas in the world. The salinity increases from 36.5 psu in Bab Al  
56 Mandab Strait to about 41 psu in the northern part of the sea. The average surface temperature is  
57 about 29°C and the maximum temperature zone (>30 °C) is located in the center of the Red Sea  
58 as a result of wind convergence (Sofianos and Johns, 2003).

59 Inorganic carbon data from the Red Sea is relatively scarce and only a few studies are  
60 focusing on carbon cycle in the open Red Sea e.g. Morcos (1970); Poisson et al. (1984); Papaud  
61 and Poisson, (1986); Metzl et al. (1989); Souvermezoglou et al. (1989); Karumgalz et al. (1990);  
62 Steiner et al. (2014); and Rushdi (2014; 2015). None of these shed lights on the temporal

63 variability of the carbon cycle, and studies of the carbon cycle in the western coastal regions of  
64 the Red Sea are completely lacking. In this paper, we use new time series data of hydrography  
65 and inorganic carbon collected between 2007 and 2013 in the Sudanese coastal water off of Port  
66 Sudan (Fig. 1). We first determine the seasonal changes of DIC and  $A_T$  and their relationship to  
67 hydrography followed by an estimate of the interannual changes. Then we discuss the drivers of  
68 the seasonal variability and attempt to reconstruct the complete seasonal cycles for DIC and  $A_T$ .  
69

## 70 **2. Data and Methods**

### 71 **2.1 Data**

72 Two coastal datasets are used in this paper; one consists of summer data from 2007 and the other  
73 consists of seasonal data in the period between 2009 and 2013 (Table 1 and Fig.1). Both datasets  
74 were collected between Port Sudan and Sanganeb atoll reef, which is situated 30 km northeast of  
75 Port Sudan. Samples were collected at 4 m depth from 10 stations in 2007 (SPS1 section; Ali,  
76 2008), while during 2009-2013 (SPS2 section) the samples were collected at 4 m depth from 5  
77 stations and at 4-60 m depth from one station. Usually, the SPS2 section was sampled during  
78 October, February, April, and July. For both datasets, sampling was performed using a Hydro-  
79 Bios water sampler (2 liters) with a thermometer attached. Analyses for DIC,  $A_T$ , salinity (S),  
80 and temperature (T) were performed using standard methods (see Table 1).

### 81 82 **2.2 Data processing**

83 For SPS2 section, spatial variation between the surface stations has been examined with respect  
84 to temperature (SST), salinity (SSS),  $A_T$ , and DIC, and no systematic biases was found.  
85 Consequently, the average of the surface stations (4 m depth) was calculated to determine the  
86 seasonal and interannual variations.

87 Further, sub-surface (>10 m) data have been obtained for station 4 in SPS2 section only,  
88 and we assume that the variability measured at this station captures that of the whole section.  
89 The negligible systematic spatial variability observed in the surface as well as in sub-surface  
90 hydrographic data of SPS2 (see Ali et al., 2017) supports this assumption.

91 The year is divided into four seasons; summer (May to August), fall (September to  
92 November), winter (December to February), and spring (March to April). It might be argued that  
93 the summer season is too extensive; however, the fall season seasons is defined based upon when

94 Port Sudan receives major parts of the annual rainfall, which is in September to November (IMR,  
95 2012).

96 A comparison between seasonal and interannual variations was performed by first  
97 determining the interannual changes as standard deviations (STDs), and then associate it with  
98 annual means of summer data and of winter data, separately. Then, for each season (winter and  
99 summer), we compared the STDs associated with the annual means with the average seasonal  
100 change. The interannual trend of DIC has been briefly examined for the surface and sub-surface  
101 water based on data from both SPS1 and SPS2 sections, and in this calculation the summer season  
102 was chosen due to best data coverage.

103

### 104 **3. Results**

#### 105 **3.1 Hydrography**

106 The surface and sub-surface temperature and salinity measurements from station 4 in SPS2  
107 section clearly show a seasonal signal (Fig. 2) in agreement with the more comprehensive coastal  
108 measurements thoroughly described by Ali et al. (2017). The surface temperature reaches its  
109 maxima close to 32 °C during summer and early autumn and minima of roughly 26 °C in winter.  
110 The surface salinity is lower during spring and summer (38.5 psu) compared to autumn and  
111 winter (39.0-39.5 psu), and the salinity lagged temperature with approximately 3 months. The  
112 seasonal density variation is roughly following that of salinity and is inverse of the temperature  
113 variation, with the densest water (26-27 kg m<sup>-3</sup>) occurring in winter while the lowest density  
114 water (24-25 kg m<sup>-3</sup>) appears in summer and autumn. During autumn and winter, the pycnocline  
115 was deeper than during summer and spring (Fig. 2c), which indicates a higher mixing during the  
116 first period.

117

#### 118 **3.2 Total alkalinity (A<sub>T</sub>)**

119 Surface A<sub>T</sub> along the SPS2 section varies with the seasons as shown in Fig. 3a. However, the A<sub>T</sub>  
120 seasonality is less clear than that observed in hydrography and there is high degree of scatter in  
121 the surface data reflecting spatial variation between the stations. Generally, the surface A<sub>T</sub> is -at  
122 maximum during autumn and winter and at minimum during spring and summer ( $\Delta A_T \approx 40 \mu\text{mol}$   
123  $\text{kg}^{-1}$ ), except for the years 2009 and 2010, when the maximum and minimum in A<sub>T</sub> occurred 2-3



124 months earlier compared to the rest of the study period. Sub-surface  $A_T$  values vary similar to  
125 those in the surface layer (Fig. 3b), but the seasonal amplitudes decrease with depth.

126 Generally, the averaged surface  $A_T$  are in good agreement with those reported for the open  
127 ocean Red Sea at a similar latitude (19.5-19.9 °N) during the GEOSECS cruise in winter 1977  
128 (2446  $\mu\text{mol kg}^{-1}$ , Weiss et al., 1983) and the MEROU-I cruise in summer 1982 (2436  $\mu\text{mol kg}^{-1}$ ,  
129 Beauverger et al., 1984a).

130 The interannual variation of surface  $A_T$  ( $\pm 28 \mu\text{mol kg}^{-1}$ ) is greater by about two times that  
131 of the seasonal surface  $A_T$  ( $\pm 15 \mu\text{mol kg}^{-1}$ ), and a similar picture is seen in the sub-surface water,  
132 however, with a smaller rate between interannual and seasonal  $A_T$  variation. Furthermore, the  
133 interannual variation was larger during summer ( $\pm 37 \mu\text{mol kg}^{-1}$ ) than during winter ( $\pm 18 \mu\text{mol}$   
134  $\text{kg}^{-1}$ ).

135  $A_T$  is a semi conservative parameter and therefore generally co-varied directly with S (Figs.  
136 2 and 3). The correlation coefficients of the implied relationship for the area of study has been  
137 determined by linear regression between  $A_T$  and S, and the result is shown in Table 2. The  
138 regression coefficients for the surface samples are in good agreement with the results of Ali  
139 (2008) who reported a similar relationship between surface  $A_T$  and SSS ( $a=28.94$ ,  $b=1300.2$ ,  $R^2$   
140  $= 0.92$ ) based on MEROU-II data collected in October 1982 along the central axis of the Red  
141 Sea (Beauverger et al., 1984b).

142

### 143 **3.3 Dissolved Inorganic Carbon (DIC)**

144 Surface DIC along the SPS2 section show a clear seasonal variation, with maximum values  
145 during winter, minimum during summer, and an approximate seasonal amplitude of 32  $\mu\text{mol kg}^{-1}$   
146 (Fig. 4a). Similar to the surface  $A_T$ , also surface DIC is scattered reflecting the spatial variation  
147 between the stations. Some seasonality is also seen in the sub-surface DIC shallower than  
148 approximately 40 m, but the seasonal amplitude was lower than observed in the surface. Below  
149 40 m, there is no clear seasonal DIC signals (Fig. 4b).

150 The interannual variations in the surface is stronger during summer ( $\pm 6 \mu\text{mol kg}^{-1}$ ) than  
151 winter, but the magnitude is much smaller than the seasonal variation ( $\pm 20 \mu\text{mol kg}^{-1}$ ). In sub-  
152 surface shallower than 40 m, a similar picture is seen but with a smaller rate between interannual  
153 and seasonal amplitude.

154 DIC is inversely correlated with SST both in surface water ( $R^2=0.52$ ), see Fig. 5, and in  
155 the sub-surface ( $R^2=0.6$ ). Further, DIC also correlates with salinity, but these relationships are  
156 weaker:  $R^2=0.16$  and  $0.26$  for surface and sub-surface, respectively.

157 The interannual trend in surface DIC based on summer values was  $2.26 \mu\text{mol kg}^{-1} \text{y}^{-1}$  ( $R^2=$   
158  $0.61$ ), however, this result was not further explored. In sub-surface, no interannual trend was  
159 seen.

160

## 161 **4. Discussion**

162 In the above, we showed that the seasonal changes of DIC and  $A_T$  are taking place more or less  
163 in concert with variations in temperature and salinity. In the following, these covariations are  
164 explored in detail with the aim to gain insight into the processes that (i) are responsible for the  
165 correlations and (ii) are most influential for the seasonal variations.

166

### 167 **4.1 DIC and $A_T$ correlation with salinity**

168 The significant and positive relationship between  $A_T$  and salinity (Table 2) observed both in  
169 surface and sub-surface, indicate that large part of the changes in  $A_T$  is driven by salinity  
170 variation. This finding is in agreement with the results of Millero et al., (1998) who reported that,  
171 in subtropical ocean, changes in salinity account for more than 80% of total variability in  $A_T$ .  
172 The correlation arises because  $A_T$  variability in the upper ocean is controlled mainly by processes  
173 that drives salinity changes (e.g. precipitation/evaporation) (Millero et al., 1998; Lee et al.,  
174 2006). For our data, however, salinity variation explains only about 36% of the  $A_T$  variations.  
175 This is most likely due to the fact that  $A_T$  is also impacted by calcium carbonate ( $\text{CaCO}_3$ )  
176 formation through pelagic and coral calcification in the Red Sea which is classified as one of the  
177 oceanic regions with the highest  $\text{CaCO}_3$  production (Takahashi et al., 2014; Steiner et al., 2014).

178 The relationship between surface  $A_T$  and salinity in our dataset (Table 2) has a much larger  
179 positive intercept and lower slope compared to the Indian Ocean surface  $A_T$  - salinity relationship  
180 ( $a=68.8$ ,  $b=-114$ ) described by Millero et al. (1998). The Indian Ocean surface waters is the  
181 source water for the Red Sea, and Takahashi et al. (2014) suggests that the large intercept in the  
182 Red Sea is due to high salinity associated with  $\text{CaCO}_3$  production. In addition, the change in  $A_T$ -  
183 SSS relationship can be explained partially by a loss of inorganic carbon by sedimentation in the  
184 deep water, as suggested by Metzl et al. (1989) and Souvermezoglou et al., (1989). In the deep

185 water,  $A_T$  will further decrease by remineralization of organic matter, which adds nitrate ( $\text{NO}_3$ )  
 186 to the water column (Elageed et al., “Oxygen and alkalinity utilization rates in the Red Sea”,  
 187 manuscript in preparation). Furthermore, calcium carbonate precipitates may occur and reduce  
 188 the deep-water alkalinity even further (Ali, 2008; Rushdi, 2014; 2015). Thus, the observed  
 189 relationship between surface  $A_T$  and SSS is a result of mixing between fresher surface water that  
 190 conform to the Indian Ocean relationship in the southern Red Sea and high salinity/low  $A_T$  deep  
 191 water of the Red Sea (e.g. Ali, 2008; Steiner et al., 2014).

192 The correlation between DIC and salinity was positive but weaker than for  $A_T$ , which is as  
 193 expected due to the chemical composition of these parameters. As for  $A_T$ , mixing also influence  
 194 DIC but in general, biological activity and air-sea gas exchange are important contributors to the  
 195 variability.

196  $A_T$  and DIC were salinity normalized ( $nA_T$  and  $nDIC$ ) according to Friis et al. (2003),  
 197 where a constant salinity of 39 psu and an intercept of 1160 (based on data in the current study)  
 198 were used, see Table 2. The salinity normalization did not change the phase of the seasonal  
 199 variation of  $A_T$  and DIC but it slightly decreased the seasonal amplitudes of  $A_T$  and DIC. Thus,  
 200 even though salinity is an important factor there are also other factors contributing to the  
 201 variability in DIC and  $A_T$ .

202

## 203 **4.2 DIC correlation with SST**

204 We have found a strong anti-correlation between DIC and SST (Fig.6). A combination of  
 205 processes can be responsible for the observed correlation. For instance, a decrease/increase in  
 206 SST induces an increase/decrease in  $\text{CO}_2$  solubility and  $\text{pCO}_2$  difference across air-sea interface,  
 207 which enhances/reduces  $\text{CO}_2$  exchange with the atmosphere. Thus, the higher the SST the lower  
 208 the DIC content of the ocean. The theoretical slope of this relationship can be computed for the  
 209 tropical ocean (see e.g. Takahashi et al., 1993) according to:

210

$$211 \quad \frac{\partial DIC}{\partial T} = \frac{DIC_m \tau}{R} = -11 \quad (1)$$

212

213 where  $\tau = \frac{\partial \ln p\text{CO}_2}{\partial DIC} = 0.0423$ ,  $R$  is the Revelle Factor (= 8 for tropical waters), and  $DIC_m$  is the  
 214 observed mean value (= 2060  $\mu\text{mol kg}^{-1}$ ). With these values, Equation (1) suggests that DIC

215 should decrease by  $11 \mu\text{mol kg}^{-1}$  for each  $1^\circ\text{C}$  temperature increase. However, the observed slope  
216 in our data is only  $-5.23 \mu\text{mol kg}^{-1}$ , which is about half of theoretical slope, so approximately  
217 half of the change in DIC can be attributed temperature variation.

218 Thorough analyses of the reason why we observe only half of the expected temperature  
219 impact on DIC is beyond the data at hand. Nevertheless, one possible explanation for the  
220 discrepancy can be biological production, which is higher in the southern central Red Sea during  
221 the winter than summer due to a strong anti-cyclonic gyre centered at  $19.5^\circ\text{N}$  (Raitsos et al.,  
222 2013). The gyre has convergence of water at the surface and downwelling in the center.  
223 Consequently, upwelling of relatively cold sub-surface water and sinking of relatively warm  
224 surface water occur across the basin, both at coastal boundaries and away from them. This results  
225 in low productivity at the core of the gyre and higher productivity at its borders. On the contrary,  
226 low primary production during summer is caused by strong seasonal stratification (Raitsos et al.,  
227 2013). Higher primary production during winter would oppose the effect of increasing DIC  
228 expected from the decreasing temperature (see Eq. 1). During summer, primary production is  
229 low and might be surpassed by remineralization of organic matter, which increases DIC and thus  
230 opposes the decreasing DIC from increased SST. Therefore, the overall result of primary  
231 production and remineralization would be to dampen the temperature driven seasonal DIC  
232 amplitude. Therefore, we suggest that the observed DIC-SST relationship is due to air-sea gas  
233 exchange modulated by biological activity.

234

### 235 **4.3 Influential processes for DIC and $A_T$ seasonal variations**

236 In the Red Sea, surface salinity, DIC, and  $A_T$  increase from south to north, while the maximum  
237 temperature zone is located and centered south of our study area (e.g. Ali, 2008). Furthermore,  
238 according to Ali et al. (2017) the geostrophic current in the coastal area off of Port Sudan directs  
239 from south to north during spring and summer bringing relatively warm and fresh water from the  
240 south, and this water is also characterized by a relatively low DIC and  $A_T$  (Ali, 2008). During  
241 autumn and winter the geostrophic current is reversed and brings waters which is colder, more  
242 saline, and richer in DIC and  $A_T$  from north. The observed seasonal changes in DIC and  $A_T$  are  
243 most likely produced by the effect of the alongshore advection superimposed on local responses  
244 to SST and SSS changes as well as biological activity.

245 Ali et al. (2017) examined the effect of across-shore advection on the seasonality of  
246 temperature and salinity and they found that at a seasonal time scale, the across-shore  
247 displacement signal has to be small. Based on this finding, we expect that the effect of across-  
248 shore advection on the seasonality of DIC and  $A_T$  is also small, and this is evidenced by a lack  
249 of east- or westward trends in the spatiotemporal variation between SPS1 and SPS2 stations.

250

#### 251 **4.4 Reconstruction of complete DIC and $A_T$ cycles**

252 Since the most influential processes for seasonal variations of DIC and  $A_T$  are identified, we  
253 attempt to reconstruct the complete seasonal patterns of DIC and  $A_T$  during the period 2009 –  
254 2014. As a first step, DIC values ( $DIC_{SST-sat}$  in Eq. 1) was calculated by applying the DIC-SST  
255 relationship (Fig. 5) and daily satellite-derived SST data ( $SST_{sat}$ ; NOAA High-resolution  
256 Blended Analysis of Daily SST, Version 2) extracted from 37.38° E, 19.62°N, close to the area  
257 of study. The  $SST_{sat}$  has the same seasonal amplitude (approximately 6°C) as the observed SST  
258 (Ali et al., 2017; their Fig. 5). The seasonality of daily  $DIC_{SST-sat}$  has the same shape as the  
259 observed DIC, higher during winter and lower during summer with a similar amplitude (~32  
260  $\mu\text{mol kg}^{-1}$ ) but slightly higher values than the observed DIC, especially during summers. The  
261 daily  $DIC_{SST-sat}$  variations is about 6  $\mu\text{mol kg}^{-1}$  in magnitude (Fig. 6) and these are a result of  
262 the SST variation of approximately 1°C (Ali et al., 2017).

263 Next, the effect of the salinity variations on DIC was included by using a multivariate  
264 regression taking into consideration the observed surface DIC, SST, and SSS:

265

$$266 \quad DIC_{Reg} = a \text{ SSS} + b \text{ SST} + c \quad (2)$$

267

268 where the multivariate regression coefficients were:  $a = 6.4814$ ,  $b = -4.9739$ ,  $c = 1959$ , and  $R^2 =$   
269  $0.54$ . To study the daily DIC variability we used Eq. 2,  $SST_{sat}$ , and surface salinity ( $SSS_{Adv}$ )  
270 computed for the area in this study based on the effect of local evaporation/precipitation and  
271 alongshore advection as described in Ali et al. (2017). The resulting  $DIC_{Reg}$  has similar  
272 amplitudes as  $DIC_{SST-sat}$ , which was based only on SST (Fig.6). However, when the regression  
273 was extended by salinity (as for  $DIC_{Reg}$ ), the correlation coefficient ( $R^2=0.54$ ) improved slightly  
274 compared to the  $DIC_{SST-sat}$  relationship ( $R^2= 0.52$ ). This is partially explained by the observed  
275 phase shift between SSS and SST of approximately 3 months, which will also introduce a slight

276 phase shift between  $DIC_{Reg}$  and  $DIC_{SST-sat}$  (Fig.6) (Ali et al., 2017). It must be emphasized,  
277 however, that comparisons performed by Ali (2017) revealed a good correspondence between  
278 satellite-derived and measured SST whereas the computed  $SSS_{Adv}$  showed a higher degree of  
279 mismatch and varied over a much greater range than the measured SSS (Ali et al., 2017; their  
280 Fig. 5). Due to this imperfection of the  $SSS_{Adv}$ , we believe that the true potential of using Eq. 1  
281 to estimate DIC in the study area is even greater than has been realized in Fig. 6.

282 The lack of a “perfect” SSS dataset also hampers the estimation of the daily variability of  
283  $A_T$  in the study area. Nevertheless, the  $A_T$ -SSS relationship (shown in Table 2) was applied on  
284  $SSS_{Adv}$  to estimate the complete seasonal cycle of surface  $A_T$  during the period of our survey.  
285 The computed  $A_T$  has a clear seasonal cycle resembling that observed in surface  $A_T$  (Fig. 7), high  
286 during winter and low during summer with average amplitude about  $40 \mu\text{mol kg}^{-1}$ . However, as  
287 expected, there is a higher degree of mismatch compared to Fig. 6 and the computed  $A_T$  vary  
288 over a larger range. Thus, it is likely to produce an improved reconstruction of the seasonal cycle  
289 of  $A_T$  if daily SSS is used along with Equation 2.

290

## 291 **5. Summary and Conclusion**

292 The seasonal and interannual variations of  $A_T$  and DIC in the coastal Red Sea have been studied  
293 using a new time series collected between Port Sudan and Sanganeb reef from 2007 to 2013.  
294 Both variables show maximum values during autumn and winter, and minimum during spring  
295 and summer associated with the maxima and minima of salinity. Temperature is at maximum  
296 during summer and at minimum during winter. There are positive linear relationships between  
297  $A_T$  and salinity and between DIC and salinity, although the latter is relatively weak, indicating  
298 that salinity is an important driver primarily for  $A_T$  variations.

299 The occurrence of high DIC values in winter, when primary production is normally at its  
300 highest, indicates that the consumption of DIC by primary production is less than the increase of  
301 DIC from air-sea gas exchange. The remaining changes of DIC are caused by alongshore  
302 advection during autumn and winter. The geostrophic current directs from north to south bringing  
303 relatively cold and saline water enrich with carbon to the area of study, while during spring and  
304 summer the current is reversed and transports warm and fresh water characterized with relatively  
305 low DIC from the south.

306 Interannual variations of  $A_T$  and DIC have been briefly examined, and for surface  $A_T$ , these  
307 changes were twice as much as the seasonal variation whereas for surface DIC, the interannual  
308 changes were found to be less than the seasonal changes.

309 In conclusion, seasonal DIC variations are governed by, in decreasing order, temperature  
310 driven air-sea gas exchange, biological activity, and alongshore advection. Whereas  $A_T$   
311 variations are driven mainly by salinity, which is directly controlled by local processes such as  
312 evaporation/precipitation in addition to alongshore advection.

313

### 314 *Acknowledgement*

315 The SPS1 survey is the fieldwork of Ali's master thesis (2008) funded by Quota program at University of Bergen,  
316 while the SPS2 survey was funded by the Norwegian Norad's Program for Master Studies (NOMA) and organized  
317 by Institute of Marine Research (IMR) at Red Sea University (RSU) in Port Sudan. All cruises operating the SPS2  
318 transect have been conducted on the Sudanese Marine Security Department ships. We are thankful to the former  
319 director of IMR: Prof. Abdel Gadir D. El Hag for his support and to all IMR staff for making the time series collection  
320 possible. Also, thanks to Solveig Kringstad the former senior engineer at Geophysical Institute - University of  
321 Bergen for her great efforts in analyzing the samples. NOAA High-resolution Blended Analysis of Daily SST,  
322 (Version 2) data were obtained from the NOAA/OAR/ESRL PSD, Boulder, Colorado, USA  
323 (<http://www.esrl.noaa.gov/psd/>). The work of A.M. Omar and I. Skjelvan has been partly supported by the Research  
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## 327 **6. References**

328 Ali, E. B., 2008. The inorganic carbon cycle in the Red Sea, Master's thesis, Geophysical  
329 Institute, University of Bergen, Bergen, Norway, 77 pp.

330 Ali, E. B., Churchill, J. H., Barthel, K., Skjelvan, I., Omar, A. M., de Lange, T. E., and Eltaib, E.  
331 B. A., 2017. Seasonal and interannual variations of hydrographic parameters in the Sudanese  
332 coast of the Red Sea, 2009-2015, submitted to Regional Studies in Marine Sciences journal.

333 Beauverger, C., Brunet, C. Poisson, A. 1984a. MD 33/MEROU a bord du "MARION  
334 DUFRESNE" 16 juin au 4 juillet 1982, 25 septembre au 4 octobre 1982. Les rapports des  
335 campagnes ala mcr, publication des T.A.A.F. No. 82-04, 181 pp.

336 Beauverger, C. 1984b. MD 33/MEROU a bord du "MARION DUFRESNE" ]6 juin au 4 juillet  
337 1982, 25 septembre au 4 octobre 1982. Les rapports des campagnes a la mer, publication des  
338 T.A.A.F. No. 82-05, 163 pp.

339 Canadell, J. G., Le Qu'ér'e, C., Raupach, M. R., Field, C. B., Buitenhuis, E. T., Ciais, P., Conway,  
340 T. J., Gillett, N. P., Houghton, R. A., and Marland, G., 2007. Contributions to accelerating  
341 atmospheric CO<sub>2</sub> growth from economic activity, carbon intensity, and efficiency of natural  
342 sinks, *P. Natl. Acad. Sci. USA*, 104, 18353– 18354.

343 Dickson, A. G., Sabine, C. L., Christian, J. R., 2007. Guide to best practices for ocean CO<sub>2</sub>  
344 measurements, p. 191. PICES Special Publication 3.

345 Elageed et al., 2017. Apparent oxygen and alkalinity utilization rates in the Red Sea, in prep.

346 Friis, K., Körtzinger, A., Wallace, D. W. R., 2003. The salinity normalization of marine inorganic  
347 carbon chemistry data, *Geophys. Res. Lett.*, 30, 1085, doi:10.1029/2002GL015898.

348 IMR, 2012. Environmental Baseline Study for Mersa El-Sheikh Ibrahim, conducted by Institute  
349 of Marine Research (IMR) in coloration with Faculty of Marine Sciences and Fishers (FMSF)  
350 at Red Sea University (RSU), Port Sudan, Sudan.

351 Johnson, K. M., Wills, K. D., Butler, D. B., Johnson, W. K., and Wong, C. S., 1993. Coulometric  
352 total carbon dioxide analysis for marine studies, *Mar. Chem.*, 44, 167–187. 1993.

353 Krumgalz, B.S., Erez, J., Chen, C.T.A., 1990. Anthropogenic CO<sub>2</sub> penetration in the northern  
354 Red Sea and in the Gulf of Elat (Aqaba), *Oceanologica Acta*, 13 (3), 283-290.

355 Lee, K., Tong, L. T., Millero, F. J., Sabine, C. L., Dickson, A. G., Goyet, C., Park, G. H.,  
356 Wanninkhof, R., Feely, R. A., Key, R. M., 2006. Global relationships of total alkalinity with  
357 salinity and temperature in surface waters of the world's oceans, *Geophys. Res. Lett.*, 33,  
358 L19605, doi:10.1029/2006GL027207.

359 Metzl, N., Moore III, B., Papaud, A., Poisson, A., 1989. Transport and carbon exchange in Red  
360 Sea inverse methodology, *Global Biogeochemical Cycle*, Vol. 3, No. 1, pages 1-26.

361 Millero, F. J., Lee, K., Roche, M., 1998. Distribution of alkalinity in the surface waters of the  
362 major oceans, *Mar. Chem.*, 60, 111 –130.

363 Morcos, S.A., 1970. Physical and chemical oceanography of the Red Sea, *Oceanography and*  
364 *Marine Biology Annual Review* 8 pp. 73–202.

365 Papaud, A., Poisson, A., 1986. Distribution of dissolved CO<sub>2</sub> in the Red Sea and correlation with  
366 other geochemical tracers, *J. Mar. Res.*, 44, 385-402.

367 Poisson, A., Morcos, S.A., Souvermezoglou, T., Papaud, A., Lvanoff, A., 1984. Some aspects of  
368 biogeochemical cycles in the Red Sea with special reference to new observations made in  
369 summer 1982, *Deep Sea Res. Part A. Oceanographic Res. Paper*, Vol.31, Issue 6-8, 707-718.



370 Raitso, D. E., Pradhan, Y., Brewin, R. J. W., Stenchikov, G., Hoteit, I., 2013. Remote Sensing  
371 the Phytoplankton Seasonal Succession of the Red Sea. PLoS ONE 8(6): e64909.  
372 doi:10.1371/journal.pone.0064909.

373 Rushdi, A. I., 2014. Calcium carbonate saturation levels of the Red Sea coast of Yemen and  
374 biogeochemical effects of excess CO<sub>2</sub>. In: International conference on the marine  
375 environment of the Red Sea, King Abdulaziz University, Jeddah, 10–13 Nov 2014 (in press).

376 Rushdi, A.I., 2015. Calcite and Aragonite Saturation States of the Red Sea and Biogeochemical  
377 Impacts of Excess Carbon Dioxide, in: Rasul, N.M.A., Stewart, I. C.F. (Eds), The Red Sea  
378 the Formation, Morphology, Oceanography and Environment of a Young Ocean Basin,  
379 Springer Earth System Sciences, ISBN 978-3 662-45201-1(eBook), DOI 10.1007/978-3-  
380 662-45201-1.

381 Sofianos, S. S., Johns, W. E., 2003. An Oceanic General Circulation Model (OGCM)  
382 investigation of the Red Sea circulation: 2. Three-dimensional circulation in the Red Sea, J.  
383 Geophys. Res., 108, NO. C3, 3066, doi:10.1029/2001JC001185.

384 Souvermezoglou, T., Metzel, N., Poisson, A., 1989. Red Sea budgets of salinity, nutrients and  
385 carbon calculated in the Strait of Bab Al Mandab during summer and winter seasons, Journal  
386 of Marine Research, 47,441-456.

387 Steiner, Z., Erez, J., Shemesh, A., Yam, R., Lazar, B., 2014. Basin-scale estimates of pelagic and  
388 coral reef calcification in the Red Sea and Western Indian Ocean, PANS, Vol. 111, No. 46,  
389 16303-16308.

390 Takahashi, T., Olafsson, J., Goddard, J., Chipman, D.W., Sutherland, S.C., 1993. Seasonal  
391 variation of CO<sub>2</sub> and nutrients in the high-latitude surface oceans: A comparative study.  
392 Glob. Biogeochem. Cycles 7, 843–878.

393 Takahashi, T., Sutherland, S.C., Wanninkhof, R., Sweeney, C., Feely, R. A., Chipman, D.W.,  
394 Hales, B., Friederich, G., Chavez, F., Sabine, C., and others. 2009. Climatological mean and  
395 decadal change in surface ocean pCO<sub>2</sub>, and net sea-air CO<sub>2</sub> flux over the global oceans. Deep  
396 Sea Research Part II 56(8–10):554–577.

397 Takahashi, T., Sutherland, S. C., Chipman, D. W., Goddard, J. G., Ho, C., Newberger, T., and  
398 others. 2014. Climatological distributions of pH, pCO<sub>2</sub>, total alkalinity, and CaCO<sub>3</sub>  
399 saturation in the global surface ocean, and temporal changes at selected locations. Mar.  
400 Chem. 164:95–125.

401 Weiss R. F., Broecker, W. S., Craig, H. B., Spencer, D. W, 1983. Geosecs Indian Ocean  
402 Expedition, 5. Hyd. data. Nat. Sc. Fond., U.S. Government Print. Off.  
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## Tables

**Table 1: Methods of analysis**

Parameters	Methods	Instrument	Accuracy
T	Temperature sensor	SAIV CTD model 204	$\pm 0.01^{\circ}\text{C}$
S	Inductive cell conductivity sensor (salinity calculated from conductivity)	SAIV CTD model 204	$\pm 0.02 \text{ mS cm}^{-1}$
$A_T$	Open potentiometric titration (Dickson et al., 2007)	Marianda VINDTA 3S	$\pm 1 \mu\text{mol}$
DIC	Gas extraction of acidified water samples followed by coulometric titration method (Johnson et al., 1993; Dickson et al., 2007)	VINDTA 3C with UIC Coulometer (CM5012)	$\pm 0.5 \mu\text{mol}^*$

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\* The accuracy was achieved by using certified reference material (CRM) supplied by prof. Andrew Dickson of Scripps Institution of Oceanography, USA.

419 **Table 2: Regression coefficients of  $A_T$ -salinity relationship**

$A_T$ -S relationship at:	a	b	$R^2$
surface	32.9	1160	0.36
Sub-surface	26.4	1417	0.36

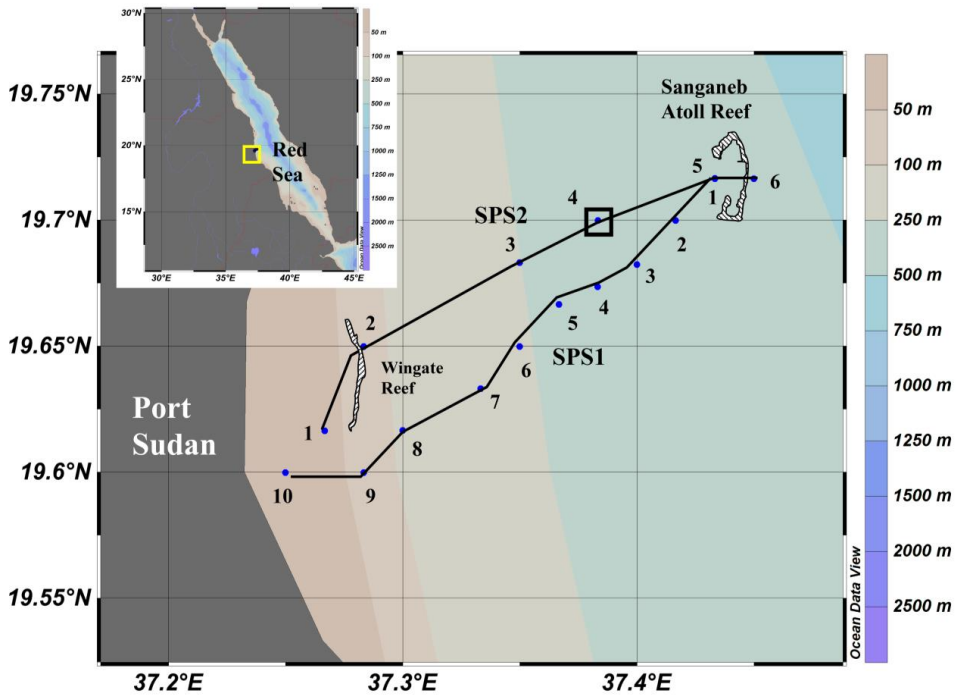
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## Figures



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426 Fig. 1: The location of the coastal stations between Port Sudan and Sanganeb atoll reef. The SPS1  
427 section is made up of ten stations (southern line) and the SPS2 section contains 6 stations  
428 (northern line). Station 1 at SPS1 section and station 5 at SPS2 section are located at the same  
429 position. Sampling depth from all stations are 4 m, except for station 4 at SPS2 section (black  
430 solid square), where data were collected from several depths between 4 and 60 m.

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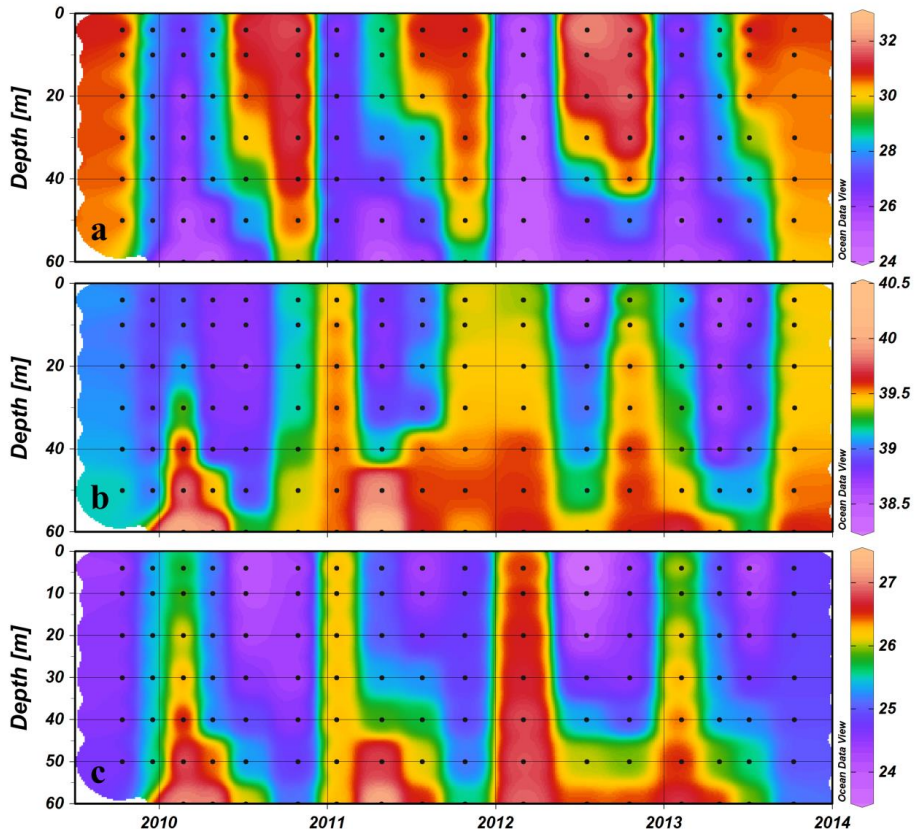
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440 Fig. 2: Seasonal and interannual variations of water column (a) temperature [ $^{\circ}\text{C}$ ], (b) salinity,  
441 and (c) density [ $\text{kg m}^{-3}$ ].

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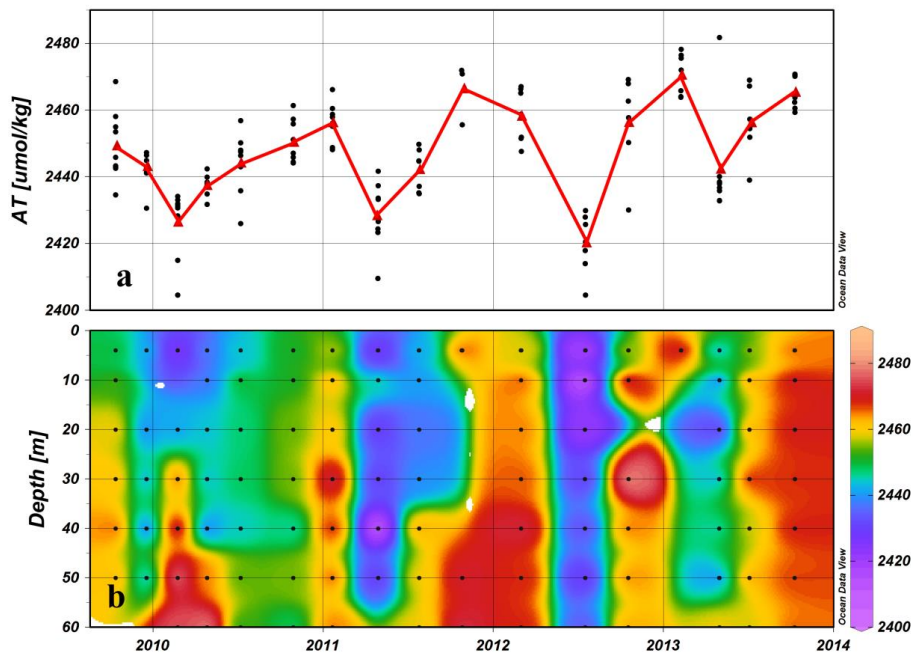
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450 Fig. 3: Alkalinity ( $A_T$ ) values [ $\mu\text{mol kg}^{-1}$ ] (a) in the surface (station 1 to 6) and (b) in the water  
 451 column (station 4). The red triangles and the line in Fig.3a indicate averaged values. All data are  
 452 from the SPS2 section.

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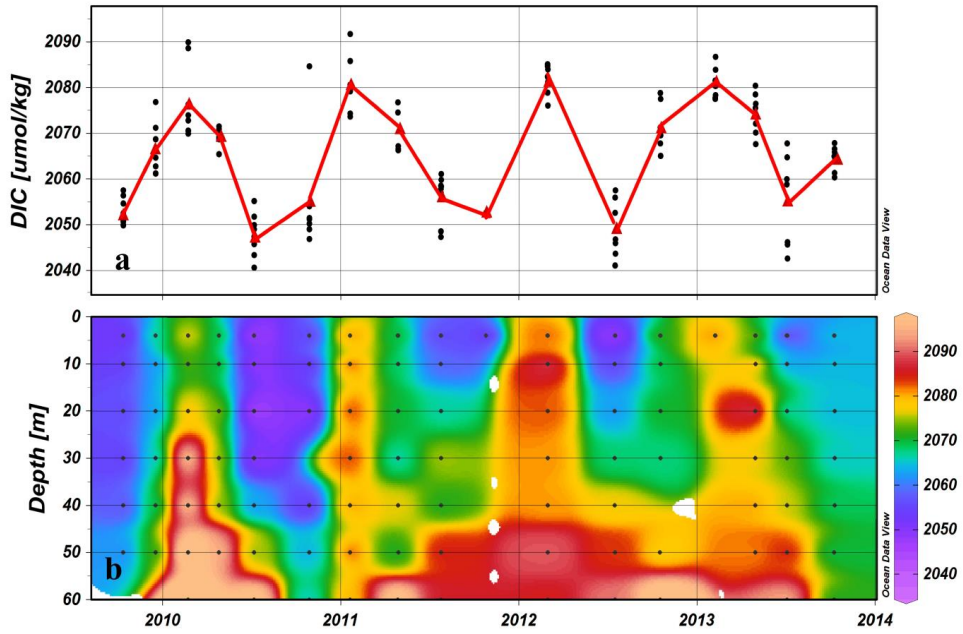
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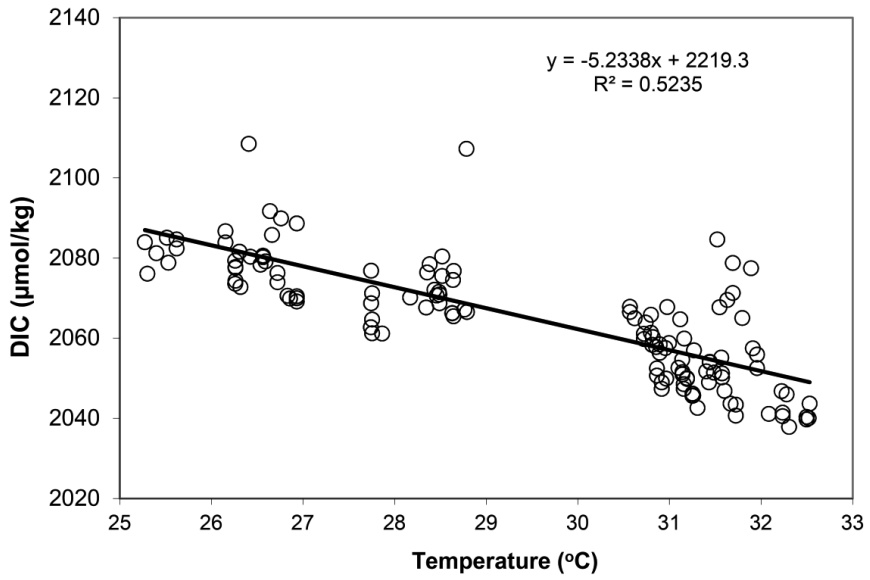
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Fig. 4: DIC values [ $\mu\text{mol kg}^{-1}$ ] (a) in the surface (station 1 to 6) and (b) in the water column (station 4). The red triangles and the line in Fig.4a indicate averaged values. All data are from the SPS2 section.

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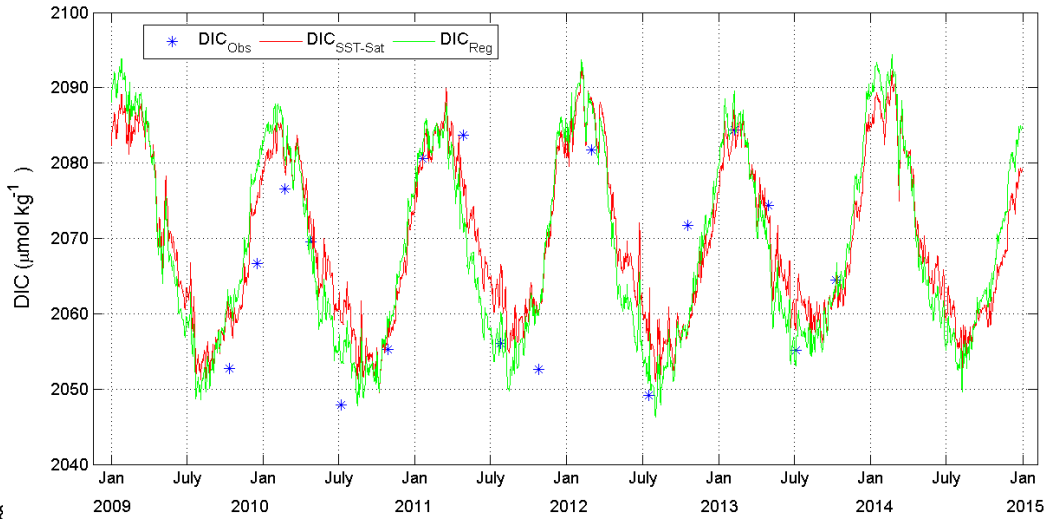


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Fig. 5: Surface DIC [ $\mu\text{mol kg}^{-1}$ ] as a function of SST [ $^{\circ}\text{C}$ ]. All data are from the SPS2 section.



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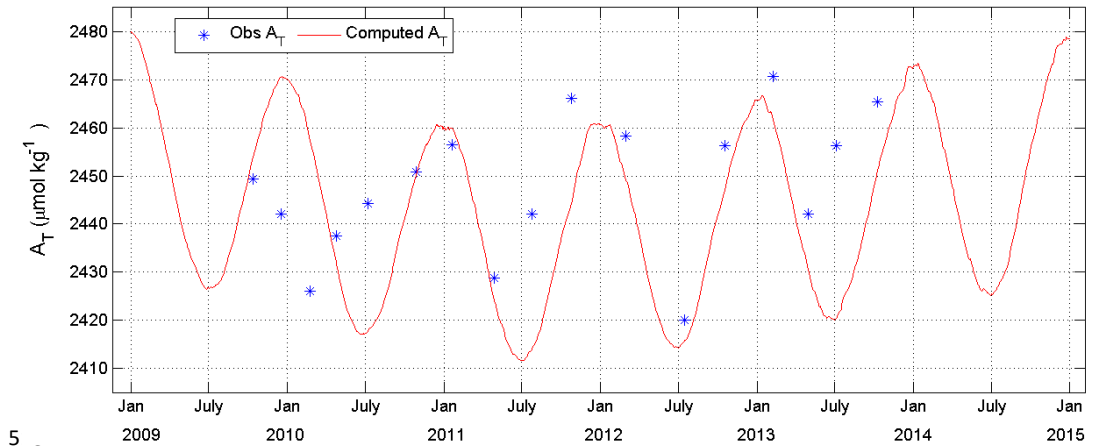
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501 Fig. 6: The seasonal variation of computed DIC:  $DIC_{SST-sat}$  (red line) is calculated based on  
502 satellite-derived SST ( $SST_{Sat}$ ) and  $DIC_{Reg}$  (green line) is calculated based on multivariate  
503 regression coefficients of the SST-SSS-DIC relationship,  $SST_{Sat}$ , and  $SSS_{Adv}$ . Blue stars indicate  
504 averaged observed surface DIC along the SPS2 section.

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519 Fig. 7: The seasonal variation of the computed daily  $A_T$  (red line) based on  $SSS_{Adv}$ . Blue stars  
520 indicate averaged observed surface  $A_T$  along the SPS2 transect.

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# Paper III

## **Sea surface $p\text{CO}_2$ variability and air sea gas exchange in the coastal Sudanese Red Sea**

Ali, E. B., I. Skjelvan, A. M. Omar, A. Olsen, T. E. de Lange, T. Johannessen, S. Elageed

*Manuscript in preparation*



# 1 Sea surface pCO<sub>2</sub> variability and air sea gas exchange in the coastal 2 Sudanese Red Sea

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5 Elsheikh B. Ali<sup>1,2</sup>, Ingunn Skjelvan<sup>3,1</sup>, Abdirahman M. Omar<sup>3,1</sup>, Are Olsen<sup>1</sup>, Tor E. de Lange<sup>1</sup>,  
6 Truls Johannessen<sup>1,3</sup>, and Salma Elageed<sup>1,2</sup>

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8 <sup>1</sup>Geophysical Institute (GFI), University of Bergen (UiB), Bergen, Norway

9 <sup>2</sup>Institute of Marine Research (IMR), Red Sea University, Port Sudan, Sudan

10 <sup>3</sup>Uni Research Climate, Bjerknes Centre, Bergen, Norway

## 11 12 **Abstract**

13 Oceanic partial pressure of carbon dioxide ( $pCO_2^w$ ) have been determined for the first time over  
14 a full annual cycle in the coastal Red Sea off Port Sudan. The measurements were obtained between  
15 October 2014 and October 2015 using moored autonomous sensors.  $pCO_2^w$  varies throughout the  
16 year with an amplitude of approximately 60  $\mu\text{atm}$ , overlaid a high frequency signal of 10  $\mu\text{atm}$ .  
17 The highest values of about 420  $\mu\text{atm}$  occur during summer-autumn and the lowest values of about  
18 360  $\mu\text{atm}$  occur during winter-spring. The seasonal  $pCO_2^w$  variation is mainly incurred by  
19 temperature changes and the remaining  $pCO_2^w$  change is controlled by along-coast advection. The  
20 area is a net annual sink for CO<sub>2</sub> of size 24.4 mmol CO<sub>2</sub> m<sup>-2</sup> y<sup>-1</sup>. During summer and autumn, the  
21 area is a source for atmospheric CO<sub>2</sub>, with CO<sub>2</sub> fluxes ( $F_{CO_2}$ ) of 0.1 to 2 mmol CO<sub>2</sub> m<sup>-2</sup> day<sup>-1</sup>, while  
22 during winter and spring, the area is a sink for atmospheric CO<sub>2</sub> of 0.02 to 5 mmol CO<sub>2</sub> m<sup>-2</sup> day<sup>-1</sup>.  
23 Based on data from the years 1977 to 2015, it is likely that during the 2000s the focus area  
24 transformed from being a net annual atmospheric source of CO<sub>2</sub> to a net annual sink for CO<sub>2</sub>.

## 25 26 **Keywords**

27 Sea surface pCO<sub>2</sub>; CO<sub>2</sub> flux; coastal Red Sea; seasonality

31 **1. Introduction**

32 Carbon dioxide (CO<sub>2</sub>) is released into the atmosphere from burning of fossil fuel and land use  
33 changes. During the last decade (2006-2015) these processes have been responsible for an input to  
34 the atmosphere of  $10.3 \pm 0.5$  Pg C y<sup>-1</sup> (Le Quéré et al., 2016), of which the ocean and terrestrial  
35 biosphere have absorbed about  $2.6 \pm 0.5$  and  $3.1 \pm 0.9$  Pg C y<sup>-1</sup>, respectively (Le Quéré et al., 2016).  
36 The global carbon budget is based on an enormous amount of data from all over the world, but  
37 there is still an ongoing effort to more precisely determine the amount of carbon exchanged  
38 between the different reservoirs.

39 The Red Sea is one of the under sampled areas of the world ocean. Situated between Africa  
40 and Asia, the Red Sea represents only approximately 0.12% of the world ocean surface area. It is  
41 one of the warmest and saltiest ocean areas in the world, and the biodiversity is rich, with numerous  
42 coral reefs along the coast. The area is exposed to both heavy ship traffic and climatic changes  
43 (IMR, 2012), but in spite of this, only a few studies have been conducted in the Red Sea with the  
44 carbon cycle as the main focus, and this is particularly true for the coastal areas. The GEOSECS  
45 cruise in 1977 (Weiss et al., 1983), the MEROU cruises in 1982 (Beauverger et al., 1984 a;b;  
46 Souvermezoglou et al., 1989; Metzl et al., 1989), and the MINERVE cruises in 1991, 1992, and  
47 1999 (Metzl et al., 1995; 2008) were all conducted along the north-south central axis of the Red  
48 Sea, and all of these reported that the Red Sea was a source of CO<sub>2</sub> to the atmosphere. This was  
49 also confirmed by Ali (2008). The only exception to this is Metzl et al. (1995), who reported that  
50 the northern part of the Red Sea appeared to be a sink for CO<sub>2</sub> during summer 1991.

51 However, the central area of the Red Sea might not be representative for the coastal part of  
52 the sea, which is clearly seen from e.g. sea surface height anomalies which varies from coast to  
53 open ocean depending on the time of year (Yao et al., 2014), and spatiotemporal variation of  
54 primary production which is slightly higher along the coast than in the open sea (Raitsos et al.,  
55 2013). Yao et al. (2014) also reported that, for parts of the year, the coastal water was slightly  
56 fresher than in the central Red Sea.

57 In this study, we use new carbon and hydrography data from moored instruments deployed  
58 off the Sudanese coast of the Red Sea. The aim of the work is to determine, for the first time in this  
59 area, the air sea flux of CO<sub>2</sub> over a full annual cycle, and further unravel its seasonal variations and  
60 drivers.

## 61 **2. Data and Methods**

### 62 **2.1 Data**

63 The data used in this study are from two moored instruments deployed in October 2014 and  
64 retrieved one year later, in October 2015. The mooring site, (37.395 °E, 19.720 °N), is located 18  
65 km north east of Port Sudan at the Sudanese coast (Fig. 1), located at the same position as one of  
66 the stations (St4) in a time series between Port Sudan and Sanganeb atoll reef presented in Ali et  
67 al. (2017a; 2017b). The moored instruments consisted of a Sea Bird MicoCat temperature  
68 (accuracy  $\pm 0.01^\circ\text{C}$ ) and conductivity (accuracy  $\pm 0.02 \text{ mS cm}^{-1}$ ) sensor (SBE 37-SM), and a  
69 SAMI-CO<sub>2</sub> instrument (Submersible Autonomous Moored Instrument CO<sub>2</sub> sensor, Sunburst  
70 Sensors, accuracy  $\pm 3 \text{ }\mu\text{atm}$ ). The bottom depth at the mooring site is about 800 m, and the  
71 MicroCat and SAMI-CO<sub>2</sub> were parked at 37 and 34 m depth, respectively. Temperature and  
72 conductivity were determined hourly, while the partial pressure of CO<sub>2</sub> ( $p\text{CO}_2$ ) was measured every  
73 three hours throughout the year of deployment. The  $p\text{CO}_2$  measurements were calibrated using  
74 discrete duplicate samples of DIC (Dissolved Inorganic carbon), and  $A_T$  (Total Alkalinity)  
75 collected from 35 and 40 m depth immediately after the mooring deployment. The DIC and  $A_T$   
76 samples were conserved using saturated HgCl<sub>2</sub> solution, kept cold and dark, and analyzed at the  
77 Geophysical Institute, University of Bergen, Norway. DIC was determined using a VINDTA 3C  
78 with UIC Coulometer (CM5012, accuracy  $\pm 0.5 \text{ }\mu\text{mol kg}^{-1}$ ) based on gas extraction of acidified  
79 water samples methods followed by coulometric titration (Johnson et al., 1993; Dickson et al.,  
80 2007), while  $A_T$  was analyzed using Marianda VINDTA 3S (accuracy  $\pm 1 \text{ }\mu\text{mol kg}^{-1}$ ). The accuracy  
81 was achieved by using certified reference material (CRM) supplied by prof. Andrew Dickson,  
82 Scripps Institution of Oceanography, USA. Averages of measured DIC and  $A_T$  from 35 and 40 m  
83 depth were used to calculate  $p\text{CO}_2$  through the CO2SYS program (Pierrot et al., 2006) with the  
84 carbon system constants of Merbach et al. (1973) refitted by Dickson and Millero (1987). The  
85 computed  $p\text{CO}_2$  were compared with the  $p\text{CO}_2$  measurements and an adjustment of  $-6 \text{ }\mu\text{atm}$  was  
86 applied to the observed  $p\text{CO}_2$  data.

87 Atmospheric  $x\text{CO}_2$  measurements from the nearest station (Sede Boker in Negev desert,  
88 Israel) have been downloaded from NOAA/CMDL ([http://cdiac.ornl.gov/trends/co2/cmdl-  
89 flask/wis.html](http://cdiac.ornl.gov/trends/co2/cmdl-flask/wis.html)), and atmospheric  $p\text{CO}_2$  values were calculated according to Körtzinger (1999).

90 The net air-sea CO<sub>2</sub> flux ( $F_{\text{CO}_2}$ ) was determined using the  $p\text{CO}_2$  difference between the  
91 seawater and atmosphere:

92

93

$$F_{Co_2} = KS(pCO_2^{seawater} - pCO_2^{atmosphere}) \quad (1)$$

94

95 where  $K$  is the gas transfer velocity and  $S$  is the solubility of  $CO_2$  in the seawater. For low wind  
96 speeds,  $K$  relationship do not have a big impact on the calculated fluxes (e.g. Sweeney et al., 2007),  
97 but at wind speeds above approximately  $5 \text{ m s}^{-1}$ , the deviation between  $K$ 's becomes notable. Here,  
98  $K$  from Nightingale et al. (2000) was used, which is relatively similar to that from Sweeney et al.  
99 (2007) for wind speeds below  $10 \text{ m s}^{-1}$ . The wind at 10 m height above sea surface,  $U_{10}$ , was  
100 calculated from daily averaged 6-hourly u- and v-wind velocity components extracted from  
101 NCEP/NCAR reanalysis 2 data at the position  $20.00^\circ \text{N}$  and  $37.50^\circ \text{E}$ .  $S$  from Weiss (1974) was  
102 used.

103

## 104 **2.2 Data processing**

105 Ali et al. (2017a) examined in detail the seasonality of the water masses with respect to temperature  
106 and salinity along their sections SPS1 and SPS2 between Port Sudan and Sanganeb atoll reef. They  
107 found that the mixed layer depth (MLD) in the area of study was approximately 40 m deep during  
108 most of the year and deeper during winter mixing. The water shallower than 40 m is part of a  
109 homogeneous surface layer, and thus, the mooring data from 37 m depth in the current study are  
110 defined as surface data.

111 At a few occasions, the moored instruments were dragged down to larger depths than 37 m,  
112 apparent from the Sea Bird MicroCat pressure values. This happened e.g. in the end of July and  
113 end of September 2015, however, no noticeable change was observed in  $pCO_2$  during these times,  
114 and consequently, these data were included in the analyses.

115 The summer season is defined as the months May – August while the winter are the months  
116 December – February as in Ali et al. (2017b). The transitions from winter to summer and from  
117 summer to winter are referred to as spring and fall, respectively, and the fall season is when Port  
118 Sudan receive the major parts of the annual rainfall.

119

120



### 121 3. Results

122 The temperature and salinity data for the period between October 2014 and October 2015 is shown  
123 in Fig. 2a and 2b and have also been described in detail by Ali et al. (2017a). The seasonal  
124 temperature change was about 6°C while that of salinity was approximately 1.1 psu. The highest  
125 temperatures occurred during autumn and the lowest at the end of January, while the summer was  
126 a transition period with relatively large variations over short time. In mid-May, the temperature  
127 abruptly increased to its average summer value of 28.5°C. The salinity (Fig. 2b) was highest  
128 (around 39.7 psu) during autumn and winter and lowest (around 39.0 psu) during spring and  
129 summer. Several abrupt and large changes in salinity occurred, e.g. in early March and in late  
130 August (Fig. 2b). The seasonal peaks and lows for the salinity lagged those for temperature by  
131 about 3 months, which, according to Ali et al. (2017a) most likely is caused by temperature  
132 variation being governed by air-sea heat exchange in addition to horizontal fluxes associated with  
133 advection of water along the coast. This process partly explain the salinity variation, but in addition,  
134 the salinity is controlled by evaporation/precipitation. However, the fresh water runoff into the Red  
135 Sea is limited and the evaporation dominates over precipitation

136 The oceanic  $pCO_2$  ( $pCO_2^w$ ) showed similar seasonal pattern as the temperature (Fig. 2c), with  
137 a seasonal amplitude of approximately 60  $\mu\text{atm}$ , and an induced higher frequency signal of about  
138 10  $\mu\text{atm}$ . The highest  $pCO_2^w$  occurred during autumn (400-420  $\mu\text{atm}$ ) when the water was warmest  
139 and saltiest, while the lowest  $pCO_2^w$  was observed in January (ca. 350  $\mu\text{atm}$ ) when the water was  
140 coldest. During spring, the  $pCO_2^w$  value was relatively stable as for temperature and to some  
141 degree also salinity. Abrupt  $pCO_2^w$  changes occurred during mid-May and early of January, where  
142 the latter had character of being an extreme event, possibly connected to both temperature and wind  
143 speed.

144 The atmospheric  $pCO_2$  ( $pCO_2^{air}$ ) (Fig. 2c) was highest during winter and spring and lowest  
145 during summer. The seasonal atmospheric amplitude was about 15  $\mu\text{atm}$ , i.e. 25% of that of  $pCO_2^w$   
146 of 60  $\mu\text{atm}$ .  $pCO_2^{air}$  was lower than  $pCO_2^w$  during summer and autumn and higher than  $pCO_2^w$   
147 during winter and spring. The seasonal cycle of  $\Delta pCO_2$  (Fig. 2d) resembled that of  $pCO_2^w$  and  
148 positive  $\Delta pCO_2$  ( $pCO_2^w > pCO_2^{air}$ ) indicates that  $CO_2$  was directed from sea to air while negative  
149 values indicates that  $pCO_2^w < pCO_2^{air}$ .

150 The daily CO<sub>2</sub> fluxes ( $F_{CO_2}$ ) are shown in Fig. 2f, and in general, the  $F_{CO_2}$  values were  
151 relatively low, which was a result of low wind speeds during parts of the year in addition to periods  
152 of nearly CO<sub>2</sub> equilibrium between ocean and atmosphere. For the period between October and  
153 November 2014, the ocean was supersaturated with CO<sub>2</sub> ( $\Delta pCO_2$  positive) and an amount of up to  
154 1.8 mmol CO<sub>2</sub> m<sup>-2</sup> day<sup>-1</sup> was degassed to the atmosphere. During winter 2015, a significant CO<sub>2</sub>  
155 undersaturation ( $\Delta pCO_2$  of -20 to -47  $\mu$ atm) developed and the area was characterized as a sink for  
156 atmospheric CO<sub>2</sub> with fluxes of up to -5 mmol CO<sub>2</sub> m<sup>-2</sup> day<sup>-1</sup> during mid-January. The strongest  
157 ingassing occurred during events of high wind speeds (Fig. 2e). From February to April 2015, the  
158 surface water was slightly undersaturated with CO<sub>2</sub> ( $\Delta pCO_2$  varied between zero and -20  $\mu$ atm)  
159 and the  $F_{CO_2}$  ranged between -0.02 and -1.2 mmol CO<sub>2</sub> m<sup>-2</sup> day<sup>-1</sup>. From mid-May until October  
160 2015, the sea surface again became a source for CO<sub>2</sub>, of size 0.13 to 1.8 mmol CO<sub>2</sub> m<sup>-2</sup> day<sup>-1</sup>, with  
161 peaks related to peaks in the wind speed.

162

## 163 **4. Discussion**

164 The main hydrographic and biogeochemical drivers influencing the annual cycle of pCO<sub>2</sub> are  
165 discussed in Section 4.1, while Section 4.2 focuses on the long-term trend in air-sea CO<sub>2</sub> exchange  
166 in the Red Sea over a period of nearly four decades from 1977 to 2015.

167

### 168 **4.1 Factors controlling $pCO_2^w$**

169  $pCO_2^w$  is affected by changes in temperature, dissolved inorganic carbon (DIC), alkalinity ( $A_T$ ),  
170 and to a minor degree salinity, whereas biological production affects  $pCO_2^w$  through changes in  
171 DIC. The seasonal variability of DIC and  $A_T$  is thoroughly described in Ali et al. (2017b) where  
172 they used a new coastal time series off Port Sudan to unravel the carbonate system in this area.  
173 They found that DIC variations were mainly governed by temperature driven air-sea gas exchange  
174 modulated by biological activity, while  $A_T$  was controlled primarily by evaporation/precipitation  
175 driven salinity changes. Most of the remainder of the DIC and  $A_T$  variations are a result of  
176 advection of water masses along the coast (Ali et al., 2017b).

177

178 **4.1.1  $pCO_2$  versus temperature**

179 Within the range of the observed seasonal temperature change (ca. 6°C), the relationship between  
180  $pCO_2^w$  and temperature is nearly linear. However, a more general exponential expression is used  
181 similar to the thermodynamic relationship of Takahashi et al. (1993). From mooring observations,  
182 daily  $pCO_2^w$  versus daily SST gives the relationship

183 
$$\ln pCO_2^w = \ln (233) + 0.018 * SST, (R^2 = 0.60) \quad (2)$$

184 The proposed relationship accounts for 60% of the variability between  $pCO_2$  and SST, and the  
185 relationship also indicate that  $pCO_2^w$  changes by 1.8% per °C. The weekly  $pCO_2^w$  versus weekly  
186 SST was also analyzed (not shown) to check if the high frequency  $pCO_2^w$  variations overlaid the  
187  $pCO_2^w$  signal did affect the correlation between  $pCO_2^w$  and SST. This exercise gave a similar  
188 relationship as in Eq. 2 but with better correlation coefficient ( $R^2= 0.67$ ) most likely due to  
189 smoothing. A theoretical value of the  $pCO_2^w$  versus temperature change was also calculated from  
190 typical coastal Red Sea data (SST = 28°C, SSS = 39 psu, DIC = 2065  $\mu\text{mol kg}^{-1}$ ,  $A_T = 2440 \mu\text{mol}$   
191  $\text{kg}^{-1}$ ), the software CO2SYS (Pierrot et al., 2006), and constants from Mehrbach et al. (1973)  
192 refitted by Dickson and Millero (1987). From this, an isochemical change of 3.9% in  $pCO_2^w$  per °C  
193 was estimated. This is more than twice the percentage change based on observed data but, on the  
194 other hand, slightly less than that of Takahashi et al. (1993), who, based on North Atlantic data,  
195 estimated a change in  $pCO_2$  of 4.23% per °C at constant DIC,  $A_T$ , and salinity condition. The above  
196 calculation demonstrates that there are also other processes than temperature, which are influencing  
197 the  $pCO_2^w$  in such a way that the  $pCO_2^w$  decreases while T increases.

198 The seasonal change of  $pCO_2^w$  versus SST seems to have an elliptical shape, where  $pCO_2$   
199 during the months September towards January are below the regression line, while  $pCO_2$  during  
200 the months February towards August are, with a few exceptions, above the regression line. The  
201 lowest  $pCO_2^w$  and SST is observed in January and highest  $pCO_2^w$  in July-August (Fig. 3).  $pCO_2^w$   
202 during early April (380-390  $\mu\text{atm}$ ), is about 25  $\mu\text{atm}$  higher than the late December value (ca. 360  
203  $\mu\text{atm}$ ) for the same temperature (approximately 27°C). Salinity varies from approximately 39.8 psu  
204 during the first period to 38.8 psu during the last. We propose that the elliptical shape is a result of  
205 super saturation of surface water  $CO_2$  during early summer and autumn, and  $CO_2$  is degassed from  
206 the sea into the atmosphere. This leads to decreasing  $pCO_2$  during late autumn and early winter,

207 although the temperature is still higher. During early April and late December, the averaged wind  
208 values are about the same, and consequently,  $F_{CO_2}$ , which is directed from atmosphere to sea at  
209 both occasions, is controlled by  $\Delta pCO_2$ . In late December, the calculated  $CO_2$  flux into the sea is  
210 approximately  $-1 \text{ mmol } CO_2 \text{ m}^{-2} \text{ day}^{-1}$ , while in early April this amounts to  $-0.3 \text{ mmol } CO_2 \text{ m}^{-2}$   
211  $\text{day}^{-1}$ . The most likely processes being responsible for this change in  $pCO_2$  are (i) the  $CO_2$  uptake  
212 by surface sea occurring during December –April (ii) and along-coast advection. Metzl et al.  
213 (1995) and Ali (2008) used MINERVE data collected in 1991, 1992, and 1999 and found that the  
214 highest surface  $pCO_2^w$  values were located in the south central Red Sea similar to the maximum  
215 SST (south of our study area). Ali et al. (2017b) observed that during early February 2015 (their  
216 Fig. 4d) the geostrophic current in the area of study was reversed from southwards along-coast to  
217 northwards, which introduced water with higher  $pCO_2$ , lower salinity, and relatively higher  
218 temperature into the coastal area.

219 Similar elliptical shapes of  $pCO_2^w$  versus SST were also observed in the North Atlantic  
220 subtropical gyre by Lefèvre and Taylor (2002) and in the Caribbean Sea by Wanninkhof et al.  
221 (2007). According to Lefèvre and Tylor (2002), the explanation for the shape is that during  
222 summertime, the sea surface is super saturated with respect to  $CO_2$  and will emit  $CO_2$  to the  
223 atmosphere, lowering the  $pCO_2$  of the water. Thus, during fall and winter, when the water cools,  
224 the  $pCO_2$  would decrease towards even lower values than during summer. For the Caribbean Sea,  
225 Wanninkhof et al. (2007) showed that also other processes, like biological production and fresh  
226 water addition, played a role making the ellipsoid form of the  $pCO_2$ -SST relationship.

227

#### 228 **4.1.2 $pCO_2$ versus DIC**

229 There are no concurrent DIC measurements to the mooring  $pCO_2$  data, and to examine the  
230 relationship between  $pCO_2^w$  and DIC we use a regression relationship developed in Ali et al.  
231 (2017b) based on discrete surface DIC, SST, and SSS from the coastal section SPS2 (see Fig. 1);  
232  $DIC_{Reg} = aSSS + bSST + c$  ( $a = 6.4814$ ,  $b = -4.9739$ ,  $c = 1959$ ,  $R^2 = 0.54$ ). In the current work,  
233  $DIC_{Reg}$  is determined by evaluating the above equation using the daily temperature and salinity  
234 from the mooring. Fig. 4 shows the seasonal cycle of the computed  $DIC_{Reg}$ . There is a relative  
235 strong negative correlation ( $R^2=0.67$ ) between  $pCO_2^w$  and  $DIC_{Reg}$  (Fig. 5), and this is a signature

236 of CO<sub>2</sub> source areas. In such areas, higher pCO<sub>2</sub> induce higher efflux of CO<sub>2</sub> from sea to air, and  
237 thus the surface water DIC is reduced. On the contrary, the  $pCO_2^w$ -DIC correlation in sink areas is  
238 positive i.e. pCO<sub>2</sub> increases with increasing DIC (Takahashi et al., 1993). According to Ali et al.  
239 (2017b), there is a negative correlation between DIC and temperature at the Sudanese coast, and  
240 this is in line with the observed positive pCO<sub>2</sub>-SST correlation and estimated negative pCO<sub>2</sub>-DIC  
241 correlation from the current work. No correlation was found between temperature normalized  
242  $pCO_2^w$  and DIC.

243

#### 244 **4.2 Air-sea gas exchange over years**

245 The Red Sea is located in the sub-tropical area, which is characterized globally as a net annual  
246 source area of CO<sub>2</sub> (Takahashi et al., 2002). The monthly average of  $pCO_2^{air}$  and  $pCO_2^w$  for the  
247 period 1977 - 2015 within the area 19 - 20°N, 37 - 39°E is shown in Fig. 6. The figure combines  
248 datasets collected in the open sea (GEOSECS 1977, Weiss et al., 1983; MEROU I&II 1982,  
249 Beauverger et al., 1984a; b; MINERVE 1991; 1992; 1999, Metzl et al. 1995; 2008) with datasets  
250 from the Sudanese coast (time series from 2007 and 2009-2013, Ali et al., 2017b; mooring data in  
251 the current study). DIC values of GEOSECS was extracted from GLODAPv2 database and the data  
252 were adjusted according to the advices on the GLODAPv2 web page (Olsen et al., 2016; Key et  
253 al., 2015). The atmospheric measurements from Sede Boker station in the Negev desert started in  
254 1995, and to estimate an atmospheric value for 1977, we first compared atmospheric  $xCO_2^{air}$  values  
255 from Mauna Loa with the atmospheric data from the Sede Boker station. The amplitude was  
256 slightly larger at the latter station, but in spite of this, the interannual trend was similar at the two  
257 stations. Thus, it is reasonable to assume that the change in xCO<sub>2</sub> between 1977 and 1995 at the  
258 Israeli station was similar to the change in xCO<sub>2</sub> observed at Mauna Loa during the same period,  
259 i.e. 27 ppm. From this, we can estimate the atmospheric xCO<sub>2</sub> value at Sede Boker to be 338 ppm  
260 in 1977, which corresponded to a  $pCO_2^{air}$  value of approximately 323 μatm (Fig. 6). Then, we  
261 could, with confidence, draw a line representing the Red Sea  $pCO_2^{air}$  over the years 1977 - 2015  
262 (Fig. 6). Over these nearly four decades, the atmospheric CO<sub>2</sub> content increased by 66 μatm, from  
263 323 μatm in winter 1977 to 389 μatm in winter 2015, which is equivalent to an annual increase of  
264 1.74 μatm y<sup>-1</sup>.

265 The open ocean data were collected during summer and autumn except GEOSECS 1977,  
266 which was collected in December 1977, while the coastal datasets cover the whole year. Fig. 6  
267 shows that, at least since 2009, the Red Sea switches over the year between being a sink and a  
268 source of atmospheric CO<sub>2</sub>. During summer and autumn,  $pCO_2^w$  (open dots) are higher than  
269  $pCO_2^{air}$  (gray dots), which indicates degassing of CO<sub>2</sub> during these seasons. During winter and  
270 spring, although we have few data, the  $pCO_2^w$  (black dots) are in general less than the  $pCO_2^{air}$   
271 values, which indicate uptake of atmospheric CO<sub>2</sub> during these seasons. Exception to this is in  
272 December 2009 and February 2010 (SPS2 data, current study). Over the year of 2014 to 2015, the  
273 area was a net sink for atmospheric CO<sub>2</sub> of size 24.4 mmol CO<sub>2</sub> m<sup>-2</sup> y<sup>-1</sup>.

274 The sub-tropics are considered net annual source areas for atmospheric CO<sub>2</sub>, (Takahashi et  
275 al., 2002), and at the entrance of the Red Sea CO<sub>2</sub> is emitted to the atmosphere during both winter  
276 and summer (Souvermezoglou et al., 1989). Thus, it is reasonable to assume that the Red Sea also  
277 should be considered a net annual source area for atmospheric CO<sub>2</sub>. The only data that could prove  
278 such an assumption is the GEOSECS data from winter 1977 (Weiss et al., 1983). We are aware  
279 that the quality of these data is subject for discussion since a relatively large data correction had to  
280 be performed (Olsen et al., 2016; Key et al., 2015). However, if we assume that the GEOSECS  
281 data are acceptable, we can include the winter value of the one central Red Sea station from 1977  
282 in Fig. 6, and further interpolate between all winter and spring pCO<sub>2</sub> data available. Based on this,  
283 we speculate that the central Red Sea has been transformed from being a source area for  
284 atmospheric CO<sub>2</sub> all year around to becoming an area, which is in CO<sub>2</sub> equilibrium with the  
285 atmosphere during parts of the year, and even, at occasions, becomes a sink for atmospheric CO<sub>2</sub>.

286 So, when could such a transformation have occurred? To answer this question, we  
287 compared the interannual  $pCO_2^w$  trend lines based on summer-autumn and winter-spring data with  
288 that for the interannual  $pCO_2^{air}$  trend in Fig. 6. It appeared that the line for the  $pCO_2^w$  winter-spring  
289 trend crossed the  $pCO_2^{air}$  trend line around the year 2002, which is referred to as the time of CO<sub>2</sub>  
290 equilibrium between atmosphere and ocean. During summer and autumn, the  $pCO_2^{air}$  is increasing  
291 with 1.95 μatm y<sup>-1</sup> (not shown), which is larger than the annual change of  $pCO_2^w$  of 1.19 μatm y<sup>-1</sup>.  
292 This indicates that during summer-autumn, the area emits a decreasing amount of CO<sub>2</sub> to the  
293 atmosphere from one year to the next. Thus, it is reasonable to assume that in the future, equilibrium  
294 is reached between atmosphere and sea surface also during summer-autumn. The coastal Red Sea  
295 might then be converted to a sink area for CO<sub>2</sub> throughout the year.

296 As mention, the amount of winter-spring data is low, and the conclusion of the Red Sea being  
297 a CO<sub>2</sub> source prior to 2002 must be drawn with care. The historical winter data is from the open  
298 ocean while the coastal data is not present until 2007. Therefore, it might be that the negative CO<sub>2</sub>  
299 fluxes are only real in the coastal area and not in the open ocean. However, the main finding from  
300 the coastal data is still valid; that the coastal Red Sea absorbs CO<sub>2</sub> during the winter of the last  
301 years.

302 This finding is compatible with the Mediterranean Sea, which was classified as a source of  
303 CO<sub>2</sub> during 1980s, while in the 2000s equilibria between  $pCO_2^w$  and  $pCO_2^{air}$  was reached, and at  
304 present the ocean is considered a minor sink of CO<sub>2</sub> (Taillandier et al., 2012). The Red Sea is  
305 similar to the Mediterranean Sea in terms of being a semi-closed sea with relatively high  
306 temperature and salinity, and, apparently, following the same transformation from being a source  
307 area for CO<sub>2</sub> to becoming sink area. As both seas are being influenced by the same monsoon system  
308 especially during summer, it is reasonable to assume that the same factors driving the  
309 transformation in the Mediterranean Sea are also responsible for the source to sink transformation  
310 in the Red Sea. A similar transformation is seen in the Baltic Sea, which also was characterized as  
311 a source area for CO<sub>2</sub> before the industrial revolution, while during the industrialization period, the  
312 Baltic Sea appears to be both a source and a sink area for atmospheric CO<sub>2</sub> (Omstedt et al., 2009).

313 Air-sea CO<sub>2</sub> exchange is also influenced by increasing SST as a result of global warming.  
314 Raitsos et al. (2011a) found that the annual mean SST in the Red Sea has increased by about 0.032  
315 °C y<sup>-1</sup> for the period between 1985 and 2007. An increasing SST leads to an increase in air-sea CO<sub>2</sub>  
316 efflux through the changes in CO<sub>2</sub> solubility (Yilmaz, 2008). Over time, this will create a loss of  
317 DIC in the Red Sea. The only source for DIC to the Red Sea is from the Gulf of Aden water, and  
318 according to Souvermezoglou et al. (1989), about 21% of the DIC budget that enters into the Red  
319 Sea from the Gulf of Aden is lost by air-sea gas exchange. Thus, an increase of the CO<sub>2</sub> flux to the  
320 atmosphere due to warming will increase the percentage of loss by air-sea gas exchange on the  
321 total DIC budget. If the supplied CO<sub>2</sub> is not balanced with the consumption, this process will lead  
322 to a larger loss by air-sea gas exchange on the total DIC budget. Quantification of this has not been  
323 further elaborated.

324 For simplicity, the increasing atmospheric CO<sub>2</sub> content from burning of fossil fuel and land  
325 use changes have been neglected in the above assumption. When the increase in atmospheric CO<sub>2</sub>  
326 content (approximately 1.8 μatm y<sup>-1</sup>) is taken into account, the picture changes slightly. During

327 wintertime, when the Sudanese part of the Red Sea is a sink for atmospheric CO<sub>2</sub>, the size of the  
328 sink is assumed to increase even more with increasing SST, while during summers, when the sea  
329 is a source for atmospheric CO<sub>2</sub>, the size of the source will be less than without warming.

330         Increasing SST will also induce more stratification in the upper ocean, which leads to less  
331 available nutrients and reduced primary production (Behrenfeld et al., 2006; Raitso et al., 2011b;  
332 Taillandier et al., 2012). Low vertical mixing in CO<sub>2</sub> source areas like the Red Sea means less CO<sub>2</sub>  
333 exported from the deep layer into the surface, and consequently decreasing CO<sub>2</sub> levels in the  
334 surface layer. This, in turn, results in CO<sub>2</sub> equilibrium with the atmosphere and transformation into  
335 a sink area for atmospheric CO<sub>2</sub>.

336

## 337 **5. Summary and conclusion**

338 Using a SAMI-CO<sub>2</sub> sensor moored at about 37 m depth in the coastal Red Sea off Port Sudan, the  
339 annual cycle of oceanic *p*CO<sub>2</sub> (*p*CO<sub>2</sub><sup>w</sup>) was studied in the period October 2014 - October 2015.  
340 During summer-autumn, the *p*CO<sub>2</sub><sup>w</sup> values were highest, while winter-spring was the period with  
341 lowest *p*CO<sub>2</sub><sup>w</sup> values, and the annual amplitude of approximately 60 μatm was overlaid a high  
342 frequency fluctuation of about 10 μatm. The area of study acted as a source for CO<sub>2</sub> during summer  
343 and autumn and a sink for CO<sub>2</sub> during winter and spring.

344         *p*CO<sub>2</sub><sup>w</sup> is relatively strongly correlated with SST ( $R^2 = 0.6$ ), which indicate that the  
345 temperature is the main driver for the changes in *p*CO<sub>2</sub><sup>w</sup>. In addition to temperature, along-coast  
346 advection, described in details in Ali et al. (2017a; b), also contribute to the observed variability  
347 in *p*CO<sub>2</sub><sup>w</sup>.

348         The area is a net annual sink for atmospheric CO<sub>2</sub> of size 24.4 mmol CO<sub>2</sub> m<sup>-2</sup> y<sup>-1</sup>, acting as a  
349 source during summer to fall and a sink during winter to spring. When taking into consideration  
350 data from the period 1977 to 2015, it is likely that the area transformed from being a net annual  
351 source for CO<sub>2</sub> to becoming a net annual sink sometimes during the 2000s, when a similar  
352 transformation was seen in the Mediterranean Sea.

353         In the current study, *p*CO<sub>2</sub> and air-sea gas exchange have been studied in coastal Red Sea.  
354 However, it is beyond doubt that additional coastal and open ocean data would have contributed to  
355 further unravelling of the carbon cycle in this part of the sub-tropical seas.



356

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365 **6. References**

- 366 Ali, E. B., 2008. The inorganic carbon cycle in the Red Sea, Master's thesis, Geophysical Institute,  
367 University of Bergen, Bergen, Norway, 77 pp.
- 368 Ali, E. B., Churchill, J. H., Barthel, K., Skjelvan, I., Omar, A. M., de Lange, T. E., and Eltaib, E.  
369 B. A., 2017a. Seasonal and interannual variations of hydrographic parameters in the Sudanese  
370 coast of the Red Sea, 2009-2015, submitted to Regional Studies in Marine Sciences journal.
- 371 Ali, E.B., Omar, A.M., Skjelvan, I., Johannessen, T., 2017b. Dissolved inorganic carbon and total  
372 alkalinity at the Sudanese coastal Red Sea, 2009-2013, submitted to Regional Studies in Marine  
373 Sciences journal.
- 374 Beauverger, C., Brunet, C. Poisson, A. 1984a. MD 33/MEROU a bord du "MARION DUFRESNE"  
375 16 juin au 4 juillet 1982, 25 septembre au 4 octobre 1982. Les rapports des campagnes ala mer,  
376 publication des T.A.A.F. No. 82-04, 181 pp.
- 377 Beauverger, C. 1984b. MD 33/MEROU a bord du "MARION DUFRESNE" ]6 juin au 4 juillet  
378 1982, 25 septembre au 4 octobre 1982. Les rapports des campagnes a la mer, publication des  
379 T.A.A.F. No. 82-05, 163 pp.
- 380 Behrenfeld, M. J., O'Malley, R., Siegel, D., McClain, C., Sarmiento, J., Feldman, G., Milligan, A.,  
381 Falkowski, P., Letelier, R., Boss, E., 2006. Climate-driven trends in contemporary ocean  
382 productivity. Nature 444, 752–755.
- 383 Dickson, A.G., Millero, F.J., 1987. A comparison of the equilibrium constants for the dissociation  
384 of carbonic acid in seawater media. Deep-Sea Research 34, 1733–1743
- 385 Dickson, A. G., Sabine, C. L., Christian, J. R., 2007. Guide to best practices for ocean CO<sub>2</sub>  
386 measurements, p. 191. PICES Special Publication 3.

387 IMR, 2012. Environmental Baseline Study for Mersa El-Sheikh Ibrahim, conducted by Institute of  
388 Marine Research (IMR) in coloration with Faculty of Marine Sciences and Fishers (FMSF) at  
389 Red Sea University (RSU), Port Sudan, Sudan.

390 Johnson, K. M., Wills, K. D., Butler, D. B., Johnson, W. K., and Wong, C. S., 1993. Coulometric  
391 total carbon dioxide analysis for marine studies, *Mar. Chem.*, 44, 167–187. 1993.

392 Key, R.M., Olsen, A., van Heuven, S., Lauvset, S. K., Velo, A., Lin, X., Schirnick, C., Kozyr, A.,  
393 Tanhua, T., Hoppema, M., Jutterström, S., Steinfeldt, R., Jeansson, E., Ishii, M., Pérez, F. F.,  
394 and Suzuki, T., 2015. Global Ocean Data Analysis Project, Version 2 (GLODAPv2),  
395 ORNL/CDIAC-162, NDP-P093. Carbon Dioxide Information Analysis Center, Oak Ridge  
396 National Laboratory, US Department of Energy, Oak Ridge, Tennessee.  
397 doi:10.3334/CDIAC/OTG.NDP093\_GLODAPv2.

398 Körtzinger, A., 1999. Determination of carbon dioxide partial pressure ( $p\text{CO}_2$ ), In: *Methods of*  
399 *Seawater Analysis* (eds.: K. Grasshoff, K. Kremling, M. Ehrhardt), Wiley-VCH, pp149-158.

400 Lefèvre, N., Taylor, A., 2002. Estimating  $p\text{CO}_2$  from sea surface temperatures in the Atlantic  
401 gyres, *Deep Sea Res., Part I*, 49, 539 – 554.

402 Le Quéré, C., Andrew, R. M., Canadell, J. G., Sitch, S., Korsbakken, J. I., Peters, G. P., Manning,  
403 A. C., Boden, T. A., Tans, P. P., Houghton, R. A., Keeling, R. F., Alin, S., Andrews, O. D.,  
404 Anthoni, P., Barbero, L., Bopp, L., Chevallier, F., Chini, L.P., Ciais. P., Currie, K., Delire, C.,  
405 Doney, S. C., Friedlingstein, P., Gkritzalis, T., Harris, I., Hauck, J., Haverd, V., Hoppema, M.,  
406 Klein Goldewijk, K., Jain, A.K., Kato, E., Körtzinger, A., Landschützer, P., Lefèvre, N., Lenton,  
407 A., Lienert, S., Lombardozi, D., Melton, J. R., Metzl, N., Millero, F., Monteiro, P.M.S., Munro,  
408 D.R., Nabel, J. E. M. N., Nakaoka, S.-i., O'Brien, K., Olsen, A., Omar, A. M., Ono, T., Pierrot,  
409 D., Poulter, B., Rödenbeck, C., Salisbury, J.E., Schuster, U., Schwinger, J., Séférian, R.,  
410 Skjelvan, I., Stocker, B. D., Sutton, A.J., Takahashi, T., Tian, H., Tilbrook, B., van der Laan-  
411 Luijks, I. T., van der Werf, G.R., Viovy, N., Walker, A. P., Wiltshire, A.J., Zaehle, S., 2016.  
412 Global carbon budget 2016, *Earth Syst. Sci. Data*, 8, 605–649,

413 Mehrbach, C. Culberson, C. H. Hawley, J. E., Pytkowicz, R. M., 1973. Measurement of the  
414 apparent dissociation constants of carbonic acid in seawater at atmospheric pressure. *Limnology*  
415 *and Oceanography* 18, 897-907.

416 Metzl, N., Moore III, B., Papaud, A., Poisson, A., 1989. Transport and carbon exchange in Red Sea  
417 inverse methodology, *Global Biogeochemical Cycle*, Vol. 3, No. 1, pages 1-26.

418 Metzl, N., Poisson, A. Louanchi, F., Brunet, C., Schauer, B., 1995. Spatio-temporal distributions  
419 of air-sea fluxes of CO<sub>2</sub> in the Indian and Antarctic Oceans: a first step. *Tellus*, 47B, 56-69.

420 Metzl, N. 2008. Sea Surface and Atmospheric fCO<sub>2</sub> data in the Mediterranean Sea, Red Sea and  
421 Indian Ocean during the R/V Marion-Dufresne Minerve-40 cruise from Marseille (France) to  
422 La Reunion, Oct. 18 - Nov. 1, 1999.

423 Nightingale, P. D., Malin, G., Law, C.S., Watson, A. J., Liss, P. S., Liddicoat, M.I., Boutin, J., R.  
424 C. Upstill-Goddard, R. C., 2000. In-situ evaluation of air-sea gas exchange parameterizations  
425 using novel conservative and volatile tracers, *Global Biogeochem. Cycles*, 14, 373-387.

426 Olsen, A., Key, R. M., van Heuven, S., Lauvset, S. K., Velo, A., Lin, X., Schirnick, C., Kozyr, A.,  
427 Tanhua, T., Hoppema, M., Jutterström, S., Steinfeldt, R., Jeansson, E., Ishii, M., Pérez, F. F.,  
428 and Suzuki, T., 2016. The Global Ocean Data Analysis Project version 2 (GLODAPv2) – an  
429 internally consistent data product for the world ocean, *Earth Syst. Sci. Data*, 8, 297-323,  
430 <https://doi.org/10.5194/essd-8-297-2016>.

431 Omstedt, A., Gustafsson, E., and Wesslander, K., 2009. Modelling the uptake and release of carbon  
432 dioxide in the Baltic Sea surface water, *Cont. Shelf Res.*, 29, 870–885.

433 Pierrot, D. E. L., Wallace, D. W. R., 2006. MS Excel Program Developed for CO<sub>2</sub> System  
434 Calculations. ORNL/CDIAC-105a. Carbon Dioxide Information Analysis Center, Oak Ridge  
435 National Laboratory, U.S. Department of Energy, Oak Ridge, Tennessee. doi:  
436 10.3334/CDIAC/otg.CO2SYS\_XLS\_CDIAC105a.

437 Raitso, D. E., Hoteit, I., Prihartato, P. K., Chronis, T., Triantafyllou, G., Abualnaja, Y., 2011a.  
438 Abrupt warming of the Red Sea, *Geophys. Res. Lett.*, 38, L14601, doi:  
439 10.1029/2011GL047984.

440 Raitso, D. E., Lavender, S. J., Maravelias, C. D., Haralabous, J., McQuatters-Gollop, A.,  
441 Edwards, M. Reid, P. C., 2011b. Macroscale factors affecting diatom abundance: A  
442 synergistic use of Continuous Plankton Recorder and satellite remote sensing data, *Int. J.*  
443 *Remote Sens.*, 32, 2081–2094, doi:10.1080/01431161003645832.

444 Raitso, D. E., Pradhan, Y., Brewin, R. J. W., Stenchikov, G., Hoteit, I., 2013. Remote Sensing the  
445 Phytoplankton Seasonal Succession of the Red Sea. *PLoS ONE* 8(6): e64909.  
446 doi:10.1371/journal.pone.0064909.

447 Sweeney, C., Gloor, E., Jacobson, A. R., Key, R. M., McKinley, G., Sarmiento, J. L., and  
448 Wanninkhof, R., 2007. Constraining global air-sea gas exchange for CO<sub>2</sub> with recent bomb C-  
449 14 measurements, *Global Biogeochem. Cy.*, 21, GB2015, doi:10.1029/2006gb002784.

450 Souvermezoglou, T., Metzl, N., Poisson, A., 1989. Red Sea budgets of salinity, nutrients and  
451 carbon calculated in the Strait of Bab Al Mandab during summer and winter seasons, *Journal*  
452 *of Marine Research*, 47,441-456.

453 Taillandier, V., D'Ortenzio, F., Antoine, D., 2012. Carbon fluxes in the mixed layer of the  
454 Mediterranean Sea in the 1980s and the 2000s. *Deep Sea Res. I* 65, 73–84.

455 Takahashi, T., Olafsson, J., Goddard, J., Chipman, D.W., Sutherland, S.C., 1993. Seasonal  
456 variation of CO<sub>2</sub> and nutrients in the high-latitude surface oceans: A comparative study. *Glob.*  
457 *Biogeochem. Cycles* 7, 843–878.

458 Takahashi, T., Feely, R.A., Weiss, R.F., Wanninkhof, R.H., Chipman, D.W., Sutherland, S.C.,  
459 Takahashi, T.T., 2002. Global sea–air CO<sub>2</sub> flux based on climatological surface ocean pCO<sub>2</sub>,  
460 and seasonal biological and temperature effects. *Deep-Sea Research II* 49 (9–10), 1601–1622.

461 Wanninkhof, R., Olsen, A., Trin˜anes, J., 2007. Air–sea CO<sub>2</sub> fluxes in the Caribbean Sea from  
462 2002–2004. *Journal of Marine Systems* 66 (1–4), 272–284.

463 Weiss, R. F., 1974. Carbon dioxide in water and sea water: the solubility of a non -ideal gas, *Mar.*  
464 *Chem.*, 2:203-215.

465 Weiss R. F., Broecker, W. S., Craig, H. B., Spencer., D. W., 1983. Geosecs Indian Ocean  
466 Expedition, 5. Hyd. data. Nat. Sc. Fond., U.S. Government Print. Off.

467 Yao, F. C., Hoteit, I., Pratt, L. J., Bower, A. S., Zhai, P., Kohl , A., Gopalakrishnan,  
468 G., 2014. Seasonal overturning circulation in the Red Sea: 1. Model validation and summer  
469 circulation. *Journal of Geophysical Research-Oceans*. 119:2238-2262.

470 Yilmaz A., De Lange G., Dupont S., Fine M., Fowler S.W., Gattuso J.P., Gazeau F., Gehlen M.,  
471 Goyet C., Jeffree R., Montagna P., Rees A.P., Reynaud S., Rodolfo-Metalpa R., Ziveri P. and  
472 F. 2008. Briand Impacts of acidification on biological, chemical and physical systems in the  
473 Mediterranean and Black Seas Menton, 1 – 4 October 2008 I - EXECUTIVE SUMMARY OF  
474 CIESM WORKSHOP 36.

475  
476

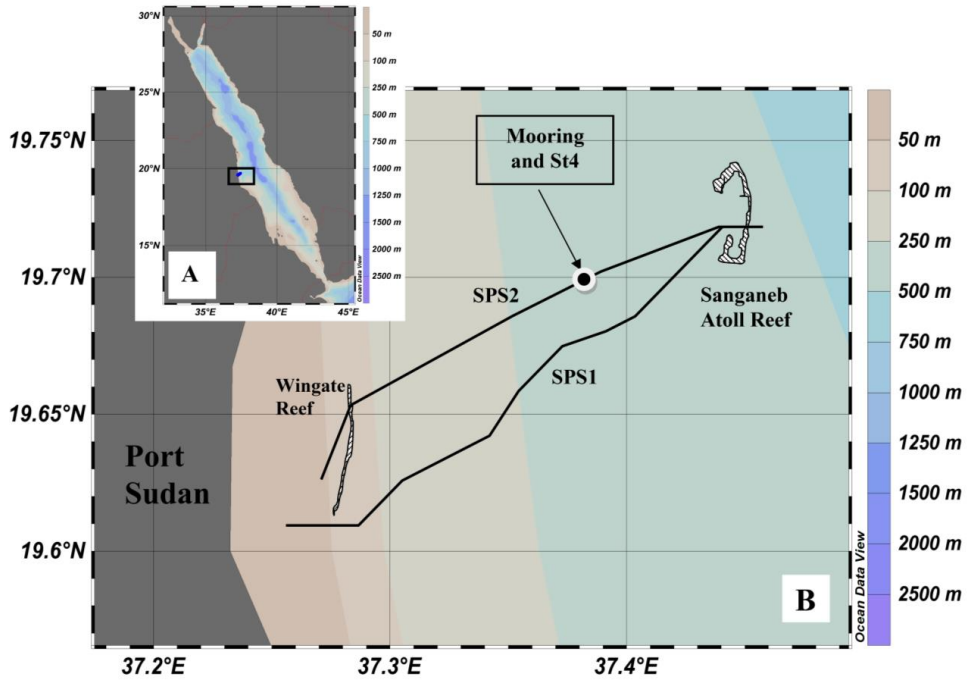


Fig. 1: The mooring location between Port Sudan and Sanganeb atoll reef. The lines SPS1 and SPS2 indicate two time series transects described in Ali et al. (2017a; 2017b) and used in the current work. Note that the mooring position is the same as that of station 4 in the time series transect SPS2.

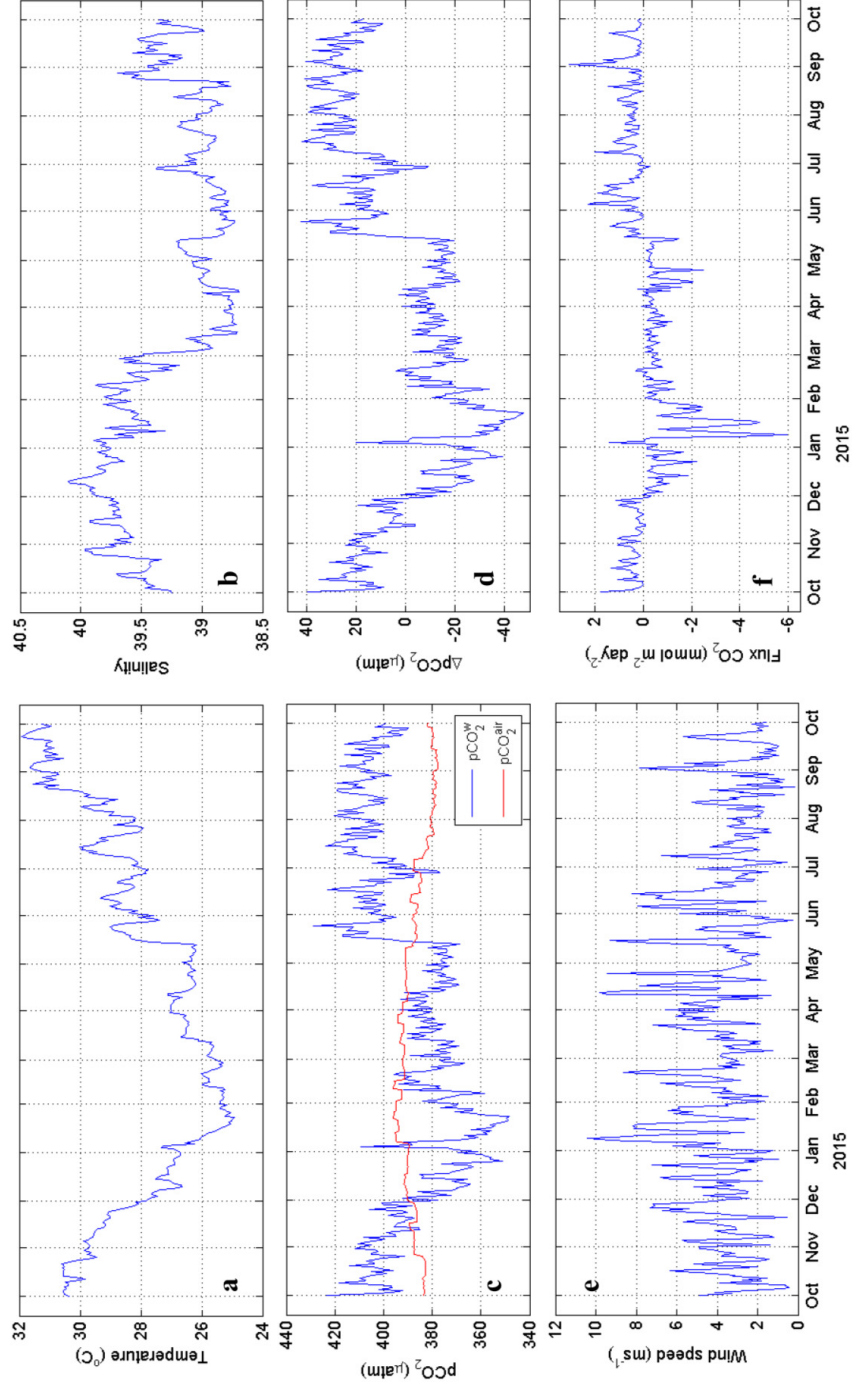


Fig. 2: Daily average of (a) temperature [ $^{\circ}\text{C}$ ], (b) salinity, (c)  $pCO_2^{\text{air}}$  and  $pCO_2^{\text{w}}$  [ $\mu\text{atm}$ ], (d)  $\Delta pCO_2$  [ $\mu\text{atm}$ ], (e) wind speed [ $\text{m s}^{-1}$ ], and (f)  $CO_2$  fluxes [ $\text{mmol CO}_2 \text{ m}^{-2} \text{ day}^{-1}$ ].

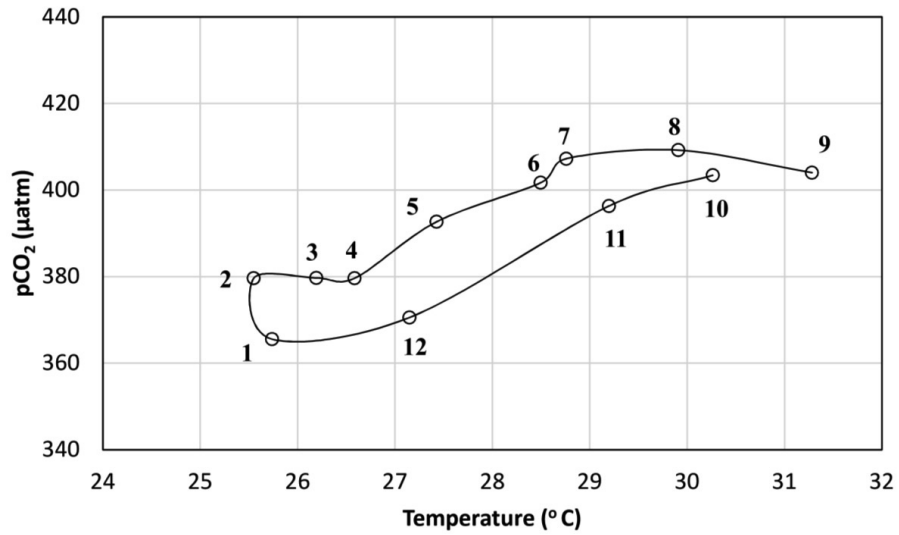


Fig. 3: Monthly averaged  $pCO_2^w$  [ $\mu\text{atm}$ ] as a function of temperature [ $^{\circ}\text{C}$ ]. The black line indicates the annual cycles of the  $pCO_2^w - SST$  changes. The number represent the month of sampling.

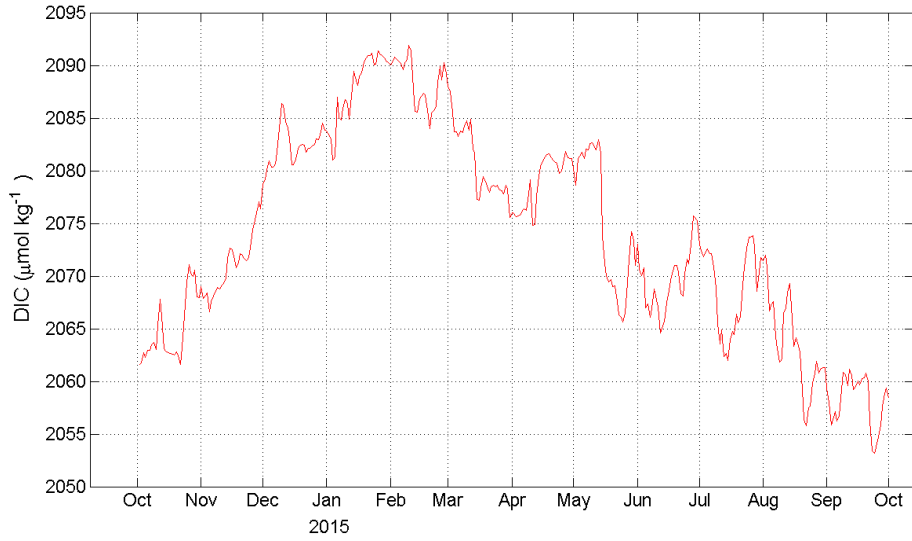


Fig. 4: The seasonal cycle of computed DIC ( $DIC_{Reg}$ ) calculated based on multivariate regression coefficients of the SST-SSS-DIC relationship from Ali et al. (2017b).



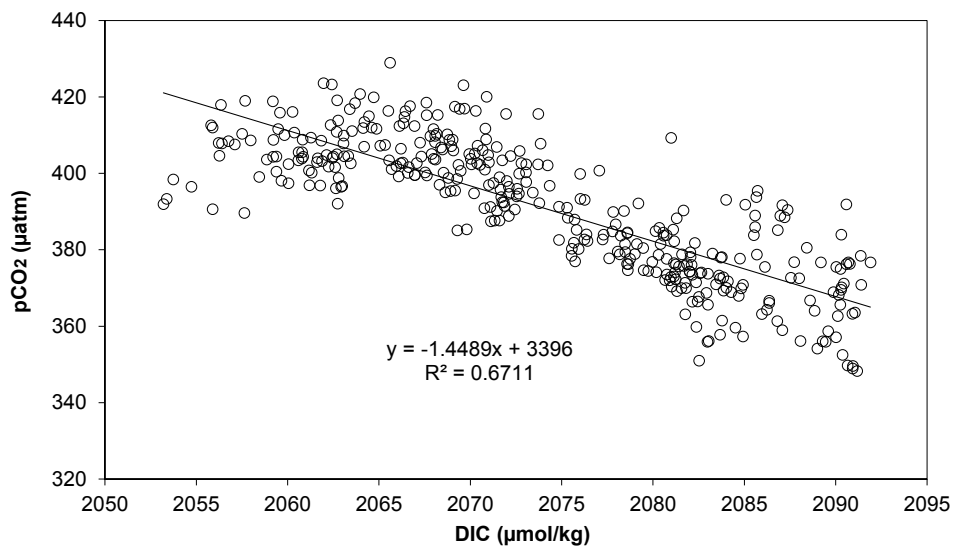


Fig.5 The relationship between  $pCO_2^w$  and  $DIC_{Reg}$

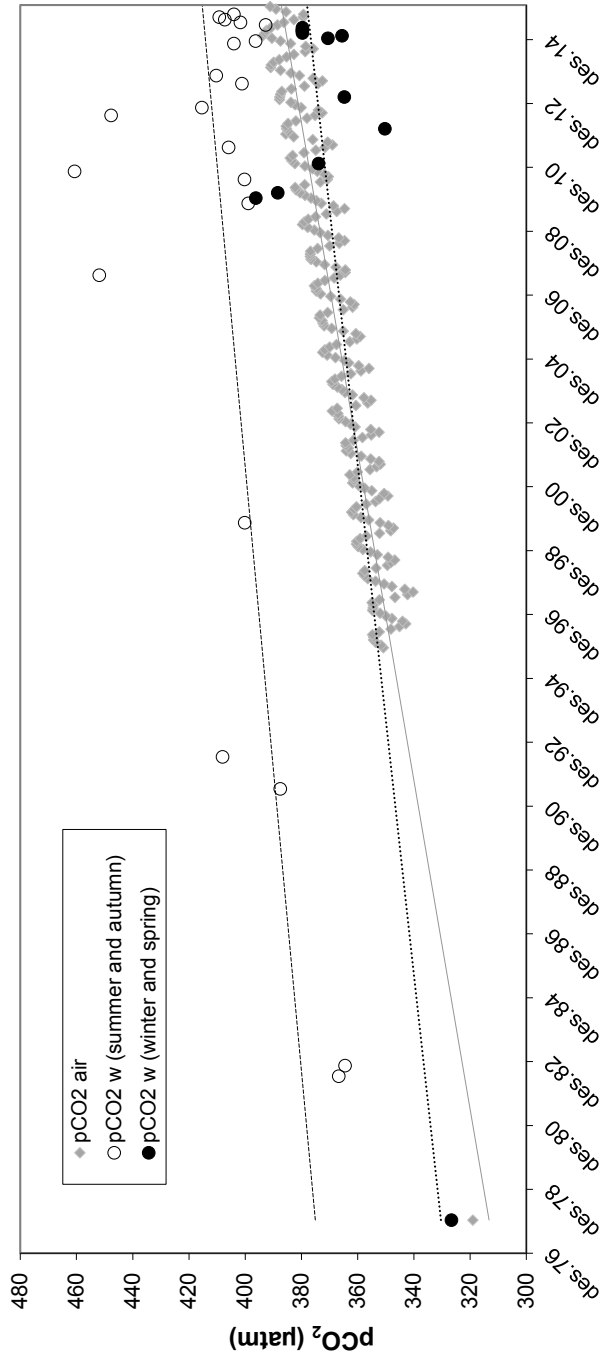


Fig. 6: Monthly average of atmospheric  $pCO_2$  data from Sede Boker station and oceanic  $pCO_2$  data from GEOSECS 1977 (Weiss et al., 1983), MEROU 1982 (I&II) (Beauverger et al., 1984), MINERVE 1991; 1992; 1999 (Metzl et al. 1995; 2008), Sanganeb 2007 (SPS1) (Ali, 2008), Sanganeb 2009-2013 (SPS2) (Ali et al., 2017b), and the mooring (current work). Trend lines are shown for  $pCO_2^w$  during summer and autumn (dashed line),  $pCO_2^w$  during winter and spring (dotted line), and atmospheric  $pCO_2$  (grey solid line).